

Federal Railroad Administration

Hazardous Materials Car Placement In A Train Consist

Office of Research and Development Washington, DC 20590

Volume I Appendices

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sisted of six task i	tems: (1) review of accident t us materials compatibility, (The review and analysis con- trends and regulations, (2) an (3) railroad operational con- endations, and (6) preparation	
data showed the reamainline freight trahigher the speed, the hazardous commodities for chemical incompal, 210 combinations oxidizing mineral acacids (Group 2) and chemicals. Consequentations established aprevent mixing of inservative (worst-case cargos, although a scenarios. It was reservations.	r quarter to be statistically in. Analysis confirmed that he more cars derailed on the (by volume movement) plus fur tibility, a total of 5,151 bi were judged incompatible. ids (ASTM Chemical Reactivity caustics (Group 10) dominance-based and risk-based ranking post-derailment separation discompatible chemicals. This caparation by 30 in-train separation may accompation may accompation.	dent Reporting System (RAIRS) y the "safest" location in a the longer the train, or the average. Also, the top 101 ming nitric acid were analyzed nary combinations. Of these, Combinations involving non- y Group 1), oxidizing mineral ate the list of incompatible ings were established. Calcu- stance of 40 meters minimum to can be accomplished by a con- cars for incompatible hazmat olish this for most accident aterials was not cited in any	
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Unclassified

METRIC CONVERSION FACTORS

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		LENGTH	_			<u></u>	· mm	millimeters	0.04	inches	in
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in	inches	*2.5	centimeters	cm	7	≣18	m	meters meters	3.3 1.1	feet	ft
ft	feet	30	centimeters	cm	· =	₽	m km	kilometers	0.6	yards miles	yd mi
yd	yards miles ~	0.9	meters	m	=	17	KIII.		0.0	1111109	mi
mi	miles	1.6	kilometers	km	<u>=</u>	<u> </u>			AREA		
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	-		•		6	15	cm ²	square centimeters	0.16	square inches	in ²
in ²	square inches	6.5	square centimeters	cm ²	- =	<u> </u>	m²	square meters	1.2	square yards	yd ²
ft ²	square feet	0.09	square meters	m²	=	<u>=</u> 14	km²	square kilometers	0.4	square miles	mi ²
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_	short tons	0.9	tonnes	t	4	=	t	tonnes (1000 kg)	1.1	short tons	
	(2000 lb)			*	_=	10					
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tsp	teaspoons	5	milliliters	ml	=	<u> </u>	ml	milliliters.	0.03	fluid ounces	fi oz
Tosp	tablespoons	15	milliliters	ml	3	Ē	1	liters	2.1	pints	pt
fi oz	fluid ounces	30	milliliters	ml	· -=	= ,	- 1	liters	1.06	quarts	qt
C	cups	0.24	liters	1		<u> </u>	١.	liters	0.26	gallons	gal
ρt	pints	0.47	liters	4		<u> </u>	m3	cubic meters	36	cubic feet	ft ³
qt	quarts	0.95	liters	ı	==	≣6	m ³	cubic meters	1.3	cubic yards	Aq ₃
gal	gallons	3.8	liters	١,	· . — —						
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APPENDIX A

FORM FRA F6180-54 (12-74)
RAIL EQUIPMENT ACCIDENT/INCIDENT REPORT

RAIL EQUIPMENT ACCIDENT/INCIDENT REPORT

FORM APPROVED OMB NO. 04R4008

420 104-101

NAME OF REPORTING RAILROAD NAME OF OTHER RAILROAD INVOLVED IN TRAIL NAME OF RAILROAD RESPONSIBLE FOR TRACK R U.S. DOT-AAR GRADE CROSSING IDENTIFICATION		NY/INCIDE	Int	Amerak Autotrain	_	1s. Alphabetic Code		1b. Rairond Ad	odint/Incident Ne.	
3. NÁME OF RAILRÓAD RESPONSIBLE FOR TRACK N		NY/INCIDE	ENT	AUTOURN	ľ	1				
	44114784-					2s. Alghabetis Code		2b. Retirond Ac	odent/Incident Ne.	
4. U.S. DOT-AAR GRADE CROSSING IDENTIFICATION	AMS I TIL	NCE /single	e entry)	·		3e. Alphabetic Code		3b. Reiross Ac	cident/Inedemt No.	
	N NUMBER					S. DATE OF ACCIDENT/INCIDENT		6. TIME OF AC	CIOENT/INCIDENT	
						manth day	Y		- [) pr
TYPE OF ACCIDENT/INCIDENT (enter number in code						<u> </u>	<u> </u>			
Derailment 3. Reer end collision Head on collision 4. Side collision		aking colli roken traii	ision n collision	7. Rail-H S. RR gr			11. Fire or viole	ent rupture 1	2. Other (specify)	
L CARE CARRYING	B C4 88 7	AMAGES	OR DERAIL		MATE	RIALS (number of)		T	ACUATED (ett.)	
	. Unna (M#49EU	OH DERAM	LED		10. CARE WHICH RELEASED HAZ	. MAT.	III. PEOPLE EV	ACUATED (ett.)	*
					LOC	TION				
2. DIVISION	13. NEAR	EST STAT	ION			14. MILEPOST (to negrest tenth)		15. STATE (Inc	ietter code)	1
				ENVIRON	MENT	AL CONDITIONS		<u></u>		
	17. VISIBÎ 1. Da	LITY (ting	le entry/ 3. Dusk	1	C006	18. WEATHER (single entry)	-			
'F	2. Da		4. Dark			1, Clear 2, Cloudy 3	. Rain 4.	Fog 5. S	leet 8. Snow	
9. METHOD				OPE	OITAF	IAL DATA				
(place X in 1 Manual block	் ₄⊑	Automa	tic black]ر	Yard	rules 10 Aut	p. trein control	13 🗌	Other (specify)	
appropriate 2 Interlocking	5	Traffic	control	8	Time	table 11 Verb	nel permission			
box(es)) 3 Ceb signel	6	Auto. tr	ain stop	9[Radi	o 12 Trair	n anders			
O. SPEED (recorded speed, if eveilable) Est.	21. TRAIN	NUMBER				22. TIME TABLE DIRECTION		-		1
MPH Recorded						1. North 2. Sout	th 3. East	4. West		
					EQUIP	MENT				
3. TRAILING TONS (gross tonnege, excluding power units)	1. Fn	OF EQUIPS eight train seenger tra	1	18T (single enti 3. Mixed 4. Work t	trein 5	Single car 7. Yard/switch Cut of cars 5. Light locals		25. WAS THE EC IN ITEM 24 I	DUIPMENT IDENTIF UNATTENDED? 2. No	
S. TRACK NUMBER OF NAME			SSIFICATIO			28. ANNUAL TRACK DENSITY (po				10
								2. Yard	4. Indust	
0. PRINCIPLE CAR/UNIT	XXx.	in	itial and Nun	riber		30b. Pedition in Train		30e.	Loaded (yet or no)	
(1) First Involved (dentiled, struck, striking, etc.)		,				·				
(2) Causing (mechanical failures)										
11. LOCOMOTIVE UNITS (no. of)	. Head End	Mid 1 b. Manual	Frain c. Remeta	Rear E	-	32. CARS (NO. of)		Londed a. Freight b. Pun.	Empty c. Freight d. Pass.	e. Cabo
(1) Total in Train						(1) Total in Equipment	Consist			
(2) Total Dereiled						(2) Total Derailed				
	PR	OPERT	Y DAMAS	GE (estimat	ed cost	including labor, to repair or	replace)		<u>-l</u>	
3. EQUIPMENT DAMAGE		J. 4	\$			M. TRACK, SIGNAL, WAY AND ST		GE	[8	
(to be reported for this equipment co	nsist only	,				(to be reported by	y reilroed in item	3 only)	•	
			ACCIDE	NT/INCID	ENT C	USE CODE				
S. PRIMARY CAUSE CODE	38. CON	TRIBUTIN	G CAUSE	CODE		37. If no code available, explain cause.				
					ASUA					
E. NUMBER OF PERSONS INJURED		30	. ESTIMATI	ED TOTAL DA	YS DISA	BLITY	40. NUMBER OF	FATALITIES		
CREW	no. of)						HOURS	ON DUTY		
	II. CONDL	CTORS	44. (erakemen		45. ENGINEER Hrs: Mins:		44. CONDUC	roa Mins:	
7. TYPED NAME AND TITLE						48. SIGNATURE		ATTE:	49. DATE	
· · · · · · · · · · · · · · · · · · ·						# URA I VIE			49. DATE	
O. NARRATIVE DESCRIPTION - Describe the coule, net	we and circ	umstances	of accident/i	incident						

FORM FRA F 6180-64 (12-74) REPLACES FORM FRA F 6180-15 (11-72) WHICH IS OBSOLETE

APPENDIX B DATA RELATED TO DERAILMENT ANALYSIS

FLACEMENT OF HAZARDOUS MATERIALS CARS IN TRAIN CONSISTS
SUMMARY ANALYSIS OF DOT/FRA ACCIDENT/INCIDENT (RAIRS) DATA
FOR CALENDAR YEARS 1982 TO 1985
DERAILMENTS OF FREIGHT OR MIXED TRAINS, MAINLINE TRACK, ONLY
"UNITS" DEFINED AS LOCOMOTIVES + CARS + CABOOSE

NUMBER OF DERAILMENTS IN DATA FILE NUMBER OF DERAILMENTS USED IN ANALYSIS NUMBER OF TIMES FIRST UNIT FIRST INVOLVED TOTAL NUMBER OF UNITS IN CONSISTS AVERAGE LENGTH OF TRAIN TOTAL NUMBER OF DERAILED UNITS AVERAGE DERAILED UNITS PER TRAIN TOTAL NUMBER OF HAZ MAT CARS IN CONSISTS HAZ MAT CARS DAMAGED IN DERAILMENTS HAZ MAT CARS RELEASING IN DERAILMENTS TOTAL NUMBER OF PEOPLE EVACUATED		6425 5562 ** 489 451220 81 44823 8.1 4691 1223 289 23259
NUMBER OF FIRST UNITS DERAILED THAT ARE LOADED EMPTY NUMBER OF CAUSING UNITS THAT ARE LOADED EMPTY	= =	3703 1859 1103 400

DERAILMENT ANALYSIS BY TRAIN SPEED --

SPEED FROM	(MPH) TO	NUMBER TRAINS	TOTAL UNITS	AVERAGE LENGTH	DERAILED UNITS	DERAILED UNITS/TRAIN
o	o	7 0	6388	91	310	4.4
1	10	1766	127951	72	9290	5.3
11	20	1111	89850	80	7022	6.3
21	30	1194	98990	82	10474	8.8
31	40	736	69400	94	8657	11.8
41	50	450	40471	89	5880	13.1
51	60	203	16229	79	2763	43.6
61	70	32	1941	60	427	13.3

ECONOMIC IMPACT --

DAMAGE TO EQUIPMENT, TRAINS WITHOUT HAZMAT	~ = 390835050
DAMAGE TO TRACK, TRAINS WITHOUT HAZMAT	= 86282755
DAMAGE TO EQUIPMENT, TRAINS WITH HAZMAT	= 78246154
DAMAGE TO TRACK, TRAINS WITH HAZMAT	= 13722037

^{*}After removal of "bad data" items.

PLACEMENT OF HAZARDOUS MATERIALS CARS IN TRAIN CONSISTS
SUMMARY ANALYSIS OF DOT/FRA ACCIDENT/INCIDENT (RAIRS) DATA
FOR CALENDAR YEARS 1982 TO 1985
DERAILMENTS OF FREIGHT OR MIXED TRAINS, MAINLINE TRACK, ONLY
"UNITS" DEFINED AS LOCOMOTIVES + CARS + CABOOSE

DERAILMENT ANALYSIS BY TRAIN LENGTH: SHORT TRAINS (LESS THAN 11 LOCOMOTIVES/CARS) --

TOTAL NL AVERAGE TOTAL NL AVERAGE TRAINS W	OF DERAILMENTS LUMBER OF UNITS I LENGTH OF TRAIN JMBER OF DERAILE DERAILED UNITS WITH DERAILED/DA OF DERAILED HAZN	IN CONSISTS N ED UNITS PER TRAIN NMAGED/RELE			111 812 7 322 2.9 6
TRAINS OF 1	.1-50 LOCOMOTIVE	ES/CARS			
TOTAL NU AVERAGE TOTAL NU AVERAGE TRAINS W	OF DERAILMENTS L IMBER OF UNITS 1 LENGTH OF TRAIN IMBER OF DERAILE DERAILED UNITS DITH DERAILED/DA OF DERAILED HAZM	N CONSISTS I ID UNITS PER TRAIN MAGED/RELE		2 2 2 2	1188 38861 32 6900 5.8 93 242
ANALYSIS	BY THIRDS			×	
TRAIN 3RD	FIRST UNIT INVOLVED	PER CENT	CAUSING UNIT	TOTAL DERAILED	PER CENT
1 2 3	513 352 323	43.2 29.6 27.2	48 68 58	2323 2 56 3 2014	33.7 37.1 29.2
ANALYSIS	BY QUARTERS				
TRAIN 4TH	FIRST UNIT INVOLVED	PER CENT	CAUSING UNIT	TOTAL DERAILED	PER CENT
1 2 3 4	408 291 247 242	34.3 24.5 20.8 20.4	52 55 44 43	1630 2070 1712 1488	23.6 30.0 24.8 21.6

FLACEMENT OF HAZARDOUS MATERIALS CARS IN TRAIN CONSISTS
SUMMARY ANALYSIS OF DOT/FRA ACCIDENT/INCIDENT (RAIRS) DATA
FOR CALENDAR YEARS 1982 TO 1985
DERAILMENTS OF FREIGHT OR MIXED TRAINS, MAINLINE TRACK, ONLY
"UNITS" DEFINED AS LOCOMOTIVES + CARS + CABOOSE

DERAILMENT ANALYSIS BY TRAIN LENGTH: TRAINS OF 51-100 LOCOMOTIVES/CARS --

TOTAL NO AVERAGE TOTAL NO AVERAGE TRAINS O NUMBER O	DF DERAILMENTS L JMBER OF UNITS 1 LENGTH OF TRAIN JMBER OF DERAILE DERAILED UNITS WITH DERAILED/DA DF DERAILED HAZM	IN CONSISTS N ED UNITS PER TRAIN AMAGED/RELE	LYSIS ASING HAZMAT CARS	# # # # #	2468 187722 76 20549 8.3 199 546
	BY MINDS				
TRAIN	FIRST UNIT	PER	CAUSING	TOTAL	FER
3RD	INVOLVED	CENT	UNIT	DERAILED	CENT
•	940	70.	20.4	=	
1	940 816	38.1 33.1	221	6613 7877	32.E.
2 3	712	28.8	260 187	7876 6060	38.3 29.5
ANALYSIS	BY QUARTERS				
TRAIN	FIRST UNIT	PER	CAUSING	TOTAL	PER
4TH	INVOLVED	CENT	UNIT	DERAILED	CENT
			* * *		,
1	737	29.9	160	4732	23.0
2 3	593 455	24.0	186	5843	28.4
	65 3.	26.5	190	5894	28.7
4	485	19.7	132	4080	19.9

FLACEMENT OF HAZARDOUS MATERIALS CARS IN TRAIN CONSISTS
SUMMARY ANALYSIS OF DOT/FRA ACCIDENT/INCIDENT (RAIRS) DATA
FOR CALENDAR YEARS 1982 TO 1985
DERAILMENTS OF FREIGHT OR MIXED TRAINS, MAINLINE TRACK, ONLY
"UNITS" DEFINED AS LOCOMOTIVES + CARS + CABOOSE

DERAILMENT ANALYSIS BY TRAIN LENGTH: LONG TRAINS (GREATER THAN 100 LOCOMOTIVES/CARS) --

TOTAL NU AVERAGE TOTAL NU AVERAGE TRAINS	OF DERAILMENTS U JMBER OF UNITS I LENGTH OF TRAIN JMBER OF DERAILE DERAILED UNITS WITH DERAILED/DA JF DERAILED HAZM	N CONSISTS D UNITS PER TRAIN MAGED/RELE		= = = = = =	1795 223825 124 17052 9.5 146 421
ANALYSIS	BY THIRDS				
TRAIN 3RD	FIRST UNIT . INVOLVED	PER CENT	CAUSING UNIT	TOTAL DERAILED	PER CENT
1 2 3	710 599 486	39.6 33.4 27.1	266 220 144	6011 6439 4602	35.3 37.8 27.0
ANALYSIS	BY QUARTERS	•			i
TRAIN 4TH	FIRST UNIT	PER CENT	CAUSING UNIT	TOTAL DERAILED	PER CENT
1 2 3 4	5,45 473 426 351	30.4 26.4 23.7 19.6	197 190 144 99	4583 4456 4541 3272	26.9 27.3 26.6 19.2

PLACEMENT OF HAZARDOUS MATERIALS CARS IN TRAIN CONSISTS
SUMMARY ANALYSIS OF DOT/FRA ACCIDENT/INCIDENT (RAIRS) DATA
FOR CALENDAR YEARS 1982 TO 1985
DERAILMENTS OF FREIGHT OR MIXED TRAINS, MAINLINE TRACK, ONLY
"UNITS" DEFINED AS LOCOMOTIVES + CARS + CABOOSE

DERAILMENT ANALYSIS BY TRAIN LENGTH, A SUMMARY: ALL TRAINS OVER 10 LOCOMOTIVES/CARS IN LENGTH --

TOTAL N AVERAGE TOTAL N AVERAGE TRAINS	OF DERAILMENTS LUMBER OF UNITS I LENGTH OF TRAIN UMBER OF DERAILE DERAILED UNITS WITH DERAILED/DA OF DERAILED HAZM	N CONSISTS D UNITS FER TRAIN MAGED/RELE		# # # # #	5451 450408 82 44501 8.2 438 1209
ANALYSI	S BY THIRDS				
TRAIN	FIRST UNIT	PER	CAUSING	TOTAL	PER
3RD	INVOLVED	CENT	UNIT	DERAILED	CENT
1	2163	39.7	555	14947	33.6
2	1767	32.4	548	16878	37.9
3	1521	27.9	389	12676	28.5
ANALYSI	B BY QUARTERS				
TRAIN	FIRST UNIT	FER	CAUSING	TOTAL	PER
4TH		CENT	UNIT	DERAILED	CENT
1	1690	31.0	409	10945	24.6
2	1357	24.9	431	12569	28.2
3	1326	24.3	378	12147	27.3
4	1078	19.8	274	8840	19.9

NOTE: THIS SUMMARY IGNORES SHORT TRAINS (< 11 UNITS IN LENGTH)

APPENDIX C

REVIEW OF SELECTED RAILROAD ACCIDENTS INVOLVING MULTIPLE HAZARDOUS MATERIALS

REVIEW OF SELECTED RAILROAD ACCIDENTS INVOLVING MULTIPLE HAZARDOUS MATERIALS

E.R. Zamejc/H. P. Beneke

May 6, 1988

The following examples of train derailment accidents which resulted in hazardous materials mixing were reviewed:

• NTSB/HZM-87-01	(Miamisburg, Ohio, 7-8-86)
• NTSB/RAR-86-04	(Pine Bluff, Arkansas, 6-9-86)
• NTSB/RAR-83-05	(Livingston, Louisiana, (9-28-82)
 NTSB/HZM-80-01 	(Paxton, Texas, 6-8-75)

The review focused on whether or not mixing of the hazardous materials caused or worsened the consequences following the accident--i.e. fire and/or explosions. For the purpose of this analysis the following assumptions were made:

- Only binary combinations of chemicals were evaluated.
- All possible binary combinations were evaluated even though in some cases the tank cars remained intact and, as such, could not have contributed to the consequences. The purpose of this approach is to determine if the situation could have become even worse if the tank car leaked or ruptured.
- The thermal effects of fire on the stability of the chemical in tank cars was considered.

Results of the evaluations are given as follows:

NTSB/HZM-87/01

The tank cars of chemicals involved in this derailment and resultant fate are:

- Yellow phosphorus spilled and burned
- Molten sulfur spilled and mixed
- Tallow spilled and mixed

Anticipated binary chemical compatibilities are as follows:

- Yellow phosphorus/molten sulfur forms P₂S₅ with the generation of heat. P₂S₅ is the chemical used in the tips of friction matches and, as such, can be easily ignited by friction and less likely by heat (autoignition temperature = 527°F). Combustion of P₂S₅ forms toxic and corrosive fumes of SO₂ and P₂O₅.
- Yellow phosphorus/tallow no reaction expected. Note: tallow is an animal fat containing C16 to C18. It is combustible with a flash point of 509°F.
- Molten sulfur/tallow no reaction expected.

Thus, no incompatible reactions are expected. Other considerations are as follows:

- Yellow phosphorus, by itself, is pyrophoric which indicates it burns spontaneously in air at 86°F or above (no-ignition source required). Combustion of yellow phosphorus produces toxic P205.
- Tallow, because of its high flash point, could have prevented the fire if dumped on top of the phosphorus.
- Based on the observations, tallow was mixed with the molten sulfur but the mixture did not burn. The tallow may have minimized or prevented the molten sulfur from burning and, as such, may have mitigated the consequences.

In conclusion, mixing of the chemicals during this train derailment

did not make matters worse as compared with the consequences of spilling the chemicals without mixing, but rather, may have somewhat mitigated the consequences.

NTSB/RAR-86/04

The tank cars of chemicals involved in this derailment and resultant fate are:

- Vinyl chloride (2 cars) insulated, remained intact, saved by fire fighters
- Polymethylene polyphenylisocyanate (2 cars) one car exploded, the other burned in resulting fire
- Polyethylene/polypropylene pellets (4 cars) burned in fire
- Butyl acrylate (2 cars) spilled and burned in fire
- Acrylic acid disposition unknown
- Ethylene oxide exploded in fire

The polyethylene/polypropylene pellets are chemically inert in that they have no residual reactive groups so incompatible reactions are not possible with the other chemicals involved in the derailment. Other anticipated binary chemical compatibilities are as follows:

- Vinyl chloride/polymethylene polyphenylisocyanate no reaction expected.
- Vinyl chloride/butyl acrylate no reaction expected.
- Vinyl chloride/acrylic acid vinyl chloride would dissolve in acrylic acid which could initiate vigorous polymerization if the mixture is concentrated in either of the components. This would result in the generation of heat (probably insufficient amount to initiate a fire). However, the low boiling point of vinyl chloride (-14°C) reduces the potential risk. This is because in an unconfined (i.e. open) liquid pool, such as would result following a tank rupture, a majority of the vinyl chloride would vaporize

before it could react.

- Vinyl chloride/ethylene oxide no reaction expected.
- Polymethylene polyphenylisocyanate/butyl acrylate no reaction expected.
- Polymethylene polyphenylisocyanate/acrylic acid addition reaction causing generation of some heat (probably insufficient amount to initiate a fire).
- Polymethylene polyphenylisocyanate/ethylene oxide no reaction expected.
- Butyl acrylate/acrylic acid butyl acrylate would dissolve in acrylic acid which could initiate polymerization if the mixture is concentrated in either of the components. This would result in the generation of heat (probably insufficient amount to initiate a fire).
- Butyl acrylate/ethylene oxide no reaction expected.
- Acrylic acid/ethylene oxide ethylene oxide could dissolve in acrylic acid (or vice versa) which would initiate acid catalyzed polymerization of ethylene oxide (potentially violent). However, the low boiling point of ethylene oxide (11°C) reduces the potential risk. The polymerization reaction would generate heat that may initiate a fire.

Thus, polymerization reactions resulting in the formation of heat may have occurred during the derailment. It is possible that the heat generated from these reactions initiated a fire as well as caused formation of flammable vapors which intensified the fire. For example, ethylene oxide vapors are heavier than air and may flash back to their origin upon ignition from a distant source away from the spill. Other considerations are as follows:

Polymethylene polyphenylisocyanate is shipped as a 50 wt.% solution of methylene diphenylisocyanate plus 50 wt.% polymer (CHRIS Hazardous Chemical Data). If the tank cars containing this mixture are exposed to excessive heat such as in the case of a fire then violent and rapid polymerization could occur. This could result in the rupture of the container (i.e. explosion) because the rate of pressure rise exceeds the capability of the relief valve's venting rate. Also, relief valves and other safety venting devices may be initially successful in relieving the over-pressure caused by the polymerization reaction but may become blocked during the course of venting. Blockage may be caused by deposition of reaction products or if the tank car had rolled over onto its side. When blocked, the container is likely to explode especially if exposed to a fire.

• The ethylene oxide explosion was probably caused by a runaway thermal decomposition reaction. Initiation of an ethylene oxide runaway can occur when it is heated to a temperature of 200°C or above in a closed system¹.

In conclusion, mixing of the chemicals during this train derailment may have worsened the consequences as compared with the consequences of spilling the chemicals without mixing. Also, because of the potential for thermal instability of chemicals such as ethylene oxide and polymerizable chemicals when heated in confined containers such as tank cars, segregation of these chemicals from other flammable chemicals in a train should be considered.

NTSB/RAR-83/05

The tank cars of chemicals involved in this derailment and resultant fate are:

- Vinyl chloride (1,241,000 lb) consumed by fire, one car exploded
- Styrene monomer (176,000 lb) consumed by fire
- Motor fuel anti-knock compound (75,000 lb) consumed by fire, one car exploded (Note: this was assumed to be tetraethyl lead.)
- Toluene diisocyanate (23,000 lb) consumed by fire

^{1 &}quot;Thermal Stability and Deflagration of Ethylene Oxide" by L.G. Britton, Paper 62d, AIChE Spring National Meeting, New Orleans, March 8, 1988.

- Phosphoric acid (2,100,000 lb) spilled and adsorbed by ground
- Hydrofluosilicic acid (200,000 lb) spilled and adsorbed by ground
- Sodium hydroxide (195,000 lb) spilled and adsorbed by ground
- Perchloroethylene (190,000 lb) spilled and adsorbed by ground
- Ethylene glycol (194,000 lb) spilled and adsorbed by ground

The anticipated binary chemical compatibilities are as follows:

- Vinyl chloride/styrene no reaction expected.
- Vinyl chloride/tetraethyl lead Tetraethyl lead, as an organometallic compound, could initiate a potentially violent polymerization of vinyl chloride. This would result in the generation of heat (may be sufficient amount to initiate a fire). However, the high volatility of vinyl chloride would minimize the consequences because most of the spilled vinyl chloride would rapidly vaporize when spilled.
- Vinyl chloride/toluene diisocyanate no reaction expected.
- Vinyl chloride/phosphoric acid Phosphoric acid could initiate the
 polymerization of vinyl chloride if an appreciable quantity
 dissolves in phosphoric acid. Same consequences as in "vinyl
 chloride/tetraethyl lead" but less likely.
- Vinyl chloride/hydrofluosilicic acid Same reaction and consequences as in "vinyl chloride/phosphoric acid".
- Vinyl chloride/sodium hydroxide Violent polymerization could result if the mixture is concentrated in either of the components. This would result in the generation of heat (probably insufficient amount to initiate a fire). The consequences would be minimized by the high volatility of vinyl chloride and low solubility of vinyl chloride in aqueous sodium hydroxide solution.
- Vinyl chloride/perchloroethylene no reaction expected.
- Vinyl chloride/ethylene glycol no reaction expected.
- Styrene/tetraethyl lead Tetraethyl lead, as an organometallic compound, could initiate a potentially violent anionic polymerization of styrene. This would result in the generation of heat (may be sufficient amount to initiate a fire).

- Styrene/toluene diisocyanate no reaction expected.
- Styrene/phosphoric acid Phosphoric acid could initiate a potentially violent or explosive cationic polymerization of styrene. A fire may be initiated.
- Styrene/hydrofluosilicic acid Same reaction and consequences as in "styrene/phosphoric acid".
- Styrene/sodium hydroxide Violent polymerization could result if the mixture is concentrated in either of the components. This would result in the generation of heat (may be sufficient amount to initiate a fire).
- Styrene/perchloroethylene no reaction expected.
- Styrene/ethylene glycol no reaction expected.
- Tetraethyl lead/toluene diisocyanate no reaction expected.
- Tetraethyl lead/phosphoric acid potential liberation of ethane gas and resulting fire.
- Tetraethyl lead/hydrofluosilicic acid Same reaction and consequences as in "tetraethyl lead/phosphoric acid".
- Tetraethyl lead/sodium hydroxide no reaction expected.
- Tetraethyl lead/perchloroethylene possible addition reaction or polymerization with the generation of heat.
- Tetraethyl lead/ethylene glycol no reaction expected.
- Toluene diisocyanate/phosphoric acid Rapid hydrolysis to the amine if water is present with subsequent reaction with remaining isocyanate functionality. Reaction may be violent.
- Toluene diisocyanate/hydrofluosilicic acid Same reaction and consequences as in "toluene diisocyanate/phosphoric acid".
- Toluene diisocyanate/sodium hydroxide nearly explosive hydrolysis of isocyanate functionality and generation of CO2.
- Toluene diisocyanate/perchloroethylene no reaction expected.
- Toluene diisocyanate/ethylene glycol potentially explosive polymerization of both reagents to form polyurethane.
- Phosphoric acid/hydrofluosilicic acid no reaction expected.
- Phosphoric acid/sodium hydroxide violent neutralization reaction.
- Phosphoric acid/perchloroethylene no reaction expected.
- Phosphoric acid/ethylene glycol exothermic elimination reaction

- and probable formation of ethylene oxide (fire hazard).
- Hydrofluosilicic acid/sodium hydroxide violent neutralization reaction.
- Hydrofluosilicic acid/perchloroethylene no reaction expected.
- Hydrofluosilicic acid/ethylene glycol exothermic elimination reaction and probable formation of ethylene oxide (fire hazard).
- Sodium hydroxide/perchloroethylene potential exothermic addition reaction or polymerization reaction.
- Sodium hydroxide/ethylene glycol slightly exothermic acid/base reaction.
- Perchloroethylene/ethylene glycol no reaction expected.
 Reactions, some of which are violent, may have occurred during the derailment. The resultant formation of considerable heat and flammable products (i.e. ethane gas) could have contributed to the initiation and intensity of the fire. Other considerations are as follows:
 - The explosion of the tank car of tetraethyl lead was probably caused by exposure to the intense fire. Tetraethyl lead begins to decompose above 230°F forming ethane gas (CHRIS Hazardous Chemical Data). If the decomposition reaction occurred rapidly then the relief valves could have been unable to prevent the pressure from exceeding the burst pressure of the tank car (500 psig for DOT Spec. 105A100W). The estimated maximum pressure rise assuming a 34,500 gallon capacity tank car containing 75,000 pounds of 50 weight percent tetraethyl lead in solvents (e.g. ethylene dichloride, ethylene dibromide, and toluene) and assuming complete decomposition of the tetraethyl lead at 230°F is about 745 psig. Relief valves may not be able to prevent the burst pressure from being exceeded because of the rapidity of the decomposition reaction.
 - Vinyl chloride could violently and rapidly polymerize when exposed to excessive heat such as in the case of a fire. This could result in the rupture of the container (i.e. explosion) because the rate of pressure rise exceeds the capability of the relief valve's venting rate².

In conclusion, mixing of the chemicals during this train derailment may have worsened the consequences as compared with the consequences of spilling the chemicals without mixing. Also, because of the potential for thermal instability of chemicals such as tetraethyl lead and polymerizable chemicals such as vinyl chloride when heated in confined containers such as tank cars, segregation of these chemicals from other flammable chemicals in a train should be considered.

NTSB/HZM-80/01

The tank cars of chemicals involved in this derailment and resultant fate are:

- Isobutylene (1 car) breached by fire and burned.
- Butadiene (4 cars) 1 car released and burned, 1 car exposed to fire and survived, 2 cars survived.
- Tetrahydrofuran (1 car) violent rupture in fire.
- Hydrogen fluoride (1 car) survived.
- Propylene glycol (3 cars) leaked.
- Dibasic ester (1 car) survived.
- Ethylene oxide (1 car) violent rupture in fire.
- Vinyl acetate (2 cars) 1 car spilled and burned, the other survived.
- Ethylene glycol (1 car) spilled.
- Methanol (1 car) leaked and burned.
- Ethyl acrylate (1 car) spilled and burned.
- Acetaldehyde (4 cars) spilled and burned.
- Plastic pellets (3 cars) disposition unknown.

² Another incident involving an explosion of a tank car of vinyl chloride is reported in "Vinyl Chloride Tank Car Incident", by R. D. Kogler, Loss Prevention Vol. 5 (1971), p26-28.

Rubber (1 car) - disposition unknown.

The plastic pellets and rubber are chemically inert in that they have no residual reactive groups so incompatible reactions are not possible with the other chemicals involved in the derailment. Other anticipated binary chemical compatibilities are as follows:

- Isobutylene/butadiene no reaction expected.
- Isobutylene or butadiene/tetrahydrofuran no reaction expected.
- Isobutylene or butadiene/hydrogen fluoride Both components are very volatile but hydrogen fluoride could initiate the polymerization of isobutylene or butadiene if the temperature was below the individual boiling points of 20°F for isobutylene, 24°F for butadiene, and 67°F for hydrogen fluoride. This would result in the generation of heat (probably insufficient amount to initiate a fire). The consequences would be minimized by the high volatility of isobutylene, butadiene and hydrogen fluoride.
- Isobutylene or butadiene/propylene glycol no reaction expected.
- Isobutylene or butadiene/dibasic ester (e.g. dialkyl phthalates) no reaction expected.
- Isobutylene or butadiene/ethylene oxide no reaction expected.
- Isobutylene or butadiene/vinyl acetate no reaction expected.
- Isobutylene or butadiene/ethylene glycol no reaction expected.
- Isobutylene or butadiene/methanol no reaction expected.
- Isobutylene or butadiene/ethyl acrylate no reaction expected.
- Isobutylene or butadiene/acetaldehyde no reaction expected.
- Tetrahydrofuran/hydrogen fluoride no reaction expected.
- Tetrahydrofuran/propylene glycol no reaction expected.
- Tetrahydrofuran/dibasic ester (e.g. dialkyl phthalates) no reaction expected.
- Tetrahydrofuran/ethylene oxide no reaction expected.
- Tetrahydrofuran/vinyl acetate no reaction expected.
- Tetrahydrofuran/ethylene glycol no reaction expected.
- Tetrahydrofuran/methanol no reaction expected.
- Tetrahydrofuran/ethyl acrylate no reaction expected.

- Tetrahydrofuran/acetaldehyde no reaction expected.
- Hydrogen fluoride/propylene glycol potentially violent and exothermic elimination reaction with formation of propylene oxide (fire hazard).
- Hydrogen fluoride/dibasic ester (e.g. dialkyl phthalates) no reaction expected.
- Hydrogen fluoride/ethylene oxide potentially violent polymerization of ethylene oxide. However, consequences are minimized because of high volatility of each component.
- Hydrogen fluoride/vinyl acetate potentially violent polymerization of vinyl acetate.
- Hydrogen fluoride/ethylene glycol Same as in "hydrogen fluoride/ propylene glycol".
- Hydrogen fluoride/methanol potentially violent and exothermic formation of methyl fluoride (toxic and fire hazard).
- Hydrogen fluoride/ethyl acrylate potentially violent polymerization of ethyl acrylate.
- Hydrogen fluoride/acetaldehyde violent polymerization and aldol condensation of acetaldehyde.
- Propylene glycol/dibasic ester (e.g. dialkyl phthalates) no reaction expected.
- Propylene glycol/ethylene oxide slow ring opening polymerization or addition of ethylene oxide.
- Propylene glycol/vinyl acetate no reaction expected.
- Propylene glycol/ethylene glycol no reaction expected.
- Propylene glycol/methanol no reaction expected.
- Propylene glycol/ethyl acrylate no reaction expected.
- Propylene glycol/acetaldehyde violent, possibly explosive, formation of acetal.
- Dibasic ester/ethylene oxide no reaction expected.
- Dibasic ester/vinyl acetate no reaction expected.
- Dibasic ester/ethylene glycol no reaction expected.
- Dibasic ester/methanol no reaction expected.
- Dibasic ester/ethyl acrylate no reaction expected.
- Dibasic ester/acetaldehyde slow condensation reaction.

- Ethylene oxide/vinyl acetate no reaction expected.
- Ethylene oxide/ethylene glycol potentially slow ring opening polymerization or addition of ethylene oxide.
- Ethylene oxide/methanol potentially slow ring opening polymerization or addition of ethylene oxide.
- Ethylene oxide/ethyl acrylate no reaction expected.
- Ethylene oxide/acetaldehyde slow formation of cyclic acetal.
- Vinyl acetate/ethylene glycol no reaction expected.
- Vinyl acetate/methanol no reaction expected.
- Vinyl acetate/ethyl acrylate no reaction expected.
- Vinyl acetate/acetaldehyde slow condensation is expected in absence of a strong base giving a beta-hydroxy ester or its hydration product.
- Ethylene glycol/methanol no reaction expected.
- Ethylene glycol/ethyl acrylate no reaction expected.
- Ethylene glycol/acetaldehyde violent or explosive reaction to form the cyclic acetal.
- Methanol/ethyl acrylate no reaction expected.
- Methanol/acetaldehyde violent or explosive reaction to form the cyclic acetal.
- Ethyl acrylate/acetaldehyde Michael additions or Claisen condensations are expected to be slow in absence of bases.

Reactions, some of which are violent, may have occurred during the derailment that resulted in the formation of heat and flammable gaseous products. These reactions could have contributed to the initiation and intensity of the fire. Of primary concern are the reactions of acetaldehyde with acids (e.g. hydrogen fluoride), glycols (e.g. propylene glycol and ethylene glycol), and alcohols (e.g. methanol) and the reactions of acids such as hydrogen fluoride with ethylene oxide, vinyl acetate, glycols, and methanol. These combinations should be avoided by segregation of tank cars. Other considerations are as follows:

The explosion of the tank car of tetrahydrofuran was probably

caused by exposure to the intense fire. Tetrahydrofuran, in the absence of inhibitors, is subject to auto-oxidation on exposure to air forming hydroperoxides. If the peroxides, which are thermally sensitive, are allowed to accumulate then an explosion hazard will be formed. Although stabilizers are normally added to tetrahydrofuran to prevent the peroxide formation, this particular batch may not have had sufficient stabilizer added or may have been stored for a prolonged period thereby allowing peroxides to form. Upon concentration, the peroxides will explode (CHRIS Hazardous Chemical Data). A second possible cause for the explosion is that the tank car had fallen over on its side such that the relief valve would be blocked. In this scenario, the tank car could explode by the vapor pressure increase as a result of heating the tank car in the fire.

 Explosion of tank cars containing ethylene oxide has been previously discussed.

In conclusion, mixing of the chemicals during this train derailment may have worsened the consequences as compared with the consequences of spilling the chemicals without mixing. Also, because of the potential for thermal instability of chemicals that can form organic peroxides (e.g. tetrahydrofuran), segregation of these chemicals from other flammable chemicals in a train should be considered.

APPENDIX D

U.S. DEPARTMENT OF TRANSPORTATION HAZMAT CAR PLACEMENT REGULATIONS

49 CFR Ch. I (10-1-87 Edition)

- § 174.81 Segregation and separation requirements for hazardous materials in rail cars.
- (a) Charged electric storage batteries must not be loaded in the same car nor stored with any Class A explosive.
- (b) Cyanides or cyanide mixtures must not be loaded or stored with acids or corrosive liquids.
- (c) Gas identification sets may be loaded and transported with all articles named in the segregation and separation chart, except those in column c.
- (d) Nitric acid, when loaded in the same car with other acids or other corrosive liquids in carboys, must be separated from the other carboys. A 2 by 6 inch plank, set on edge, should be

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- nailed across the floor at least 12 inches from the nitric acid carboys, and the space between the plank and the carboys of nitric acid should be filled with sand, sifted ashes, or other incombustible absorbent material.
- (e) Smokeless powder for small arms in quantities not exceeding 100 pounds net weight in one car shall be classed as flammable solid for purposes of transportation when examined for this classification by the Bureau of Explosives and approved by the Director, OHMT.
- (f) Hazardous materials may not be loaded, transported, or stored together, except as provided in the following table:

Segregation and Separation Chart							Low explosives or black powder, a									
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Poisonous gases er liquids in tank car tanks, projectives or bombs, poison gas labels	57****** /15/ X		الثا	171	1 1	11	1	1 17	1 ^	1 ^	ו^.	J ^	1	L^1		J

(49 U.S.C. 1803, 1804, 1808; 49 CFR 1.53, App. A to Part 1)
[Amdt. 174-45, 49 FR 7385, Feb. 29, 1984; 49 FR 9188, Mar. 12, 1984, as amended by Amdt. 174-46, 50 FR 806, Jan. 7, 1985; Amdt. 174-59, 51 FR 5974, Feb. 18,

Subpart D—Handling of Placarded Cars

§ 174.83 Switching of cars containing hazardous materials.

- (a) In switching operations where the use of hand brakes is necessary, a loaded placarded tank car, or a draft which includes a loaded placarded tank car, may not be cut off until the preceding car or cars clear the ladder track and the draft containing the loaded placarded tank car, or a loaded placarded tank car, shall in turn clear the ladder before another car is allowed to follow. In switching operations where hand brakes are used, it must be determined by trial whether a loaded placarded car, or a car occupied by a rider in a draft containing a placarded car, has its hand brakes in proper working condition before it is cut off.
- (b) Any car placarded "EXPLOSIVE A" or "POISON GAS" and any Class DOT-113 tank car placarded "FLAM-MABLE GAS" may not be:
 - (1) Cut off while in motion,
- (2) Coupled into with more force than is necessary to complete the coupling, or
- (3) Struck by any car moving under its own momentum.
- (c) When transporting a car placarded "EXPLOSIVES A" in a terminal, yard, or on a side track, or siding, it must be separated from the engine by at least one non-placarded car.
- (d) The doors of each closed car placarded "EXPLOSIVES A" must be closed, securely fastened, and the lading securely braced before it is moved.
- (49 U.S.C. 1803, 1804, 1808; 49 CFR 1.53, App. A to Part 1)

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26A, 41 FR 40685, Sept. 20, 1976; Amdt. 74-43, 48 FR 27699, June 16, 1983; 48 FR 50440, 50442, Nov. 1, 1983]

- § 174.84 Switching of flatcars carrying placarded trailers, freight containers, portable tanks or IM portable tanks.
- (a) A placarded flatcar or a flatcar carrying a placarded trailer, freight container, portable tank or IM porta-

ble tank under this subchapter may not be cut off while in motion.

- (b) No rail car moving under its own momentum may be permitted to strike any placarded flatcar or any flatcar carrying a placarded trailer, freight container, portable tank or IM portable tank.
- (c) No placarded flatcar or any flatcar carrying a placarded trailer, freight container, portable tank or IM portable tank may be coupled into with more force than is necessary to complete the coupling.

[Amdt. 174-40, 46 FR 9894, Jan. 29, 1981]

§ 174.85 Placement of freight cars placarded "EXPLOSIVES A" in yards, on sidings, or side tracks.

A rail car placarded "EXPLOSIVES A" while in a yard or on a siding or side track must be placed so that it will be safe from all probable danger of fire. A car so placarded may not be placed under a bridge or overhead highway crossing, not in or alongside a passenger shed or station except for loading or unloading purposes.

§ 174.86 Position in train of cars placarded "EXPLOSIVES A" or "POISON GAS" when accompanied by cars carrying guards or technical escorts.

A rail car placarded "EXPLOSIVES A" or "POISON GAS" in a moving or standing train must be next to and ahead of any car occupied by the guards or technical escorts accompanying this car. However, if a car occupied by guards or technical escorts is equipped with a lighted heater or stove, it must be the fourth car behind any car requiring "EXPLOSIVES A" placards.

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26A, 41 FR 40685, Sept. 20, 1976]

- § 174.87 Placarded cars prohibited in passenger trains, limited in mixed trains.
- (a) A placarded rail car may not be transported in a passenger train. However, it may be transported in a mixed train, but only at such times and between such points that freight train service is not in operation and subject to the following limitations:

- (1) A placarded car (other than one placarded "COMBUSTIBLE") may not be transported next to an occupied caboose or a car carrying passengers in mixed trains, except as provided in § 174.86.
- (2) When a car containing hazardous materials requiring labels is moved in a mixed train and it is not occupied by an employee of the carrier, placards must be applied to the car as required by Subpart F of Part 172 of this subchapter.

§ 174.88 Position in train of cars placarded "EXPLOSIVES A".

In a moving or standing train, a car placarded "EXPLOSIVES A" may not be placed nearer than the sixth car from the engine or an occupied caboose. However, when the length of the train will not permit this car to be so placed, it must be placed as near the middle of the train as possible, but not less than the second car from the engine or occupied caboose.

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26B, 41 FR 57071, Dec. 30, 1976]

§ 174.89 Position in train of cars placarded "RADIOACTIVE".

In a moving or standing train, a car placarded "RADIOACTIVE" may not be placed next to any other loaded placarded car (other than one placarded "COMBUSTIBLE"), an engine, occupied caboose, or carload of undeveloped film. Cars placarded "RADIOACTIVE" may be placed next to each other.

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26A, 41 FR 40685, Sept. 20, 1976; Amdt. 174-26B, 41 FR 57071, Dec. 30, 1976]

- § 174.90 Separating cars placarded "EX-PLOSIVES A" or "POISON GAS" from other cars in trains.
- (a) In a moving or standing train, a car placarded "EXPLOSIVES A" or "POISON GAS" may not be placed next to:
- (1) A passenger car or combination car that may be occupied except as provided in § 174.86;
- (2) Any loaded placarded car other than a car placarded with the same

placard or one placarded "COMBUS-TIBLE".

- (3) An engine;
- (4) A wooden underframe car (except on narrow gauge railroads);
- (5) A loaded flatcar, except that loaded cars placarded "EXPLOSIVES A" may be placed next to each other. A flatcar equipped with permanently attached ends of rigid construction is considered to be an open-top car. (See paragraph (a)(6) of this section.)
- (6) An open-top car when any of the lading protrudes beyond the car ends or when any of the lading extending above the car ends is liable to shift so as to protrude beyond the car ends;
- (7) A car with automatic refrigeration or heating apparatus in operation, or a car with open-flame apparatus in service, or with an internal combustion engine in operation;
- (8) A car containing lighted heaters, stoves, or lanterns;
- (9) A car occupied by any person, including any attendant for the cargo contained therein; or
- (10) An occupied caboose, except as provided in § 174.86.
- (b) In a moving or standing train, a car placarded "EXPLOSIVES A" may not be placed next to a car placarded "POISON GAS".

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26A, 41 FR 40685, Sept. 20, 1976]

§ 174.91 Position in train of loaded placarded tank car other than car placarded "COMBUSTIBLE".

Except for a tank car placarded "COMBUSTIBLE", a loaded placarded tank car in a moving or standing train may not be nearer than the sixth car from the engine, occupied caboose, or passenger car. However, when the length of the train will not permit a loaded placarded tank car to be so placed, it must be placed as near the middle of the train as possible and not nearer than the second car from the engine, occupied caboose, or passenger car.

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26A, 41 FR 40685, Sept. 20, 1976]

- § 174.92 Separating loaded placarded tank cars other than cars placarded "COM-BUSTIBLE" from other cars in trains.
- (a) In a moving or standing train a loaded placarded tank car, other than one placarded "COMBUSTIBLE", may not be placed next to:
- (1) A passenger car or combination car, other than a car occupied by technical escorts and authorized personnel accompanying shipments;
- (2) Any car placarded "EXPLOSIVES A", "RADIOACTIVE", or "POISON GAS":
 - (3) An engine or occupied caboose;
- (4) A wooden underframe car (except on narrow gauge railroads):
- (5) A loaded flatcar, other than a specially equipped car in trailer-on-flatcar or container-on-flatcar service or a flatcar loaded with vehicles secured by means of a device designed for that purpose and permanently installed on the flatcar, and of a type generally accepted for handling in interchange between railroads subject to the following:
- (i) A flatcar equipped with permanently attached ends of rigid construction is considered to be an open-top car (see paragraph (a)(6) of this section); and
- (ii) This exception for cars in traileron-flatcar service does not apply to loaded flatbed trucks, loaded flatbed trailers, loaded open-top trailers, or loaded trucks or trailers without securely closed doors;
- (6) An open-top car when any of the lading protrudes beyond the car ends or when any of the lading extending above the car ends is liable to shift so as to protrude beyond the car ends;
- (7) A car with automatic refrigeration or heating apparatus in operation or a car with open-flame apparatus in service or with an internal combustion engine in operation;
- (8) A car occupied by any person, including any attendant for the cargo contained therein.

[Amdt. 174-26, 41 FR 16092, Apr. 15, 1976, as amended by Amdt. 174-26A, 41 FR 40685, Sept. 20, 1976]

§ 174.93 Position in train of a tank car displaying RESIDUE placards.

Except for a tank car placarded COMBUSTIBLE—RESIDUE, a tank

car displaying RESIDUE placards in a moving or standing train may not be placed nearer than the second car from an engine or occupied caboose.

[Amdt. 174-60, 51 FR 23079, June 25, 1986]

APPENDIX E

CANADIAN TRANSPORT COMMISSION HAZMAT CAR PLACEMENT REGULATIONS

SUBPART E—HANDLING OF CARS BY CARRIERS

- § 74.575 Restrictions. (a) When local conditions make the acceptance, transportation, or delivery of commodities hazardous, local restrictions may be imposed by the carrier.
 - (b) All carriers must report to the Commission, for publication, full information as to restrictions which may be imposed against the acceptance, delivery, or transportation of dangerous commodities, over any portion of their lines.
- § 74.578 Flammable vapors. (a) A box car or container car placarded flammable liquid, division 3.1 or 3.2 or known to contain flammable liquids, must not be entered with a lighted open-flame lantern, torch, or other fire, until both car doors have been opened and sufficient time allowed for ventilation and escape of any vapors.
- § 74.580 Violations must be corrected. (a) All violations discovered must be corrected before forwarding shipments.
- § 74.581 Routing shipments. (a) Before any shipment of explosives, division 1.1 or division 1.2, destined to a point beyond the lines of the initial carrier, is accepted from a shipper, the initial carrier must ascertain that the shipment can go forward by the route designated, and to avoid delays en route the initial carrier must be in possession of full rate information, before forwarding the shipment.
- § 74.582 Movement to be expedited. (a) Carriers must forward shipments of dangerous commodities promptly and within 48 hours. Saturdays. Sundays, and holidays excluded after acceptance at originating point or receipt at any yard, transfer station, or interchange point, except that where bi-weekly or weekly service only is performed, shipments of dangerous commodities must be forwarded on the first available train.
 - (b) No car shall be received and held at any point so as to defeat the purpose of this section or of § 74.560.
- **‡ 74.589 Handling cars.** (a) Definitions. (1) "Person" means any individual, partnership, corporation, association, joint stock company, business trust or other organized group of persons, or any trustee, receiver, assignee, or personal representative, and includes any department or agency of the Government of Canada, any Province, or any other political, governmental or legal entity;
- (2) "Railroad" means any person engaged in transportation as a common carrier by rail and includes its agent or employees;
- (3) "Engine" means any locomotive, propelled by any form of energy, used by a railroad;
- (4) "Freight car" means any vehicle used for the transportation of property by rail;
- (5) "Passenger car" means any vehicle used for the transportation of passengers by rail;
- (6) "Combination car" means any vehicle used for the transportation of both property and passengers by rail;
- (7) "Occupied caboose" means any vehicle used by railroad employees, caretakers, or other authorized to ride therein;
- (8) "A train" is one or more engines coupled together with or without cars displaying markers;
- (9) "Freight train" means one or more engines coupled with one or more freight cars, displaying markers;

- (10) "Passenger train" means one or more engines coupled with one or more passenger cars carrying passengers, displaying markers;
- (11) "Mixed train" means one or more engines coupled with one or more freight cars and passenger cars carrying passengers, displaying markers;
 - (b) Placards on cars.
- (1) Cars, containers and trailers requiring car certificates pursuant to § 74.525 (d) and placards pursuant to the Transportation of Dangerous Goods Regulations shall not be transported unless they are at all times placarded and certified as required. Placards and car certificates lost in transit shall be replaced at the next inspection point, and those not required shall be removed at the next terminal where the train is classified.
- (2) At points where trains are inspected, cars placarded with a placard with a square background and adjacent cars shall be inspected. Such cars shall continue in transit only when inspection shows them to be in condition for safe transportation.
 - (c) Switching of cars containing dangerous commodities.
- (1) In switching operations where the use of hand brakes is necessary, a loaded, placarded tank car, or a draft which includes a loaded, placarded tank car shall not be cut off until the preceding car or cars clear the ladder track, and the draft containing the loaded, placarded tank car, or a loaded, placarded tank car, shall in turn clear the ladder before another car is allowed to follow. In switching operations where hand brakes are used, it must be determined by trial whether a loaded, placarded car, or a car occupied by a rider in a draft containing a placarded car has its hand brakes in proper working condition before it is cut off.
- (2) No car bearing a placard with a square background shall be cut off while in motion or coupled into with more force than is necessary to complete the coupling. No car moving under its own momentum shall be allowed to strike any car placarded with a placard with a square background.
- (3) When transporting a car placarded with a placard with a square background in a terminal, or yard, or on a side track or siding, the car shall be separated from the engine by at least one non-placarded car.
- (4) The doors of each closed car placarded with a placard with a square background shall be closed, securely fastened, and the lading securely braced before the car is moved.
- (5) A placarded flatcar or a flatcar carrying a placarded trailer or freight container shall not be cut off while in motion.
- (6) No rail car moving under its own momentum shall be permitted to strike any placarded flatcar or any flatcar carrying a placarded trailer or freight container.
- (7) No placarded flatcar or flatcar carrying a placarded trailer or freight container shall be coupled into with more force than is necessary to complete the coupling.
- (8) No spec. 112 or 114 tank car shall be coupled with any other car at a relative coupling speed in excess of four miles per hour, unless
 - (i) the tank is empty, and
 - (ii) the tank has been cleaned and purged.
 - (d) Placement of freight cars placarded with a placard with a square background in yards, on sidings, or side tracks. A rail car placarded with a placard with a square background shall be placed while in a yard, on a siding or side track so that it

will be safe from all probable danger of fire. A car so placarded shall not be placed under a bridge or overhead highway crossing, nor in or alongside a passenger shed or station.

- (e) Notice to train crews of placarded cars.
- (1) Train crews shall have in their possession a document indicating the position in the train of each loaded placarded car containing dangerous commodities, except when the position is changed or the placarded car is placed in the train by a member of the train crew. A train consist may be used to meet this requirement.
- (2) One member of a train crew of a train transporting dangerous commodities shall have in his possession a copy of all Emergency Response Forms required pursuant to § 73.429.
 - (f) Position of cars in trains. A car carrying dangerous goods shall not be marshalled next to a car described in a column other than column 3 or 4 of the table shown on page 173 if an X appears in that column on the line corresponding to type of car carrying the dangerous goods (as described in column 1) and the placard groups of the dangerous goods (as indicated in column 2).

POSITION IN FREIGHT OR MIXED TRAIN OF CARS CONTAINING DANGEROUS COMMODITIES

		3	4	5	6	7	1	9	10	н			
-	,	When train length permits must not be nearer than 6th from engine, occupied	When train length does not permit must be near middle of train but not nearer than 2nd from engine. occupied	Open-top car when lading protrudes beyond car or when lading		Any car piggyback or container with automatic heating or refrigera- tion, lighted heater, stoves							
1	2	caboose	caboose or	above car	Loaded	lanterns or		_		•.			
		or	occupied	end is	flat	internal		Car Placarded					
Type of Car	Placard group no.	occupied car	car (d)	liable to shift	CBT (a)	combustion	Group 1	Group 2	Group 3	Group 4			
Any car (includes Flat cars, carrying Trailers, or Containers)	Group 1	x	x	x	x	x		x	x	x			
Tank Cars	Group 2	Х(б)	X(b)	х	х	х .	х	(c)	×	x			
Other than Tank Cars	Group 2						х		х	х			
Any Car	Group 3	х	х	x	х	x	x	x		х			
Any Car	Group 4	х	х				x	х	х				
Tank Car	Group 5	Must not be placed next to engine or occupied caboose (b)											
Amy Car	Group 6	NO MARSHALLING REQUIRED											

PLACARD GROUP

- 1. Group 1 consists of Division 1.1 and 1.2
- 2. Group 2 consists of Divisions 1.3, 1.4, 1.5, 2.1, 2.2, 2.3, 2.4, 3.1, 3.2, 4.1, 4.2, 4.3, 5.1, 5.2, 6.1, 6.2, and Class 8.
- Group 3 consists of Cyanogen Chloride, Hydrogen Cyanide, Nitric Oxide, Phosgene, Nitrogen Dioxide, Phosphine, Diborane or Diborane Mixtures, Arsine, Boron Trifluoride, Carbonyl Sulfide, Cyanogen, Nitric Oxide and Nitrogen Tetroxide Mixtures, Nitrogen Oxides n.o.s. and Nitrogen Trioxide.
- 4. Group 4 consists of Class 7.
- Group 5 consists of tank cars carrying Division 3.3 flammable liquids or tank cars placarded "RESIDUE" or "EMPTY"
- 6. Group 6 consists of Class 9, and of Class 6 bearing St. Andrews cross plac-

FOOTNOTES:

- (a) Except trailer-on-flat-car, container-on-flat-car, tri-level and bi-level cars and any other car specially equipped with tie down devices for handling vehicles. Permanent end bulk head flat cars considered the same as an open-top car.
- (b) Except when train consists only of placarded tank cars.
- (c) Tank cars containing flammable gases Division 1 of Class 2 must be separated from tank car shipments of Chlorine, Anhydrous Ammonia and Sulphur Dioxide by 5 cars.
- (d) When train length does not permit means not having sufficient buffer cars to provide the 5 car separation.

Revised July, 1986.

APPENDIX F HAZARDOUS MATERIALS DEFINITIONS

U.S. Department of Transportation

HAZARDOUS MATERIALS TRANSPORTATE

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Research and Special Programs Administration

HAZARDOUS MATERIALS DEFINITIONS

The following definitions have been abstracted from the Code of Federal Regulations, Title 49-Transportation, Parts 100-177. Refer to the referenced sections for complete details. NOTE: The plus (+) in Column (1), Sec. 172.101, Hazardous Materials Table) fixes the propershipping name and hazard class for that entry without regard to whether the material meets the definition of that class. [Sec. 172.101(a)(1)]

<u>HAZARDOUS MATERIAL</u> - A substance or material which has been determined by the Secretary of Transportation to be capable of posing an unreasonable risk to health, safety and property when transported in commerce, and which has been so designated. (Sec. 171.8).

 $\underline{\text{MULTIPLE HAZARDS}}$ - A material meeting the definition of more than one hazard class is classed according to the provisions set forth in Sec. 173.2(a) and (b).

HAZARD CLASS	UN No	DEFINITIONS
		An Explosive - Any chemical compound, mixture, or device, the primary or common purpose of which is to function by explosion, i.e., with substantially instantaneous release of gas and heat, unless such compound, mixture, or device is otherwise specifically classified in Parts 171-179. (Sec. 173.50)
CLASS A EXPLOSIVE	1	Detonating or otherwise of maximum hazard. The nine types of Class A explosives are defined in Sec. 173.53.
CLASS B EXPLOSIVE	1	In general, function by rapid combustion rather than detonation and include some explosive devices such as special fireworks, flash powders, etc. (Sec. 173.88)
CLASS C EXPLOSIVE	1	Certain types of manufactured articles containing Class A or Class B explosives, or both, as components but in restricted quantities, and certain types of fireworks. (Sec. 173.100)
BLASTING AGENT	. 1	A material designed for blasting which has been tested in accordance with Sec. 173.114a(b) and found to be so insensitive that there is very little probability of accidental initiation to explosion or of transition from deflagration to detonation. [Sec. 173.114a(b)]
	,	Compressed Cas - Any material or mixture having in the container a pressure EXCEEDING 40 psia at 70°F., or a pressure exceeding 104 psia at 130°F.; or any liquid flammable material having a vapor pressure exceeding 40 psia at 100°F. [Sec. 173.300(a)]
		Non-liquefied compressed gas is a gas, other than gas in solution, which under the charged pressure is entirely gaseous at a temperature of 70°F.
	-	Liquefied compressed gas is a gas which, under the charged pressure, is partially liquid at a temperature of 70°F.
		Compressed gas in solution is a nonliquefied compressed gas which is dissolved in a solvent.
FLAMMABLE GAS	2	Any compressed gas meeting the requirements for lower flammability limit, flammability limit range, flame projection, or flame propagation criteria as specified in Sec. 173.300(b).
NONFLAMMABLE GAS	2	Any compressed gas other than a flammable compressed gas.

HAZARD CLASS	UN No	DEFINITIONS
COMBUSTIBLE LIQUID	3	Any liquid having a flash point at or above 100°F. and below 200°F. as determined by tests listed in Sec. 173.115(d). Exceptions are found in Sec. 173.115(b).
FLAMMABLE LIQUID	3	Any liquid having a flash point below 100°F. as determined by tests listed in Sec. 173.115(d). For exceptions, see Sec. 173.115(a). Pyroforic Liquid - Any liquid that ignites spontaneously in dry or moist air at or below 130°F. [Sec. 173.115(c)]
FLAMMABLE SOLID	4	Any solid material, other than an explosive, which is liable to cause fires through friction, retained heat from manufacturing or processing, or which can be ignited readily and when ignited, burns so vigorously and persistently as to create a serious transportation hazard. Included in this class are spontaneously combustible and water-reactive materials. (Sec. 173.150) Spontaneously Combustible Material (Solid) - A solid substance (including sludges and pastes) which may undergo spontaneous heating or self-ignition under conditions normally incident to transportation or which may, upon contact with the atmosphere, undergo an increase in temperature and ignite. (Sec. 171.8)
		Water Reactive Material (Solid) - Any solid substance (including sludges and pastes) which, by interaction with water, is likely to become spontaneously flammable or to give off flammable or toxic gases in dangerous quantities. (Sec. 171.8)
ORGANIC PEROXIDE	5	An organic compound containing the bivalent -0-0 structure and which may be considered a derivative of hydrogen peroxide where one or more of the hydrogen atoms have been replaced by organic radicals must be classed as an organic peroxide unless[See Sec. 173.151(a) for details].
OXIDIZER	5	A substance such as chlorate, permanganate, inorganic peroxide, or a nitrate, that yields oxygen readily to stimulate the combustion of organic matter. (See Sec. 173.151)
POISON A	2	Extremely Dangerous Poisons - Poisonous gases or liquids of such nature that a very small amount of the gas, or vapor of the liquid, mixed with air is dangerous to life. (Sec. 173.326)
POISON B	6	Less Dangerous Poisons - Substances, liquids or solids (including pastes and semi-solids), other than Class A or Irritating materials, which are known to be so toxic to man as to afford a hazard to health during transportation; or which, in the absence of adequate data on human toxicity, are presumed to be toxic to man. (Sec. 173.343)
IRRITATING MATERIAL	6	A liquid or solid substance which, upon contact with fire or when exposed to air, gives off dangerous or intensely irritating fumes, but not including any poisonous material, Class A. (Sec. 173.381)
ETIOLOGIC AGENT	6	An "etiologic agent" means a viable micro-organism, or its toxin, which causes, or may cause, human disease. (Sec. 173.386)
RADIOACTIVE MATERIAL	7	Any material, or combination of materials, that spontaneously emits ionizing radiation, and having a specific activity greater than 0.002 microcuries per gram. (Sec. 173.389) [See Sec. 173.389(a) through (1) for details]

HAZARD CLASS	UN No.	DEFINITIONS
CORROSIVE MATERIAL	8	Any liquid or solid that causes visible destruction or irreversible alterations in human skin tissue or a liquid that has a severe corrosion rate on steel. [See Sec. 173.240(a) and (b) for details]
ORM - OTHER REGULATED MATERIALS		(1) Any material that may pose an unreasonable risk to health and safety or property when transported in commerce; and (2) Does not meet any of the definitions of the other hazard classes specified in this subpart; or (3) Has been reclassed an ORM (specifically or permissively) according to this subchapter. [Sec. 173.500(a)] NOTE: A material with a flashpoint of 100°F. to 200°F. must be classed as a combustible rather than as an ORM if it is a hazardous waste or is offered in a packaging having a rated capacity of more than 110 gallons.
ORM-A	9	A material which has an anesthetic, irritating, noxious, toxic, or other similar property and which can cause extreme annoyance or discomfort to passengers and crew in the event of leakage during transportation. [Sec. 173.500(b)(1)]
ORM-B	9	A material (including a solid when wet with water) capable of causing significant damage to a transport vehicle from leakage during transportation. Materials meeting one or both of the following criteria are ORM-B materials: (1) A liquid substance that has a corrosion rate exceeding 0.250 inch per year (IPY) on aluminum (nonclad 7075-T6) at a test temperature of 130°F. An acceptable test is described in NACE Standard TM-01-69; and/or (2) Specifically designated by name in Sec. 172.101. [Sec. 173.500(b)(2)]
ORM-C	9	A material which has other inherent characteristics not described as an ORM-A or ORM-B but which makes it unsuitable for shipment, unless properly identified and prepared for transportation. Each ORM-C material is specifically named in Sec. 172.101. [Sec.173.500(b)(3)]
ORM-D	9	A material such as a consumer commodity which, though otherwise subject to the regulations of this subchapter, presents a limited hazard during transportation due to its form, quantity and packaging. They must be materials for which exceptions are provided in Sec. 172.101. A shipping description applicable to each ORM-D material or category of ORM-D materials is found in Sec. 172.101. [Sec. 173.500(b)(4)]
ORM-E	9	A material that is not included in any other hazard class, but is subject to the requirements of this subchapter. Materials in this class include (1) hazardous waste and (2) hazardous substance, as defined in Sec. 171.8. [Sec. 173.500(b)(5)]

THE FOLLOWING ARE OFFERED TO EXPLAIN SOME OF THE ADDITIONAL TERMS USED IN PREPARATION OF HAZARDOUS MATERIALS FOR SHIPMENT. (Sec. 171.8)

CONSUMER COMMODITY (See ORM-D above)	A material that is packaged or distributed in a form intended and suitable for sale through retail sales agencies or instrumentalities for consumption by individuals for purposes of personal care of household use. This term also includes drugs and medicines. (Sec. 171.8)
FLASH POINT	The minimum temperature at which a substance gives off flammable vapors which, in contact with a spark or flame, will ignite. For liquids, see Sec. 173.115; for solids, see Sec. 173.150.

TERM	DEFINITION
FORBIDDEN	Material is prohibited from being offered or accepted for transportation. This prohibition does not apply if these materials are diluted, stabilized, or incorporated in devices and they are classed in accordance with the definitions of hazardous materials. [Sec. 172.101(d)(1)]
HAZARDOUS SUBSTANCE	For transportation purposes, a material and its mixtures or solutions, that is identified by the letter "E" in Column (1) of the Hazardous Materials Table, Sec. 172.101, when offered for transportation in one package, or in one transport vehicle if not packaged, and when the quantity of the material therein equals or exceeds the reportable quantity (RQ). For details, refer to Sec. 171.8 and Sec. 172.101 (Hazardous Materials Table).
HAZARDOUS WASTE	Any material that is subject to the hazardous waste manifest requirements of the Environmental Protection Agency specified in the CFR, Title 40, Part 262 or would be subject to these requirements in the absence of an interim authorization to a State under Title 40, CFR, Part 123, Subpart F. (Sec. 171.8). Questions regarding EPA hazardous waste regulations, call Toll Free: (800) 424-9065 or in Washington: 554-1404.
LIMITED QUANTITY	The maximum amount of a hazardous material as specified in those sections applicable to the particular hazard class for which there is a specific labeling and packaging exception from the requirements. See Sec. 173.118, 173.118(a), 173.153, 173.244, 173.306, 173.345, 173.364 and 173.391.
REPORTABLE QUANTITY	The quantity of hazardous substance specified in the Hazardous Materials Table (Sec. 172.101) and identified by the letter "E" in Column (1). (Sec. 171.8)

* THIS HANDOUT IS DESIGNED AS A TRAINING AID FOR ALL INTERESTED PARTIES WHO MAY BECOME INVOLVED WITH HAZARDOUS MATERIALS. IT DOES NOT RELIEVE PERSONS FROM COMPLYING WITH THE DEPARTMENT OF TRANSPORTATION'S HAZARDOUS MATERIALS REGULATIONS. SPECIFIC CRITERIA FOR HAZARD CLASSES AND RELATED DEFINITIONS ARE FOUND IN THE CODE OF FEDERAL REGULATIONS (CFR), TITLE 49, PARTS 100-177.

Information Services Division, DMT-11 Office of Operations and Enforcement Materials Transportation Bureau Department of Transportation Washington, D.C. 20590

APPENDIX G TABLE OF POSITION IN TRAIN REQUIREMENTS

		POSITIO	N IN T	RAIN HAZA	APPE OF PL RDOU	ACARI	CII DED CA TERIAL	S		
	//	PLACARD APPLIED ON CAR				/		A A A A A A A A A A A A A A A A A A A	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	
		TYPE OF CAR	\$		Orti	erio /	A 18	Ja Strain		' 3
		RESTRICTIONS								
WHEN TRAIN LENGTH PERMITS		Must not be nearer than 6th from engine, occupied caboose or passenger car	√	√	_		V			
WHEN TRAIN LENGTH DOES NOT PERMIT	Must be near middle of train but not		√	√			√			POOTS:OTES
	1	Loaded flat car. A flatcar equipped with permanently attached ends of rigid construction is considered to be an open-top car.	v [©]	√	√ .		√ ②			FOOTNOTES: D Loaded cars placarded "EXPLOSIVES A" may be placed next to each other.
	i	An open-top car when any of the lading protrudes beyond the car ends or when any of the lading extending above the car ends is liable to shift so as to protrude beyond the car ends:	√	V	v ∕		V	·	,	② A specially equipped car in trailer-on-flatcar or container-on- flatcar service or a flatcar loaded with vehicles secured by means of
	Γ	ENGINE	√	•	V	V	•		v '	device designed for that purpose and permanently installed on the flatcar, and of a type generally accepted for handling in
M	1	Except as provided in lines 10 and 11, a car occupied by any person or a passenger car or combination car that may be occupied. OCCUPIED CASCOSE		√ ³	√ ^③	V	√	©	√	interchange between railroads ma be placed next to these placarded loaded tank cars subject to the following: this exception for cars i trailer-on-flatcar service does not
MUST :				√	v [⊙]	V	√		√	apply to loaded flatbed trucks. loaded flatbed trailers, loaded oper top trailers, or loaded trucks or trailers without securely closed
ZÓF B		OCCUPIED GUARD CAR	√ ³	√ ³	√ 0		V			doors. ② A rail car placarded "EXPLOSIVES A" or "POISON
E		UNDEVELOPED Flum				V				GAS" in a moving or standing train must be next to and ahead o
I A C E D	1	A car with automatic refrigeration or heating apparatus in operation, or a car with open-flame apparatus in service, or with an internal combustion engine in operation:	√	v	√ /.		•			any car occupied by the guards or technical escorts accompanying this car. However, if a car occupie by guards or technical escorts is equipped with a lighted heater or stove, it must be the fourth car
NEXT		A car containing lighted heaters, stoves, or lanterns;	√	√	V					behind any car requiring "EXPLOSIVES A" placards.
Ť	CAR	EXPLOSIVES A		•	√	V	V .	√		Applies only in mixed train service, see section 174.87
	1	POISON GAS	√			√	V	V		
	ACARON	Loaded placarded car, other than a car placarded with the same placard or the "combustible" placard.	√	•	v	V			·	
	D	EADIOACTIVE	√	√	V		V	√		

NOTE: This chart was extracted from Union Pacific Railroad Company Timetable No. 1 (2-12-78).

APPENDIX H CHEMICAL REACTIVITIES OF BINARY COMBINATIONS

APPENDIX H

Chemical Reactivities of Binary Combinations

The results from the binary chemical evaluations are given in the tables on pages H-7 through H-111. The tables contain the following information:

1st Column: ASTM Chemical Reactivity Group for the 1st Chemical,

2nd Column: Name of the 1st Chemical,

3rd Column: STC Number of the First Chemical.

4th Column: ASTM Chemical Reactivity Group for the 2nd Chemical,

5th Column: Name of the 2nd Chemical,

6th Column: STC Number of the 2nd Chemical,

7th Column: General description of the compatibility or consequences

of the combination,

8th Column: General comments on the reaction,

9th Column: Reference Number (See Page H- ??? for the Reference

List).

The chemicals are arranged in alphabetic order with the lower ASTM Chemical Reactivity Number being the 1st Chemical. For example information on the compatibility of chlorine and hydrocyanic acid can be found by searching the 2nd Column for Hydrocyanic Acid and the 5th column for chlorine. Hydrocyanic acid is the 1st chemical because it has a lower ASTM Group number than chlorine (Group 1 as opposed to Group 104). To facilitate the location of combinations, Table H1 on pages H-2 through H-4 provides an alphabetic list of chemicals and their corresponding ASTM Chemical Reactivity Groups. A comparison between the ASTM and Coast Guard Chemical Reactivity Groups is given in Table H2 on pages H-5 and H-6.

TABLE H1. List of Chemical Names and Their Corresponding ASTM Reactivity Group

Chemical Name	STC R Number V	-	Hazaro Class				
Acetal dehyde	4907210	41	FL	l	352Ø	•	Aldehydes
Acetic Acid, Glacial	4931303	48	CM	1	2953	3	Organic Acids
Acetic Anhydride	4931304	68	CM		1533	107	Water Reactive Substances
Acetone	[4908105]	46	FL	1	2971	19	Ketones
Acrylic Acid	4931405	88	CM	1	1031) 3	Organic Acids
Acrylonitrile	4906420	36] FL	1	4811	26	Nitriles
Alcoholic Beverage	4910102	77	[CL	I	1276	4	Alcohols and Glycols
Alkaline Liquid, n.o.s.	4935220	92	CM	1	975	10	Caustics
Ammonium Nitrate Solution	4918774	65	OM	1	1860	106	Water and Mixtures Containing Water
Anhydrous Ammonia	4904210	4	NG	1 4	42526] 10	Caustics
Aniline Oil, Liquid	4921410	85	PB	1	1088	7	Aliphatic and Aromatic Amines
Benzene	4908110	33	ļ FL	i	5067	16	Aromatic Hydrocarbons
Butadiene	4905704	25	FG	1	7025		Unsaturated, Aliphatic Hydrocarbons
Butadiene, Inhibited	4905703	9Ø	FG	1	1005	•	Unsaturated, Aliphatic Hydrocarbons
Butane	4905702	18	j FG	i	9473		Saturated Aliphatic Hydrocarbons
Butane	14905706	13	i FG	i :	12215		Saturated Aliphatic Hydrocarbons
Butyl Acrylate (Corr. L., n.o.s.)	4912215	63	i CL	i	1957	•	Esters
Butyl Alcohol	4909129	87	i FL	i	1052	•	Alcohols and Glycols
Butyraldehyde	4908119	98	I FL	i	87Ø	•	Aldehydes
Carbon Dioxide, Refrigerated Liquid	4904509	20	i NG	i	8172	•	1
Carbon Disulfide	4908125	89	l FL	i	1013		Mercaptans and Other Organic Sulfides
Carbon Tetrachloride	4940320	72	I DA	i	1343		Halogenated Compounds
Chlorine	4904120	2	i NG	i.	46686		Strong Oxidizing Agents
Chloroprene, Inhibited	4907223	97	l FL	ì	889		Halogenated Compounds
Coal Tar Distillate	4915363	78	i CL	i	1259		Combustible and Flammable Materials
Combustible Liquid, n.o.s.	4915185	45	CL	1	3139		[Combustible and Flammable Materials
Compound, Cleaning, Liquid	4915147	75	i CL	1	1281		Combustible and Flammable Materials
Corrosive Liquid, n.o.s.	[4936540]	5Ø	i CM	l ì	2916	•	Contract Canal C
Crude Oil Petroleum	[4930348]	17	j FL	1	9995	•	 Combustible and Flammable Materials
Cyclohexane	•		•	i i		-	•
	4908132	61	FL	1	2089		Saturated Aliphatic Hydrocarbons
Denatured Alcohol	4909141	44	FL	!	3140	•	Alcohols and Glycols
Denatured Alcohol	4909151	19	FL	!	8947	•	Alcohols and Glycols
Dinitrotoluene, Liquid	[4963120]	100	OE	1	866		Nitro Compounds
Distillate Fuel Oil	2911315	26	CL	1	6549		Combustible and Flammable Materials
Ethyl Acetate	4909160	86	FL	1	1066	•	Esters
Ethyl Acrylate, inhibited	4907215	76	FL	!	1281	•	Esters
Ethyl Alcohol	4909159	99	FL]	867	•	Alcohols and Glycols
Ethylene Oxide	4906610	27	FL		6535		Epoxides
Ferric Chloride Solution	4932342	64	CM	l			Strong Oxidizing Agents
Flammable Liquid, n.o.s.	4910185	59	FL	I			Combustible and Flammable Materials
Formaldehyde Solution	4913144	95	CL		901	•	Aldehydes
Formaldehyde Solution	4913168	6Ø	CL	1	2157		Aldehydes
Fuel Oil	4915112	11	CL	1			Combustible and Flammable Materials
Fuel Oil	4915113	24	CL	1	7443	101	Combustible and Flammable Materials
Fuel Oil	4915111	12	CL	1	14038	101	Combustible and Flammable Materials

TABLE H1. List of Chemical Names and Their Corresponding ASTM Reactivity Group (Cont.)

Chemical Name		_	-	Movements per Year		
			· 	·· ·		
Fuel, Aviation, Turbine Engine	4909215	35	FL	4874	101	Combustible and Flammable Materials
Fuel, Aviation, Turbine Engine	[4915167]	74	CL	1283	101	Combustible and Flammable Materials
Gasoline	4908178	22	FL	7676	101	Combustible and Flammable Materials
Gasoline, nec	2911190	70	FL	1410	101	Combustible and Flammable Materials
Glycol Ethers, nec (C.L., n.o.s.)	4913194	94] CL	902	14	Ethers
Hexamethylene Diamine Solution	4935645	31	{ CM	5580	7	Aliphatic and Aromatic Amines
Hexane .	4908183	69	FL	1454	29	Saturated Aliphatic Hydrocarbons
Hydrochloric Acid	4930228	14	[CM	10973	1	Non-oxidizing Mineral Acids
Hydrocyanic acid	4920125	8Ø	Į PA	1226	1 1	Non-oxidizing Mineral Acids
Hydrofluorosilicic Acid	4930026	93	[CM	930	1	Non-oxidizing Mineral Acids
Hydrogen Fluoride	4930024	58] CM	2464	1	Non-oxidizing Mineral Acids
Hydrogen Peroxide Solution	4918335	67	[DM	1646	104	Strong Oxidizing Agents
Hydroiodic Acid	4930226	96	j CM	897	1	Non-oxidizing Mineral Acids
Isopropano I	4909205	52	FL	2848	•	Alcohols and Glycols
Liquefied Petroleum Gas	4905752	. 5	FG	37043	28	Unsaturated, Aliphatic Hydrocarbons
Liquefied Petroleum Gas (Butene)	4905707	53	FG	2806		Unsaturated, Aliphatic Hydrocarbons
Liquefied Petroleum Gas (Butylene)	4905711	84	j FG	1139	•	Unsaturated, Aliphatic Hydrocarbons
_iquefied Petroleum Gas (Isobutane)	4905747	23	I FG	7582	•	Saturated Aliphatic Hydrocarbons
Liquefied Petroleum Gas (Isobutane)	4905750	32	FG	5360	•	Saturated Aliphatic Hydrocarbons
Liquefied Petroleum Gas (Isobutylene)	4905748	79	FG	1258		Unsaturated, Aliphatic Hydrocarbons
Liquefied Petroleum Gas (Propane)	4905781		FG	21793		Saturated Aliphatic Hydrocarbons
Liquefied Petroleum Gas (Propylene)	4905782	3Ø	i FG	5707	-	Unsaturated, Aliphatic Hydrocarbons
Maleic Anhydride	4941161	81	I DA	1168		Water Reactive Substances
Methyl Alcohol	4909237	54	FL	2630	1 4	Alcohols and Glycols
Methy! Alcohol	4909230	10	j FL	18109		Alcohols and Glycols
Methyl Chloride	4905761	82	, FG	1168	•	Halogenated Compounds
Methyl Ethyl Ketone	4909243	73	I FL	1318		Ketones
Methyl Methacrylate Monomer, inhib.	4907250	49	i FL	2927	1 13	Esters
Motor Fuel Antiknock Compound	4921445		i PB	1000	24	Toxic Metals and Metal Compounds
Nitric Acid, Fuming	4918529		CM	441	•	Oxidizing Mineral Acids
Octyl Alcohol (C.L., n.o.s.)	4913158		CL	1671	•	[Alcohols and Glycols
Oil	4915245		CL	•	-	Combustible and Flammable Materials
Oleum	4930030		CM	3243	•	Oxidizing Mineral Acids
Petroleum Naphtha	4910259		CL	•	•	Combustible and Flammable Materials
Petroleum Naphtha	4915259		CL	-	•	[Combustible and Flammable Materials
Petroleum Naphtha	4915257		CL	•	•	Combustible and Flammable Materials
Petroleum Residual Fuel Dil	2911715		CL	•	-	Combustible and Flammable Materials
Petroleum, Partially Refined	2911735		CL	•	-	Combustible and Flammable Materials
Phenol	[4921220]		PB	5974	-	Phenois and Cresois
Phosphoric Ácid	4921220		CM	28008	•	Non-oxidizing Mineral Acids
	4916141		I FS	•	-	Strong Reducing Agents
Phosphorus, White or Yellow			CM	2622	•	Caustics
Potassium Hydroxide, Liq. or solution	14935230		CM		•	Epoxides
Propylene Oxide	[4906620]	34	í Lr	4913	[34	[Thoy i dea

TABLE H1. List of Chemical Names and Their Corresponding ASTM Reactivity Group (Cont.)

Chemical Name	Number N		• •		•	vements r Year	•		•
Sodium Hydroxide, liq. or solution	4935243	1		CM		48367	1	10	Caustics
Sodium Hydroxide, Liq. or solution	4935240	7	1	CM	1	27274	1	10	Caustics
Sodium Metal	4916456	101	1	FS	1	855	1	21	Elemental Alkali, Alkaline Earth Metals
Styrene Monomer, Inhibited	4907265	15	1	FL	1	1Ø859	Ι	16	Aromatic Hydrocarbons
Sulfur Dioxide	4904290	47	1	NG	1	2959	1	1Ø5	Strong Reducing Agents
Sulfuric Acid	4930040	3	١	CM	1	46265	١	2	Dxidizing Mineral Acids
Sulfuric Acid, spent	4930042	37	Ì	CM	Ì	4809	ĺ		Oxidizing Mineral Acids
Toluene	4909305	62	i	FL	İ	2084	i		Aromatic Hydrocarbons
Toluene Diisocyanate	[4921575]	56	į	PB	i	2610	į	18	Isocyanates
Vinyl Acetate	4907270	38	i	FL	i	4431	i	13	Esters
Vinyl Chloride	4905792	9	i	FG	i	20081	i	17	
Zylene	4909351	51	i	FL	i	2869	i		Aromatic Hydrocarbons
Xy lene	149Ø935Ø1	57	i	FL	i	2546	i		Aromatic Hydrocarbons

TABLE H2. A Comparison of the ASTM and Coast Guard Chemical Reactivity Groups

$\overline{}$			
1		Coast	
ASTM		Guard	
No.	ASTM Group	No.	Coast Guard Cargo Group
		0	Unassigned Cargoes
1	Non-oxidizing Mineral Acids	1	Non-oxidizing Mineral Acids
2	Oxidizing Mineral Acids	2	Sulfuric Acids
2	Oxidizing Mineral Acids	3	Nitric Acid
3	Organic Acids	4	Organic Acids
10	Caustics	5	Caustics
10	Caustics	6	Ammonia
7	Aliphatic and Aromatic Amines	7	Aliphatic Amines
7	Aliphatic and Aromatic Amines	8	Alkanolamines
7	Aliphatic and Aromatic Amines	9	Aromatic Amines
6	Amides	10	Anides
107	Water Reactive Substances	11	Organic Anhydrides
18	Isocyanates	12	Isocyanates
13	Esters	13	Vinyl Acetate
14	Ethers	13	Vinyl Acetate
103	Polymerizable Compounds	14	Acrylates
13	Esters	14	Acrylates
26	Nitriles	15	Substituted Allyls
17	Halogenated Organics	15	Substituted Allyls
4	Alcohols and Glycols	15	Substituted Allyls
34	Epoxides	16	Alkylene Oxides
103	Polymerizable Compounds	16	Alkylene Oxides
34	Epoxides	17	Epichlorohydrin
17	Halogenated Compounds	17	Epichlorohydrin
19	Ketones	18	Ketones
5	Aldehydes	19	Aldehydes
4	Alcohols and Glycols	20	Alcohols,Glycols
31	Phenols and Cresols	21	Phenols and Cresols
-		22	Caprolactum Solution
28	Unsaturated, Aliphatic	30	Olefins
	Hydrocarbons		
29	Saturated Aliphatic	31	Paraffins
j	Hydrocarbons		
16	Aromatic Hydrocarbons	32	Aromatic Hydrocarbons
101	Combustible and Flammable	33	Miscellaneous Hydrocarbon
	Materials		Mixtures
13	Esters	34	Esters
103	Polymerizable Compounds	35	Vinyl Halides
17	Halogenated Organics	35	Vinyl Halides
17	Halogenated Organics	36	Halogenated Hydrocarbons
26	Nitriles	37	Nitriles
20	Mercaptans and Other Organic	38	Carbon Disulfide
ا ۲۷	Sulfides	30	Carbon bisurriac
	Juli lucs	39	Sulfolane
<u>-</u> 14	Ethers	40	Glycol Ethers
7.4		10	

TABLE H2. A Comparison of the ASTM and Coast Guard Chemical Reactivity Groups (Cont.)

No.	ASTM Group	Coast Guard No.	Coast Guard Cargo Group
4	Alcohols and Glycols	40	Glycol Ethers
14	Ethers	41	Ethers
27	Nitro Compounds	42	Nitro compounds
106	Water and Mixtures Containing	43	Miscellaneous Water Solutions
	Water		
8	Azo and Diazo Compounds,	-	
	Hydrazine		
9	Carbamates	- 1	
11	Cyanides	-	
12	Dithiocarbamates	-	
15	Inorganic Fluorides	-	
21	Elemental Alkali, Alkaline	- ,	
_ ′ [Earth Metals	ļ	·
22	Other Elemental	-	
	<pre>Metals(Foam,Sponge,etc.)</pre>		
23	Other Elemental	-	·
ŀ	<pre>Metals(Sheets,Rods,etc.)</pre>		
24	Toxic Metals and Metal	-	
Ì	Compounds	Ì	
25	Nitrides	-	
30	Organic Peroxides and	-	·
	Hydroperoxides	1	
32	Organophosphates,	-	
	Phosphothiates	ł	
33	Inorganic Sulfides		
106	Explosives	-	
104	Strong Oxidizing Agents	-	
105	Strong Reducing Agents	-	

Chemical Reactivity of Binary Systems

ASTM GRP	Chemical Name	STC A		STC Number Compatibility		RE
5 A c	ceta i dehyde	4907210	107 Acetic Anhydride	49313 0 4 Heat	Condensation reaction (abnormal Perkins reaction) is probably slow. Forms vinyl acetate.	1
5 Ac	ceta I dehyde	4907210	19 Acetone	4908105 Compatible	No reaction expected.	1
5 Ac	ceta I dehyde	49 07210	10 Alkaline Liquid, n.o.s.	4935220 Heat/Polymerization	Violent aldol condensation forms 3-hydroxybutanal and polymerization through carbonyl group.	1
5 Ac	ceta I dehyde	[4907210]	10 Anhydrous Ammonia	4904210 Heat/Polymerization	Potentially violent reaction forms ethaneimine and vigorous polymerization.	1
5 Ac	ceta i dehyde	4907210	7 Aniline Oil, Liquid	4921410 Heat/Polymerization	Reacts vigorously with probable polymerization and imine formation - N-ethylidenebenzeneamine.	
5 Ac	ceta i dehyde	4907210	104 Chlorine	49 04 12 0 Heat/Fire	Violent reaction forming chloral under basic or acidic conditions. Also, oxidation of aldehyde group	
5 A	cetaldehyde	4907210	27 Dinitrotoluene, Liquid	4963120 Heat/Explosion	Potentially vigorous reaction under basic conditions.	1
5 A	ceta i dehyde	4907210	34 Ethylene Oxide	4906610 Compatible	Slow cyclic acetal formation at ambient temperatures.	-
5 j A	ceta i dehyde	4907210	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Evolves heat and forms a black spongy material. Dilution of acetaldehyde in water moderates reaction	1
5 A	ceta i dehyde	4907210	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Compatible	1	
5 Ac	ceta i dehyde	4907210	7 Hexamethylene Diamin Solution	e 4935645 Heat/Polymerization	Polymerization and formation of N,N'-diethylidene-1,6-hexanediam ine moderated by presence of water.	-

	TM P Chemical Name	STC / Number 0	•	STC Number Compatibility	General Comments	: .	EF- #
1	5 Acetaldehyde	4907210	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Forms extremely explosive poly(ethylidene) peroxide - accelerated by acidic or basic conditions.		2
 	5 Aceta I dehyde	4907210	31 Pheno I	4921220 Heat/Fire	Reacts vigorously forming 1,1-bis(4-hydroxyphenyl) ethane. Promoted by acid.		
	5 Aceta dehyde	[4907210]	105 Phosphorus, White or Yellow	4916141 Heat	Reacts vigorously.	1	5
 	5 Aceta I dehyde	4907210	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Polymerization	Violent aldol condensation forms 3-hydroxybutanal and polymerization through carbonyl group.	 I	1
 I	5 Aceta dehyde	4907210	34 Propylene Oxide	4906620 Compatible	Slow reaction at ambient temperature - slower than ethylene oxide reaction.	!	
 !	5 Aceta i dehyde	4907210	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Polymerization	Violent aldol condensation forms 3-hydroxybutanal and polymerization through carbonyl group.	:- -	1
I	5 (Aceta I dehyde		10 Sodium Hydroxide, Liq. or solution		Violent aldol condensation forms 3-hydroxybutanal and polymerization through carbonyl group.		1
 I	5 (Aceta I dehyde	[4907210]	21 Sodium Metal	4916456 Heat/Fire	Very vigorous reaction forms pinacol dianion.	}	
 I	5 Aceta I dehyde	4907210	105 Sulfur Dioxide	4904290 Unknown	1	l	
 I	5 Acetaldehyde	[4907210]	18 Toluene Diisocyanate	4921575 Heat	Potential reaction of enol.	1	5
 	3 Acetic Acid, Glacia	49313Ø3	5 Aceta I dehyde	4907210 Heat/Polymerization	Potentially violent/explosive polymerization of acetaldehyde.	l	1
1			107 Acetic Anhydride				2

	TM P Cher	mical	STC / Name Number (•	STC Number Compatibility	General Comments	REF
	3 Acetic	Acid,	Glacial 49313Ø3	26 Acrylonitrile	4906420 Heat/Polymerization	Exotherm observed when mixture was heated between 100 and 200 degrees C.	12
_	3 Acetic	Acid,	Glacia 4931303	10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Violent neutralization.	
-	3 Acetic	Acid,	Glacia 4931303	106 Ammonium Nitrate Solution	4918774 Heat	May ignite when warmed, especially if concentrated. Reaction mitigated by water content.	} 1
	3 Acetic	Acid,	Glacia 4931303	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Violent neutralization forms ammonium acetate.	I
	3 Acetic	Acid,	Glacia 49313Ø3	7 Aniline Oil, Liquid	4921410 Heat/Neutralization	Minimal heat generated.	12
	3 Acetic	Acid,	Glacia 49313Ø3	5 Butyraldehyde	4908119 Heat/Polymerization	Exotherm observed when mixture/ was heated between 200 and 300 degrees C.	-
	3 Acetic	Acid,	Glacia 4931303	104 Chlorine	4904120 Heat/Fire	Potentially vigorous/violent reaction.	{
_	3 Acetic	Acid,	Glacia 49313Ø3	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potential vigorous/violent polymerization initiated by acetic acid.	1
••	3 Acetic	Acid,	Glacia 4931303	34 Ethylene Oxide	4906610 Heat/Polymerization	Potentially vigorous polymerization with heat liberation will occur in contac with acids.	1 t
				5 Formaldehyde Solutio		Polymerization of formaldehyde initiated by acetic acid.	
-	3 Acetic	Acid,	Glacia 4931303	5 Formaldehyde Solutio	on 4913144 Heat/Polymerization	Polymerization of formaldehyde initiated by acetic acid.	1
	3 Acetic	Acid,	Glacia1 4931303	7 Hexamethylene Diamin Solution	ne 4935645 Heat/Neutralization	Heat of neutralization.	1
			Glacia 4931303			Mixing and heating of dilute solutions will form heat and peracetic acid which explodes a 110 C.	1

	TM STC /	•	STC Number Compatibility		REF #
-	3 Acetic Acid, Glacial 4931303	105 Phosphorus, White or Yellow	4916141 Compatible	No reactivity found below 60 degrees C.	12
	3 Acetic Acid, Glacial 4931303	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Neutralization	Violent neutralization.	1
	3 Acetic Acid, Glacial 4931303	34 Propylene Oxide	4906620 Heat/Polymerization	Exotherm observed when mixture was heated between 100 and 200 degrees C. Potential polymerization.	12
-	3 Acetic Acid, Glacial 49313Ø3	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Neutralization	Excessive heat generated.	12
	3 Acetic Acid, Glacial 4931303	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Excessive heat generated.	12
-	3 Acetic Acid, Glacial 4931303	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Vigorous hydrogen evolution with possible explosion.	 i
	3 Acetic Acid, Glacial 4931303	18 Toluene Diisocyanate	4921575 Heat/Gas	Vigorous reaction forming ArNHC(0)0C(0)CH3, CO2 gas and some heat - polymerization if water present.	ļ.
	19 Acetone 4908105	104 Chlorine	4904120 Heat/Fire	Potentially vigorous/violent oxidation reaction with possible ignition.	
	19 Acetone 4908105	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Readily forms dimeric, trimeric or polymeric peroxides which are explosive.	
	19 Acetone 4908105	105 Phosphorus, White or Yellow		No reaction expected.	1
	19 Acetone 4908105	21 Sodium Metal		[Potentially violent reaction reduces acetone to give 40 to 50% pinacolate dianion.	!
	3 Acrylic Acid 4931405	5 Aceta dehyde	4907210 Heat/Polymerization	polymerization of acetaldehyde (same as with acetic acid).	1
	3 Acrylic Acid 4931405	•		Violent acid catalyzed hydrolysis of acetic anhydride if mixed with aqueous acrylic acid (need H2O).	1 :

AST		STC AS Number GR	•	STC Number Compatibility		REF #
]	3 Acrylic Acid	4931405	26 Acrylonitrile	4906420 Compatible	No exotherm was observed when the mixture was heated up to 200 degrees C.	12
I _	3 Acrylic Acid	49314Ø5	10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Potentially vigorous polymerization of acrylic acid as well as neutralization reaction.	 !
 	3 Acrylic Acid	49314Ø5 1	96 Ammonium Nitrate Solution	4918774 Heat	Mixture may ignite by analogy to acetic acid/ammonium nitrate mixture. Reaction mitigated by water.	
1	3 Acrylic Acid	49314Ø5	10 Anhydrous Ammonia	4904210 Heat/Polymerization	Violent polymerization.	5
	3 Acrylic Acid	49314Ø5	7 Aniline Oil, Liquid	4921410 Heat/Polymerization	Violent polymerization.	5
	3 Acrylic Acid	49314Ø5	5 Butyraldehyde	4908119 Compatible	No reaction observed when mixture was heated up to 300 degrees C.	12
 	3 Acrylic Acid	49314Ø5 1	Ø4 Chlorine	4904120 Heat/Fire	Probably more vigorous than acetic acid/chlorine mixture.	
1	3 Acrylic Acid	49314Ø5 	17 Chioroprene, Inhibited	4907223 Heat/Polymerization	Expected polymerization of chloroprene initiated by acrylic acid.	·
1	3 Acrylic Acid	4931405	34 Ethylene Oxide	4906610 Heat/Polymerization	Potentially vigorous polymerization with heat liberation may occur in contact with acids.	
1			Solution		•	
i i	3 Acrylic Acid	4931405	5 Formaldehyde Solution	n 4913168 Heat/Polymerization	Polymerization of formaldehyde initiated by acrylic acid.	i i
	3 Acrylic Acid	4931405	5 Formaldehyde Solution	n 4913144 Heat/Polymerization	Polymerization of formaldehyde initiated by acrylic acid.	
	3 Acrylic Acid					5

AS GR	•	STC / Number C	•	STC Number Compatibility	
	3 Acrylic Acid	4 9314 0 5	104 Hydrogen Peroxide Solution	4918335 Heat/Explosion	Will form explosive peracid which should have explosivity similar to peracetic acid.
1	3 Acrylic Acid	4931405	105 Phosphorus, White or Yellow	4916141 Compatible	No exotherm observed when 12 mixture heated to 60 degrees C.
 	3 Acrylic Acid	49314Ø5	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Polymerization/ Neutralization	Moderate heat generation. Also, potential anionic polymerization by analogy to acrolein/hydroxide.
 [3 Acrylic Acid	4931405	34 Propylene Oxide	4906620 Compatible	No exotherm observed when 12 mixture heated to 200 C.
 I	3 Acrylic Acid	4931405	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Polymerization/ Neutralization	Moderate heat generation. Also, 12 potential anionic polymerization by analogy to acrolein/hydroxide.
1	3 Acrylic Acid	49314 0 5	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Polymerization/ Neutralization	Moderate heat generation. Also, 12 potential anionic polymerization by analogy to acrolein/hydroxide.
 	3 Acrylic Acid	4931405	21 Sodium Metal	4916456 Heat/Polymerization/ Flammable Gas/Fire	(Vigorous hydrogen evolution with) possible explosion.
1	3 Acrylic Acid	49314Ø5 	18 Toluene Diisocyanate	4921575 Heat/Gas	Vigorous reaction forming ArNHC(0)OC(0)CH=CH2, CO2 gas, and heat - polymerization if water present.
	-	•		4931304 Compatible	No exotherm observed when 12 mixture heated up to 300 degrees C.
 			108 Ammonium Nitrate Solution	4918774 Compatible	Minimal reaction anticipated.
1	26]Acrylonitrile	4906420	194 Chlorine	49 04120 Heat/Fire	Potential explosion because, by analogy, bromine has caused acrylonitrile to explode.
i	26 Acrylonitrile	4906420			Possible vigorous polymerization of acrylonitrile by FeCl3.

ASTM GRP Chemics	al Name	STC /	•	STC Number Compatibility	General Comments	REF #
26 Acrylonit	rile	4906420	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Epoxidation of vinyl group (accelerated by base) and resulting exotherm.	1
26 Acrylonit	rile	4906420	31 Pheno I	4921220 Heat/Polymerization	Possible polymerization of acrylonitrile by phenol	
26 Acrylonit	rile	49Ø642Ø	105 Phosphorus, White or Yellow	4918141 Compatible	No exotherm observed when mixture heated up to 60 degrees C.	12
26 Acrylonit	rile	4906420	105 Sulfur Dioxide	49Ø429Ø Unknown	I .	l
4 Alcoholic	Beverage	491 Ø1Ø2	5 Aceta I dehyde	4907210 Heat	Reacts vigorously forming acetal but reaction will be significantly moderated by water content.	
4 Alcoholic	Beverage	4910102	5 Butyraİdehyde	49Ø8119 Heat	Potential vigorous reaction that is significantly moderated by water content.	 -I
4 Alcoholic	Beverage	4910102	104 Chlorine	4904120 Heat/Fire/Explosion	Forms ethyl hypochlorite which can explode on exposure to heat or sunlight - moderated by water.	1
. 4[Alcoholic	Beverage	4910102	34 Ethylene Oxide	4906610 Heat/Polymerization	ISlow reaction at room temperature, moderated by water content.	
	_		104 Hydrogen Peroxide Solution	· · · · · · · · · · · · · · · · · · ·	Addition of alcohols to conc. H202 forms hydroperoxides that can be detonated by shock or heat.	2
4 Alcoholic	Beverage	4910102	34 Propylene Oxide	4906620 Heat/Polymerization	temperature by analogy to alcoholic beverage/ethylene oxide reaction.	1
		[4910102]	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	hydrogen.	2
4 Alcoholic	Beverage				[Vigorous or violent urethane formation plus polymerization.	I

AST GRF		STC ASTM Number GRP Chemical Name	STC Number Compatibility	General Comments	RE	. :
1	Ø Alkaline Liquid, n.o.s.	4935220 107 Acetic Anhydride	49313Ø4 Heat	Violent hydrolysis reaction with potentially explosive boiling. Forms acetate salt.	1	· -
	Ø Alkaline Liquid, n.o.s.	4935220 19 Acetone	4908105 Heat	Slow aldol condensation forms 4-hydroxy-4-methyl-2-pentanone.	 	
1	Ø Alkaline Liquid, n.o.s.	4935220 26 Acrylonitrile	4906420 Heat/Polymerization	Violent polymerization occurs if alkaline is concentrated.	 	1
	LØ]Alkaline Liquid, n.o.s.	4935220 106 Ammonium Nitrate Solution	4918774 Heat	Will form heat.	 	
 : :	lØ Alkaline Liquid, n.ο.s.	4935220 17 Carbon Tetrachloride	9 4940320 Heat/Toxic Gas	Forms CC 30H which converts to phosgene and HCL gas.	1	
1 1	Ø Alkaline Liquid, n.o.s.	4935220 17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potentially vigorous polymerization. Slight solubility in water will limit polymerization rate.	 	
:	lØ Alkaline Liquid, n.o.s.	4935220 27 Dinitrotoluene, Liquid	4963120 Heat/Explosion	Presence of alkali reduces decomposition temperature. Requires temperature >~ 150 C for explosion.	1	2
;	LØ Alkaline Liquid, n.o.s.	[4935220] 13 Ethy Acetate	4909160 Heat	Potentially violent base initiated hydrolysis forming acetic acid and ethanol.	1	- -
 	LØ Alkaline Liquid, n.o.s.	4935220 34 Ethylene Oxide	4906610 Heat/Polymerization Fire	polymerization with heat liberation will occur in contact with bases.	1	1
1	10 Alkaline Liquid, n.o.s.	4935220 107 Maleic Anhydride	4941161 Heat	Decomposes explosively in presence of alkalis forming maleic acid salt.		1
 [10 Alkaline Liquid, n.o.s.	4935220 17 Methy Chioride	4905761 Heat	[Rapid formation of methanol and chloride salt. May be violent.	1	
		4935220 19 Methy! Ethyl Ketone		Slow aldol reaction.	1	

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	General Comments	REF #
10 Alkaline Liquid, n.o.s.	4935220 105 Phosphorus, White or Yellow	4916141 Heat/Flammable Gas/Toxic Gas	Yields phosphine gas when the mixture is boiled.	 1
10 Alkaline Liquid, n.o.s.	4935220 34 Propylene Oxide	4906620 Heat/Polymerization/ Fire	Base catalyzed polymerization may be violent or explosive.	2
18 Alkaline Liquid, n.o.s.	4935220 21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Forms hydrogen gas and heat.	8
10 Alkaline Liquid, n.o.s.	4935220 16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potential polymerization mitigated by low solubility of styrene in water.	 I 1
10 Alkaline Liquid, n.o.s.	4935220 105 Sulfur Dioxide	4904290 Heat/Neutralization	Reacts to form alkali sulfites in solution or bisulfites if excess SO2 present.	1 1
10 Alkaline Liquid, ก.o.s.	4935220 18 Toluene Diisocyanate	4921575 Heat/Polymerization/ Gas	Exothermic hydrolysis to carbamate followed by polymerization will occur in contact with bases.	2
10 Alkaline Liquid, n.o.s.	4935220 13 Viny Acetate	4907270 Heat	Potentially violent base initiated hydrolysis forming acetic acid and acetaldehyde.	1 1
10 Alkaline Liquid, n.o.s.	4935220 17 Vinyl Chloride	4905792 Heat/Polymerization/ Flammable Gas	Potentially vigorous polymerization with acetylene formation. Mitigated by low solubility in water.	1 1
106 Ammonium Nitrate Solution	4918774 107 Acetic Anhydride	4931304 Heat/Fire	Probably ignites by analogy to acetic acid/ammonium nitrate. Forms acetic acid.	i !
106 Ammonium Nitrate Solution	[4918774] 107 Maleic Anhydride	4941161 Heat	Reaction of maleic anhydride with cold water is slow and non-hazardous. Forms maleic acid.	11
10 Anhydrous Ammonia	4904210 107 Acetic Anhydride	49313 0 4 Heat	Vigorous exothermic reaction with possible explosive boiling forms acetamide and ammonium acetate.	1 1

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility		REF #
10 Anhydrous Ammonia	4904210 19 Acetone	4 9 0 81 0 5 Heat	Possibly violent reaction forms 2-propaneimine Moderated by volatility of NH3.	!
10 Anhydrous Ammonia	4904210 26 Acrylonitrile	4906420 Heat/Polymerization	Violent polymerization of acrylonitrile upon contact with amines but moderated by volatility of NH3.	5
10 Anhydrous Ammonia	4904210 106 Ammonium Nitrate Solution	4918774 Compatible	No reaction expected.	
10 Anhydrous Ammonia	4904210 17 Carbon Tetrachloride	4940320 Heat	Possible violent chloride displacement reactions.	 l
10 Anhydrous Ammonia	4904210 104 Chlorine	4904120 Heat/Explosion	Excess of chlorine in ammonia explodes due to formation of extremely unstable NC13.	1
10 Anḥydrous Ammonia	4904210 17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Possible vigorous polymerization reaction initiated by NH3.	 I
10 Anhydrous Ammonia	4904210 27 Dinitrotoluene, Liquid	4963120 Heat	· · · · · · · · · · · · · · · · · · ·	
10 Anhydrous Ammonia	4904210 13 Ethyl Acetate	4909160 Heat	Vigorous reaction forming acetamide and ethanol.	
10 Anhydrous Ammonia	4904210 34 Ethylene Oxide	4906610 Heat/Polymerization Fire	/ Violent polymerization with heat liberation. Reaction moderated by volatility of NH3.	2
10 Anhydrous Ammonia	4904210 107 Maleic Anhydride	4941161 Heat	Decomposes explosively in amines forming maleic acid monoamide and aminosuccinic anhydride.	
	4904210 17 Methyl Chloride	Gas/Toxic Gas	Potentially violent displacement reaction forming CH3NH2 and HC1 plus exothermic neutralization.	
	4904210 19 Methyl Ethyl Ketone		Exothermic formation of 2-butaneimine.	 J 1
10 Anhydrous Ammonia	4904210 24 Motor Fuel Antiknock Compound	: 4921445 Heat/Flammable Gas	Potential ethane formation.	

ASTM GRP Chemical Name	STC Number	, ·	STC Number Compatibility	General Comments	REF #
10 Anhydrous Ammonia	4904210	34 Propylene Oxide]4906620 Heat/Polymerization/ Fire	/ Violent polymerization reaction similar to ethylene oxide/ammonia reaction.	 [
10 Anhydrous Ammonia	49 0 421 0	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Reacts slowly to give NaNH2 and H2. Sodium amide is very reactive with water or air.	4
10 Anhydrous Ammonia	4 9 0 421 0	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Vigorous/violent reaction which may be mitigated by low volatility of NH3.	[·
10 Anhydrous Ammonia	4904210	105 Sulfur Dioxide	4904290 Heat/Neutralization	SO2 is soluble in liquid NH3. Reacts to form amides such as NH4SO2NH2.	
10 Anhydrous Ammonia	4904210	18 Toluene Diisocyanate	4921575 Heat	Violent reaction forms Ar-NHC(0)NH2.	9
10 Anhydrous Ammonia	4904210	13 Vinyl Acetate]4907270 Heat/Polymerization	Vigorous reaction forming acetamide and acetaldehyde.	I .
10 Anhydrous Ammonia	4904210	17 Vinyl Chloride	4905792 Heat/Polymerization/ Flammable Gas	/ Potentially violent polymerization. Acetylene can be formed.]
7 Aniline Oil, Liqui	d 4921410	107 Acetic Anhydride	49313Ø4 Heat	Violent reaction with excessive heat generated forming >>> N-phenylacetamide.	12
7¦Aniline Oil, Liqui	,		4906420 Heat/Polymerization	acrylonitrile upon contact with amines.	 5
7 Aniline Oil, Liqui	d 4921410	17 Carbon Tetrachloride		Substitution reaction may occur at unknown rate.	*
7 Aniline Oil, Liqui	d 4921410	104 Chlorine	4904120 Heat/Fire	Violent chloramine formation.	1
7 Aniline Oil, Liqui	d 4921410	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	polymerization.	l
				Expected rapid polymerization by analogy to ammonia/ethylene oxide.	

AST GRP	•	Chemi	ical	Name	STC / Number 0	•	STC Number Compatibility	 General Comments	 REF #
	7 Ani	iline	Oil,	Liquid	[4921410]	104 Ferric Chloride Solution	4932342 Heat	Weak complexation.	
	7 A n i	line	Oil,	Liquid	492141 0	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Exothermic conversion to hydroxylamine. Potential explosion of H202 caused by heating.	
	7 A n i	iline	Oil,	Liquid	4921410	107 Maleic Anhydride	4941161 Heat	Decomposes explosively in presence of amines forming N-phenylmaleic acid monoamide.	 1
	7 A ni	iline	Oil,	Liquid	4921410	17 Methyl Chloride	490 5761 Heat	Rapid, potentially violent alkylation of aniline forms N-methylaniline hydrochloride.	
	7 Ani	iline	0i1,	Liquid	4921410	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas	Potential ethane formation.	I I
	7 A n i	iline	Oil,	Liquid	4921410	105 Phosphorus, White or Yellow	4916141 Compatible	No exotherm observed when mixture was heated to 60 degrees C.	12
	7 A n i	iline	Oil,	Liquid	492141 <i>0</i>	34 Propylene Oxide	4906620 Heat/Polymerization	Exotherm observed when mixture was heated between 200 and 300 C. Possible polymerization.	12
	7 An	iline	Dil	, Liquid	4921410	21 Sodium Metal	[4916456 Heat/Flammable Gas/Fire	Violent electron transfer expected analagous to Birch reduction. Hydrogen evolution expected.	· 1 I
	7 An	iline	Oil	, Liquid	4921410	18 Toluene Diisocyanate	4921575 Heat	Violent reaction with urea formation.	12
	7 An	iline	0il	Liquid	4921410		Flammable Gas	Potentially violent anionic polymerization. May form acetylene.	
 :	 16 Bei	nzene			4908110	104 Chiorine	4904120 Heat/Fire	Explosion of benzene vapors and chlorine was initiated by light.	
\	 16 Ber	nzene			4908110	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	May form combinations that are sensitive to heat or shock.	10

	STM RP Chemica	STC / Name Number 0	,	STC Number Compatibility		8EF #
	28 Butadiene	4905704	196 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	<u>\</u>
	28 Butadiene	4905704	104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1
	28 Butadiene	[4905704]	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Analagous to AlCl3 in non-aqueous environment being used to polymerize butadiene.	
	28 Butadiene	4905704	104 Hydrogen Peroxide Solution	4918335 Heat/Polymerization/ Explosion	Very reactive polymerization initiator with trace of Fe+2 (Fenton's reagent). Epoxidation may occur.	 I
	28 Butadiene	4905704	31 Pheno I	4921220 Heat/Polymerization	Pressure increased when combined in presence of a catalyst and solvent causing bottle to rupture.	1
-	28 Butadiene	[4905704]	105 Phosphorus, White or Yellow	[4916141 Compatible	[No exotherm observed when mixture was heated to 60 degrees C.	12
	28 Butadiene,	Inhibited 4905703	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 i
	28 Butadiene,	Inhibited 4905703	104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1
	28 Butadiene,	Inhibited 4905703	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Probable slow polymerization as with non-inhibited material.	
!	28 Butadiene,		104 Hydrogen Peroxide Solution	Explosion	Exotherm from epoxidation and polymerization may cause H2O2 to explode.	
	28 Butadiene,		31 Pheno I		Pressure increased when combined in presence of a catalyst and solvent causing bottle to rupture.	

	STM RP Chemical Name	STC Number	•	STC Number Compatibility		R⊟ #	•
	28 Butadiene, Inhibite	d 49Ø57Ø3	105 Phosphorus, White or Yellow	4916141 Compatible	No exotherm observed when mixture was heated to 60 degrees C.	1 1	- 2
	29 Butane	49 Ø 57 Ø 2	104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	 	- 1
	29 Butane	[4905706]	184 Hydrogen Peroxide Solution	4918335 Compatible	No reaction anticipated.	 	- <i>'</i>
<u>-</u> .	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 	-
- 	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215	104 Chlorine	49 0 412 0 Heat/Fire	Potentially vigorous/violent oxidation	 	- !
 	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Possible polymerization.	ľ	. <u> </u>
 [13 Butyl Acrylate (Corr. L., n.o.s.)	4912215	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Epoxidation (or polymerization) can lead to exotherm causing explosion of H2O2.		
l	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	1	 -
ı	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215	21 Sodium Metal	4916456 Heat/Polymerization, Fire	/ Probable vigorous/violent polymerization with acyloin formation - 5-hydroxy-4-octanone.		1
1	4 Buty! Alcohol		5 Acetal dehyde		Reacts vigorously forming 1,1-di(n-butoxy) ethane.	1	5
- [4 Butyl Alcohol		5 (Butyra I dehyde	•	Potentia! vigorous reaction forming 1,1-di(n-butoxy) butane.	l	 I
- ا					Forms butyl hypochlorite which can violently explode on exposure to heat or sunlight.		1
1				4906610 Heat/Polymerization		\ \ :	

AS	•	STC ASTM Number GRP Chemical Name	STC Number Compatibility	REF General Comments #
	4 Butyl Alcohol	4909129 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Addition of alcohols to conc. 2 H2O2 forms hydroperoxides that can be detonated by shock or heat.
1	4 Butyl Alcohol	4909129 34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.
1	4 Butyl Alcohol	4909129 21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Hydrogen evolution.
	4 Butyl Alcohol	49 0 9129 18 Toluene Diisocyana	te 4921575 Heat	Violent urethane formation (ArNHCD2C8H17).
1	5 Butyraldehyde	[4908119] 107 Acetic Anhydride	49313Ø4 Heat	Condensation reaction (abnormal 1 Perkins reaction) is probably slow. Forms vinyl acetate.
1	5 Butyra dehyde	4908119 10 Alkaline Liquid, n.o.s.	4935220 Heat/Polymerization	Potentially violent aldol condensation forms 2-ethyl-3-hydroxybutanal and polymerization.
	5 Butyraldehyde	4908119 10 Anhydrous Ammonia	4904210 Heat/Polymerization	Potentially violent butaneimine formation and polymerization by analogy to acetaldehyde/ammonia.
1	5 Butyra I dehyde	4908119 7 Aniline Cil, Liqui	d 4921410 Heat/Polymerization	Minimal heat generated. Possible 12 polymerization and imine formation - N-butylidenebenzeneamine.
1	5 Butyraldehyde	4908119 104 Chlorine	4904120 Heat/Fire	Probable rapid oxidation.
	5 Butyra dehyde	4908119 27 Dinitrotoluene, Liquid	4963120 Heat/Explosion	Potentially vigorous reaction under basic conditions.
	5 Butyra I dehyde	4908119 34 Ethylene Oxide	4906810 Compatible	May form cyclic acetal but reaction is expected to be slow at ambient temperature.
!	5 Butyra I dehyde	4908119 104 Ferric Chloride Solution	4932342 Heat/Polymerization	Possible vigorous cationic polymerization by analogy to acetaldehyde/FeCl3.

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	
5 Butyra I dehyde	4908119 7 Hexamethylene Diamine Solution	e 4935645 Heat/Polymerization	Potentially vigorous polymerization and imine formation - N,N'-dibutylidene-1,6-hexanediam ine.
5 Butyra I dehyde	4908119 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Alpha-hydroperoxides will be formed which are explosive.
5 Butyraldehyde	4908119 31 Phenol	4921220 Heat/Fire	Reacts vigorously forming 1,1-bis(4-hydroxyphenyl) ethane. Promoted by acid.
5 Butyraldehyde	4908119 105 Phosphorus, White or Yellow	4916141 Compatible	No exotherm observed when 12 mixture heated to 60 degrees C.
5 Butyraldehyde	4908119 10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Polymerization	Potentially violent aldol condensation forms 2-ethyl-3-hydroxybutanal and polymerization.
5 Butyra I dehyde	4908119 34 Propylene Oxide	4906620 Compatible	[Possible cyclic acetal [formation. Slow reaction at ambient temperature.
5 Butyra I dehyde.	4908119 10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Polymerization	Potentially violent aldol condensation forms 2-ethyl-3-hydroxybutanal and polymerization.
5]Butyraldehyde	4908119 10 Sodium Hydroxide, liq. or solution	4935243 Heat/Polymerization	condensation forms 2-ethyl-3-hydroxybutanal and polymerization.
	4908119 21 Sodium Metal		Very violent reaction forms pinacol dianion formation.
5 Butyraldehyde	4908119 18 Toluene Diisocyanate	4921575 Heat	Potential reaction of enol.
20 Carbon Disulfide	4908125 104 Chlorine	4904120 Heat/Fire/Toxic Gas	Liquid chlorine added to CS2 in 1 presence of an iron surface resulted in an explosion.
	4908125 34 Ethylene Oxide		

	STM Cher	nical Name	STC / Number (•	STC Number Compatibility	 General Comments	:	REF #
_	20 Carbon	Disulfide	4900125	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Possible polymerization of CS2.	.1	
-	20 Carbon	Disulfide	49Ø8125	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Toxic Gas/Explosion		Ì	
•	20 Carbon	Disulfide	4908125	34 Propylene Oxide	4906620 Compatible	Exotherm observed when mixture was heated between 200 and 300 degrees C.	1	12
	20 Carbon	Disulfide	4908125	21 Sodium Metal	4916456 Heat/Fire	Mixtures of sodium and CS2 are shock-sensitive explosives.	 	2
_	20 Carbon	Disulfide	49Ø8125	105 Sulfur Dioxide	[4904290 Compatible	[SO2 is soluble in CS2 when heated, separates out when cooled.	 [6
	17 Carbon	Tetrachlori	de 494Ø32Ø	20 Carbon Disulfide	49Ø8125 Heat	1	 	
-	17 Carbon	Tetrachlori	de 494Ø32Ø	104 Chlorine	4904120 Compatible	Probably no reaction.		
-	17 Carbon	Tetrachlori	de 494Ø32Ø	104 Hydrogen Peroxide Solution	4918335 Compatible	No reaction expected.	 	
	17 Carbon	Tetrachlorio	de 494Ø32Ø	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	1	
-	17 Carbon	Tetrachlori	de 494Ø32Ø	21 Sodium Metal	4916456 Heat	Reacts violently with hot	1	1
	1 04 Chiori	n e	4904120	107 Acetic Anhydride	4931304 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	 	1
-	104 Chioria	ne	4904120	106 Ammonium Nitrate Solution		Forms chloramine nitrate, nitrosyl chloride, NHC12 and heat that may cause NH4NO3 to explode.	1	
_	104 Chlori		4904120	107 Maleic Anhydride	4941161 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.		1
	1 04 Chlori				4916141 Heat/Fire/Toxic Gas	White phosphorus burns spontaneously or reacts explosively in gaseous chlorine forming PC13.		1

ASTM GRP Chemical Name	STC / Number (STC Number Compatibility	 General Comments	RE	. :
17 Chloroprene, Inhibited	[4907223]	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.		· -
17 Chloroprene, Inhibited	4907223	20 Carbon Disulfide	4908125 Compatible	Probably no reaction.	 	
17 Chioroprene, Inhibited	4907223	1 04 Chlorine	4904120 Heat/Fire	Potentially violent 1,2- and 1,4- chlorination of diene.	1	
17 Chloroprene, Inhibited	4907223	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Possible rapid, violent polymerization mitigated by water content.	1	 1
17 Chloroprene, Inhibited	4907223	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Probable polymerization initiator. Heat could lead to H2O2 decomposition causing an explosion.	1	
17 Chloroprene, Inhibited	4907223	24 Motor Fuel Antiknock Compound	4921445 Heat/Polymerization	Potential anionic polymerization.	 	
17 Chloroprene, Inhibited	4907223	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	ľ	 J
17 Chloroprene, Inhibited	4907223	21 Sodium Metal	4916456 Heat/Polymerization/ Fire	/ Sodium forms explosive mixtures with chlorinated hydrocarbons. Violent anionic polymerization.	Ĭ	1
101 Coal Tar Distillate	4915363	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.		
101 Coal Tar Distillate			4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.		
101 Coal Tar Distillate	4915363	104 Hydrogen Peroxide Solution		Heat evolution if alkenes, aldehydes, ketones, or alcohols are present - peroxidation/epoxidation.		
		106 Ammonium Nitrate Solution		[Ammonium nitrate solutions containing significant water are essentially not flammable.	 	

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	
101 Combustible Liquid, n.o.s.	4915185 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Combustible Liquid, n.o.s.	4915185 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Heat evolution if alkenes, aldehydes, ketones, or alcohols are present - peroxidation/epoxidation.
101 Compound, Cleaning, Liquid	4915147 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.
101 Compound, Cleaning, Liquid	4915147 104 Chiorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Compound, Cleaning, Liquid	4915147 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	
101 Crude Oil Petroleum	4910165 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.
101 Crude Oil Petroleum	[4910165 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Crude Oil Petroleum	4910165 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Heat evolution if contains 10 alkenes. May form combinations that are sensitive to heat or shock.
29 Cyclohexane	4908132 104 Chlorine -	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
29 Cyclohexane	4908132 104 Hydrogen Peroxide Solution	4918335 Compatible	May form combinations that are 10 sensitive to heat or shock.
4 Denatured Alcohol	4909151 5 Acetaldehyde	4907210 Heat	Reacts vigorously forming acetal 5 (b.p. = 103 C).

AST GRP	•	STC ASTM Number GRP Chemical Name	STC Number Compatibility	
- 	4 Denatured Alcohol	4909141 , 5 Aceta dehyde	4907210 Heat	Reacts vigorously forming acetal 5 (b.p. = 103 C).
J 	4 Denatured Alcohol	4909151 5 Butyraldehyde	4908119 Heat	Potential vigorous reaction forming 1,1-diethoxybutane.
 	4 Denatured Alcohol	4909141 5 Butyraldehyde	4 908119 Heat	Potential vigorous reaction forming 1,1-diethoxybutane.
1	4 Denatured Alcohol	4909151 104 Chlorine	4904120 Heat/Fire/Explosion	Forms ethyl hypochiorite which 1 can violently explode on exposure to heat or sunlight.
I 	4 Denatured Alcohol	4909141 104 Chlorine	4904120 Heat/Fire/Explosion	Forms ethyl hypochlorite which 1 can violently explode on exposure to heat or sunlight.
I	4 Denatured Alcohol	4909151 34 Ethylene Oxide	4906610 Heat/Polymerization	Slow reaction at room 1 temperature in absence of acidic or basic catalysts.
1	4 Denatured Alcohol	4909141 34 Ethylene Oxide	4906610 Heat/Polymerization	Slow reaction at room 1 temperature in absence of acidic or basic catalysts.
I	4 Denatured Alcohol	4909151 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Addition of alcohols to conc. 2 H2O2 forms hydroperoxides that can be detonated by shock or heat.
i	4 Denatured Alcohol	4909141 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Addition of alcohols to conc. 2 H202 forms hydroperoxides that can be detonated by shock or heat.
	*	4909151 34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.
!	•	4909141 34 Propylene Oxide	,	Slow reaction at room temperature in absence of acidic or basic catalysts.
 -	4 Denatured Alcohol	4909151 21 Sodium Metal		Violent reaction evolves 2 hydrogen.

AST) GRP	•	STC		STC Number Compatibility		[REF]
	Denatured Alcohol	4909141	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Violent reaction evolves hydrogen.	 2
 } 4	4 Denatured Alcohol	4909151	18 Toluene Diisocyanate	4921575 Heat/Fire	Potentially violent urethane formation (ArNHCO2C8H17) with excessive heat generation.	 12
		4909141	18 Toluene Diisocyanate	4921575 Heat/Fire	Potentially violent urethane formation (ArNHCO2C8H17) with excessive heat generation.	 12
27	7 Dinitrotoluene, Liquid	4963120	107 Acetic Anhydride	49313Ø4 Heat	Slow acetylation of methyl group - catalyzed by bases.	i
27	7 Dinitrotoluene, Liquid	4963120	164 Chiorine	4904120 Heat/Explosion	Potentially violent chlorination (oxidation) of methyl groups.	<u> </u>
27	7 Dinitrotoluene, Liquid	4963120	104 Ferric Chloride Solution	[4932342 Compatible	Reaction not self-evident.	I I
27	7 Dinitrotoluene, Liquid	4963120	104 Hydrogen Peroxide Solution	4918335 Heat/Explosion	Have activated methyl hydrogen atoms for abstraction by hydroxy! radical with potential exotherm.	i 1
27	7 Dinitrotoluene, Liquid	4963120	107 Maleic Anhydride	4941161 Heat/Polymerization	Slow conjugate addition and consequent polymerization.	I
2	7 Dinitrotoluene, Liquid	4963120	105 Phosphorus, White or Yellow	4916141 Heat/Explosion	ļ	1
10:	1 Distillate Fuel Oil	2911315	108 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	
10:	1 Distillate Fuel Oil	2911315	184 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1
10	1 Distillate Fuel Oil	2911315	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	If unsaturation exists, heat evolution can occur. May form combinations sensitive to heat or shock.	 10
13	3 Ethyl Acetate	4909160	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	

STM	STC ASTM	STC	I	[REF
RP Chemical Name	Number GRP Chemical Name	Number Compatibility	General Comments	#
13 Ethyl Acetate	4909160 104 Chlorine	4904120 Heat/Fire	Potentially violent oxidation reaction.	١
13 Ethy! Acetate	4909180 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	[Possible formation of perester {CH3C(0)00CH2CH3} can lead to exotherm causing explosion of H202.	l
13 Ethyl Acetate	4909160 105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	l
13 Ethyl Acetate	4909160 21 Sodium Metal	4916456 Heat/Fire	Potentially violent acyloin reaction forming 3-hydroxy-2-butanone.	I
13 Ethyl Acrylate, inhibited	4907215 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 e
13 Ethyl Acrylate, inhibited	4907215 104 Chlorine	4904120 Heat/Fire	Potentially violent addition and oxidation reactions.	4
13 Ethyl Acrylate, inhibited	4907215 104 Ferric Chloride Solution	4932342 Heat/Polymerization	Possible polymerization.	l
13 Ethyl Acrylate, inhibited	4907215 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Epoxidation (or polymerization) can lead to exotherm causing explosion of H202.	1
13 Ethyl Acrylate, inhibited	4907215 105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	1
13 Ethyl Acrylate, inhibited	4907215 21 Sodium Metal	4916456 Heat/Polymerization/ Fire	acyloin condensation forms 4-hydroxy-1,5-hexadien-3-one.	1
4 Ethyl Alcohol		4907210 Heat	Reacts vigorously forming aceta (b.p. = 103 C).	11 :
4 Ethyl Alcohol	4909159 5 Butyraldehyde	4908119 Heat	Potential vigorous reaction forming 1,1-diethoxybutane.	ľ
	4909159 104 Chlorine			

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	REF General Comments #
4 Ethyl Alcohol	4909159 34 Ethylene Oxide	4906610 Heat/Polymerization	Slow reaction at room ! 1 temperature in absence of acidic or basic catalysts.
4 Ethyl Alcohol	4909159 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Addition of alcohols to conc. 2 H202 forms hydroperoxides that can be detonated by shock or heat.
4 Ethyl Alcohol	4909159 34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.
4 Ethyl Alcohol	49 0 9159 21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Violent reaction evolves 2 hydrogen.
4 Ethyl Alcohol	4909159 18 Toluene Diisocyanate	4921575 Heat/Fire	Potentially violent urethane 12 formation (ArNHCO2C8H17) with excessive heat generation.
34 Ethylene Oxide	4906610 106 Ammonium Nitrate Solution	4918774 Compatible	Slow, non-hazardous reaction 11 forms ethylene glycol in water.
34 Ethylene Oxide	4906610 104 Chiorine	49 04120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.
34 Ethylene Oxide	4906610 104 Ferric Chloride Solution	4932342 Heat/Polymerization	Polymerization with heat 1 liberation may occur in contact with aqueous iron chlorides.
34 Ethylene Oxide	4906610 104 Hydrogen Peroxide Solution		Alpha positions are activated to attack by hydroxyl radicals generating hydroperoxides - explosive.
34 Ethylene Oxide	4906610 105 Phosphorus, White or Yellow		1
34 Ethylene Oxide		4904290 Unknown	1
104 Ferric Chloride Solution	4932342 107 Acetic Anhydride		Vigorous hydrolysis forms a 6 solution that is electrically conductive.

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility		REF #
104 Ferric Chloride Solution	4932342 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium salts dissolve in aqueous solution with cooling.	i
104 Ferric Chloride Solution	4932342 1 04 Hydrogen Peroxide Solution	4918335 Heat/Explosion	Potentially violent release of oxygen by Fe+3 catalyzed decomposition of H202.	
104 Ferric Chloride Solution	4932342 107 Maleic Anhydride	4941181 Heat	Potentially vigorous hydrolysis of anhydride catalyzed by FeCl3 foriming maleic acid.	,
104 Ferric Chloride Solution	4932342 105 Phosphorus, White or Yellow]4916141 Heat	Forms phosphoric acid and ferrous chloride.	6
! 101 Flammable Liquid, n.o.s.	4910185 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 -
101 Flammable Liquid, n.o.s.	4910185 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1
101 Flammable Liquid, n.o.s.	4910185 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Heat evolution if liquid contains unsaturated hydrocarbons, aldehydes, ketones, alcohols.	I
5 Formaldehyde Solution	4913144 107 Acetic Anhydride	49313 04 Heat	Abnormal Perkin reaction may occur slowly (catalyzed in basic conditions). Also,hydrolysis reaction.	1
5 Formaldehyde Solution	4913168 107 Acetic Anhydride	4931304 Heat	Abnormal Perkin reaction may occur slowly (catalyzed in basic conditions). Also,hydrolysis reaction.	!
5 Formaldehyde Solution	4913168 10 Alkaline Liquid, n.o.s.	4935220 Heat/Polymerization	polymerization. Also, Cannizzaro reaction forming methanol and formic acid.	
5 Formaldehyde Solution	4913144 10 Alkaline Liquid, n.o.s.	4935220 Heat/Polymerization	Potentially violent polymerization. Also, Cannizzaro reaction forming methanol and formic acid.	1

AST GRP	•	STC / Number 0	·	STC Number Compatibility	 General Comments	RE	. :
1	5 Formaldehyde Solution	4913168	10 Anhydrous Ammonia	4904210 Heat/Polymerization	Condensation reaction forming methaneimine or hexamethylene tetramine. Potential polymerization.	1	.3
	5 Formaldehyde Solution	[4913144]	10 Anhydrous Ammonia	4904210 Heat/Polymerization	Condensation reaction forming methaneimine or hexamethylene tetramine. Potential polymerization.	1	 13
	5 Formaldehyde Solution	[4913168]	7 Aniline Oil, Liquid	4921410 Heat/Polymerization	[Vigorous reaction forms N-methylidenebenzeneamine. Mitigated by water content.	l	. <u> </u>
 	5 Formaldehyde Solution	4913144	7 Aniline Oil, Liquid	4921410 Heat/Polymerization	Vigorous reaction forms N-methylidenebenzeneamine. Mitigated by water content.	İ	. <u> </u>
 	5 Formaldehyde Solution	[4913168]	104 Chlorine	4904120 Heat/Fire	May be violent.	···	 -
 !	5 Formaldehyde Solution	4913144	104 Chlorine	4904120 Heat/Fire	May be violent.	· 	 I
1	5 Formaldehyde Solution	4913168	27 Dinitrotoluene, Liquid	4963120 Compatible	No reaction expected.	ì	-
 	5 Formaldehyde Solution	4913144	27 Dinitrotoluene, Liquid	4963120 Compatible	No reaction expected.	. 1	۱-
 I	5 Formaldehyde Solution	4913168	34 Ethylene Oxide	4906610 Heat	Acetal formation may be faster than acetaldehyde/ethylene oxion rate but moderated by water content.		
 I	5 Formaldehyde Solution	4913144	34 Ethylene Oxide	49Ø661Ø Heat	Acetal formation may be faster than acetaldehyde/ethylene oxion rate but moderated by water content.		 I
J	5 Formaldehyde Solution	4913168	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Polymerization of formaldehyde moderated by water content.	Î	
 	5 Formaldehyde Solution	4913144	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Polymerization of formaldehyde moderated by water content.		. <u>-</u>

ASTM GRP	Chemical Name	STC A Number G		STC Number Compatibility	 General Comments	: .	EF #
•	maldehyde ution	4913144	7 Hexamethylene Diamine Solution	4935645 Heat/Polymerization	[Polymerization of formaldehyde and formation of amine N-methylols and imines.	l ,	
•	maldehyde ution	[4913168]	7 Hexamethylene Diamine Solution	e 4935645 Heat/Polymerization	Polymerization of formaldehyde and formation of amine N-methylols and imines.	1	 I
-	maldehyde ution	4913144	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Impact sensitive compounds may be formed depending upon hydroxyhydroperoxides concentration.	1	3
	maldehyde ution	[4913168]	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Impact sensitive compounds may be formed depending upon hydroxyhydroperoxides concentration.	1	3
	maldehyde ution	4913144	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	 	
	maldehyde ution	4913168	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected	I	
-	maldehyde ution	4 913144	10 Potassium Hydroxide, Liq. or solution	493523Ø Heat/Polymerization	Vigorous polymerization and Cannizzaro reaction forming methanol and formic acid.	1	
-	maldehyde ution	4913168	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Polymerization	Vigorous polymerization and Cannizzaro reaction forming methanol and formic acid.	l	!
	maidehyde ution	4913144	34 Propylene Oxide	4906620 Compatible	Probably slow rate of formation of cyclic acetal at ambient temperature.	1	
j 5 For	maldehyde ution		34 Propylene Oxide		Probably slow rate of formation of cyclic acetal at ambient temperature.		
Sol	maldehyde ution		10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Polymerization	Vigorous polymerization and Cannizzaro reaction forming methanol and formic acid.		-
5 For	maldehyde ution		10 Sodium Hydroxide, liq. or solution		Vigorous polymerization and Cannizzaro reaction forming methanol and formic acid.	 	

ASTM GRP Chemical Name	STC ASTM Number GRP Chem	mical Name	STC Number Compatibility	General Comments	REF #
5 Formaldehyde Solution	4913168 10 Sodium Liq. on	Hydroxide, r solution	4935240 Heat/Polymerization	Vigorous polymerization and Cannizzaro reaction forming methanol and formic acid.	
5 Formaldehyde Solution	4913168 10 Sodium liq. or	Hydroxide, r solution	4935243 Heat/Polymerization	Vigorous polymerization and Cannizzaro reaction forming methanol and formic acid.	
5 Formaldehyde Solution	4913144 21 Sodium	Metal .	[4916456 Heat/Flammable Gas/Fire	Hydrogen evolution.	
5 Formaldehyde Solution	4913168 21 Sodium	Metal	4916456 Heat/Flammable Gas/Fire	Hydrogen evolution.	1 1
5 Formaldehyde Solution	4913168 18 Toluend	e Diisocyanate	4921575 Heat	Potential reaction of enol.	
5 Formaldehyde Solution	4913144 18 Toluene	e Diisocyanate	4921575 Heat	Potential reaction of enol.	" .
101 Fuel Oil	4915113 106 Ammonio Solutio		4918774 Compatible	Ammonium nitrate solutions containing significant water an essentially not flammable.	 re
101 Fuel Oil	4915112 106 Ammoniu Solutio		4918774 Compatible	Ammonium nitrate solutions containing significant water an essentially not flammable.	 re -
 101 Fuel Dil	4915111 106 Ammonio 		4918774 Compatible	Ammonium nitrate solutions containing significant water a essentially not flammable.	re
101 Fuel Oil	4915112 104 Chlori		4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	h 1
10/1 Fuel Oil		ne	49 0 412 0 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	h 1
101 Fuel Oil				Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	h 1
·	4915112 104 Hydrog Solutio	en Peroxide on		If alkenes, aldehydes or keton are present, heat will be evolved from peroxidation/epoxidation.	 es

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	
101 Fuel Oil	[4915111] 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	If alkenes, aldehydes or ketones are present heat will be evolved from peroxidation/epoxidation.
101 Fuel Oil	4915113 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	If alkenes, aldehydes or ketones are present, heat will be evolved from peroxidation/epoxidation.
101 Fuel, Aviation, Turbine Engine	4915167 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.
101 Fuel, Aviation, Turbine Engine	4909215 108 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.
101 Fuel, Aviation, Turbine Engine	4909215 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
{ 101 Fuel, Aviation, Turbine Engine	4915167 104[Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Fuel, Aviation, Turbine Engine	4909215 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Minimal effect expected unless fuel contains unsaturation, aldehydes, ketones or alcohols.
101 Fuel, Aviation, Turbine Engine	4915167 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Minimal effect expected unless fuel contains unsaturation, aldehydes, ketones or alcohols.
101 Gasoline	4908178 106 Ammonium Nitrate Solution		Ammonium nitrate solutions containing significant water are essentially not flammable.
101 Gasoline	4908178 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Gasoline	4908178 104 Hydrogen Peroxide Solution		Minimal effect due to lack of 10 functionalization. May form combinations sensitive to heat or shock.

ASTM GRP Ch	nemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	
101 Gaso	ine, nec	2911190 106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.
101 Gaso	ine, nec	2911198 184 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Gaso	ine, nec	2911190 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Minimal effect due to lack of 10 functionalization. May form combinations sensitive to heat or shock.
	ol Ethers, nec	4913194 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons, causing ignition.
_	ol Ethers, nec , n.o.s.)	4913194 34 Ethylene Oxide	49Ø661Ø Compatible	No reaction expected.
	ol Ethers, nec	4913194 104 Ferric Chloride Solution	4932342 Heat	Complexation occurs.
	ol Ethers, nec	4913194 104 Hydrogen Peroxide Solution	4918335 Explosive Mixture	Ethers are used as solvents for H202 reactions but unstable hydroperoxides may form if Fe+2 present.
-	ol Ethers, nec , n.o.s.)	4913194 34 Propylene Oxide	4906620 Compatible	No reaction expected (unless alcohol function is present)
	ol Ethers, nec	4913194 21 Sodium Metal	4916456 Compatible	No reaction expected if alcohol functionality is not present. Ether may be used as solvent for Na.
	ol Ethers, nec , n.o.s.)	4913194 18 Toluene Diisocyanate	4921575 Compatible	No reaction expected (unless alcohol function is present)
-	nethylene ine Solution	4935645 107 Acetic Anhydride	49313 04 Heat	Potentially violent formation of N,N'-diacety -1,6-hexanediamine with explosive boiling.
·	nethylene ine Solution	4935645 26 Acrylonitrile	4906420 Heat/Polymerization	Violent polymerization upon 5 contact with amines.

AST GRF	•	STC A Number G	•	STC Number Compatibility	 General Comments	[RE
	7 Hexamethylene Diamine Solution	4935645	17 Carbon Tetrachloride	4940320 Heat	Potential vigorous displacement or reaction.	l
	7[Hexamethylene Diamine Solution	493 5645	1Ø4 Chlorine	4904120 Heat/Fire	Vigorous chloramine formation moderated by water content.	1
	7 Hexamethylene Diamine Solution	4935645	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potentially vigorous or violent polymerization.	Ι
	7 Hexamethylene Diamine Solution	4935645	34 Ethylene Oxide	4906610 Heat/Polymerization	Potentially violent ring opening and polymerization.	ıl
	7 Hexamethylene Diamine Solution	[4935645]	104 Ferric Chloride Solution	4932342 Heat	Complexation.	1
	7 Hexamethylene Diamine Solution	4935645	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Forms hydroxylamines and other oxidation products. Potential explosion of H2O2 caused by heating.	1
	7[Hexamethylene Diamine Solution	4935645	107 Maleic Anhydride	4941161 Heat	Decomposes explosively in presence of amines.	i
	7 Hexamethylene Diamine Solution	4935645	17 Methyl Chloride	49 0 5761 Heat	Potentially violent displacement reaction.	;
1	7 Hexamethylene Diamine Solution]4935645	24 Motor Fuel Antiknock Compound	4921445 Heat	Probable complexation	Ι
	7 Hexamethylene Diamine Solution	4935645	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	1
	7 Hexamethylene Diamine Solution		34 Propylene Oxide		Potentially vigorous or violent ring opening and polymerization	-
	7 Hexamethylene Diamine Solution	4935645	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Hydrogen evolution possible.	
	7 Hexamethylene Diamine Solution	[4935645]	18 Toluene Diisocyanate	4921575 Heat/Polymerization	Violent polymerization.	1
 					/ Potentially vigorous/violent polymerization. Acetylene could be generated.	

	TM Chemical Name	STC / Number	•	STC Number Compatibility	· ·	REF H
	29 Hexane	4908183	184 Chlorine	4904120 Heat/Fire	Chilorine reacts vigorously with most hydrocarbons causing ignition of some.	1
	29 Hexane	4908183	104 Hydrogen Peroxide Solution	4918335 Compatible	No reaction anticipated.	
	1 Hydrochloric Acid	4930228	5 Aceta I dehyde	4907210 Heat/Polymerization	Vigorous polymerization in acids and aldol condensation forms 2-butenal and 3-hydroxybutanal.	2
	1 Hydrochloric Acid	4930228	3 Acetic Acid, Glacial	4931303 Heat	Heat generated from hydration of glacial acetic acid.	5
	1 Hydrochloric Acid	493Ø228	107 Acetic Anhydride	49313Ø4 Heat/Fire	Mixing acetic anhydride with water can be explosive in aqueous mineral acids. Forms acetic acid.	 1
	1 Hydrochloric Acid	493Ø228	19 Acetone	4908105 Heat	Acid catalyzed aldol condensation generates some heat and diacetone alcohol (Slow reaction).	8
	1 Hydrochloric Acid	493Ø228	3 Acrylic Acid	4931405 Heat/Polymerization	Heat could initiate slow polymerization but primarily a hydration reaction.	- -
	1 Hydrochloric Acid	4930228	26 Acrylonitrile	4906420 Heat/Polymerization	Potentially a vigorous reaction with strong acids - hydrolysis plus polymerization.	5
	1 Hydrochloric Acid	493Ø228	4 Alcoholic Beverage	4910102 Heat/Flammable Gas/Toxic Gas	Very slow displacement reaction generates ethyl chloride (b.p.=12.5 C), water and some heat.	8
	1 Hydrochloric Acid	493Ø228	10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Violent neutralization reaction.	5
	1 Hydrochloric Acid	493Ø228	106 Ammonium Nitrate Solution	4918774 Compatible	May form NOC1 if heated.	
,	1 Hydrochloric Acid	493Ø228	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Violent neutralization generates NH4Cl and heat that volatilizes excess NH3.	

AST GRP	•	STC A Number G	•	STC Number Compatibility	
	1 Hydrochloric Acid	493Ø228	7[Aniline Oil, Liquid	4921410 Heat/Neutralization	Vigorous acid/base 12 neutralization reaction generates heat and aniline hydrochloride.
	1 Hydrochloric acid	493Ø228	28 Butad i ene	4905704 Heat/Polymerization	Moderate acid catalyzed cationic polymerization mitigated by low solubility of butadiene in water.
 I	1 Hydrochloric acid	493Ø228 -	28 Butadiene, Inhibited	4905703 Heat/Polymerization	Moderate acid catalyzed cationic polymerization mitigated by low solubility of butadiene in water.
 	1 Hydrochloric acid	4930228	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215 Heat/Polymerization	Potentially vigorous [polymerization and hydrolysis reaction forms acrylic acid and butanol.
- - -	1 Hydrochloric Acid	4930228	4 Butyl Alcohol	4909129 Heat	Slow displacement reaction 8 generates butyl chloride, water, and some heat.
 !	1 Hydrochloric Acid	493Ø228	5 Butyraldehyde	4908119 Heat/Polymerization	Minimal heat generated but 12 potential polymerization through carbonyl group or aldol condensation.
 	1 Hydrochloric Acid	4930228	20 Carbon Disulfide	4908125 Compatible	No reaction expected because CS2 is insoluble in aqueous solutions.
-	1 Hydrochioric Acid	4930228	17 Carbon Tetrachloride	4940320 Compatible	No reaction anticipated.
	1 Hydrochloric Acid	4930228	184 Chlorine	49 0 412 0 Heat	Liquid chlorine will be volatilized by warmer HCl solution to give chlorine gas and transpired HCl.
I	1 Hydrochloric Acid	493Ø228	17 Chloroprene, Inhibited	4907223 Heat/Polymerization/ Fire	Potentially vigorous/violent polymerization mitigated by low solubility of chloroprene in water.

AS Gri	TM P Chemical Name	STC Number	•	STC Number Compatibility	. General Comments	•	REF #
	1 Hydrochloric Acid		101 Coal Tar Distillate	·	Minimal heat generated.		
	1 Hydrochloric Acid		101 Combustible Liquid, n.o.s.		Minimal heat generated.	Į	
	1 Hydrochloric Acid	4930228	101 Compound, Cleaning, Liquid	•	Minimal heat generated.	1	
	1 Hydrochloric Acid	4930228	101 Crude Oil Petroleum	4910165 Heat	Minimal heat generated.	 	
	1 Hydrochloric Acid	493Ø228	4 Denatured Alcohol	4909151 Heat/Flammable Gas/Toxic Gas	Slow displacement reaction generates ethyl chloride (b.p.=12.5 C), water, and some heat.	 -	8
-	1 Hydrochloric Acid	[493Ø228	4 Denatured Alcohol	4909141 Heat/Flammable Gas/Toxic Gas	Slow displacement reaction generates ethyl chloride (b.p.=12.5 C), water, and some heat.	I,	8
	1 Hydrochloric Acid	493Ø228	101 Distillate Fuel Cil	2911315 Heat	Minimal heat generated.	1	
	1 Hydrochloric Acid	493Ø228	13 Ethyl Acetate	4909160 Heat	Potentially vigorous acid catalyzed hydrolysis reaction forms acetic acid and ethanol.		·
	1 Hydrochloric Acid	493Ø228	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization	Potentially vigorous polymerization and hydrolysis reaction forms acrylic acid and ethanol.	·	
	1 Hydrochloric Acid	493Ø228	4 Ethyl Alcohol	49 0 9159 Heat/Flammable Gas/Toxic Gas	Slow displacement reaction generates ethyl chloride (b.p.=12.5 C), water, and some heat.	 	8
-	1 Hydrochloric Acid	493Ø228	34 Ethylene Oxide	4906610 Heat/Polymerization	contact with acids. Violent reaction by analogy to HCI/propylene oxide.	 1	1
	1 Hydrochloric Acid	493Ø228	101 Flammable Liquid, n.o.s.	4910185 Heat	Minimal heat generated.	1	
	1 Hydrochloric Acid	493Ø228 	5 Formaldehyde Solutio	n 4913168 Heat/Polymerization	Potentially vigorous polymerization moderated by water content.	1	

ASTI GRP	•	STC /		STC Number Compatibility	•	REF
	L Hydrochloric Acid	4930228	5 Formaldehyde Solutio	n 4913144 Heat/Polymerization	Potentially vigorous polymerization moderated by water content.	
 Í	1 Hydrochloric Acid	4930228	101 Fuel Óil	4915111 Heat	Minimal heat generated.	 I
 	1 Hydrochloric Acid	4930228	101 Fuel Oil	4915113 Heat	Minimal heat generated.	
:	1 Hydrochloric Acid	4930228	101 Fuel 0il	4915112 Heat	Minimal heat generated.	 I
I	1 Hydrochloric Acid	4930228	101 Fuel, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.	i
1	1 Hydrochloric Acid	4930228	101 Fuel, Aviation, Turbine Engine	49 0 9215 Heat	Minimal heat generated.	
!	1 Hydrochloric Acid	493Ø228	101 Gasoline	4908178 Heat	Minimal heat generated.	- -
	1 Hydrochloric Acid	4930228	101 Gasoline, nec	2911190 Heat	Minimal heat generated.	
\ \	1 Hydrochloric Acid	4930228	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat/Flammable Gas/Toxic Gas	Generates 1,2-dichloroethane (b.p.= 83.5 C), water, and some heat.	1 1
1	1 Hydrochloric Acid	4930228	7 Hexamethylene Diamin Solution	e 4935645 Heat/Neutralization	Vigorous acid/base neutralization reaction generates heat and hexamethylene diamine dihydrochloride.	1
I	1 Hydrochloric Acid	4930228	104 Hydrogen Peroxide Solution	4918335 Heat/Toxic Gas/Explosion	Vigorous reaction forms chlorine, oxygen and water.	(
l	1 Hydrochloric Acid	4930228	4 Isopropanol	4909205 Heat/Flammable Gas/Toxic Gas	Slow displacement reaction generates isopropyl chloride (b.p.=34.8 C), water and some heat.	
1	1 Hydrochloric Acid	4930228	107 Maleic Anhydride	4941161 Heat/Polymerization	Violent acid catalyzed hydrolysis of anhydride plus polymerization. Forms maleic acid.	
1	1 Hydrochloric Acid	493Ø228	4 Methyl Alcohol	4909237 Heat/Flammable Gas/Toxic Gas	Moderate displacement reaction generates methyl chloride (b.p.=-23.7 C), water and some heat.	1

AST) GRP	•	STC / Number	•	STC Number Compatibility	 General Comments	REF #
	1 Hydrochloric Acid	4930228	4 Methyl Alcohol	4909230 Heat/Flammable Gas/Toxic Gas	Moderate displacement reaction generates methyl chloride (b.p.=-23.7 C), water and some heat.	8
:	1 Hydrochloric Acid	4930228	17 Methyl Chloride	4905761 Compatible	No reaction anticipated.	Ι
1	1 Hydrochforic Acid	4930228	19 Methy! Ethy! Ketone	4909243 Heat	Acid catalyzed aldol reaction (slow) with minimal heat generation.	12
1 1	1 Hydrochloric Acid	4930228	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat/Polymerization	Potentially vigorous polymerization and hydrolysis reaction forms methacrylic acid and methanol.	
 1	1 Hydrochloric Acid	493Ø228	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas/Fire	Potential formation of ethane	1
1	L Hydrochloric Acid	4930228	2 Nitric Acid, Fuming	4918529 Heat/Toxic Gas	[Forms aqua regia. A large excess of HCL forms NOCL and chlorine gas. 3HCL + HNO3 -> NOCL + CL2 H2O	
1	1 Hydrochloric Acid	4930228	4 Octy! Alcohol (C.L., n.o.s.)	4913158 Heat	Slow displacement reaction generates octyl chloride, water and some heat.	8
:	1 Hydrochloric Acid	493Ø228	1Ø1 0i	4915245 Heat	Minimal heat generated.	
1	1 Hydrochloric Acid	4930228	2 Oleum	4930030 Heat/Toxic Gas	Violent expulsion of HCl gas as oleum dissolves in HCl solution Also forms chlorosulfonic acid.	•
				4910259 Heat	_	
	1 Hydrochloric Acid	[493Ø228]	101 Petroleum Naphtha	4915259 Heat	Minima! heat generated.	1
:	1 Hydrochloric Acid	4930228	101 Petroleum Naphtha	4915257 Heat	Minimal heat generated.	
 :	1 Hydrochloric Acid	4930228	101 Petroleum Residual Fuel Oil	2911715 Heat	Minimal heat generated.	1
_ _			101 Petroleum, Partially Refined	2911735 Heat	Minimal heat generated.	

AS GR	•	STC A Number G		STC Number Compatibility	 General Comments	•	 EF #
 	1 Hydrochloric Acid	4930228	31 Pheno I	4921220 Heat	Weak complexation.		,
1	1 Hydrochloric Acid	493Ø228	105 Phosphorus, White or Yellow	Gas/Toxic Gas	Forms phosphine gas and PCI3 at 200 C. Potential explosion by analogy to Hydroiodic acid/phosphorus.	I	6
 	1 Hydrochloric Acid	4930228			Violent eruption of acid when added to KOH.		 2
 ľ	1 Hydrochloric Acid	4930228	34 Propylene Oxide	4906620 Heat/Polymerization	Violent reaction.	1	5
 	1 Hydrochloric Acid	4930228	101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 Heat	Minimal heat generated.	 [İ
 	1 Hydrochloric Acid	4930228	10 Sodium Hydroxide, liq. or solution		Violent neutralization reaction	 .	5
 I	1 Hydrochloric Acid	4930228	10 Sodium Hydroxide, Liq. or solution		Violent neutralization reaction	.	 5
I	1 Hydrochloric Acid	4930228	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Sodium explodes on contact with aqueous hydrochloric acid, slowly if anhydrous. Hydrogen evolution.	1	2
	1 Hydrochloric Acid	493Ø228	16 Styrene Monomer, Inhibited	49 072 85 Heat/Polymerization	Violent polymerization and addition forming a-chloroethyl benzene. Mitigated by insolubility.	1	 }
<u></u>	1 Hydrochloric Acid	493Ø228	2 Sulfuric Acid	4930040 Heat/Toxic Gas	Violent solubilization of sulfuric acid in HCl solution releases 250 vol. of hydrogen chloride gas.	1	5
 	1 Hydrochloric Acid	493Ø228	2 Sulfuric Acid, spent		Violent solubilization moderate by both HCl and H2SO4 being in aqueous solutions.	d	. - -
	1 Hydrochloric Acid	493 ø 228		4921575 Heat/Polymerization Gas	/ Potentially vigorous acid catalyzed hydrolysis of isocyanate leading to polymerization.	1	9

AST GRP	•	STC / Number C	ASTM GRP Chemical Name	STC Number Compatibility	General Comments	REF
	1 Hydrochloric Acid	4930228	13 Vinyl Acetate	4907270 Heat/Polymerization	Violent polymerization and addition reaction forms acetic acid and acetaldehyde.	[[
	1 Hydrochloric Acid	493Ø228	17 Vinyl Chloride	4905792 Heat/Polymerization	Potentially vigorous polymerization mitigated by volatility of vinyl chloride.	1
	1 Hydrocyanic acid	4920125	5 Aceta I dehyde	4907210 Heat/Polymerization	Reaction can be violent. Cyanohydrin (alpha-hydroxypropionitrile) will form in liquid HCN.	1 1
	1 Hydrocyanic acid	4920125	107 Acetic Anhydride	4931304 Heat/Fire	Vigorous reaction forms acetonitrile and acetic acid.	1
	1 Hydrocyanic acid	4920125	19 Acetone	4908105 Heat	Potentially vigorous cyanohydrin	i]
	1 Hydrocyanic acid	492Ø125	26 Acrylonitrile	4906420 Heat/Polymerization	Expected vigorous polymerization of acrylonitrile catalyzed by HCN.	۰ <u></u> ۱۱
	1 Hydrocyanic Acid	4920125	4 Alcoholic Beverage	491 0 102 Heat	Displacement reaction generates propionitrile (b.p.=97.4 C), water, and some heat.] (
	1 Hydrocyanic acid	4920125	10 Alkaline Liquid, n.o.s.	4935220 Heat/Polymerization	HCN can polymerize or decompose explosively particularly in contact with alkaline materials.	·
	1 Hydrocyanic acid	4920125	106 Ammonium Nitrate Solution	4918774 Heat/Polymerization	Water in ammonium nitrate can catalyze self-polymerization of HCN.	1
	1 Hydrocyanic Acid	4920125	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Neutralization reaction.	
	1 Hydrocyanic acid	492Ø125	28 Butadiene	4905704 Heat/Polymerization	Moderate polymerization rate possible mitigated by low boiling point of butadiene (-5 C).	i
	1 Hydrocyanic acid	4920125	28 Butadiene, Inhibited	4905703 Heat/Polymerization	Moderate polymerization rate possible mitigated by low boiling point of butadiene (-5 C).	1

•	TM RP Chemical Name	STC A	-	STC Number Compatibility	General Comments	RE	. :
	1 Hydrocyanic acid	4920125	13 Butyl Acrylate (Corr. L., n.o.s.)	. 4912215 Heat/Polymerization	Potentially vigorous polymerization, substitution or addition forms acrylonitrile and butanol.	 	
	1 Hydrocyanic Acid	4920125	4 Butyl Alcohol	j49Ø9129 Heat	Displacement reaction generates pentanenitrile (b.p.=141 C), water, and some heat.	i	 8
 j	1 Hydrocyanic Acid	4920125	5 Butyraldehyde	4908119 Heat	Potentially vigorous reaction forms butyraldehyde cyanohydrin and significant quantities of heat.		 I
 	1 Hydrocyanic Acid	492Ø125	20 Carbon Disulfide	4908125 Compatible	No reaction expected because both are reducing agents.	1	
 	1 Hydrocyanic acid	4920125	17 Carbon Tetrachloride	4940320 Heat	Formation of nitriles (CCI3CN - b.p.=83 degrees C) with some heat. Probably a slow reaction.	 	 8
I	1 Hydrocyanic acid	492Ø125	104 Chlorine	4904120 Heat/Toxic Gas	Reacts to form cyanogen chloride, a lachrymator.	1	 1
]	1 Hydrocyanic acid	4920125	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potentially vigorous polymerization.	i	
 	1 Hydrocyanic Acid	4920125	4 Denatured Alcohol	4909151 Heat	Displacement reaction generates propionitrile (b.p.=97.4 C), water, and some heat.	Ι	 8
 I	1 Hydrocyanic Acid	[4920125]	4 Denatured Alcohol	4909141 Heat	Displacement reaction generates propionitrile (b.p.=97.4 C), water, and some heat.	 	8
 			13 Ethyl Acetate	4909160 Heat	Potentially vigorous substitution reaction forms acetonitrile and ethanol.		 1
 I	1 Hydrocyanic acid	492Ø125	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization	polymerization or addition reaction.	1	 \
 	1 Hydrocyanic acid			4909159 Heat	Displacement reaction generates propionitrile (b.p.=97.4 C), water, and some heat.		 8 [

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	· ·	REF
1 Hydrocyanic acid	4920125 ³ 34 Ethylene Oxide	4906610 Heat/Polymerization	Vigorous polymerization of ethylene oxide with heat liberation may occur in contact with acids.	1
1 Hydrocyanic acid	4920125 104 Ferric Chloride Solution	4932342 Heat	Anhydrous FeCl3 combines with HCN in excess solvent to form FeCl3-2HCN crystals - Mitigation.	6
1 Hydrocyanic Acid	4920125 5 Formaldehyde Solution	n 4913144 Heat	Potentially vigorous reaction moderated by presence of water. Forms formaldehyde cyanohydrin.	
1 Hydrocyanic Acid	4920125 5 Formaldehyde Solution	n 4913168 Heat	Potentially vigorous reaction moderated by presence of water. Forms formaldehyde cyanohydrin.	
1 Hydrocyanic Acid	4920125 14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat	Minimal cleavage and heat. However,hydroperoxides formed if previously exposed to air are explosive.	 I
1 Hydrocyanic acid	4920125 104 Hydrogen Peroxide Solution	4918335 Heat/Toxic Gas/Explosion	HCN is readily oxidized and may react explosively forming NOx gas.	 8)
1 Hydrocyanic Acid	4920125 4 Isopropanol	49 0 92 0 5 Heat	Displacement reaction generates 2-cyanopropane, water, and some heat.	 8
1 Hydrocyanic acid	4920125 107 Maleic Anhydride	4941161 Heat/Polymerization	Potential ring opening of anhydride or polymerization (less violent than with HF).	
1 Hydrocyanic Acid	4920125 4 Methyl Alcohol	4909230 Heat	Displacement reaction generates acetonitrile (b.p.=82 C), water, and some heat.	8 J
1 Hydrocyanic Acid	[4920125] 4 Methyl Alcohol	4909237 Heat	Displacement reaction generates acetonitrile (b.p.=82 C), water, and some heat.	 8
1 Hydrocyanic acid	4920125 17 Methyl Chloride	4905761 Heat	Formation of acetonitrile (CH3CN) with some heat.	 8

AŚT GRF	•	STC A Number G	•	STC Number Compatibility	 General Comments	Ri	EF ₩
	1 Hydrocyanic acid	4920125	19 Methyl Ethyl Ketone	49 0 9243 Heat	Potentially vigorous cyanohydrin		
	1 Hydrocyanic acid	4920125	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat/Polymerization	Potentially vigorous polymerization, addition or substitution forms methacrylonitrile and ethanol.	l	
	1 Hydrocyanic acid	492Ø125	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas/Fire	Potential formation of ethane gas.	1	
	1 Hydrocyanic Acid	4920125	2 Nitric Acid, Fuming	4918529 Heat	Reacts violently with cyanides.	1	5
	1 Hydrocyanic acid	492Ø125	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat	Displacement reaction generates nonanenitrile (C8H17CN), water, and some heat.	 	 8
	1 Hydrocyanic Acid	4920125	2 0 eum	4930030 Heat/Toxic Gas	Reacts to form CO gas and possible formation of NOx by analogy of HCN/H2O2.	1	
 I	1 Hydrocyanic Acid	4920125	31 Pheno	4921220 Compatible	No reaction expected.	1	(
	1 Hydrocyanic acid	4920125	10 Potassium Hydroxide, Liq. or solution	493523Ø Heat/Polymerization	HCN can polymerize or decompose explosively particularly in presence of alkaline materials.	1	5
 	1 Hydrocyanic acid	492Ø125	34 Propylene Oxide	4906620 Heat/Polymerization	Vigorous polymerization of propylene oxide.	l	
· 	1 Hydrocyanic acid	492 Ø125	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Polymerization	[HCN can polymerize or decompose explosively particularly in the presence of alkaline materials.		5
	1 Hydrocyanic acid	4920125	10 Sodium Hydroxide, liq. or solution		[HCN can polymerize or decompose explosively particularly in the presence of alkaline materials.		5
 	1 Hydrocyanic acid	4920125	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Reacts to form hydrogen gas and heat.	 	8
 -	1 Hydrocyanic acid	[4920125]	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potentially vigorous polymerization and addition reaction forms alpha-cyanoethyl benzene.	1	

AST	•	STC / Number (·	STC Number Compatibility	General Comments	•	ĒF ∰
	1 Hydrocyanic acid	492Ø125	185 Sulfur Dioxide	4904290 Compatible	No reaction expected because both are reducing agents.	1	I
	1 Hydrocyanic Acid	4920125	2 Sulfuric Acid	4930040 Heat/Toxic Gas	Reacts to form CO, NH3 and H2O in concentrated sulfuric acid.	\	
	1 Hydrocyanic Acid	4920125	2 Sulfuric Acid, spent	4930042 Heat/Toxic Gas	Reacts to form CO gas.	1	
	1 Hydrocyanic acid	4920125	18 Toluene Diisocyanate	4921575 Heat	Potentially vigorous reaction forming ArNHC(0)CN. Possible polymerization if water present.	 	 9
	1 Hydrocyanic acid	492Ø125	13 Vinyl Acetate	4907270 Heat/Polymerization	Potentially violent polymerization, addition or substitution forms acetonitrile and acetaldehyde.	1	 I
	1 Hydrocyanic acid	4920125	17 Vinyl Chloride	4905792 Heat/Polymerization	[Potentially vigorous polymerization reaction.	1	
	1 Hydrofluorosilicio	: 4930026	5 Aceta dehyde	4907210 Heat/Polymerization	Potentially violent polymerization and aldol condensation to 2-butenal.	1	2
	1 Hydrofluorosilicio Acid	: 4930026	3 Acetic Acid, Glacial	4931303 Heat	Heat generated from hydration o glacial acetic acid.	 f	
	1 Hydrofluorosilicio	4930026	107 Acetic Anhydride	4931304 Heat/Fire	Mixing acetic anhydride with water can be explosive in aqueous mineral acids. Forms acetyl fluoride.	l	1
	1 Hydrofluorosilicio Acid	: 4930026	19 Acetone	4908105 Heat	Slow acid catalyzed aldol condensation forms diacetone alcohol.	 	8
 !	1 Hydrofluorosilicio Acid	 - 4930026		4931405 Heat/Polymerization	Heat could initiate slow polymerization but primarily a hydration reaction.		·
 -	Acid			4906420 Heat/Polymerization	Potentially vigorous polymerization.	 !	 ,
			4 Alcoholic Beverage		Vigorous displacement reaction generates heat and ethyl fluoride (b.p.=-37.7 C, low toxicity).	ı	8

AST)	•	STC / Number C	•	STC Number Compatibility	 General Comments	REF #
:	Hydrofluorosilicic Acid	4930026	10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Forms SiO2 and NaF in excess alkaline or possibly SiF4 and HF vapor boil-off if acid in excess.	
:	1 Hydrofluorosilicic Acid	4930026	106[Ammonium Nitrate Solution	4918774 Heat/Toxic Gas	Modest reaction expected with release of gaseous HF and SiF4.	1
	1 Hydrofluorosilicic Acid	4930026	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Forms (NH4)SiF6. Heat of neutralization causes boil-off of liquid ammonia (boiling point = -33 C).	
	1 Hydrofluorosilicic Acid	4930026	7 Aniline Oil, Liquid	4921410 Heat/Neutralization	Vigorous acid/base neutralization reaction generates heat and aniline hydrofluoride.	8
 :	1 Hydrofluorosilicic Acid	4930026	28 Butadiene	4905704 Heat/Polymerization	Acid catalyzed cationic polymerization mitigated by low solubility of butadiene in water.	Ι,
	1 Hydrofluorosilicic Acid	4930026	28 Butadiene, Inhibited	4905703 Heat/Polymerization	Acid catalyzed cationic polymerization mitigated by low solubility of butadiene in water.	
	1 Hydrofluorosilicic Acid	4930026	13 Butyl Acrylate (Corr L., n.o.s.)	. 4912215 Heat/Polymerization	Potential vigorous/violent polymerization.	1
 	1 Hydrofluorosilicic Acid			49 09 129 Heat	Vigorous displacement reaction generates heat and butyl fluoride.	8
 	1 Hydrofluorosilicic Acid	4930026	5 Butyraldehyde	,		1
	Acid	4930026	20 Carbon Disulfide		No reaction expected - CS2 soluble in liquid.	
					Possible formation of CI3CF (b.p.=23.7 C).	

ASTI GRP	•	STC A	•	STC Number Compatibility	General Comments	REF #
:	Hydrofluorosilicic Acid	4930026	104 Chlorine	4904120 Compatible	No reaction - H2SiF6 is solvent for chlorine.	1
:	1 Hydrofluorosilicic Acid	4930026	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potentially violent polymerizatoin.	1
:	1 Hydrofluorosilicic Acid	4930026	101 Coal Tar Distillate	4915363 Heat	Minimal heat generated.	I
	1 Hydrofluorosilicic Acid	[4930026]	101 Combustible Liquid,	4915185 Heat	Minimal heat generated.	1
:	1 Hydrofluorosilicic Acid	4930026	101 Compound, Cleaning, Liquid	4915147 Heat	Minimal heat generated.	I
:	1 Hydrofluorosilicic Acid	4930026	101 Crude Oil Petroleum	4910165 Heat	Minimal heat generated.	Ι
 :	1 Hydrofluorosilicic Acid	4930026	4 Denatured Alcohol	4909151 Heat/Flammable Gás	Vigorous displacement reaction: generates heat and ethyl fluoride (b.p.=-37.7 C, low toxicity).	8
] :	1 Hydrofluorosilicic Acid	4930026	4 Denatured Alcohol	4909141 Heat/Flammable Gas	Vigorous displacement reaction generates heat and ethy! fluoride (b.p.=-37.7 C, low toxicity).	8
 ':	1 Hydrofluorosilicic Acid	4930026	101 Distillate Fuel Oil	2911315 Heat	Minimal heat generated.	1
 :	1 Hydrofluorosilicic Acid			4909160 Heat	Potentially vigorous/violent hydrogen bonding.	
:	1 Hydrofluorosilicic Acid	4930026	13 Ethyl Acrylate, inhibited	49 0 7215 Heat/Polymerization	Potentially vigorous/violent polymerization.	į
 :	1 Hydrofluorosilicic Acid	493ØØ26	4 Ethyl Alcohol	49 0 9159 Heat/Flammable Gas	Vigorous displacement reaction generates heat and ethyl fluoride (b.p.=-37.7 C, low toxicity).	8
]				4906610 Heat/Polymerization	Potentially violent polymerization with heat liberation.	1

AST	•	STC ASTM Number GRP Chemic	STC cal Name Number	Compatibility	 General Comments	RE	٠.
	1 Hydrofluorosilicic Acid	4930028 101 Flammable n.o.s.	= Liquid, 491Ø185 }		Minimal heat generated.	1	- [
	1 Hydrofluorosilicic Acid	4930026 5 Formalde	hyde Solution 4913144 I	Heat/Polymerization	Polymerization possible but presence of water would mitigate reaction.	 ;	-
	1 Hydrofluorosilicic Acid	4930026 5 Formaldeh	hyde Solution 4913168	Heat/Polymerization	Polymerization possible but presence of water would mitigate reaction.	 	-
	1 Hydrofluorosilicic Acid	4930026 101 Fue! Cil	4915113	Heat	Minimal heat generated.	i	
	1 Hydrofluorosilicic Acid	4930026 101 Fuel Oil	4915111	Heat	Minima! heat generated.	1	 1
	1 Hydrofluorosilicic Acid	4930026 101 Fue Cil	4915112	Heat	Minimal heat generated.		 l
	1 Hydrofluorosilicic Acid	4930026 101 Fuel, Av Turbine	•	Heat	Minimal heat generated.	1	- -
	1 Hydrofluorosilicic Acid	4930026 101 Fuel, Av Turbine !		Heat	Minimal heat generated.	1	
	1 Hydrofluorosilicic Acid	4930026 101 Gasoline	4908178	Heat	Minimal heat generated.	 	
	1 Hydrofluorosilicic Acid	4930026 101 Gasoline	, nec 2911190	Heat	Minimal heat generated.	1	
	Acid	4930026 14 Glycol E (C.L., n	.o.s.)	Heat/Flammable Gas/Toxic Gas	Potential Zeisel type ether cleavages forming 1,2-difluoroethane (b.p. = 30.7 C).		
	1 Hydrofluorosilicic Acid	4930026 7 Hexameth Solution	ylene Diamine 4935645	Heat/Neutralization	Vigorous acid/base neutralization reaction generates heat and hexamethylen diamine dihydrofluoride.	e e	8
		4930026 104 Hydrogen Solution	Peroxide 4918335		Hydrolysis of H2SiF6. Trace metals catalyze oxygen release but not sudden.	1	

AST GRP		STC / Number 0	'	STC Number Compatibility	 General Comments	:	₹
	1 Hydrofluorosilicic Acid	4930026	4 Isopropanol	4909205 Heat	[Vigorous displacement reaction generates heat and isopropyl fluoride.	1	 8
 I	1 Hydrofluorosilicic Acid	4930026	107 Maleic Anhydride	4941181 Heat/Polymerization	Potential violent/explosive rin opening of anhydride or polymerization. Forms maleoyl fluoride.	gļ	1
I	1 Hydrofluorosilicic Acid	4930026	4 Methyl Alcohol	4909230 Heat/Flammable Gas	Vigorous displacement reaction generates heat and methyl fluoride (b.p.=-78.2 C, low toxicity).	1	8
i	1 Hydrofluorosilicic Acid	4 93ØØ26	4 Methyl Alcohol	4909237 Heat/Flammable Gas	Vigorous displacement reaction generates heat and methyl fluoride (b.p.=-78.2 C, low toxicity).	1	8
1	1 Hydrofluorosilicic Acid	49 3ØØ26	17 Methyl Chloride	4905761 Heat/Flammable Gas/Toxic Gas	Potential vigorous exchange reaction to form CH3F and HCL. Mitigated by volatlility of CH3CL.	I	
i	1 Hydrofluorosilicic Acid	4930026	19 Methyl Ethyl Ketone	49 0 9243 Heat	Slow acid catalyzed aldol condensation.		8
i	1 Hydrofluorosilicic Acid	4930026	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat/Polymerization	Potentially vigorous/violent polymerization.	, I	
 	1 Hydrofluorosilicic Acid	4930026	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas/Fire	Potential formation of ethane gas.	1	
1	Acid	•		4918529 Heat/Toxic Gas	Violent reaction with SiF4 in solution liberating HF gas.		
1	1 Hydrofluorosilicic Acid	4930026	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat	Vigorous displacement reaction generates heat and octyl fluoride.	i	
 I	1 Hydrofluorosilicic Acid	4930026	101 0	4915245 Heat	Minimal heat generated.	ļ	!
 					Reacts with water forming HF an H2SO4 mist and possibly fluorosulfonic acid - SO2(OH)F.	nd	

AST GRE	•	STC A Number C	•	STC Number Compatibility		RE	. :
	1[Hydrofluorosilicic Acid	4930026	101 Petroleum Naphtha	491Ø259 Heat	Minimal heat generated.	 	- I I
	1 Hydrofluorosilicic Acid	493ØØ26	101 Petroleum Naphtha		Minimal heat generated.	 I	-
	1 Hydrofluorosilicic Acid	493ØØ28	·	4915257 Heat	Minimal heat generated.	 [-
	1 Hydrofluorosilicic Acid	4930026	101 Petroleum Residual Fuel Oil		Minimal heat generated.	 1	-
 	1 Hydrofluorosilicic Acid	4930026	101 Petroleum, Partially Refined	2911735 Heat	Minimal heat generated.	 	-
 	1 Hydrofluorosilicic Acid	4930026	31 Pheno I	4921220 Heat	Complexation reaction.	 	
-	1 Hydrofluorosilicic Acid	4930026	105 Phosphorus, White or Yellow	4916141 Heat	Possibly soluble. Minimal reaction expected with PF3 forming but slowly hydrolyzes to 2HF and POF.		 l
 I	1 Hydrofluorosilicic Acid	4930026	10 Potassium Hydroxide, . Liq. or solution	4935230 Heat/Neutralization	Eruption of acid when added to KOH forms KF and SiO2.		 2
 	1 Hydrofluorosilicic Acid	4930026	34 Propylene Oxide	4906620 Heat/Polymerization	Potentially violent polymerization with heat liberation.	1	
1	1 Hydrofluorosilicic Acid	4930026	101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 Heat	Minimal heat generated.	!	,
	Acid		10 Sodium Hydroxide, Liq. or solution		Violent neutralization reaction forms NaF and SiO2.		
	1 Hydrofluorosilicic Acid	4930026	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Violent neutralization reaction forms NaF and SiO2.	i	
	1 Hydrofluorosilicic Acid	4930026	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire		ł	
				4907265 Heat/Polymerization	Potentially violent polymerization and addition reaction forms alpha-fluoroethy benzene.	I	

STM RP Chemical Name	STC / Number (STC Number Compatibility	General Comments	RE
1 Hydrofluorosilicic Acid	4930026	2 Sulfuric Acid	4930040 Heat/Toxic Gas	Liberates HF and SiF4 gas.	 Ì
1 Hydrofluorosilicic Acid	4930026	2 Sulfuric Acid, spent	4930042 Heat/Toxic Gas	Forms SiO2 precipitate and liberates HF gas.	
1 Hydrofluorosilicic Acid	4930026	18 Toluene Diisocyanate	4921575 Heat/Polymerization	Potentially vigorous/violent reaction forming ArNHCOF. Possible polymerization if water present.	1
1 Hydrofluorosilicic Acid	4930026	13 Vinyl Acetate	4907270 Heat/Polymerization	Potential violent polymerization or addition of HF forms alpha-fluorovinyl acetate.	1
1 Hydrofluorosilicic Acid	4930026	17 Vinyl Chloride	4905792 Heat/Polymerization	Potentially violent polymerization mitigated by volatility of vinyl chloride.	i
1 Hydrogen Fluoride	4930024	5 Aceta i dehyde	4907210 Heat/Polymerization	Violent polymerization in presence of acids and aldol condensation to 2-butenal.	[
1 Hydrogen Fluoride	4930024	3 Acetic Acid, Glacial	4931303 Heat/Toxic Gas	Can react violently - hydrogen bonding to acetic acid. Possible formation of volatile CH3CDF.	
1 Hydrogen Fluoride	493ØØ24	107 Acetic Anhydride	4931304 Heat	Potentially violent reaction if water added. Hydrogen bonding. Forms acetyl fluoride.	
1 Hydrogen Fluoride	4930024	19 Acetone	49 0 81 0 5 Heat	Slow acid catalyzed aldol condensation forms 4-methyl-3-penten-2-one.	
1 Hydrogen Fluoride	4930024	3 Acrylic Acid	4931405 Heat/Polymerization	Can react violently with possible formation of acryl fluoride.	
1 Hydrogen Fluoride	4930024	26 Acrylonitrile	4906420 Heat/Polymerization	Potentially vigorous polymerization.	
1 Hydrogen Fluoride	4930024	4 Alcoholic Beverage	4910102 Heat/Gas	Violent dissolution of anhydrous HF and formation of ethyl fluoride (b.p.=~37.7 C, low toxicity).	I

AS GRI		STC A Number G	•	STC Number Compatibility	 General Comments	RE	F; ₩ ;
	1 Hydrogen Fluoride	4930024	10/Alkaline Liquid, n.o.s.	493522Ø Heat/Neutralization	Violent neutralization reaction forms fluoride salts.	1	{
	1 Hydrogen Fluoride	4930024	106 Ammonium Nitrate Solution	4918774 Heat	No reaction expected. Heat from solubilization of HF.	1	
	1 Hydrogen Fluoride	4930024	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Violent neutralization reaction forming NH4F and NH4HF2 salts, possibly as aerosols.	1	1
	1 Hydrogen Fluoride	4930024	7 Aniline Oil, Liquid	4921410 Heat/Neutralization	Vigorous acid/base neutralization reaction generates aniline hydrofluoride		8
	1 Hydrogen Fluoride	4930024	28 Butadiene	4905704 Heat/Polymerization	Acid catalyzed cationic polymerization mitigated by low boiling point of butadiene (-5 C).		
	1 Hydrogen Fluoride	4930024	28 Butadiene, Inhibited	4905703 Heat/Polymerization	Acid catalyzed cationic polymerization mitigated by low boiling point of butadiene (-5 C).	1	1
 	1 Hydrogen Fluoride	4930024	13 Butyl Acrylate (Corr. L., n.o.s.)	4912215 Heat/Polymerization	Potential violent polymerization.	 	
	1 Hydrogen Fluoride	[4930024]	4 Butyl Alcohol	49 0 9129 Heat	Vigorous displacement reaction forms butyl fluoride.	 	<u>-</u> 8
 	1 Hydrogen Fluoride	4930024	5 Butyraldehyde	4908119 Heat/Polymerization	Violent polymerization in presence of acids and aldol condensation to 2-ethyl-2-hexenal.	l	(
- -			20 Carbon Disulfide		No reaction expected - both reducing agents and immiscible at ambient temperatures.	1	·
 I	1 Hydrogen Fluoride	4930024	·	4940320 Heat/Toxic Gas	Potentially vigorous reaction forming Cl3CF (b.p.=23.7 C).		.
 I	1 Hydrogen Fluoride	4930024	104 Chlorine	4904120 Compatible	No reaction expected.	۱	
 				4907223 Heat/Polymerization	Potentially violent polymerization.	ſ	

AST GRP	•	STC / Number 0	•	STC Number Compatibility	 General Comments	•	==- REF #
	1 Hydrogen Fluoride	4930024	101 Coal Tar Distillate	4915363 Heat	Minimal heat generated.	!	·\
	1 Hydrogen Fluoride	4930024	101 Combustible Liquid, n.o.s.	4915185 Heat	Minimal heat generated.		1
	1 Hydrogen Fluoride	4930024	101 Compound, Cleaning, Liquid	4915147 Heat	Minimal heat generated.		·
	1 Hydrogen Fluoride	4930024	101 Crude Oil Petroleum	4910165 Heat	Minimal heat generated.	 I	
	1 Hydrogen Fluoride	4930024	4 Denatured Alcohol	4909141 Heat/Gas	Vigorous displacement reaction generates heat and ethyl fluoride (b.p.=-37.7 C, low toxicity).	1	8
	1 Hydrogen Fluoride	4930024	4 Denatured Alcohol	4909151 Heat/Gas	Vigorous displacement reaction generates heat and ethyl fluoride (b.p.=-37.7 C, low toxicity).	1	8
	1 Hydrogen Fluoride	4930024	101 Distillate Fuel Oil	2911315 Heat	Minimal heat generated.	<i>-</i>	
	1 Hydrogen Fluoride	4930024	13 Ethyl Acetate	4909160 Heat	Potential vigorous hydrogen bonding.	1	
	1 Hydrogen Fluoride	4930024	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization	Potential vigorous/violent polymerization~*	1	
	1 Hydrogen Fluoride	4930024	4 Ethy! Alcohol	4909159 Heat/Gas	Vigorous displacement reaction generates heat and ethyl fluoride (b.p.=-37.7 C, low toxicity).	1	8
					Potentially violent polymerization with heat liberation in contact with HF.	 	1
 	1 Hydrogen Fluoride	4930024	101 Flammable Liquid, n.o.s.	•	Minimal heat generated.	 ا	
	1 Hydrogen Fluoride	4930024	5 Formaldehyde Solutio	on 4913168 Heat/Polymerization	Potentially vigorous polymerization mitigated by water content.	 	 I
 I				on 4913144 Heat/Polymerization	Potentially vigorous polymerization mitigated by water content.	 	

AST GRP	•	STC A Number G	•	STC Number Compatibility	General Comments	REF #
[1 Hydrogen Fluoride	4930024	101 Fuel Cil	4915111 Heat	Minimal heat generated.	1 1
	1 Hydrogen Fluoride	4930024	101 Fuel Oil	4915112 Heat	Minimal heat generated.	
	1 Hydrogen Fluoride	4930024	101 Fuel Oil	4 915113 Heat	Minimal heat generated.	
	1 Hydrogen Fluoride	4930024	101 Fuel, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.	
	1 Hydrogen Fluoride	4930024	101 Fuel, Aviation, Turbine Engine	4909215 Heat	Minimal heat generated.	
	1 Hydrogen Fluoride	[4930024]	1Ø1 Gasoline	4908178 Heat	(Minimal heat generated.	 [
	1 Hydrogen Fluoride	4930024	101 Gasoline, nec	2911190 Heat	Minimal heat generated.	l ,
	1 Hydrogen Fluoride	4930024	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat/Toxic Gas	Potential Zeisel type ether cleavages forming 1,2-difluoroethane (b.p. = 30.7 C).	
	1 Hydrogen Fluoride	4930024	7 Hexamethylene Diamin Solution	ne 4935645 Heat/Neutralization	Vigorous acid/base neutralization reaction generates hexamethylene diamine dihydrofluoride.	8
	1 Hydrogen Fluoride	493ØØ24	104 Hydrogen Peroxide Solution	4918335 Heat/Toxic Gas	Vigorous reaction forms fluorine.	[6
	1 Hydrogen Fluoride	4930024	4 Isopropanol	4909205 Heat	Vigorous displacement reaction generates heat and isopropyl fluoride.	8
	1 Hydrogen Fluoride				Potential violent/explosive ring opening of anhydride or polymerization (less violent than H2SiF6).	
	1 Hydrogen Fluoride	4930024	4 Methyl Alcohol _	4909230 Heat/Gas	Vigorous displacement reaction generates heat and methyl fluoride (b.p.=-78.2 C, low toxicity).	8
 			4 Methyl Alcohol		Vigorous displacement reaction generates heat and methyl fluoride (b.p.=-78.2 C, low toxicity).	

AST GRP	•	STC ASTM Number GRP	Chemical Name	STC Number Co	mpatibility	 	General Comments	R	#
-	1 Hydrogen Fluoride	4930024 17 We	ethyl Chioride	49 6 5761 Heat/ -	Toxic Gas	forming g	ly vigorous reaction aseous CH3F and HCL. by volatility of	1	
	1 Hydrogen Fluoride	4930024 19 Me	ethyl Ethyl Ketone	4909243 Heat		Slow acid condensat	catalyzed aldol ion.	1	8
	1 Hydrogen Fluoride		ethyl Methacrylate onomer, inhib.	4907250 Heat/	Polymerization	Potential polymeriz	- ·	1	
	1 Hydrogen Fluoride		otor Fuel Antiknock ompound	4921445 Heat/	Gas	Potential gas.	formation of ethane		
	1 Hydrogen Fluoride	4930024 2 Ni	itric Acid, Fuming	4918529 Compa	tible	Possible fluorine.	exidation of HF to		
	1 Hydrogen Fluoride		ctyl Alcohol (C.L., .o.s.)	4913158 Heat			displacement reaction octyl fluoride and	1	8
<u></u>	1 Hydrogen Fluoride	4930024 101 0	il	4915245 Heat		Minimal h	eat generated.	1	 1
	1 Hydrogen Fluoride	4930024 2 0 	leum	4930030 Heat		fluosulfo	react to form nic acid (SO2(DH)F) rolyzes to H2SO4 and F	i iF	
 !	1 Hydrogen Fluoride	4930024 101 Pe	etroleum Naphtha	4910259 Heat		Minimal h	eat generated.	· 1	- - -
 i	1 Hydrogen Fluoride	4930024 101 Pe	etroleum Naphtha	4915257 Heat		Minimal h	eat generated.	 	
	1 Hydrogen Fluoride	· ·	etroleum Naphtha	4915259 Heat		-	eat generated.	· 	
 	1 Hydrogen Fluoride	4930024 101 Pe	etroleum Residual uel Oil	2911715 Heat		Minimal h	eat generated.	1	
 	1 Hydrogen Fluoride	4930024 101 Pe Re	etroleum, Partially efined	2911735 Heat		[Minimal h	eat generated.	1	
 	1 Hydrogen Fluoride	4930024 31 Ph	heno l	4921220 Heat		Complexat	ion reaction.	i	
 I		4930024 105 Ph	hosphorus, White or ellow	4916141 Heat/	Toxic Gas	Phosphine from mixi	(PH3) and PF3 may for		

AST GRP	•	STC / Number 0	•	STC Number Compatibility	General Comments		REF
	1 Hydrogen Fluoride	4930024	10 Potassium Hydroxide, Liq. or solution	493523Ø Heat/Neutralization	Violent eruption of acid when added to KOH forms KF salt.	1	2
	1 Hydrogen Fluoride	4930024	34 Propylene Oxide	4906620 Heat/Polymerization	Potentially violent polymerization with heat liberation.	!	5
 	1 Hydrogen Fluoride	4930024	101 Pulp Mill Liquid (F.L., n.o.s.)	491 0 32 0 Heat	Potentially vigorous formation of heat.	- - -	. <u></u>
 	1 Hydrogen Fluoride	4930024	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Violent neutralization reaction forms NaF salt.	1	5
 I	1 Hydrogen Fluoride	4930024	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Neutralization	Violent neutralization reaction forms NaF salt.	1	 5
 -	1 Hydrogen Fluoride	493ØØ24	21 Sodium Metal	4916456 Heat/Gas	Aqueous solution of HF reacts explosively with sodium, slowly if anhydrous. Hydrogen generated.		 2
- 	1 Hydrogen Fluoride	4930024	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potentially violent/explosive polymerization and addition reaction forms alpha-fluoroethy benzene.	1	 I
 i	1 Hydrogen Fluoride	493ØØ24	2 Sulfuric Acid	4930040 Compatible	No reaction expected with anhydrous HF. Release of HF vapor if aqueous HF.	1	 1
 	1 Hydrogen Fluoride	493 002 4	2 Sulfuric Acid, spent	4930042 Compatible	Some dissolution of hydrogen fluoride with generation of hea with rest of HF volatilized.	l it	
l	1 Hydrogen Fluoride	4930024	18 Toluene Diisocyanate	4921575 Heat	Potentially vigorous/violent reaction forming ArNHCOF. Possible polymerization if wate present.		9
	1 Hydrogen Fluoride	4930024	13 Vinyl Acetate	4907270 Heat/Polymerization	Potential vigorous/violent polymerization and addition of HF forms alpha-fluorovinyl acetate.	 ۱	5 }
I	1 Hydrogen Fluoride	493ØØ24	17 Vinyl Chloride	4905792 Heat/Polymerization	Potentially vigorous reaction mitigated by low boiling point of vinyl chloride - flash off.		

ASTM GRP Chemical Name	STC ASTM Number[GRP	Chemical Name	STC Number Compatibility	REF General Comments #
104 Hydrogen Peroxide Solution	4918335 107 A	cetic Anhydride	49313 0 4 Heat/Explosion	Presence of an excess of acetic 1 anhydride forms diacety! peroxide which explodes readily.
104 Hydrogen Peroxide Solution	4918335 106 Ai	mmonium Nitrate olution	4918774 Explosive Mixture	Added oxidizer could make combination more sensitive to sudden decomposition of NH4NO3.
104 Hydrogen Peroxide Solution	4918335 107 M	aleic Anhydride	4941161 Heat/Explosion	Potential epoxidation of maleic anhydride causing exotherm and forming maleic peroxide.
104 Hydrogen Peroxide Solution		hosphorus, White or ellow	4916141 Heat/Fire/Explosion	Ignition of phosphorus can occur 2 at the air-solution interface leading to a violent reaction.
1 Hydroiodic Acid	493Ø226 5 A	ceta i dehyde	4907210 Heat/Polymerization	Violent polymerization and aldol 2 condensation to 2-butenal and 3-hydroxybutanal.
1 Hydroiodic Acid	493Ø226 3 A	cetic Acid, Glacial	4931303 Heat	Heat generated from hydration of glacial acetic acid.
1 Hydroiodic Acid	4930226 107 A	cetic Anhydride	49313Ø4 Heat/Fire	Mixing acetic anhydride with 1 water can be explosive in aquéous mineral acids. Forms acetyl iodide.
1 Hydroiodic Acid	4930226 19 A	cetone	49 0 81 0 5 Heat	Slow acid catalyzed aldol 8 condensation forms diacetone alcohol.
1 Hydroiodic Acid	493Ø226 3 A	crylic Acid	4931405 Heat/Polymerization	Heat could initiate slow polymerization but primarily a hydration reaction.
1 Hydroiodic Acid	493Ø226 26 A	crylonitrile	4906420 Heat/Polymerization	Potentially vigorous reaction by analogy to hydrochloric acid/acrylonitrile reactions.
1 Hydroiodic Acid	493Ø226 4 A	Icoholic Beverage	4910102 Heat	[Displacement reaction generates 6 ethyl iodide (b.p.=72.4 C) and some heat.
1 Hydroiodic Acid		Ikaline Liquid, .o.s.	4935220 Heat/Neutralization	Violent neutralization reaction forms iodide salts.

AS GRI	•	STC A Number G	•	STC Number Compatibility	 General Comments	REF
 -	1 Hydroiodic Acid	4930226	106 Ammonium Nitrate Solution	4918774 Heat	[HI may be oxidized by nitrate forming I2 which reacts with NH4 ion to form NI3 (may explode if dry).	1
	1 Hydroiodic Acid	493Ø226	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Violent neutralization reaction forms NH4I.	1
1	1 Hydroiodic Acid	493Ø226	7 Aniline Oil, Liquid	4921410 Heat/Neutralization	Violent acid/base neutralization reaction generates aniline hydroiodide.	8
1	1 Hydroiodic Acid	4930226	28 Butadiene	4905704 Heat/Polymerization	Potential violent polymerization. Reaction mitigated by low solubility of butadiene in water.	i
1	1 Hydroiodic Acid	493Ø226	28 Butadiene, Inhibited	4905703 Heat/Polymerization	Potential violent polymerization. Reaction mitigated by low solubility of butadiene in water.	
i	1 Hydroiodic Acid	493Ø226	13 Butyl Acrylate (Corr L., n.o.s.)	. 4912215 Heat/Polymerization	Potentially vigorous acid catalyzed hydrolysis and polymerization forms acrylic acid and butanol.	·
1	1 Hydroiodic Acid	4930226	4 Butyl Alcohol	4909129 Heat	Displacement reaction generates butyl iodide (b.p.=130 C) and some heat.	8
 	1 Hydroiodic Acid	493Ø226	5 Butyraldehyde	4908119 Heat/Polymerization	Violent polymerization and aldol condensation to 2-ethyl-2-hexenal and 3-hydroxy-2-ethylhexanal.	
			20 Carbon Disulfide	4908125 Compatible	No reaction expected because both are reducing agents.	1
	• •		17 Carbon Tetrachloride	4940320 Heat	May form Cl3CI (b.p. = 142 C).	
 	1 Hydroiodic Acid	493Ø226	194 Chlorine	4904120 Heat/Toxic Gas	Potentially vigorous reaction forms HCI and iodine.	6
l					Potentially vigorous/violent polymerization mitigated by low solubility of chloroprene in water.	 I

AST GRP		STC / Number 0	•	STC Number Compatibility	 General Comments	•	#
	1 Hydroiodic Acid	4930226	101 Coal Tar Distillate	4915363 Heat	Minimal heat generated.	1	1
1	1 Hydroiodic Acid	4930226	101 Combustible Liquid, n.o.s.	4915185 Heat	Minimal heat generated.]	
- 	1 Hydroiodic Acid	4930226	101 Compound, Cleaning, Liquid	4915147 Heat	Minimal heat generated.	!	
,- - - 	1 Hydroiodic Acid	4930226	101 Crude Oil Petroleum	4910165 Heat	Minimal heat generated.		
	1 Hydroiodic Acid	493Ø226	4 Denatured Alcohol	49 0 9151 Heat	Displacement reaction generates ethyl iodide (b.p.=72.4 C) and some heat.	1	8
- 	1 Hydroiodic Acid	4930226	4 Denatured Alcohol	4909141 Heat	Displacement reaction generates ethyl iodide (b.p.=72.4 C) and some heat.		8
 l	1 Hydroiodic Acid	4930226	101 Distillate Fuel Gil	2911315 Heat	Minimal heat generated.	1	
	1 Hydroiodic Acid	493Ø226	13 Ethyl Acetate	4909160 Heat	Potentially vigorous acid catalyzed hydrolysis forms acetic acid and ethanol.	1	 l
-	1 Hydroiodic Acid	4930226	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization	Potentially vigorous/violent polymerization and hydrolysis reaction forms acrylic acid and ethanol.		
 	1 Hydroìodic Acid	4930226	4 Ethyl Alcohol	4909159 Heat	Displacement reaction generates ethyl iodide (b.p.=72.4 C) and some heat.	 	8
 	1 Hydroiodic Acid		34 Ethylene Oxide	4906610 Heat/Polymerization	liberation may occur in contact with acids - potentially violent.		1
 	1 Hydroiodic Acid	493Ø226	101 Flammable Liquid, n.o.s.	4910185 Heat	Minimal heat generated.	 	
 1				on 4913144 Heat/Polymerization	Potentially vigorous polymerization mitigated by water content.	1	·

ASTI	•	STC A Number C	•	STC Number Compatibility	 General Comments	REF #
	[Hydroiodic Acid	493Ø226	5 Formaldehyde Solutio	on 4913168 Heat/Polymerization	Potentially vigorous polymerization mitigated by water content.	
	1 Hydroiodic Acid	[4930226]	101 Fuel Oil	4915111 Heat	Minimal heat generated.]
1	1 Hydroiodic Acid	4930226	101 Fuel Oil	4915112 Heat	Minimal heat generated.	1 1
	1 Hydroiodic Acid	4930226	101 Fuel Oil	4915113 Heat	Minimal heat generated.	
i	1 Hydroiodic Acid	4930226	101 Fuel, Aviation, Turbine Engine	4909215 Heat	Minimal heat generated.	
1	1 Hydroiodic Acid	4930226	101 Fuel, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.	1
1	1 Hydroiodic Acid	4930226	101 Gasoline	4908178 Heat	Minimal heat generated.	1
1	1 Hydroiodic Acid	4930226	101 Gasoline, nec	2911190 Heat	Minimal heat generated.	
1	1 Hydroiodic Acid	493Ø226	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat	Cleavage to 1,2-diiodoethane (b.p. = 200 C).	
1	1 Hydroiodic Acid	4930226	7 Hexamethylene Diami Solution	ne 4935645 Heat/Neutralization	Violent acid/base neutralization reaction generates hexamethylend diamine dihydroiodide.	
	1 Hydroiodic Acid	493Ø226	104 Hydrogen Peroxide Solution	4918335 Heat/Toxic Gas	Vigorous reaction forms iodine, possibly as a vapor.	6
1	1 Hydroiodic Acid	493Ø226	4 Isopropanol	4909205 Heat	Displacement reaction generates isopropyl iodide and some heat.	8
 I	1 Hydroiodic Acid	493Ø226	107 Maleic Anhydride	4941161 Heat/Polymerization	Potential explosive acid catalyzed hydrolysis of anhydride plus polymerization. Forms maleic acid.	1
 	1 Hydroiodic Acid	493Ø226	4 Methyl Alcohol	4909230 Heat/Toxic Gas	Vigorous displacement reaction generates heat and methyl iodid (b.p. = 42.4 C - not flammable)	e
1	1 Hydroiodic Acid	4930226	4 Methy! Alcohol	4909237 Heat/Toxic Gas	Vigorous displacement reaction generates heat and methyl iodid (b.p. = 42.4 C - not flammable)	8 e

AS GR	•	STC / Number C		STC Number Compatibility	 General Comments	REF
 	1 Hydroiodic Acid	493Ø226	17 Methyl Chloride	4905781 Heat	Minimal displacement reaction forming CH3I moderated by immiscibility of methyl chloride/water.	I
 	1 Hydroiodic Acid	4930226	19 Methy! Ethy! Ketone	49 0 9243 Heat	Slow acid catalyzed aldol condensation.	8
- -	1 Hydroiodic Acid	493Ø226	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat/Polymerization	Potentially vigorous/violent polymerization and hydrolysis forms methacrylic acid and methanol.	i
	1 Hydroiodic Acid	[4930226]	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas/Fire	Potential formation of ethane gas.	
 I	1 Hydroiodic Acid	493ØØ26	2 Nitric Acid, Fuming	4918529 Heat	Dilute solution of hydrogen iodide reduces nitric acid to '' nitrous acid and iodine.	6
 I	1 Hydroiodic Acid	4930226	4 Octyl Alcohol (C.L., n.o.s.)	[4913158 Heat	Displacement reaction generates octyl iodide and some heat.	8
 I	1 Hydroiodic Acid	4930226	1Ø1 0il	4915245 Heat	Minimal heat generated.	l
<u></u>	1 Hydroiodic Acid	[493Ø226]	2 0 eum	4930030 Heat/Toxic Gas	Forms SO2, I2 in slight excess of HI; S, I2 in greater excess of HI; H2S, I2 in large excess of HI.	6
 I	1 Hydroiodic Acid	493Ø226	101 Petroleum Naphtha	4910259 Heat	Minimal heat generated.	1
 I	1 Hydroiodic Acid	4930226	101 Petroleum Naphtha	4915257 Heat	Minimal heat generated.	1
1			·		Minimal heat generated.	
 I	1 Hydroiodic Acid	4930226	101 Petroleum Residual Fuel Oil	2911715 Heat	Minimal heat generated.	I
i	1 Hydroiodic Acid	4930226	101 Petroleum, Partially Refined	2911735 Heat	Minimal heat generated.	I
 	1 Hydroiodic Acid	4930226	31 Phenol	4921220 Heat	Minimal heat generated.	1
- -		493Ø226	105 Phosphorus, White or Yellow	4916141 Heat/Flammable	Phosphine formed from mixing ma explode. Phosphonium iodide may form - reacts violently with water.	y :

ASTI GRP	Chemical Name	STC A	· · · · · · · · · · · · · · · · · · ·	STC Number Compatibility	 General Comments	•	EF
	1 Hydroiodic Acid	4930226	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Neutralization	Eruption of acid when added to KOH forms KI salt.	1	2
	1 Hydroiodic Acid	4930226	34 Propylene Oxide	4906620 Heat/Polymerization	Potentially violent polymerization by analogy to HCL/propylene oxide reaction.	1	,
	1 Hydroiodic Acid	4930226	101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 Heat	Minimal heat generated.		
	1 Hydroiodic Acid	4930226	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Neutralization	Violent neutralization reaction forms NaI salt.		5
	1 Hydroiodic Acid	4930226	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Violent neutralization reaction forms NaI salt.	1	5
	1 Hydroiodic Acid	4930226	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Potentially explosive reaction forms hydrogen gas, sodium iodide, and heat.	 	
~	1 Hydroiodic Acid	493Ø226	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Violent polymerization and addition forms alpha-iodomethyl benzene. Mitigated by insolubility.	1	
	1 Hydroiodic Acid	4930226	2 Sulfuric Acid	4930040 Heat/Toxic Gas	Forms SO2, I2 in slight excess of HI; S, I2 in greater excess of HI; H2S, I2 in large excess of HI.	1	6
	1 Hydroiodic Acid	4930226	2 Sulfuric Acid, spent	4930042 Heat/Toxic Gas	Forms SO2, I2 in slight excess of HI; S, I2 in greater excess of HI; H2S, I2 in large excess of HI.	1	 6
- !	1 Hydroiodic Acid	493Ø226	18 Toluene Diisocyanate	4921575 Heat/Polymerization, Gas	/ Potentially vigorous/violent hydrolysis and subsequent polymerization.		9
1	1 Hydroiodic Acid	493 6 226	13 Vinyl Acetate	4907270 Heat/Polymerization	Potentially violent polymerization and hydrolysis reaction forms acetic acid and acetaldehyde.		, i
 	1 Hydroiodic Acid	493Ø226	17 Vinyl Chloride	4905792 Heat/Polymerization	Potentially vigorous/violent reaction mitigated by low boiling point of vinyl chloride	 	

AST GRF	·	STC / Number C	•	STC Number Compatibility	 General Comments	:	# #
	4 Isopropanol	4909205	5 Aceta I dehyde	4907210 Heat	Reacts vigorously forming di(isopropoxy) ethane.		 5
	4 Isopropanol	4909205	5 Butyra I dehyde	490 8119 Heat	Potential vigorous reaction forming di(isopropoxy) butane.		 I
	4 Isopropanol	49Ø92Ø5	104 Chlorine	4904120 Heat/Fire/Explosion	Forms 2-propyl hypochlorite which can violently explode on exposure to heat or sunlight.		1
 	4 Isopropanol	4909205	34 Ethylene Oxide	4906610 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.	1	1
 	4 Isopropanol	49Ø92Ø5	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Addition of alcohols to conc. H202 forms hydroperoxides that can be detonated by shock or heat.	1	2
<u></u>	4 Isopropanol	49 0 92 0 5	34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic `or basic catalysts.	 :	1
 i	4 Isopropanol	4909205	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Hydrogen evolution.	 	<u></u>
 	4 Isopropanol	4909205	18 Toluene Diisocyanate	4921575 Heat/Fire	Potentially violent urethane formation (ArNHC02C8H17).	 	
2	28 Liquefied Petroleum Gas	4905752	104 Chlorine	49 04 12 0 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	<u>-</u> -	1
:	Gas		104 Hydrogen Peroxide Solution		No reaction anticipated in absence of alkene functionality		,
י ו	28 Liquefied Petroleum Gas (Butene)	4905707	104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1	1
:	28 Liquefied Petroleum Gas (Butene)	4905707	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	May form combinations sensitive to heat or shock. Exotherm from epoxidation of alkene.	l	10
 :			104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.		

AS	TM RP Chemical Name	STC ASTM Number GRP Chemid	cal Name	STC Number	Compatibility	General	Comments	REF #
\ \	28 Liquefied Petroleum Gas (Butylene)	4905711 104 Hydrogen Solution	Peroxide	4918335 He	at/Fire/Explosion	May form combinat to heat or shock epoxidation of al	. Exotherm from	10
1	29 Liquefied Petroleum Gas (Isobutane)	4905750 104 Chlorine		4904120 He	at/Fire	Chlorine reacts was most hydrocarbons ignition of some.	causing	1
	29 Liquefied Petroleum Gas (Isobutane)	4905747 104 Chlorine		4904120 He	at/Fire	Chlorine reacts was most hydrocarbons ignition of some.	causing	1
1	29 Liquefied Petroleum Gas (Isobutane)	4905750 104 Hydrogen Solution	Peroxide	4918335 Co	mpatible	No reaction antic	•	
<u></u> ا	29 Liquefied Petroleum Gas (Isobutane)	4905747 104 Hydrogen Solution		4918335 Co	mpatible	No reaction antic	•	 1 1
	28 Liquefied Petroleum Gas (Isobutylene)	4905748 104 Chlorine		4 9 Ø4 12 Ø He	at/Fire	Chlorine reacts v most hydrocarbons ignition of some	s causing	1
 	28 Liquefied Petroleum Gas (Isobutylene)	4905748 104 Hydrogen Solution		4918335 He	at/Fire/Explosion	May form combinat to heat or shock epoxidation of a	. Exotherm from	
1	29 Liquefied Petroleum Gas (Propane)	4905781 104 Chlorine	+	49 0 412 0 He	at/Fire	Chlorine reacts most hydrocarbon: ignition of some	s causing	[1
1	29 Liquefied Petroleum Gas (Propane)	4905781 104 Hydrogen Solution		4918335 Cd	mpatible	No reaction antional absence of alkeno		1
1	28 Liquefied Petroleum Gas (Propylene)			4904120 He		Chlorine reacts most hydrocarbon ignition of some	s causing	
1	28 Liquefied Petroleum Gas (Propylene)	Solution	Peroxide	4918335 He	eat/Fire/Explosion		tions sensitive . Exotherm from	10
1	4 Methyl Alcohol	4909237 5 Acetalde				Reacts vigorous 1,1-dimethoxyeth will be the most alcohols.	ane. Methanol	5 ⁻

AST GRP	•	STC A Number G	· ·	STC Number Compatibility	
	4 Methyl Alcohol	4909230	5 Aceta dehyde	4907210 Heat -	Reacts vigorously forming 5 1,1-dimethoxyethane. Methanol will be the most reactive of all alcohols.
	4 Methyl Alcohol	4909230	5 Butyraldehyde	4908119 Heat	Methanol will be the most reactive of all alcohols with butyraldehyde. Forms 1,1-dimethoxybutane.
	4 Methyl Alcohol	4909237	5 Butyraldehyde	4908119 Heat	Methanol will be the most reactive of all alcohols with butyraldehyde. Forms 1,1-dimethoxybutane.
	4 Methyl Alcohol	4909230	184 Chlorine	4904120 Heat/Fire/Explosion	Forms methyl hypochlorite which 1 can violently explode on exposure to heat or sunlight.
	4 Methyl Alcohol	49Ø9237 -	104 Chlorine	4904120 Heat/Fire/Explosion	Forms methyl hypochlorite which 1 can violently explode on exposure to heat or sunlight.
	4 Methyl Alcohol	4909237	34 Ethylene Oxide	4906610 Heat/Polymerization	Slow reaction at room 1 temperature in absence of acidic or basic catalysts.
	4 Methyl Alcohol	4909230	34 Ethylene Oxide	4906610 Heat/Polymerization	Słow reaction at room 1 temperature in absence of acidic or basic catalysts.
	4 Methyl Alcohol		104 Ferric Chloride Solution	4932342 Heat	Forms formaldehyde after mixing { when exposed to light.
	4 Methyl Alcohol	4909230	104 Ferric Chloride Solution	4932342 Heat	Forms formaldehyde after mixing 6
 	4 Methyl Alcohol	4909237	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	concentrated H2O2 forms explosives that can be detonated by shock or heat.
			•	4918335 Heat/Fire/Explosion	Addition of alcohols to : concentrated H2O2 forms explosives that can be detonated by shock or heat.

AST GRF	•	STC ASTM Number GRP Chemical Name	STC Number Compatibility	General Comments	REF ∦
	4 Methyl Alcohol	[4909237] 34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.	
	4 Methyl Alcohol	[4909230] 34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.	
	4 Methyl Alcohol	4909237 21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Violent hydrogen evolution.	1
!	4 Methyl Alcohol	4909230 21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Violent hydrogen evolution.	1
 !	4 Methyl Alcohol	4909237 18 Toluene Diisocyanate	4921575 Heat/Fire	[Potentially violent urethane formation. Methanol is most reactive alcohol with the diisocyanate.	
	4 Methyl Alcohol	4909230 18 Toluene Diisocyanate	4921575 Heat/Fire	Potentially violent urethane formation. Methanol is most reactive alcohol with the diisocyanate.	1
 : :	17 Methyl Chloride	4905761 107 Acetic Anhydride	49313 04 Compatible	No reaction expected.	1
 :	17 Methyl Chloride	4905761 20 Carbon Disulfide	4908125 Compatible	Minimal reaction.	
:	17 Methyl Chloride	4905761 104 Chlorine	49 04120 Heat/ F,ire	Potential fire.	1
:	17 Methyl Chloride	4905761 104 Ferric Chloride Solution	4932342 Heat	Potentially vigorous hydrolysis reaction mitigated by water content and methyl chloride volatility.	ì
:	17 Methyl Chloride	4905761 104 Hydrogen Peroxide Solution	4918335 Heat/Explosion	Alkyl chlorides are very slowly attacked by H2O2 (under basic conditions). Potential explosion.	I '
	17 Methyl Chloride	4905761 105 Phosphorus, White or Yellow		No reaction expected.	1
1	17 Methy! Chloride	4905761 21 Sodium Metal	4918458 Heat	Sodium reacts explosively with methyl chloride.	

AS GR	TM P { Chemica! Name	STC / Number	•	STC Number Compatibility		. :
	19 Methyl Ethyl Ketone	[49Ø9243]	20 Carbon Disulfide	4908125 Compatible	No reaction expected.	·-I
 -	19 Methyl Ethyl Ketone	49Ø9243	104 Chiorine	4904120 Heat/Fire	Potentially vigorous/violent oxidation reaction with possible ignition.	. <u>-</u>
	19 Methyl Ethyl Ketone	4909243	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Slow polymerization possible.	1
 	19 Methyl Ethyl Ketone	4909243	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Forms polymeric as well as dimeric and trimeric peroxides.	1
	19 Methyl Ethyl Ketone	[4909243]	105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.	
	19 Methyl Ethyl Ketone	4909243	21 Sodium Metal	4916456 Heat	Potentially violent reaction forms pinacolate dianion.	
	13 Methyl Methacrylate Monomer, inhib.	4907250	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	. 1
 I	13 Methyl Methacrylate Monomer, inhib.	4907250	104 Chlorine	49 0 412 0 Heat/Fire	Chlorine can react explosively with all hydrocarbons, especially those which are unsaturated.	
!	13 Methyl Methacrylate Monomer, inhib.	4907250	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Possible polymerization.	
	13 Methyl Methacrylate Monomer, inhib.	4907250	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Epoxidation (or polymerization) can lead to exotherm which could lead to detonation of H2O2.	
 	Monomer, inhib.		Yellow	4916141 Compatible		 1
 	13 Methyl Methacrylate Monomer, inhib.	4907250	21 Sodium Metal	4916456 Heat/Polymerization/ Fire	/ Violent polymerization and acyloin condensation forms 4-hydroxy-2,5-dimethyl-1,5-hexad ien-3-one.	ľ
 	24 Motor Fuel Antiknoc Compound	k 4921445	106 Ammonium Nitrate Solution		Ammonium nitrate solutions containing significant water are essentially not flammable.	

AS GR	•	al	Name	STC A Number G	•	STC Number Compatibility	General Comments	:	==
	24 Motor Fue Compound	el A	ntiknock	[4921445]	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	 		
	2 Nitric Ac	 cid,	Fuming	4 918529	5 Aceta dehyde	4907210 Heat/Toxic Gas/Explosion	Potential formation of enol nitrate (CH2=CH0N02) and oxidation to acetic acid evolving NO gas.	1	1
	2 Nitric Ad	 cid,	Funning	4918529	3 Acetic Acid, Glacial	49313#3 Heat/Explosion	Acetic acid/HN03 can explode if not kept cold. Forms acetyl nitrate. May be used as a nitrating mix.	1	1)
	2 Nitric Ad	cid,	Fuming	[4918529]	107 Acetic Anhydride	4931304 Heat/Explosion	Mixtures containing more than 50 wt.% HN03 may detonate. Forms acety! nitrate or C(NO2)4.	 Ø	2
	2 Nitric Ad	cid,	Fuming	4918529	19 Acetone	4908105 Heat/Explosion	Acetone decomposes violently in HN03/H2S04 mixtures. May form enol nitrate ester.		1
	2 Nitric A	- cid,	Fuming	4918529	3 Acrylic Acid	49314Ø5 Heat	Minimal temperature/pressure rise observed when mixed. May form acryl nitrate.	1	12
	2 Nitric A	cid,	Fuming	4918529	26 Acrylonitrile	4906420 Heat/Explosion	Acrylonitrile and 70% nitric acid detonates at 70 C.	1	1
	2 Nitric A	- -	Fuming	4918529	4 Alcoholic Beverage	4910102 Heat/Explosion	Stirring concentrated HNO3 and ethanol results in a reaction that accelerates to an explosion.	1	1
 I					10 Alkaline Liquid, n.o.s.		Violent neutralization reaction	•	8
-	2 Nitric A	cid,	Fuming	4918529	106 Ammonium Nitrate Solution	4918774 Heat	Heat generated from dissolution of nitric acid in aqueous ammonium nitrate solution.	1	
1					10 Anhydrous Ammonia		Ammonia gas burns in an atmosphere of nitric acid.		
1	2 Nitric A	cid,	Fuming		7 Aniline Oil, Liquid	4921410 Heat/Fire	Aniline ignites spontaneously i the presence of red fuming HNO3 Nitramine and amineoxide formation		1

 AST GRP		mical	Name	STC /	•	STC Number Compatibility	 General Comments	REF #
	2 Nitric	Acid,	Fuming	4918529	16 Benzene	4908110 Heat/Explosion	High sensitivity to detonation by mixtures close to stoichiometric of about 84% acid.	
]	2 Nitric	Acid,	Fuming	4918529	28 Butadiene	4905704 Heat/Fire	Dienes are hypergolic in contact with concentrated nitric acid. Ignition delay of 1 ms.	
Į.	2 Nitric	Acid,	Fuming	4918529	28 Butadiene, Inhibited	4905703 Heat/Fire	Dienes are hypergolic in contact with concentrated nitric acid. Ignition delay of 1 ms.	 2
1	2 Nitric	Acid,	Fuming	4918529	29 Butane	4905706 Compatible	Forms nitromethane and nitroethane at elevated temperatures (> 400 C).	 14
 	2 Nitric	Acid,	Fuming	4918529	29 Butane	4905702 Compatible	Forms nitromethane and nitroethane at elevated temperatures (> 400 C).	 14
1	2 Nitric	Acid,	Funning	4918529	13 Butyl Acrylate (Corr L., n.o.s.)	4912215 Heat/Fire	Possible vigorous polymerization.	
I	2 Nitric	Acid,	Fuming	4918529	4 Butyl Alcohol	4909129 Heat/Explosion	May form the explosive butyl nitrate.	
1	2 Nitric	Acid,	Fuming	4918529	5 Butyraldehyde	4908119 Heat/Toxic Gas/Explosion	Excessive temperature and pressure rise observed when mixed. May form enol nitrate ester.	 12
1	2 Nitric	Acid,	Fuming	4918529	20 Carbon Disulfide	4908125 Heat	Reactive when heated between 100 and 200 C.	12
 					17 Carbon Tetrachloride		1	1 1
1					104 Chiorine	4904120 Compatible	Forms chloric acid and nitrogen peroxide when maintained in contact for one month.	6
 	2 Nitric	Acid,	Fuming	4918529	17 Chloroprene, Inhibited	4907223 Heat/Polymerization/ Fire	Dienes are hypergolic in contact with concentrated nitric acid. Possible polymerization.	, 2

ASTI		emical	Name	STC / Number C	•	STC Number Compatibility	 General Comments	[RE∓
	2 Nitri	Acid,	, Fuming	4918529	101 Coal Tar Distillate	4915363 Heat/Fire	 	
	2 Nitrio	Acid,	, Fuming	4918529	101 Combustible Liquid, n.o.s.	4915185 Heat/Fire	 	
	2 Nitri	c Acid	, Fuming	4918529	101 Compound, Cleaning, Liquid	4915147 Heat/Fire	 	l
 	2 Nitri	c Acid	, Fuming	4918529	101 Crude Oil Petroleum	491Ø185 Heat/Fire	 	
 !	2 Nitri	c Acid	, Fuming	[4918529]	29 Cyclohexane	4908132 Compatible	 	1
 I	2 Nitri	c Acid	, Fuming	4918529	4 Denatured Alcohol	4909151 Heat/Explosion	Stirring concentrated HNO3 and ethanol results in a reaction that accelerates to an explosion.	1
1	2 Nitri	c Acid	, Fuming	4918529	4 Denatured Alcohol	4909141 Heat/Explosion	Stirring concentrated HNO3 and ethanol results in a reaction that accelerates to an explosion.	1 1
 [2 Nitri	c Acid	, Fuming	4918529	27 Dinitrotoluene, Liquid	4963120 Heat	Minimal reaction forming small amounts of TNT in absence of concentrated H2SO4.	1
1	2 Nitri	c Acid	, Fuming	4918529	101 Distillate Fuel Oil	2911315 Heat/Fire	1	
l	2 Nitri	c Acid	, Fuming	4918529	13 Ethyl Acetate	4909160 Heat	Minimal temperature/pressure rise observed when mixed.	1:
 	2 Nitri	c Acid	, Fuming	4918529	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization/ Fire	Possible vigorous polymerization.	
 I			-				[Stirring concentrated HN03 and ethanol results in a reaction that accelerates to an explosion.	
	2 Nitri	c Acid	, Fuming	4918529	34 Ethylene Oxide	4906610 Heat/Polymerization/ Fire	Violent polymerization mitigate by volatility of ethylene oxide	ed e.
1							Potential oxidation to chlorine gas.	

AST GRP	•	emical	Name	STC	•	STC Number Compatibility		REF
	2 Nitri	c Acid	, Fuming	4918529	101 Flammable Liquid, n.o.s.	4910185 Heat/Fire	I	<u> </u>
 	2 Nitri	c Acid	, Funning	4918529	5 Formaldehyde Solutio	n 4913168 Heat/Toxic Gas	Oxidation to formic acid evolves NO gas. Reacts even when cold.	6
	2 Nitri	c Acid	, Fuming	4918529	5 Formaldehyde Solutio	n 4913144 Heat/Toxic Gas	Oxidation to formic acid evolves NO gas. Reacts even when cold.	6
 	2 Nitri	c Acid	, Fuming	4918529	101 Fuel Oil	4915111 Heat/Fire	!	 ! !
	2 Nitri	c Acid	, Fuming	4918529	101 Fuel Dil	4915113 Heat/Fire		
 	2 Nitri	c Acid	, Fuming	4918529	101 Fuel Oil	4915112 Heat/Fire	1	
 	2 Nitri	c Acid	, Fuming	4918529	101 Fuel, Aviation, Turbine Engine	4915187 Heat/Fire	1	1
 I	2 Nitri	c Acid	, Fuming	4918529	101]Fuel, Aviation, Turbine Engine	4909215 Heat/Fire	1	
 I	2 Nitri	c Acid	, Fuming	4918529	101 Gasoline	4908178 Heat/Fire	1	 I
 I	2 Nitri	c Acid	, Fuming	4918529	101 Gasoline, nec	2911190 Heat/Fire	1	
 	2 Nitri	c Acid	, Fuming	4918529	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat/Explosion	Mixtures of HNO3 with ethylene glycol are easily detonated by heat, impact or friction.	2
 	2 Nitri	c Acid	, Fuming	4918529	7 Hexamethylene Diamin Solution	e 4935645 Heat/Fire/Explosion	Formation of hexamethylene dinitramine, an explosive with a low ignition temperature.	I
 	2 Nitri	c Acid	, Fuming	4918529	29 Hexane	4908183 Compatible		
1	2 Nitri	c Acid	, Fuming	4918529	104 Hydrogen Peroxide Solution	4918335 Heat/Explosion	[Mix is unstable when > 50% HN03 is present due to supercritical amounts of peracids and peroxides.	6
 	2 Nitri	c Acid	, Fuming	4918529	4 Isopropanol	4909205 Heat/Explosion	May form the explosive isopropyl	-
 	2 Nitri	c Acid	, Fuming	4918529	28 Liquefied Petroleum [,] Gas	4905752 Heat/Fire	1	 }

ASTI GRP	•	nical	Name	STC / Number 0	· ·	STC Number Compatibility	General Comments	REF
:	2 Nitric	Acid,	Funing	4918529	28 Liquefied Petroleum Gas (Butene)	4905707 Heat/Polymerization	Formation of 2-nitrobutane and possible polymerization.	
	2 Nitric	Acid	, Fuming	4918529	28 Liquefied Petroleum Gas (Butylene)	4905711 Heat/Polymerization	Formation of 2-nitrobutane and possible polymerization.	1 1
;	2 Nitric	Acid	, Fuming	4918529	29 Liquefied Petro-leum Gas (Isobutane)	4905747 Compatible	Reacts rapidly in vapor phase () 400 C) forming a complex mix of nitro derivatives.	14]
: :	 2 Nitric	Acid	, Fuming	4918529	29 Liquefied Petroleum Gas (Isobutane)	[4905750 Compatible	[Reacts rapidly in vapor phase () 400 C) forming a complex mix of nitro derivatives.	14
:	 2 Nitric	Acid	, Fuming	4918529	28 Liquefied Petroleum Gas (Isobutylene)	49Ø5748 Heat/Polymerization	Nitration reaction forms a small amount of C4H7NO2 (nitroisobutylene). Possible polymerization.	14
	 2 Nitric	Acid	, Fuming	4918529	29 Liquefied Petroleum Gas (Propane)	4905781 Compatible	Reacts rapidly in vapor phase () 400 C) forming a complex mix of nitro derivatives.) 14
 	2 Nitric	Acid	, Fuming	4918529 	28 Liquefied Petroleum Gas (Propylene)	4905782 Heat/Polymerization	Polymerization and nitration.	1
 	2 Nitric	Acid	, Fuming	4918529	107 Waleic Anhydride	4941161 Heat/Explosion	Probable formation of maley dinitrate.	1
	2 Nitric	Acid	, Fuming	4918529	4 Methyl Alcohol	4909230 Heat/Explosion	Readily forms the explosive methyl nitrate.	2
 	 2 Nitric	Acid	, Fuming	[4918529 	4 Methy! Alcohol	4909237 Heat/Explosion	Readily forms the explosive methyl nitrate.	2
 1	2 Nitric	: Acid	, Fuming	4918529	17 Methyl Chloride	4905761 Heat/Fire/Toxic Gas		
` 					19 Methyl Ethyl Ketone	4909243 Heat	Minimal temperature/pressure rise observed when mixed. May form enol nitrate ester.	12
	2 Nitrio	: Acid	, Fuming	4918529	Monomer, inhib.	4907250 Heat/Polymerization Fire	polymerization.	
 [2 Nitrio	. Acid	 , Fuming	4918529	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas	Formation of ethane gas.	

AST GRP	•	emical	Name	STC	•	STC Number Compatibility	 General Comments	:	==
	 2 Nitri	c Acid	, Fuming	4918529	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat/Explosion	May form the explosive octyl nitrate	1	!
	2 Nitri	c Acid	, Fuming	4918529	101 0 ;	4915245 Heat/Fire	 	<u>-</u> -	
	2 Nitri	c Acid	, Fuming	4918529	2 01eum	4930030 Heat	Forms addition compounds such as 2H2NO3-HS2O7.	 ;	6
 	2 Nitri	c Acid	, Fuming	4918529	101 Petroleum Naphtha	4915257 Heat/Fire	I	1	
	2 Nitri	c Acid	, Funing	4918529	101 Petroleum Naphtha	4915259 Heat/Fire		1	 -
	2 Nitri	c Acid	, Fuming	4918529	101 Petroleum Naphtha	4910259 Heat/Fire	 		
	2 Nitri	c Acid	, Fuming	4918529	101 Petroleum Residual Fuel Qil	2911715 Heat/Fire	 	1	
	 2 Nitri	c Acid	, Fuming	4918529	101 Petroleum, Partially Refined	2911735 Heat/Fire	, <u>.</u>	 	
	 2 Nitri	c Acid	, Fuming	4918529	31 Pheno i	4921220 Heat	Forms mono-nitrophenol accompanied by considerable oxidation.	 	14
	 2 Nitri	c Acid	, Fuming	4918529	105 Phosphorus, White or Yellow	4916141 Heat/Fire	Phosphorus ignites in a vapor of nitric acid and burns with an intense light forming NOx and H3PO4.	f]	1
 !	 2 Nitri	c Acid	, Fuming	4918529	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Neutralization	Violent neutralization reaction	. [8
 	 2 Nitri	c Acid	, Fuming	4918529	34 Propylene Oxide	4906620 Heat/Polymerization/ Fire	Temperature and pressure increased when mixed.	1	1
					101 Pulp Mill Liquid (F.L., n.o.s.)			 	
 !		c Acid	, Fuming	4918529	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Excessive temperature and pressure increases observed when mixed.	n	12
 	2 Nitri					493524Ø Heat/Neutralization		1	12

ASTI GRP	•	Name	STC A Number G	STM RP Chemical Name	STC Number Compatibility	General Comments	REF #
	2 Nitric Acid,	Fuming	4918529	21 Sodium Meta	4916456 Heat/Fire	Sodium ignites spontaneously in contact with nitric acid having a specific gravity > 1.056.	
	2 Nitric Acid,	, Fuming	4918529	16 Styrene Monomer, Inhibited	4907265 Heat/Fire	High temperature and pressure increases when mixed. Violent polymerization and ring nitration.	9
:	2 Nitric Acid,	, Fuming	4918529	105 Sulfur Dioxide	4904290 Heat/Toxic Gas	Initially slow followed by explosive reaction forms nitrosylsulfuric acid or NO and SO3 gas.	6
:	2 Nitric Acid	, Funning	4918529	16 Toluene	4909305 Heat/Fire	Oxidation and some nitration may cause uncontrolled reaction causing a "fume-off" - not explosive.	13
:	2 Nitric Acid	, Fuming	4918529	18 Toluene Diisocyanate	4921575 Heat/Toxic Gas	Violent acid-catalyzed hydrolysis (if water present) with evolution of CO2, and NOx.	1
 	2 Nitric Acid	, Fuming	4918529	13 Vinyl Acetate	4907270 Heat/Polymerization/ Fire	Temperature and pressure increased when mixed. Ester hydrolysis and probable violent polymerization.	1
 	2 Nitric Acid	, Fuming	4918529	17 Vinyl Chloride	4905792 Heat/Polymerization/ Fire	Possible polymerization or ignition of vinyl chloride.	1
 	2 Nitric Acid	, Fuming	4918529	16 Xylene	4909350 Heat/Fire	Violent nitration reaction.	1
 	2 Nitric Acid	, Fuming	4918529	16 Xylene	4909351 Heat/Fire	Violent nitration reaction.	
- 	4 Octyl Alcoho n.o.s.)	ol (C.L.	, 4913158	5 Aceta I dehyde	4907210 Heat	Probably reacts only moderately fast as a result of steric hindrance. Forms 1,1-di(octoxy) ethane.	I
 I	4 Octyl Alcoh	ol (C.L.	, 4913158	5 Butyraldehyde	4908119 Heat	Probably reacts only moderately fast as a result of steric hindrance. Forms 1,1-di(octoxy) butane.	I

ASTI GRP	•	STC /	•	STC Number Compatibility	General Comments	REF #
	4 Octyl Alcohol n.o.s.)	(C.L., 4913158	104 Chiorine	4904120 Heat/Fire/Explosion	Forms 1-octyl hypochlorite which can violently explode on exposure to heat or sunlight.	1
	4 Octyl Alcohol n.o.s.)	(C.L., 4913158	34 Ethylene Oxide	4906610 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.	1
	4 Octyl Alcohol n.o.s.)	(C.L., 4913158	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Addition of alcohols to concentrated H202 forms explosives that can be detonated by shock or heat.	2
	4 Octyl Alcohol n.o.s.)	(C.L., 4913158	34 Propylene Oxide	4906620 Heat/Polymerization	Slow reaction at room temperature in absence of acidic or basic catalysts.	 1 1
	4 Octyl Alcohol n.o.s.)	(C.L., 4913158	21 Sodium Metal	[4916456 Heat/Flammable Gas/Fire	Exothermic reaction forms hydrogen and alkoxide.	4
}	4 Octyl Alcohol n.o.s.)	(C.L., [4913158]	18 Toluene Diisocyanate]4921575 Heat	Potentially violent urethane formation (ArNHCO2C8H17).	
10	 1 0i	4915245 <i> </i>	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	
10	1 0i	4915245	104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	[1
10	 1 0il		104 Hydrogen Peroxide Solution		May form combinations sensitive to heat or shock.	
 I	2 0 eum		5 Aceta I dehyde	4907210 Heat/Polymerization/ Fire	(oxidation) of acetaldehyde in presence of concentrated acid.	
 I	2 0 eum	4930030	3 Acetic Acid, Glacial		Violent chemical reaction with fire because H2SO4/acetic acid is reported to ignite.	5
1	2 0 eum	4930030	107 Acetic Anhydride	4931304 Heat/Fire	[Violent chemical reaction. Explosion if any water is present (i.e. rain).	5

ASTM GRP Chemical Name	STC A	•	STC Number Compatibility	General Comments	REF #
2¦Oleum	4930030	19 Acetone	4908105 Heat/Fire	Potentially violent/explosive reaction.	2
2 0leum	4930030	3 Acrylic Acid	4931405 Heat/Polymerization/ Fire	Potentially violent polymerization.	5
2 Oleum	4930030	26 Acrylonitrile	4906420 Heat/Polymerization/ Fire/Toxic Gas	Violent reaction may form NOx and SO2 gas. Acetonitrile/oleum has been shown to readily explode.	5
2 0 eum	4930030	4 Alcoholic Beverage	4910102 Heat/Flammable Gas/Fire	Vigorous reaction forms ethylene, diethyl ether, acetaldehyde. Reaction moderated by water content.	
2 0leum	4930030	10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Violent neutralization reaction	. 5
2 0 eum	493ØØ3Ø _.	106 Ammonium Nitrate Solution	4918774 Heat/Explosion	Forms HNO3 and heat. Solid (NH4)2SO4-2NH4NO3 is known to explode.	
2 0 eum	4930030	10 Anhydrous Ammonia	49 0 4210 Heat/Fire	Violent neutralization reaction forms amine sulfates.	1
2 0 eum	4930030	7 Aniline Oil, Liquid	4921410 Heat/Fire/Toxic Gas	Ignites or explodes on contact. May form NOx gas from exhaustive oxidation.	
2 0 eum	4930030	16 Benzene	4908110 Heat/Fire	Sulfonation of benzene may occu instantaneously and violently forming benzenesulfonic acid.	 r
2 01eum	493ØØ3Ø	28 Butadiene	Fire	/ Cyclopentadiene (contains a 1,3 diene) explodes in H2SO4 - polymerization/oxidation.	-1
2 0 eum	4930030	28 Butadiene, Inhibited	Fire	/ Cyclopentadiene (contains a 1,3 diene) explodes in sulfuric aci - polymerization/oxidation.	
2 Cleum		29 Butane		Winimal heat generated.	
2 0 eum				Minimal heat generated.	 !

AS GR	TM P Chemical Name	STC / Number 0	-	STC Number Compatibility	 General Comments	:	EF #
	2 0 eum	4 93ØØ3Ø	13 Butyl Acrylate (Corr L., n.o.s.)	4912215 Heat/Polymerization, Fire	/ Potentially violent/explosive because oleum/ethyl acetate is a violent reaction.		1
	2 01eum	4930030	4 Butyl Alcohol	4909129 Heat/Flammable Gas/Fire	Potentially violent/explosive reaction forms butene, dibutyl ether, and butyraldehyde.	1	
	2 0 eum	4930030	5 Butyra I dehyde	4908119 Heat/Polymerization/ Fire	/ Violent chemical reaction.	 I	 5
 	2 0leum	4930030	17 Carbon Tetrachloride	4940320 Heat/Toxic Gas	At 150 C, forms HC and phosgene. Probably moderate or slow reaction at ambient temperature.	l	 6
	2 0 eum	4930030	17 Chloroprene, Inhibited	4907223 Heat/Polymerization, Toxic Gas/Fire	/ Potential explosive polymerization and oxidation forms SO2 gas.	I	 I
	2 0 eum	4930030	101 Coal Tar Distillate	4915363 Heat	Winimal heat generated.		
	2 0 eum	4930030	101 Combustible Liquid, n.o.s.	4915185 Heat	Minimal heat generated.	<i>-</i>	
	2 0 eum	4930030	101 Compound, Cleaning, Liquid	4915147 Heat	Minimal heat generated.	ľ	
	2 0 eum	[4930030]	101 Crude Oil Petroleum	4910165 Heat	Minimal heat generated.	<i>-</i>	,
	2 0 eum	4930030	29 Cyclohexane	4908132 Heat	Minimal heat generated.	 	
	2 0 eum	493ØØ3Ø	4 Denatured Alcohol	4909151 Heat/Flammable Gas/Fire	Potentially violent/explosive reaction forms ethylene, diethy ether, and acetaldehyde.	1	
	2 Cleum		4 Denatured Alcohol	Gas/Fire	Potentially violent/explosive reaction forms ethylene, diethy ether, and acetaldehyde.		[
- -	2 Dieum		27 Dinitrotoluene, Liquid	4963120 Heat/Fire/Explosion	Sulfonation of p-Chloronitrotoluene with oleum led to an explosion.	ı	2
 I	2 0 eum	149300301	101 Distillate Fuel Cil	2911315 Heat	Minimal heat generated.	 I	

AST GRP	•	STC [A Number C	ASTM GRP Chemical Name	STC Number Compatibility	General Comments	RE	
 	2 0 eum	4930030	13 Ethy! Acetate	4909160 Heat/Fire	Violent chemical reaction.	1	- (5
 }	2 0 eum	4930030	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization/ Fire	/ Potentially violent/explosive because oleum/ethyl acetate is a violent reaction.	i	-
	2 0 eum	4930030	4 Ethyl Alcohol	4909159 Heat/Flammable Gas/Fire	Potentially violent/explosive reaction forms ethylene, diethylether, and acetaldehyde.	1	-
 	2 Cleuma	[4930030]	34 Ethylene Oxide	4906610 Heat/Polymerization, Fire	/{Polymerization with heat liberation may occur in contact with acids. Explodes on contact.		1
 	2 0 eum	[4930030]	101 Flammable Liquid, n.o.s.	4910185 Heat	Minimal heat generated.	1	1
 	2 0 eum	4930030	5 Formaldehyde Solution	n 4913144 Heat/Polymerization	Potentially vigorous/violent reaction moderated by water.	1	
 I	2 0 eum	4930030	5 Formaldehyde Solution	4913168 Heat/Polymerization	Potentially vigorous/violent reaction moderated by water.	\	. <u> </u>
 	2 0leum	4930030	101 Fuel Oil	4915112 Heat	Minimal heat generated.	1	. <u> </u>
1	2 0 eum	4930030	101 Fuel Oil	4915111 Heat	Minimal heat generated.	ı	. <u>-</u>
 (2 Oleum	4930030	101 Fuel Cil	4915113 Heat	Minimal heat generated.	ŀ	1
1	2 0leum	4930030	101 Fuel, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.		 I
1	2]01eum	4930030	1 0 1 Fuel, Aviation, Turbine Engine	49 0 9215 Heat	Minimal heat generated.	1	- -
 	2 0 eum		101 Gasoline	4908178 Heat	Minimal heat generated.		
 	2 0 eum		101 Gasoline, nec	2911190 Heat	Minimal heat generated.	1	
1	2 Oleum	4930030	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat/Fire	Heat could detonate hydroperoxides formed from prio air oxidation, otherwise possible fire.	 1 r	 1
 	2 0leum	4930030	7 Hexamethylene Diamin Solution	e 4935645 Heat/Neutra I i zation	Violent acid/base neutralization reaction. May form NOx gas from exhaustive oxidation.		

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	General Comments	REF
2 01eum	4930030 29 Hexane	4968183 Heat	Minimal heat generated.	
2 0 eum	4930030 104 Hydrogen Peroxide Solution	e 4918335 Heat/Explosion	Heat of dissolution of oleum in hydrogen peroxide solution.	1
2 0 ieum	4930030 4 Isopropano	4909205 Heat/Flammable . Gas/Fire	Potentially violent/explosive reaction forms propene, diisopropyl ether, and acetone.	 5
2 0 euna	4930030 28 Liquefied Petrole Gas	eum 4905752 Heat	Minimal heat generated.	
2 Oleum	4930030 28 Liquefied Petrole Gas (Butene)	eum 4905707 Heat/Fire/Gas	Moderate heat and CO2 generated but mitigated by volatility of butene.	
2 0 leum	4930030 28 Liquefied Petrole Gas (Butylene)	eum 4905711 Heat/Fire/Gas	Moderate heat and CD2 generated but mitigated by volatility of butylene.	•
2 0 eum .	4930030 29 Liquefied Petrole Gas (Isobutane)	eum 4905750 Heat	Minimal heat generated.	1
2 0 eum	4930030 29 Liquefied Petrole Gas (Isobutane)	eum 4905747 Heat	Minimal heat generated.	1
2 0leum	4930030 28 Liquefied Petrole Gas (Isobutylene)		Moderate heat and CO2 generated but mitigated by volatility of isobutylene.	
2 Dieum	4930030 29 Liquefied Petrole Gas (Propane)	eum 4905781 Heat	Minimal heat generated.	
2 Oleum	4930030 28 Liquefied Petrole Gas (Propylene)	eum 4905782 Heat/Fire/Gas	Moderate heat and CO2 generated but mitigated by volatility of propylene.	
2 0leum	4930030 107 Maleic Anhydride	4941161 Heat/Polymerization	Potential violent/explosive polymerization and sulfonation forms sulfonatomaleic acid.	
2 Dleum	4930030 4 Methyl Alcohol	4909230 Heat/Flammable Gas/Fire/Toxic Gas	Potentially violent/explosive reaction forms dimethyl ether, and formaldehyde.	1

AST GRP	•	STC A Number C	•	STC Number Compatibility	General Comments	RE	
	2 0 eum	4930030	4 Methyi Alcohol	4909237 Heat/Flammable Gas/Fire/Toxic Gas -	Potentially violent/explosive reaction forms dimethyl ether, and formaldehyde.	1	·-I
	2 0 eum	4930030	17 Methyl Chloride	4905761 Heat/Fire/Toxic Gas	Potential oxidation forms SO2 and HCl gas.	1	8
	2 0 eum	4930030	19 Methyl Ethyl Ketone	4909243 Heat/Fire	Potentially violent/explosive reaction.	1	5
!	2 Oleum	4930030	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat/Polymerization/ Fire	Potentially violent/explosive because oleum/ethyl acetate is a violent reaction.	1	 -
	2 Oleum	4930030	24 Motor Fuel Antiknock Compound	[4921445 Heat/Flammable Gas	Potential formation of ethane.		
 I	2 0 eum	4930030	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat/Fire	Potentially violent/explosive reaction forms octene, dioctyl ether, and octanal.		
 I	2 Oleum	4930030	101 0il	4915245 Heat	Minimal heat generated.		
 	2 Oleum	4930030	101 Petroleum Naphtha	4915257 Heat	Minimal heat generated.	1	
 -	2 01eum	4930030	101 Petroleum Naphtha	491Ø259 Heat	Winimal heat generated.	\	
 	2 0 eum	4930030	101 Petroleum Naphtha	4915259 Heat	Minimal heat generated.	1	
 i	2 0 eum	4930030	101 Petroleum Residual Fuel Oil	2911715 Heat	Minimal heat generated.	1	
 	2 0 eum	4930030	101 Petroleum, Partially Refined	2911735 Heat	Minimal heat generated.	 	
 I	2 0 eum	4930030	31 Pheno I		Sulfonation of phenol should be extremely rapid and violent.	1	
 	2 0 eum	4930030	105 Phosphorus, White or Yellow		Yellow phosphorus ignites when placed in boiling sulfuric acid		1
 [2jOleum	4930030	Liq. or solution	4935230 Heat/Neutralization	Extremely violent eruption of acid when added to KOH.	1	2
 	2 0 eum	493ØØ3Ø	34 Propylene Oxide	4906620 Heat/Polymerization Fire	/ Violent/explosive chemical reaction - since sulfuric acid spent/propylene oxide explodes.	 	5

ASTM GRP	•	l Name	STC A	STM RP Chemic	al Name	STC Number	Compatibility	General Comments	REF #
101	Petroleum	Naphtha	4910259	106 Ammonium Solution	Nitrate	491877 <u>4</u> Coi	mpatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	ı
101	. Petroleum	Naphtha	4915259	104 Chlorine		49 0 412 0 He	at/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1
101	Petroleum	Naphtha	4910259	104 Chlorine		49 0 412 0 He	at/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	
101	. Petroleum	Naphtha	4915257	104 Chlorine		49 Ø412 Ø He	at/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	 1
101	Petroleum	Naphtha	4915259	104 Hydrogen Solution	Peroxide	4918335 He	at/Fire/Explosion	May form combinations sensitive to heat or shock.	10
101	Petroleum	Naphtha	4910259	104 Hydrogen Solution	Peroxide	4918335 He	at/Fire/Explosion	May form combinations sensitive to heat or shock.	10
101	Petroleum	Naphtha	4915257	104 Hydrogen Solution	Peroxide	4918335 He	at/Fire/Explosion	May form combinations sensitive to heat or shock.	10
101	l Petroleum Fuel Oil	Residual	2911715	196 Ammonium Solution	Nitrate	4918774 Co	mpatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	
101	l Petroleum Fuel Oil	Residual	2911715	104 Chlorine		4 9Ø412Ø He	at/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1
101	Fuel Oil			104 Hydrogen Solution				Reaction expected only if material contains alkenes, aldehydes, ketones, or alcohols.	10
101	l Petroleum, Refined	, Partiall	y 2911735	106 Ammonium Solution	Nitrate	4918774 Co	mpatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 e
101				104 Chlorine		49Ø412Ø He		(Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1

	STM RP Chemical Name	STC / Number (•	STC Number Compatibility	•	Œ #
	Ø1 Petroleum, Partiall Refined	y 2911735	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Reaction expected only if material contains alkenes, aldehydes, ketones, or alcohols.	!
 }	31 Pheno I	[4921220]	108 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 1
 	31 Pheno I	4921220	104 Chlorine	4904120 Heat/Fire	Potentially violent/explosive due to exothermic chlorination of phenol.	
 I	31 PhenoI	4921220	34 Ethylene Oxide	4906610 Heat/Polymerization/ Fire	Potentially violent reaction due to phenol catalyzed ring opening polymerization.	1
l	31 Pheno í	492122 0	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Possible oxidation if trace of Fe+2 is present, potential explosion of H202 due to exotherm.	
 	31 Pheno I	4921220	105 Phosphorus, White or Yellow	4916141 Heat/Flammable Gas	l I	
I I	31 Phenol	4921220	34 Propylene Oxide	4906620 Heat/Polymerization/ Fire	Potentially violent reaction due to phenol catalyzed ring opening polymerization.	
 	1 Phosphoric Acid	4930247	5 Aceta i dehyde	4907210 Heat/Polymerization	Vigorous polymerization and aldol condensation to 2-butenal and 3-hydroxybutanal.	2
1	1 Phosphoric Acid	4930247	3 Acetic Acid, Glacial		Generation of heat.	
 I	1 Phosphoric Acid	4930247	107 Acetic Anhydride	4931304 Heat/Fire	water can be explosive in aqueous mineral acids.	1
1	1 Phosphoric Acid	[493Ø247]	19 Acetone	4 9 0 81 0 5 Heat	Slow acid catalyzed aldol condensation forms diacetone alcohol.	8
.	1 Phosphoric Acid	493Ø247	3 Acrylic Acid	4931405 Heat/Polymerization	Vigorous reaction may occur as in the hydrochloric acid/acrylic acid reaction.	

	TM P Chemical Name	STC A Number G	•	STC Number Compatibility	 General Comments	RE	
	1 Phosphoric Acid	4930247	26 Acrylonitrile	4906420 Heat/Polymerization	[Probably only a moderate reaction because phosphoric acid is a weak acid - pKa = 2.	1	-
	1 Phosphoric Acid	4930247	4 Alcoholic Beverage	4910102 Heat	Minimal heat generated.	1	-
	1 Phosphoric Acid	4930247	10/Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Violent neutralization reaction forms phosphate salts.	1	-
	1 Phosphoric Acid	493Ø247	106 Ammonium Nitrate Solution	4918774 Heat	Forms HN03, heat and (NH4)2P04 which is a fire retardant.		-
	1 Phosphoric Acid	493ø247	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Violent neutralization reaction releases heat which vaporizes any excess NH3 and forms phosphate.		-
	1 Phosphoric Acid	4930247	7 Aniline Oil, Liquid	4921410 Heat/Neutralization	Vigorous acid/base neutralization reaction generates aniline phosphate.	1	8
	1 Phosphoric acid	4930247	28 Butadiene	4905704 Heat/Polymerization	Minimal acid catalyzed cationic polymerization mitigated by low solubility of butadiene in water.	1	
	1 Phosphoric Acid	4930247	28 Butadiene, Inhibited	4905703 Heat/Polymerization	Minimal acid catalyzed cationic polymerization mitigated by low solubility of butadiene in water.	1	
	1 Phosphoric acid	493Ø247	13 Butyl Acrylate (Corr L., n.o.s.)	. 4912215 Heat	Slow hydrolysis reaction forms acrylic acid and butanol.		
~-	1 Phosphoric Acid	4930247	4 Butyl Alcohol	4909129 Heat	Minimal heat generated.		
	1 Phosphoric Acid	493 Ø247	5 Butyraldehyde	4908119 Heat/Polymerization	Vigorous polymerization and aldol condensation to 2-ethyl-2-hexenal and 3-hydroxy-2-ethylhexanal.	1	
1	1 Phosphoric Acid	4930247	20 Carbon Disulfide	4908125 Compatible	Immiscible at ambient temperature. At elevated temperatures CS2 can be oxidized/H3PO4 reduced.	 	

AST GRP	•	STC ASTM Number GRP		STC Number Compatibility	 General Comments	RE	 ∓ ∤
 	1 Phosphoric Acid	4930247 17	Carbon Tetrachloride	4940320 Compatible	No reaction anticipated.	 	·-1
	1 Phosphoric Acid	4930247 104	Chlorine	4904120 Heat -	Chlorine may dissolve in H3PO4 but reaction is not anticipated	 	-
!	1 Phosphoric Acid	4930247 17	Chloroprene, Inhibited	4907223 Heat/Polymerization	Potentially vigorous/violent polymerization mitigated by low solubility of chloroprene in water.	1	1
1	1 Phosphoric Acid	4930247 101	Coal Tar Distillate	4915363 Heat	Minimal heat generated.		. <u>-</u>
	1 Phosphoric Acid	4930247 101	Combustible Liquid, n.o.s.	4915185 Heat	Minimal heat generated.	1	:-
1	1 Phosphoric Acid	4930247 101	Compound, Cleaning, Liquid	4915147 Heat	Minimal heat generated.	1	-1
]	1 Phosphoric Acid	4930247 101	Crude Oil Petroleum	4910165 Heat	Minimal heat generated.	i	 !
 	1 Phosphoric Acid	4930247 4	Denatured Alcohol	4909141 Heat	Minimal heat generated.		
	1 Phosphoric Acid	4930247 4	Denatured Alcohol	4909151 Heat	Minimal heat generated.	 	
 	1 Phosphoric Acid	4930247 101	Distillate Fuel Oil	2911315 Heat	Minimal heat generated.		
	1 Phosphoric Acid	4930247 13	Ethyl Acetate	4909160 Heat	Slow hydrolysis reaction forms acetic acid and ethanol.	1	
I	1 Phosphoric Acid	4930247 13	Ethyl Acrylate, inhibited	4907215 Heat	Slow hydrolysis reaction forms acrylic acid and ethanol.	I	
 	1 Phosphoric Acid		Ethy Alcohol	4909159 Heat	Minimal heat generated.	i	
 I	1 Phosphoric Acid	[4930247] 34	Ethylene Oxide	[4906810[Heat/Polymerization	Polymerization with heat liberation may occur in contact with acids.	1	1
 	1 Phosphoric Acid	4930247 101	Flammable Liquid, n.o.s.		Minimal heat generated.	1	
 	1 Phosphoric Acid	493Ø247 5	Formaldehyde Solution	4913168 Heat/Polymerization	Potentially vigorous polymerization moderated by presence of water.	1	

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	 General Comments	RE	. :
1 Phosphoric Acid	4930247 5 Formaldehyde Soluti	on 4913144 Heat/Polymerization	Potentially vigorous polymerization moderated by presence of water.	!	-i
1 Phosphoric Acid	4930247 101 Fuel Oil	4915112 Heat	Minimal heat generated.		-
1 Phosphoric Acid	4930247 101 Fue 0i	4915111 Heat	Minimal heat generated.	1	-
1 Phosphoric Acid	4930247 101 Fuel Oil	4915113 Heat	Minimal heat generated.		-
1 Phosphoric Acid	4930247 101 Fuel, Aviation, Turbine Engine	49 0 9215 Heat	Minimal heat generated.	l	- -
1 Phosphoric Acid	4930247 101 Fuel, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.	l	- ا
1 Phosphoric Acid	4930247 101 Gasoline	4908178 Heat	Minimal heat generated.		-1
1 Phosphoric Acid	4930247 101 Gasoline, nec	2911190 Heat	Minimal heat generated.		. <u>-</u>
1 Phosphoric Acid	4930247 14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat	Minimal cleavage and heat. However, hydroperoxides formed from prior air oxidation are explosive.		
1 Phosphoric Acid	4930247 7 Hexamethylene Diami Solution	ine 4935645 Heat/Neutralization	Vigorous acid/base neutralization reaction generates hexamethylene diamine phosphate.	l	8
1 Phosphoric Acid	4930247 104 Hydrogen Peroxide Solution	4918335 Heat/Gas	[Heat of solution and formation of peroxyphosphoric acid. Trace metals could catalyze 02 release.	i	
1 Phosphoric Acid	4930247 4 Isopropano		Minimal heat generated	1	
	4930247 107 Maleic Anhydride	· ·	Potential vigorous acid catalyzed hydrolysis (less violent than with HCl) forms maleic acid.	 	
1 Phosphoric Acid	4930247 4 Methy! Alcohol	4909237 Heat	Minimal heat generated.	1	
	4930247 4 Methy! Alcohol		Minimal heat generated		

AST	•	STC /	ASTM GRP Chemical Name	STC Number Compatibility	 General Comments	•	EF #
	1 Phosphoric Acid	4930247	17 Methyl Chloride	4905761 Compatible	No reaction anticipated.	1	
1	1 Phosphoric Acid	4930247	19 Methy Ethy Ketone	4909243 Heat	Slow acid catalyzed aldol condensation.		8
 [1 Phosphoric Acid	4 93 0 247	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat	Slow hydrolysis reaction forms methacrylic acid and methanol.	1	
1	1 Phosphoric Acid	4930247	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas/Fire	Potential formation of ethane gas.	1	
 I	1 Phosphoric Acid	4930247	2 Nitric Acid, Fuming	[4918529 Compatible	1	1	
 (1{Phosphoric Acid	493Ø247	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat	Minimal heat generated.	1	
 l	1 Phosphoric Acid	4930247	101 Oil	4915245 Heat	Minimal heat generated.	1	
	1 Phosphoric Acid	493Ø247	101 Petroleum Naphtha	4915259 Heat	Minimal heat generated.	<u>`</u>	
 !	1 Phosphoric Acid	493Ø247	101 Petroleum Naphtha	4915257 Heat	Minimal heat generated.		
 	1 Phosphoric Acid	4930247	101 Petroleum Naphtha	4910259 Heat	Minimal heat generated.		
1	1 Phosphoric Acid	493Ø247	101 Petroleum Residual Fuel Dil	2911715 Heat	Minimal heat generated.	l	
	1 Phosphoric Acid	493Ø247	101 Petroleum, Partially Refined	2911735 Heat	Minimal heat generated.	·	
<i>-</i>	1 Phosphoric Acid	493Ø247	31 Phenoi	4921220 Heat	Minimal heat generated.	 	
	1 Phosphoric Acid	[4930247]	105 Phosphorus, White or Yellow	4916141 Compatible		١	
 !	1 Phosphoric Acid	493Ø247	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Neutralization	Violent eruption of acid when added to KOH forms phosphate salt.	1	2
<i>-</i>	1 Phosphoric Acid	493Ø247	34 Propylene Oxide	4906620 Heat/Polymerization	Potentially vigorous/violent polymerization.	 	
	1 Phosphoric Acid	493Ø247	101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 Heat	Minimal heat generated.		-,-

	TM[RP Chemical Name	STC A Number C	· ·	STC Number Compatibility	 General Comments		F\ #
 	1 Phosphoric Acid	4936247	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Violent neutralization reaction forms phosphate salt.	}	। 5
۱	1 Phosphoric Acid	4930247	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Neutralization	Violent neutralization reaction forms phosphate salt.	1	5
	1 Phosphoric Acid	4930247	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Sodium ignites in phosphoric acid. Hydrogen evolution.	1	4
 	1 Phosphoric Acid	4930247	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potentially vigorous polymerization.	1	
 	1 Phosphoric Acid	4930247	18 Toluene Diisocyanate	4921575 Heat/Polymerization/ Gas	Potentially vigorous/violent hydrolysis followed by polymerization.		9
I	1 Phosphoric Acid	493Ø247	13 Vinyl Acetate	49 07270 Heat/Polymerization	Potentially vigorous/violent hydrolysis reaction forms acetic acid and acetaldehyde.	 :	
i	1 Phosphoric Acid	4930247	17 Vinyl Chloride	4905792 Heat/Polymerization	Potentially vigorous/violent reaction mitigated by low boiling point of vinyl chloride.	 	
:	LØ5 Phosphorus, White Yellow	or 4916141	107 Acetic Anhydride	49313Ø4 Unknown	 	1	
:	LØ5 Phosphorus, White Yellow	or 4916141	106 Ammonium Nitrate Solution	4918774 Heat/Explosion	The mixture can explode by percussion.	1	1
:	LØ5 Phosphorus, White Yellow	or 4916141	107 Maleic Anhydride	4941161 Unknown		1	
 :	105 Phosphorus, White Yellow		·	4904290 Compatible	Liquid SD2 dissolves white phosphorus. If heated it will react to form phosphorus and sulfur oxides.	1	· [
i	10 Potassium Hydroxid Liq. or solution	le, 493523Ø	107 Acetic Anhydride	[49313 04 Heat	Violent hydrolysis reaction with potential explosive boiling. Forms potassium acetate.	hļ	I
1	10 Potassium Hydroxid	le, 4935230	19 Acetone	4908105 Heat	Slow aldol condensation forms 4-hydroxy-4-methyl-2-pentanone.	.1	

•	STM STC / RP Chemical Name Number	•	STC Number Compatibility	 General Comments	:	#
	10 Potassium Hydroxide, 4935230 Liq. or solution	26 Acrylonitrile	4906420 Heat/Polymerization	Violent polymerization occurs if alkaline is concentrated.	1	1
	10 Potassium Hydroxide, 4935230 Liq. or solution	106 Ammonium Nitrate Solution	4918774 Heat	Will form heat.	I	
1	10 Potassium Hydroxide, 4935230 Liq. or solution	17 Carbon Tetrachloride	4940320 Heat/Toxic Gas	Moderate heat generated from displacement reaction forming CCL30H which converts to phosgene and HCL	1	 -
İ	10 Potassium Hydroxide, 4935230 Liq. or solution	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potential vigorous polymerization mitigated by low solubility of chloroprene in water.		
<u>-</u> .	10 Potassium Hydroxide, 4935230 Liq. or solution	27 Dinitrotoluene, Liquid	4963120 Heat/Explosion	o-Nitrotoluene in the presence of NaOH exploded - DNT is expected to be more explosive:	1	2
- 	10 Potassium Hydroxide, 4935230 Liq. or solution	13 Ethyl Acetate	4909160 Heat	Potentially violent base initiated hydrolysis forming acetic acid and ethanol.	1	
- 	10 Potassium Hydroxide, 4935230 Liq. or solution	34 Ethylene Oxide	4906610 Heat/Polymerization/ Fire	Polymerization will occur in contact with bases. Propylene oxide/NaOH known to explode.	1	1
-	10 Potassium Hydroxide, 4935230 Liq. or solution	107 Maleic Anhydride	4941161 Heat	Decomposes explosively in presence of alkalis forming maleic acid potassium salt.	1	1
 	10 Potassium Hydroxide, 4935230 Liq. or solution		4905761 Heat	Very rapid displacement reaction forms methanol.	וי	
 	10 Potassium Hydroxide, 4935230 Liq. or solution	19 Methyl Ethyl Ketone]4909243 Heat	Slow aldol condensation	1	
]	10 Potassium Hydroxide, 4935230 Liq. or solution	105 Phosphorus, White or Yellow	4916141 Heat/Flammable Gas/Toxic Gas	mixture is boiled.	1	1
1	10 Potassium Hydroxide, 4935230 Liq. or solution			[Violent base catalyzed polymerization. Propylene oxide/sodium hydroxide is known to explode.		

	STM RP Chemical Name	STC / Number (ASTM GRP Chemical Name	STC Number Compatibility	 General Comments	REF ∦
	10 Potassium Hydroxide, Liq. or solution	4935230	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Hydrogen evolution.	1
	10 Potassium Hydroxide, Liq. or solution	4935230	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potential polymerization mitigated by low solubility of styrene in water.	
-	10 Potassium Hydroxide, Liq. or solution	, 493523Ø	105 Sulfur Dioxide	4904290 Heat/Neutralization	Reacts to form alkali sulfites in solution or bisulfites if excess SO2 present.	
-	16 Potassium Hydroxide, Liq. or solution	, 4935230	18 Toluene Diisocyanate	4921575 Heat/Polymerization/ Gas	Potentially vigorous exothermic hydrolysis/polymerization will occur in contact with bases.	2
_	10 Potassium Hydroxide, Liq. or solution	4935230	13 Vinyl Acetate	49 07 27 0 Heat	Potentially violent base initiated hydrolysis forming acetic acid and acetaldehyde.	1
_	10 Potassium Hydroxide, Liq. or solution	, 493523Ø	17 Vinyl Chloride	4905792 Heat/Polymerization/ Flammable Gas	Potential vigorous polymerization with possible acetylene formation. Mitigated by low solubility.	
	34 Propylene Oxide	4906620	106 Ammonium Nitrate Solution	4918774 Compatible	Minimal reaction anticipated.	1
-	34 Propylene Oxide	4906620	104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	n
- 	34 Propylene Oxide	4906620	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Potential violent polymerization FeCl3 is used for propylene oxide polymerization).	on
1	34 Propylene Oxide	49Ø662Ø	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Alpha positions are activated t attack by hydroxyl radicals generating hydroperoxides - explosive.	tol
- I	34 Propylene Oxide	4906620	105 Phosphorus, White or Yellow	4916141 Heat/Polymerization	1	
	101 Pulp Mill Liquid (F.L., n.o.s.)	4910320	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water an essentially not flammable.	 re

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	REF General Comments #
101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 104 Chlorine	4904120 Heat/Fire	Chlorine reacts vigorously with 1 most hydrocarbons causing ignition of some.
101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	May form combinations that are 10 sensitive to heat or shock.
10 Sodium Hydroxide, liq. or solution	4935243 107 Acetic Anhydride	49313Ø4 Heat	Violent hydrolysis reaction with 12 potential explosive boiling. Forms sodium acetate.
10 Sodium Hydroxide, Liq. or solution	4935240 107 Acetic Anhydride	49313Ø4 Heat	Violent hydrolysis reaction with 12 potential explosive boiling. Forms sodium acetate.
10 Sodium Hydroxide, Liq. or solution	4935240 19 Acetone	[49 0 81 0 5 Heat	Slow aldol condensation forms 4-hydroxy-4-methyl-2-pentanone.
10 Sodium Hydroxide, liq. or solution	[4935243] 19 Acetone	4908105 Heat	Slow aldol condensation forms:
10 Sodium Hydroxide, Liq. or solution	4935240 26 Acrylonitrile	- 4906420 Heat/Polymerization	Violent polymerization occurs if 1 alkaline is concentrated.
10 Sodium Hydroxide, liq. or solution	4935243 26 Acrylonitrile	4906420 Heat/Polymerization	Violent polymerization occurs if 1 alkaline is concentrated.
10 Sodium Hydroxide, Liq. or solution	4935240 106 Ammonium Nitrate Solution	4918774 Heat	Will form heat.
10 Sodium Hydroxide, liq. or solution	4935243 106 Ammonium Nitrate Solution	4918774 Heat	Will form heat.
10 Sodium Hydroxide, liq. or solution	4935243 17 Carbon Tetrachioride		Moderate heat from displacement reaction forming CCI30H which converts to phosgene and HCL.
10 Sodium Hydroxide, Liq. or solution	4935240 17 Carbon Tetrachloride	4940320 Heat/Toxic Gas	Moderate heat from displacement reaction forming CCl30H which converts to phosgene and HCL.
10 Sodium Hydroxide, liq. or solution	4935243 17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potential vigorous polymerization mitigated by low solubility of chloroprene in water.

	STM RP Chemical Name	STC A	•	STC Number Compatibility	 General Comments	REF
	10 Sodium Hydroxide, Liq. or solution	4935240	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	Potential vigorous polymerization mitigated by low solubility of chloroprene in water.	
	10 Sodium Hydroxide, liq. or solution	4935243	27 Dinitrotoluene, Liquid	4963120 Heat/Explosion	o-Nitrotoluene in the presence of NaOH exploded - DNT is expected to be more explosive.	2
- ا	10 Sodium Hydroxide, Liq. or solution	4935240	27 Dinitrotoluene, Liquid	4963120 Heat/Explosion	o-Nitrotoluene in the presence of NaDH exploded - DNT is expected to be more explosive.	2
1	10 Sodium Hydroxide, liq. or solution	4935243	13 Ethyl Acetate	4909160 Heat	Potential base initiated hydrolysis. Exotherm observed when mixture heated between 100 and 200 C.	12
1	10 Sodium Hydroxide, Liq. or solution	493524#	13 Ethyl Acetate	4909160 Heat	Potential base initiated hydrolysis. Exotherm observed when mixture heated between 100 and 200 C.	12
I	10 Sodium Hydroxide, liq. or solution	[4935243]	34 Ethylene Oxide	4906610 Heat/Polymerization/ Fire	/ Violent polymerization with head liberation will occur. Propylend oxide/NaOH known to explode.	
1	10 Sodium Hydroxide, Liq. or solution	[4935240]	34 Ethylene Oxide	4906610 Heat/Polymerization/ Fire	/[Violent polymerization with head liberation will occur. Propylend oxide/NaOH known to explode.	
!	10 Sodium Hydroxide, Liq. or solution	[4935240]	107 Maleic Anhydride	4941161 Heat	[Decomposes explosively in presence of NaOH forming maleic acid sodium salt.	1
- -	10 Sodium Hydroxide, liq. or solution	4935243	107 Maleic Anhydride	4941161 Heat	Decomposes explosively in presence of NaOH forming maleic acid sodium salt.	1
1	10 Sodium Hydroxide, Liq. or solution	4935240	17 Methyl Chloride	4905761 Heat	Very rapid displacement forming methanol.	1
1	10 Sodium Hydroxide, liq. or solution	4935243	17 Wethyl Chloride	4905761 Heat	Very rapid displacement forming methanol.	l

AST GRF	•	STC /	•	STC Number Compatibility	General Comments	•	REF)
1	Ø Sodium Hydroxide, Liq. or solution	4935240	19 Methyl Ethyl Ketone	49 0 9243 Heat	Slow aldo! condensation with minimal heat generation.	1	12
1	Ø Sodium Hydroxide, liq. or solution	4935243	19 Methy Ethy Ketone	49 0 9243 Heat	Slow aldo! condensation with minima! heat generation.	1	12
1	Ø Sodium Hydroxide, Liq. or solution	4935240	105 Phosphorus, White or Yellow	4916141 Heat/Flammable Gas/Toxic Gas	Yields phosphine gas when the mixture is boiled.	1	1
1	Ø Sodium Hydroxide, liq. or solution	4935243	105 Phosphorus, White or Yellow	4916141 Heat/Flammable Gas/Toxic Gas	Yields phosphine gas when the mixture is boiled.	 !	1
1	Ø Sodium Hydroxide, Liq. or solution	4935240	34 Propylene Oxide	4906620 Heat/Polymerization/ Fire	Violent base catalyzed polymerization. Propylene oxide/NaOH is known to explode.	 -	2
1	Ø Sodium Hydroxide, liq. or solution	4935243	34 Propylene Oxide	4906620 Heat/Polymerization/ Fire	Violent base catalyzed polymerization. Propylene oxide/NaOH is known to explode.	ı	- 2
1	Ø Sodium Hydroxide, liq. or solution	[4935243]	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potential polymerization mitigated by low solubility of styrene in water.	1	
1	Ø Sodium Hydroxide, Liq. or solution	4935240	16 Styrene Monomer, Inhibited	4907265 Heat/Polymerization	Potential polymerization mitigated by low solubility of styrene in water. ***	1	 1
1	Ø Sodium Hydroxide, Liq. or solution	[4935240]	105 Sulfur Dioxide	4904290 Heat/Neutralization	[Reacts to form alkali sulfites' in solution or bisulfites if excess SO2 present.	 I	 I
	Ø Sodium Hydroxide, liq. or solution		105 Sulfur Dioxide		Reacts to form alkali sulfites in solution or bisulfites if excess SO2 present.	 !	1
1	Ø Sodium Hydroxide, Liq. or solution		18 Toluene Diisocyanate	4921575 Heat/Polymerization/ Gas	Exothermic hydrolysis to carbamate followed by polymerization will occur in contact with bases.]	2
₁ 1	Ø Sodium Hydroxide, liq. or solution	4935243	18 Toluene Diisocyanate	4921575 Heat/Polymerization/ Gas	Exothermic hydrolysis to carbamate followed by polymerization will occur in contact with bases.	 -	2

•	STM RP Chemical Name	STC A	•	STC Number Compatibility	General Comments	REF
	10 Sodium Hydroxide, liq. or solution	4935243	13 Vinyl Acetate	4907270 Heat	Potentially violent base initiated hydrolysis forming acetic acid and acetaldehyde.	
 	10 Sodium Hydroxide, Liq. or solution	4935240	13 Vinyl Acetate	4907270 Heat	Potentially violent base initiated hydrolysis forming acetic acid and acetaldehyde.	
I	10 Sodium Hydroxide, liq. or solution	4935243	17 Vinyl Chloride	4905792 Heat/Polymerization/ Flammable Gas	Potential vigorous polymerization with possible acetylene formation mitigated by low solubility.	 y
 ا	10 Sodium Hydroxide, Liq. or solution	4935240	17 Vinyl Chloride	4905792 Heat/Polymerization/ Flammable Gas	Potential vigorous polymerization with possible acetylene formation mitigated b low solubility.	ا پ
ı	21 Sodium Metal	4916456	107 Acetic Anhydride	4931304 Heat	May be extremely reactive since acetic anhydride is more reactive than esters which reac violently.	•
- 	21 Sodium Metal	4 916456	26 Acrylonitrile	4906420 Heat/Polymerization/ Flammable Gas/Fire	Sodium could initiate violent o explosive polymerization.	r
1	21 Sodium Metal	4916456	106 Ammonium Nitrate Solution	4918774 Heat/Flammable Gas/Fire/Explosion	React through a series of reductions to form an explosive substance believed to be disodium nitrite.	1 :
- !	21 Sodium Metal	4916456	·		Sodium catalyzes polymerization of butadiene (potentially violent without solvent).	1 4
- ا	21 Sodium Metal	4916456	28 Butadiene, Inhibited		Sodium catalyzes polymerization of butadiene (potentially violent without solvent).	
- ا	21 Sodium Metal	[4916456]	104 Chlorine	4904120 Heat/Fire	Sodium and chlorine vapors read	:t
!				4963120 Heat/Explosion	DNT is prone to accept an electron in making a stable radical anion - may form explosive compounds.	l

•	TM P Chemical Name	STC /	•	STC Number Compatibility		RE	•
 	21 Sodium Metal	4916456	34 Ethylene Oxide	4906610 Heat/Polymerization	Potentially explosive by analogy to ethylene oxide/potassium.	1	·- I
1	21 Sodium Metal	4916456	104 Ferric Chloride Solution	4932342 Heat/Flammable Gas/Fire	Violent reaction occurs if a mixture of sodium and ferric chloride is shocked. H2 formation.	1	1
 !	21 Sodium Metal	4916456	184 Hydrogen Peroxide Solution	4918335 Heat/Flammable Gas/Fire/Explosion	Reacts violently. Hydrogen evolution.	}	1
 I	21 Sodium Metal	4916456	107 Maleic Anhydride	4941181 Heat/Polymerization	Decomposes explosively in presence of alkali metals. Anionic Polymerization.		1
 	21 Sodium Meta!	4916456	105 Phosphorus, White or Yellow	4916141 Heat/Fire	Reacts vigorously with sodium forming phosphide, or in air, phosphate with ignition.	!	4
 !	21 Sodium Metal	4916456	34 Propylene Oxide	4906620 Heat/Polymerization	Potentially explosive by analogy to ethylene oxide/potassium.	 }	۱
 I	21 Sodium Metal	4916456	101 Pulp Mill Liquid (F.L., n.o.s.)	4910320 Heat/Flammable Gas/Fire	[Hydrogen evolution possible (if water present).	 	
 I	21 Sodium Wetal	4916456	105 Sulfur Dioxide	4904290 Heat/Fire	SO2 reacts violently with sodium at its melting point. Heated potassium is known to burn in SO2.		
 I	16 Styrene Monomer, Inhibited	4907265	106 Ammonium Nitrate Solution	[4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	ţ	 - !
 }	16 Styrene Monomer, Inhibited			4904120 Heat/Fire	Chlorine reacts vigorously with most hydrocarbons causing ignition of some.	1	1
 I	16 Styrene Monomer, Inhibited	4907265	104 Ferric Chloride Solution		Typically anionic or radical initiators are used for styrene polymerization		
 	16 Styrene Monomer, Inhibited	4907265	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Epoxidation exotherm may initiate explosion of H202.	1	I

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	REF General Comments #
16 Styrene Monomer, Inhibited	4907285 105 Phosphorus, White or Yellow	4916141 Compatible	No reaction expected.
18 Styrene Monomer, Inhibited	4907285 21 Sodium Metal	4916458 Heat/Polymerization	Efficient anionic polymerization. Probable explosion.
105 Sulfur Dioxide	4904290 107 Acetic Anhydride	49313Ø4 Unknown	
2 Sulfuric Acid	4930040 5 Acetaldehyde	4907210 Heat/Polymerization, Fire	/ Violent polymerization of : acetaldehyde in presence of concentrated acid.
2 Sulfuric Acid	4930040 3 Acetic Acid, Glacial	4931303 Heat/Fire	Ignites or explodes on contact.
2 Sulfuric Acid	4930040 107 Acetic Anhydride	49313 <i>0</i> 4 Heat/Fire	Violent chemical reaction. Explosion if any water is present (i.e. rain).
2 Sulfuric Acid	4930040 19 Acetone	4908105 Heat/Polymerization	Potentially violent polymerization of enol plus aldol condensation followed by elimination.
2 Sulfuric Acid	[4930040] 3 Acrylic Acid	4931405 Heat/Polymerization	Potentially violent 1 polymerization but minimal heat generation reported.
2 Sulfuric Acid	4930040 26 Acrylonitrile	4906420 Heat/Polymerization Fire/Toxic Gas	/ Vigorous exothermic reaction may form NOx gas. Methacrylonitrile/ H2SO4 has been shown to explode.
2 Sulfuric Acid	[4930040] 4 Alcoholic Beverage	4910102 Heat/Flammable Gas/Fire	Potentially violent reaction forms ethylene and diethy! ether. Reaction moderated by water content.
2 Sulfuric Acid	4930040 10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Violent neutralization reaction.
2 Sulfuric Acid	Solution	4918774 Heat/Explosion	Forms HNO3 and heat. Solid (NH4)2SO4-2NH4NO3 is known to explode.
2 Sulfuric Acid	4930040 10 Anhydrous Ammonia		Violent neutralization reaction.

AST GRP		STC / Number	•	STC Number Compatibility	General Comments	REF
	2 Sulfuric Acid	4930040	7 Aniline Oil, Liquid	4921410 Heat/Fire/Toxic Gas	Ignites or explodes on contact. May form NOx gas from exhaustive oxidation.	5
	2 Sulfuric Acid	4930040	16 Benzene	4908110 Heat/Fire	Sulfonation of benzene will proceed vigorously or violently forming benzenesulfonic acid.	I
	2 Sulfuric Acid	4930040	28 Butadiene	4905704 Heat/Polymerization/ Fire	Potential violent/explosive polymerization with excessive heat generation.	12
	2 Sulfuric Acid	4930040	28 Butadiene, Inhibited	4905703 Heat/Polymerization/ Fire	Potential violent/explosive polymerization with excessive heat generation.	12
	2 Sulfuric Acid	4930040	13 Butyl Acrylate (Corr L., n.o.s.)	. 4912215 Heat/Polymerization/ Fire	/ Potentially violent/explosive reaction.	
	2 Sulfuric Acid	4930040	4 Butýl Alcohol	4909129 Heat/Flammable Gas/Fire	Potentially violent reaction forms butene, dibutyl ether and heat.	1
	2 Sulfuric Acid	4930040	5 Butyraldehyde	4908119 Heat/Polymerization/ Fire	/ Violent chemical reaction.	5
 	2 Sulfuric Acid	4930040	17 Carbon Tetrachloride	4940320 Heat/Toxic Gas	Forms HCl and phosgene at 150 C.	6
 i	2 Sulfuric Acid	4930040	17 Chloroprene, Inhibited	4907223 Heat/Polymerization/ Toxic Gas/Fire	Potentially explosive	1
 	2 Sulfuric Acid	4930040	101 Coal Tar Distillate	4915363 Heat	Minimal heat generated.	l
			n.o.s.		,	i
	2 Sulfuric Acid	4930040	101 Compound, Cleaning, Liquid	4915147 Heat		l,
	2 Sulfuric Acid	[4930040]	101 Crude Oil Petroleum	4910165 Heat	Minimal heat generated.	I
				4909151 Heat/Flammable	Potentially violent reaction forms ethylene and diethyl ether.	

ASTM GRP	Chemical Name	STC A Number G	•	STC Number Compatibility	 General Comments	REF
2 Sul	furic Acid	4930040	4 Denatured Alcohol	4909141 Heat/Flammable Gas/Fire	Potentially violent reaction forms ethylene and diethyl ether.	I (
 2 Sul	furic Acid	[4930040]	27 Dinitrotoluene, Liquid	4963120 Heat/Fire/Explosion	Since sulfuric acid/p-nitrotoluene mixtures detonate at 160 C, 2,4-DNT may explode at lower temps.	1 ·
2 Su	furic Acid	4930040	101 Distillate Fuel Oil	2911315 Heat	Minimal heat generated.	1
2 Sul	furic Acid	4936646	13 Ethyl Acetate	4909180 Heat	Minimal heat generated.	12
 2 Sul	furic Acid	[4930040]	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization	Some heat generation with potential polymerization.	9
 2 Sul	furic Acid	4930040	4 Ethyl Alcohol	4909159 Heat/Flammable Gas/Fire	Potentially violent reaction forms ethylene, diethyl ether and excessive heat.	9
 2 Sul	furic Acid	4930040	34 Ethylene Oxide	4906610 Heat/Polymerization Fire	/ Polymerization/explosion with heat liberation may occur. Analagous to sulfuric acid/propylene oxide.	ļ 1
2 Su	lfuric 'Acid	4930040	101 Flammable Liquid, n.o.s.	4910185 Heat	Minimal heat generated.	l
2 Su	lfuric Acid	4930040	5 Formaldehyde Solut	ion 4913168 Heat/Polymerization	Potentially vigorous/violent moderated by water.	i
2 Sul	lfuric Acid	4930040	5 Formaidehyde Solut	ion 4913144 Heat/Polymerization	Potentially vigorous/violent moderated by water.	1
				4915111 Heat		
] 2]Su1	lfuric Acid	4930040	101 Fuel Oil	4915113 Heat	Minimal heat generated.	
2 Sul	lfuric Acid	4930040	101 Fuel Oil	4915112 Heat	Minimal heat generated.	1
2 Su1	lfuric Acid.	4930040	101 Fuel, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.	I
			101 Fuel, Aviation, Turbine Engine	•	Minimal heat generated.	

AST GRP	· ·	STC / Number	•	STC Number Compatibility		REF
 	2 Sulfuric Acid	[4930040]	101 Gasoline	4908178 Heat	Minimal heat generated.	
 	2 Sulfuric Acid	4930040	101 Gasoline, nec	2911190 Heat	Minimal heat generated.	
I	2 Sulfuric Acid	4930040	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat/Fire	Heat could detonate hydroperoxides formed from prior air oxidation, otherwise possible fire.	
	2 Sulfuric Acid	[4930040]	7 Hexamethylene Diamine Solution	: 4935845 Heat/Neutralization	Vigorous/violent acid/base neutralization reaction. May form NOx gas from exhaustive oxidation.	
	2 Sulfuric Acid	4930040	104 Hydrogen Peroxide Solution	4918335 Heat	Heat of dissolution of concentrated sulfuric acid in hydrogen peroxide solution.	
 	2 Sulfuric Acid	4930040	4 Isopropanol	4909205 Heat/Flammable Gas/Fire	Potentially violent reaction forms propene and diisopropyl ether.	
- 	2 Sulfuric Acid	4930040	107 Maleic Anhydride	4941161 Heat/Polymerization	Potential violent/explosive_acid catalyzed hydrolysis and polymerization.	
1	2 Sulfuric Acid	4930040	4 Methyl Alcohol	4909230 Heat/Flammable Gas/Fire	Potentially violent reaction forms dimethyl ether.	
 	2 Sulfuric Acid	4930040	4 Methyi Alcohol	4909237 Heat/Flammable Gas/Fire	Potentially violent reaction forms dimethy! ether.	
 	2 Sulfuric Acid	4930040	17 Methyl Chloride	4905761 Heat/Fire/Toxic Gas	Potential oxidation forms SO2 and HCl gas.	8
 i	2 Sulfuric Acid	4930040			Potential polymerization of enol plus aldo! condensation followed by elimination.	12
 	2 Sulfuric Acid	4930040		4907250 Heat/Polymerization/ Fire	Potentially vigorous polymerization.	
1	2 Sulfuric Acid	4930040	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas	Potential formation of ethane.	

AST GRP	•	STC / Number C	•	STC Number Compatibility	General Comments	RE	
	2 Sulfuric Acid	4930040	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat/Fire	Potentially violent reaction forms octene and dioctyl ether.	1	-
	2 Sulfuric Acid	4930040	1Ø1 0;	4915245 Heat	Minimal heat generated.	1	. <u>-</u>
	2 Sulfuric Acid	4930040	101 Petroleum Naphtha	4915259 Heat	Minimal heat generated.	1	. <u>-</u>
	2 Sulfuric Acid	4930046	101 Petroleum Naphtha	4915257 Heat	Minimal heat generated.	1	۱-
	2 Sulfuric Acid	4930040	101 Petroleum Naphtha	4910259 Heat	Winimal heat generated.	1	
	2 Sulfuric Acid	4930040	101 Petroleum Residual Fuel Oil	2911715 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid	4930040	101 Petroleum, Partially Refined	2911735 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid	4930040	31 Pheno I	4921220 Heat	Sulfonation of phenol should be extremely rapid and violent.	1	
	2 Sulfuric Acid	4930040	105 Phosphorus, White or Yellow	4916141 Heat/Fire/Toxic Gas	Potential explosion by analogy to chlorosulfuric acid/phosphorus which explodes.	1	2
	2 Sulfuric Acid	4930040	10 Potassium Hydroxide, Liq. or solution	4935230 Heat/Neutralization	Violent eruption of acid when added to KOH.		2
	2 Sulfuric Acid	4930040	34 Propylene Oxide	4906620 Heat/Polymerization, Fire	/ Ignites or explodes on contact.		5
	2 Sulfuric Acid	493ØØ4Ø	101 Pulp Mill Liquid (F.L., n.o.s.)	[4910320 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid	4930040	10 Sodium Hydroxide, liq. or solution		Violent neutralization reaction	.	5
			10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Neutralization	Violent neutralization reaction		5
	2 Sulfuric Acid	4930040	21 Sodium Metal	4916456 Heat	Mild reaction if anhydrous, otherwise reacts explosively. Hydrogen generated.	İ	
					/ Ignites or explodes on contact.		

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AST	•	STC /	•	STC Number Compatibility	 General Comments	:	EF #
	2 Sulfuric Acid	4930040	16 Toluene	4909305 Heat/Fire	Toluene should be sulfonated very rapidly forming p-toluenesulfonic acid.	1	
 	2 Sulfuric Acid	4930040	18 Toluene Diisocyanate	4921575 Heat/Gas	Violent oxidation with evolution of CO2 and acid catalyzed hydrolysis if water present.	i	9
 	2 Sulfuric Acid	4930040	13 Vinyl Acetate	4907270 Heat/Polymerization/ Fire	/ Ignites or explodes on contact.	1	 5
 	2 Sulfuric Acid	[4930040]	17 Vinyl Chloride	[4905792 Heat/Polymerization/ Toxic Gas/Fire	' Potentially violent polymerization.	\ \	
 	2 Sulfuric Acid	4930040	16 Xy I ene	4909351 Heat/Fire	Sulfonation and oxidation forming xylenesulfonic acid.		
 I	2 Sulfuric Acid	4930040	16 Xy ene	4909350 Heat/Fire	Sulfonation and oxidation forming xylenesulfonic acid.	1	
 	2 Sulfuric Acid,	spent 4930042	5 Aceta I dehyde	4907210 Heat/Polymerization	Vigorous polymerization of acetaldehyde moderated by water content.		2
1	2 Sulfuric Acid,	spent 4930042	3 Acetic Acid, Glacial	4931303 Heat	Potentially vigorous reaction moderated by water content.	1	
1	2 Sulfuric Acid,	spent 4930042	107 Acetic Anhydride	49313 04 Hea t	Mixing acetic anhydride with water can be explosive especially when mineral acids are present.	1	1
!	2 Sulfuric Acid,				Potential polymerization of eno plus aldol condensation followed by elimination.	i	, ,
1		spent 4930042		4931405 Heat/Polymerization		1	
1				Fire/Toxic Gas	/ Vigorous exothermic reaction, ignites or explodes on contact and may form NOx.		
<u></u> .			4 Alcoholic Beverage		Potentially vigorous reaction forms ethylene, diethyl ether and heat. Reaction moderated by water.		

ASTM GRP		STC A me Number C	•	STC Number Compatibility	REF
2	Sulfuric Acid,	spent 4930642	10 Alkaline Liquid, n.o.s.	4935220 Heat/Neutralization	Violent neutralization reaction. E
2	Sulfuric Acid,	spent 4930042	106 Ammonium Nitrate Solution	4918774 Heat	Forms HNO3 and heat. Presence of excess water moderates heat build-up.
2	Sulfuric Acid,	spent 4930042	10 Anhydrous Ammonia	4904210 Heat/Neutralization	Vigorous neutralization reaction forming ammonium sulfate salts. Reaction moderated by water.
2	Sulfuric Acid,	spent 4930042	7 Aniline Oil, Liquid	4921410 Heat/Neutralization	Violent acid/base neutralization reaction moderated by water content.
2	Sulfuric Acid,	spent 4930042	16 Benzene	[49Ø811Ø Heat	Potentially violent sulfonation reaction forming benzenesulfonic acid. Moderated by water content.
2	Sulfuric Acid,	spent 4930042	28 Butadiene	4905704 Heat/Polymerization	Potentially vigorous polymerization mitigated by volatility of butadiene and presence of water.
2	Sulfuric Acid,	spent 4930042	28 Butadiene, Inhibited	[4905703]Heat/Polymerization	Potentially vigorous { polymerization mitigated by volatility of butadiene and presence of water.
2	Sulfuric Acid,	spent 4930042	L., n.o.s.)	. 4912215 Heat/Polymerization	Potentially vigorous/violent reaction mitigated by water content.
2			4 Butyl Alcohol	Gas/Fire	Potentially vigorous reaction forms butene, dibuty! ether, and heat. Reaction moderated by water.
2	Sulfuric Acid,	spent 4930042	5 Butyraldehyde	4908119 Heat/Polymerization	butyraldehyde moderated by water content.
2			17 Carbon Tetrachloride	4940320 Heat/Toxic Gas	[Forms HCl and phosgene at 150 C. Probably moderate or slow reaction at ambient temperatures.

TM STC / P Chemical Name Number (•	STC Number Compatibility	 General Comments	RE
2 Sulfuric Acid, spent 4930042	17 Chloroprene, Inhibited	4907223 Heat/Polymerization	polymerization reaction moderated by water content.	1
 2 Sulfuric Acid, spent 4930042	101 Coal Tar Distillate	4915363 Heat	Minimal heat generated.	1
 2 Sulfuric Acid, spent 4930042	101 Combustible Liquid, n.o.s.	4915185 Heat	Minimal heat generated.	1
 2 Sulfuric Acid, spent 4930042	101 Compound, Cleaning, Liquid	4915147 Heat	Minimal heat generated.	1
 2 Sulfuric Acid, spent 4930042	101 Crude Oil Petroleum	4910185 Heat	Minimal heat generated.	1
 2 Sulfuric Acid, spent 4930042	4 Denatured Alcohol	4909151 Heat/Flammable Gas/Fire	Potentially vigorous reaction forms ethylene, diethyl ether and heat. Reaction moderated by water.	
 2 Sulfuric Acid, spent 4930042	4 Denatured Alcohol	4909141 Heat/Flammable Gas/Fire	Potentially vigorous reaction forms ethylene, diethyl ether and heat. Reaction moderated by water.	
 2 Sulfuric Acid, spent 4930042	27 Dinitrotoluene, Liquid	4963120 Heat/Fire/Explosion	Since sulfuric acid/p-nitrotoluene mixtures detonate at 160 C, 2,4-DNT may explode at lower temps.	1
 2 Sulfuric Acid, spent 4930042	101 Distillate Fuel Oil	2911315 Heat	Minimal heat generated.	!
 2 Sulfuric Acid, spent 4930042	13 Ethyl Acetate	4909160 Heat	Potentially vigorous reaction mitigated by water content.	
 2 Sulfuric Acid, spent 4930042	13 Ethyl Acrylate, inhibited	4907215 Heat/Polymerization	Potentially vigorous/violent reaction mitigated by water content.	
 2 Sulfuric Acid, spent 4930042	4 Ethyl Alcohol	4909159 Heat/Flammable Gas/Fire	Potentially vigorous reaction forms ethylene, diethyl ether and heat. Reaction moderated by water.	
 2 Sulfuric Acid, spent 4930042	34 Ethylene Oxide	4906610 Heat/Polymerization/ Fire	Polymerization with heat liberation. Analogous to sulfuric acid/propylene oxide-explodes on contact.	1

AS	TM STC P Chemical Name Number	•	STC Number 'Compatibility	 General Comments	-	:== :EF #
	2 Sulfuric Acid, spent 4930042	101/Flammable Liquid, n.o.s.	4910185 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid, spent 4930042	5 Formaldehyde Solutio	on 4913144 Heat/Polymerization	Potentially vigorous/violent moderated by water content.	1	
	2 Sulfuric Acid, spent 4930042	5 Formaldehyde Solution	on 4913168 Heat/Polymerization	[Potentially vigorous/violent moderated by water content.	 	
	2 Sulfuric Acid, spent 4930042	101 Fuel Oil	4915111 Heat	(Minimal heat generated.	- - -	
	2 Sulfuric Acid, spent 4930042	101 Fuel Oil	4915113 Heat	Minimal heat generated.		
	2 Sulfuric Acid, spent 4930042	101 Fuel-Oil	4915112 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid, spent 4930042	101 Fue!, Aviation, Turbine Engine	4915167 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid, spent 4930042	101 Fuel, Aviation, Turbine Engine	49Ø9215 Heat	Minimal heat generated.	\ \	
	2 Sulfuric Acid, spent 4930042	101 Gasoline	4908178 Heat	Minimal heat generated.		
	2 Sulfuric Acid, spent 4930042	101 Gasoline, nec	2911190 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid, spent 4930042	14 Glycol Ethers, nec (C.L., n.o.s.)	4913194 Heat	Potential cleavage of ether group in aqueous solution containing H2SO4 forming heat.	1	·
	2 Sulfuric Acid, spent 4930042	7 Hexamethylene Diamin Solution	ne 4935645 Heat/Neutralization	Violent acid/base neutralizatio reaction moderated by water content.	on J	
	2 Sulfuric Acid, spent 4930042		4909205 Heat/Flammable Gas/Fire	Potentially vigorous reaction forms propene, disopropyl ether, and heat. Reaction moderated by water.	1	
	2 Sulfuric Acid, spent 4930042	107 Maleic Anhydride		Potential violent/explosive aci catalyzed hydrolysis and polymerization.	id	
	2 Sulfuric Acid, spent 4930042	4 Methyl Alcoho!	4909230 Heat/Flammable Gas/Fire	Potentially vigorous reaction forms dimethyl ether and heat. Reaction moderated by water.	1	

AST GRP	•	Chemi	ca!	Nam	 e	STC Number	ASTM GRP Chemical Name	STC Number Compatibility	 General Comments	RE
	2 Su	furic	Aci	id,	spent	4930042	4 Methyl Alcohol	4909237 Heat/Flammable Gas/Fire	Potentially vigorous reaction forms dimethyl ether and heat. Reaction moderated by water.	I
	2 Sul	furic	Aci	id,	spent	493ØØ42	17 Methyl Chloride	4905761 Heat/Toxic Gas	Potential oxidation forms SO2 and HCL gas.	l
,	2 Sul	furic	Aci	id,	spent	4930042	19 Methy Ethy Ketone	4909243 Heat/Polymerization	Potential polymerization of enol plus aldol condensation followed by elimination.	
	2 Sul	furic	, A ci	id,	spent	4930042	13 Methyl Methacrylate Monomer, inhib.	4907250 Heat/Polymerization	Potentially vigorous/violent polymerization reaction mitigated by water content.	1
	2 Sul	furic	Aci	id,	spent	4930042	24 Motor Fuel Antiknock Compound	4921445 Heat/Flammable Gas	Potential formation of ethane.	1
	2 Sul	furic	Ac i	id,	spent	4930042	4 Octyl Alcohol (C.L., n.o.s.)	4913158 Heat	Potentially vigorous reaction & forms octene, dioctyl ether, and heat. Reaction moderated by water.	
	2 Sui	furic	Aci	id,	spent	4930042	1Ø1 0i!	4915245 Heat	Minimal heat generated.	
	2 Su	furic	Aci	id,	spent	4930042	101 Petroleum Naphtha	4910259 Heat	[Minimal heat generated.	1
	2 Sul	furic	Aci	id,	spent	4930042	101 Petroleum Naphtha	4915257 Heat	Minimal heat generated.	l
	2 Su	furic	Aci	id,	spent	4930042	101 Petroleum Naphtha	4915259 Heat	Minimal heat generated.	1
	2 Su	furic	Ac	id,	spent	4930042	101 Petroleum Residual Fuei Oi!	2911715 Heat	Minimal heat generated.	
	·						101 Petroleum, Partially Refined	•	Minimal heat generated.	1
	2 Su	furic	: Ac	id,	spent	4930042	31 Pheno I	4921 220 Heat	Potentially vigorous sulfonation	n
	2 Sul	furic	: Ac	id,	spent	4930042	105 Phosphorus, White or Yellow	4916141 Heat/Fire	Yellow phosphorus ignites when placed in boiling sulfuric acid	ا
									Eruption of acid when added to KOH. Reaction moderated by water.	

AS		STM[RP Chemical Name	STC Number Compatibility	 General Comments	:	:== !EF #
	2 Sulfuric Acid, spent 4930042	34 Propylene Oxide	4906620 Heat/Polymerization/ Fire	Ignites or explodes on contact.	1	5
	2 Sulfuric Acid, spent 4930042	101 Pulp Mill Liquid (F.L., n.o.s.)	491 0 32 0 Heat	Minimal heat generated.	1	
	2 Sulfuric Acid, spent 4930042	10 Sodium Hydroxide, liq. or solution	4935243 Heat/Neutralization	Vigorous neutralization reaction moderated by water content.	ין	
	2 Sulfuric Acid, spent 4930042	10 Sodium Hydroxide, Liq. or solution	4935240 Heat/Neutralization	Vigorous neutralization reaction moderated by water content.	 1	
	2 Sulfuric Acid, spent 4930042	21 Sodium Metal	4916456 Heat/Flammable Gas/Fire	Aqueous sulfuric acid reacts explosively with sodium. Hydrogen generated.	I	2
	2 Sulfuric Acid, spent 4930042	16 Styrene Monomer, Inhibited	4907285 Heat/Polymerization	Reaction moderated by water content.	1	 I
	2 Sulfuric Acid, spent 4930042	16 Toluene	4909305 Heat	Potentially violent sulfonation/oxidation forming p-toluenesulfonic acid. Moderated by water.	1	
	2 Sulfuric Acid, spent 4930042	18 Toluene Diisocyanate	4921575 Heat/Gas	Violent hydrolysis with evolution of CO2 - reaction severity increased because of water content.	1	1
!	2 Sulfuric Acid,\spent 4930042	13 Vinyl Acetate	4907270 Heat/Polymerization	Potentially vigorous/violent polymerization reaction mitigated by water content.		
	2 Sulfuric Acid, spent 4930042	17 Vinyl Chloride	4905792 Heat/Polymerization	Potentially violent polymerization.	1	
	2 Sulfuric Acid, spent 4930042	16 Xylene	4909350 Heat	Potentially violent sulfonation/oxidation forming xylenesulfonic acid. Moderated by water content.	1	
 	2 Sulfuric Acid, spent 4930042	16 Xylene	4 9 0 9351 Heat	Potentially violent sulfonation/oxidation forming xylenesulfonic acid. Moderated by water content.	 I	

	-		GRP Chemical Name			REF
- -	16 Toluene	49Ø93Ø5	194 Chiorine	,	Chlorine reacts vigorously with hydrocarbons. Toluene is susceptible to free radical chlorination.	
_	16 Toluene	4909305	184 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Oxidation of methyl group could lead to exotherm and explosion or detonation.	2
_		e 4921575	106 Ammonium Nitrate Solution	4918774 Heat/Gas	Non-violent reaction with water forms CO2 gas and an organic base.	
1	18 Toluene Diisocyanate	4921575	104 Chlorine		Potentially violent oxidation.	1 1
	18 Toluene Diisocyanate	4921575	34 Ethylene Oxide	4906610 Heat/Polymerization	Isocyanate moiety may be nucleophilic enough to initiate vigorous/violent ring opening.	·
_	18 Toluene Diisocyanate	4921575	104 Ferric Chloride Solution	4932342 Compatible	No reaction self-evident.	
, - .	18 Toluene Diisocyanate	4921575	184 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	[Hydrolysis and polymerization of TDI will cause significant exotherm leading to H2O2 decomposition.	'l I
	18 Toluene Diisocyanate	4921575	105 Phosphorus, White or Yellow	4916141 Heat/Flammabie Gas		i I
_	18 Toluene Diisocyanate	4921575	34 Propylene Oxide	4906620 Heat/Polymerization	Potential polymerization.	
_	18 Toluene Diisocyanate	4921575	21 Sodium Metal	4916456 Heat/Polymerization/ Flammable Gas/Fire		I
	,		105 Sulfur Dioxide	49Ø429Ø Unknown	 	
1			106 Ammonium Nitrate Solution		Ammonium nitrate solutions containing significant water are essentially not flammable.	
- 	13 Vinyl Acetate	4907270	104 Chlorine	4904120 Heat/Fire	[Potential explosion.	
- 				4932342 Heat/Polymerization		i I
_						

AS GR	TM P Chemical Name	STC A	•	STC Number Compatibility	 General Comments	:	 EF! #
	13 Vinyl Acetate	4907270	104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Vinyl acetate in contact with peroxides may explode violently.	•	1
	13 Vinyl Acetate	4907270	105 Phosphorus, White or Yellow	[4916141 Compatible	No reaction expected.	۱	 (
	13 Vinyl Acetate	4907270	21 Sodium Metal	4916456 Heat/Polymerization/ Fire	Potentially violent acyloin condensation forms 3-hydroxy-2-butanone and acetaldehyde.	i	
	17 Vinyl Chloride	49Ø5792	106 Ammonium Nitrate Solution	4918774 Compatible	Ammonium nitrate solutions containing significant water are essentially not flammable.	 	
	17 Vinyl Chloride	4905792	20 Carbon Disulfide	[4908125]Compatible	No reaction anticipated.		
	17 Vinyl Chloride	4 9Ø5792	184 Chlorine	4904120 Heat/Fire	Potential explosion because of very exothermic chlorine addition.	1	۱
	17 Vinyl Chloride	4905792	104 Ferric Chloride Solution	4932342 Heat/Polymerization	Potentially vigorous/violent polymerization.		
	17 Vinyl Chloride	4905792	104 Hydrogen Peroxide Solution	4918335 Heat/Polymerization/ Fire/Explosion	Potential explosion due to exotherm from epoxidation and polymerization.	1	<u></u> !
 I	17 Vinyl Chloride	4905792	24 Motor Fuel Antiknock Compound	4921445 Heat/Polymerization	Potential anionic polymerization.		
 	17 Vinyl Chloride		105 Phosphorus, White or Yellow		No reaction expected.	1	
1		4905792	21 Sodium Metal	4916456 Heat	Sodium forms explosive mixtures with chlorinated hydrocarbons. Violent anionic polymerization.	l	1
 	16 Xy lene	4909350		4904120 Heat/Fire	reaction because xylene has two activated methyl groups.	1	1
 			104 Chlorine	4904120 Heat/Fire	Potentially violent/explosive reaction because xylene has two activated methyl groups.	{	1

ASTM GRP Chemical Name	STC ASTM Number GRP Chemical Name	STC Number Compatibility	
18 Xylene	4909350 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Oxidation of methyl groups could 2 lead to exotherm which causes H202 to detonate.
16 Xylene	4909351 104 Hydrogen Peroxide Solution	4918335 Heat/Fire/Explosion	Qxidation of methyl groups could 2 lead to exotherm which causes H2O2 to detonate.

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APPENDIX I RESULTS FROM THE CONSEQUENCE CALCULATIONS

APPENDIX I

Results from the Consequence Calculations

The results from the consequence calculations are given on pages I- 2 through I-62. The tables contain the following information:

- Table I1 Distances to IDLH levels (pages I-2 to I-3),
- Table I2 Toxic Emission Consequence Calculation Summary (pages I-4 to I-34),
- Table I3 Fireball Consequence Calculation Summary (pages I-35 to I-43),
- Table I4 Unconfined Vapor Cloud Explosion Consequence Calculation Summary (pages I-44 to I-56),
- Table I5 Condensed Phase Explosion Consequence Calculation Summary (pages I-57 to I-60),
- Table I6 Tank Car Explosion Consequence Calculation Summary (pages I-61 to I-62).

Given below are IDLH levels used for toxic products formed from incompatible binary reactions not already shown in Table I1:

Chemical Name	IDLH mg/m3
4	
Carbon Monoxide	1750
Cyanogen chloride	49
Ethyl chloride	53758
Fluorine	40
Fluorotrichloromethane	57083
Methyl amine	129
Methyl iodide	4733
Nitrogen dioxide	192
Phosgene	8.2
Phosphine	283
Phosphorus trichloride	286

TABLE I1. Distances to IDLH and TLV Values Following a Spill

			1	Total			Surface	l	Total				Total		
			1	Release	Distan	ce to	Area	1	Release	Distar	ice to		Release	Distan	ce to
	IDLH	TLV	Rel.	kg or	(km) a	t 25C	Covered	Rel.	kg or	(km) a	t 5ØC	Rel.	kg or	(km) at	1000
Chemical Name	mg/m3	mg/m3	Type	kg/sec	IDLH	TLV	sq.m.	Туре	kg/sec	IDLH	TLV	Туре	kg/sec		TLV
Acetaldehyde	18354	184	I	90720	1.1	5.1	95Ø33	I	4536Ø	0.9	4.0	I	22680	0.7	3.2
Acetic Acid, Glacial	2502	25	C	2.0	0.1	1.2	785	C	9.0	0.3	2.6	I	2268Ø	1.3	6.9
Acetic Anhydride	4254	21	C	1.1	⟨∅.1	1.0	⟨785	C	5.6	0.2	2.2	C	44.1	0.4	7.7
Acetone	48400	1815] C	46.4	0.1	Ø.7	785	I	45360	0.6	1.9	I	22680	0.5	1.5
Acrylic Acid	?	10	C	1.1	· -	0.8	-] C	4.0	-	1.6	I	2268Ø	-	6.3
Acrylonitrile	8843	4	C	13.2	0.2	11.1	3142	C	41.9	0.3	26.5	I	22680	0.9	16.4
Alcoholic Beverage (1)	?	?	C	0.1	-	-	-	C	0.3	-	-	C	0.8	-	-
Anhydrous Ammonia	355	18	I	90720	4.1	16.3	1320254	I	45360	3.2	11.6	I	2268Ø	2.6	8.2
Aniline Oil, Liquid	388	8	C	0.2	0.1	0.6	785	C	Ø.6	Ø.2	1.2	C	9.4	0.7	4.7
Benzene	6510	66	C	17.6	0.2	1.8	3142	C	56.3	Ø.4	3.3	I	2268Ø	1.0	3.9
Butadiene	45Ø75	2254	I	90720	0.8	2.2	50265	I	45360	0.6	1.7	I	2268Ø	0.5	1.4
Butadiene, Inhibited	45075	2254	I	90720	0.8	2.2	50265	I	45360	0.6	1.7	·I	2268Ø	0.5	1.4
Butane	?	?	I	90720	_	_	_	I	4536Ø	-	_	I	2268Ø	-	
Butyl Acrylate	?	53	C	1.3	-	Ø.7	-	C	6.1	-	1.5	C	26.4	_	3.6
Buty! Alcohol	24707	618	, c	Ø.9	⟨∅.1	Ø.2	⟨785	į c	7.4	(0.1	Ø.5	i I	2268Ø	Ø.6	2.1
Butyra!dehyde	?	?	j C	31.8	-	_	_	i I	45360	_	_	I I	2268Ø	_	
Carbon Dioxide	91667	9167	I	90720	Ø.6	1.4	28274	i I	45360	_	-	ÌΙ	2268Ø	_	
Carbon Disulfide	1586	32	•	68.6	Ø.9	6.2		iı	45360	2.0	8.6		2268Ø	1.6	6.1
Carbon Tetrachloride	1923		•	36.0	Ø.6	4.6	28274	•	116.9	1.1	3.5	•	2268Ø	1.5	6.1
Chlorine	74		•	90720	8.0	40.0		İ	45360	5.7	28.1	•	2268Ø	4.3	19.8
Chloroprene, Inhibited	1276		•	47.6	Ø.8	5.5		iI	45360	2.1	8.6	•	2268Ø	1.7	6.1
Coal Tar Distillate	32500		•	Ø.7	(0.1	Ø.2		ic	2.7		Ø.4	•	13.1	Ø.1	0.9
Crude Oil Petroleum	?	?	•	Ø.7			•	C	2.3	-	_		13.4	-	
Cyclohexane	35067		•	24.0	Ø.1	Ø.7		C	61.8	0.2	1.0	iI	2268Ø	0.6	1.8
Denatured Alcohol	?	?	i c	8.7		-		ic	26.1	-	-		22680	_	
Dinitrotoluene, Liquid	200		ic	0.0	⟨∅.1	⟨∅.1		ic	Ø	(0.1	⟨∅.1	•	0.03	0.1	0.6
Distillate Fuel Oil	?		•	12.9	-	-		ic	32.3	-	-	ic	28.0	_	
Ethy! Acetate	36713		•	20.1	0.1	Ø.5		ic	67.1	0.2	Ø.9	•	22680	Ø.5	1.6
Ethyl Acrylate, inhibited	8343	21	•	7.7	Ø.1	2.6		i c	28.5	Ø.3	5.1	•	22680		7.5
Ethyl Alcohol	?		•	8.7			-	1 C	26.1		-	I	22680	•	
Ethylene Oxide	1468		•	90720		7.2		II	45360		5.1	•	2268Ø		4.6
Formaldehyde Solution (2)	125			0.04		Ø.5		C	Ø.2			•	1.5		
Fuel Oil	?	_	•	Ø.9		v .5	-	C	2.7			C			0.0
Fuel, Aviation	: ?		•	Ø.3 Ø.7		-		1 0	2.1		-	•	17.1		
•			•			- a 7		C			1 2	•	13.2		,
Glycol Ethers (3) Hexamethylene Diamine (4)	6341		•	2.2		Ø.7		•	7.4		1.3		58.6		3.7
•	17054		•	Ø. Ø1		1.0		I	Ø. Ø5		-		Ø.6		
Hexane	17954		•	32.1		1.8			93.8		3.1		2268Ø		
Hydrochloric Acid (5)	152		•	Ø. Ø2		Ø.2		-	Ø.12		Ø.5	•	2.4		
Hydrocyanic acid	61		•	90720			6082123		45360				2268Ø		
Hydrofluorosilicic Acid (8)			•	0.004					Ø. Ø7		0.4		2.3		
Hydrogen Fluoride	65		•	90720			5674502		45360		15.5		2268Ø		
Hydrogen Peroxide Solution	106		•	0.2					Ø.9		3.5		9.9		26.
Hydroiodic Acid (7)	?		1	0.01		-		•	Ø.2		-	-	10.7		
Isopropanol	50083			6.0		Ø.3			28.3		0.7	•	2268Ø		
Liquefied Petroleum Gas	33250	1750	I	90720	0.6	1.2	28274	I	45360	0.7	1.9	I	22680	Ø.6	1.

TABLE II. Distances to IDLH and TLV Values Following a Spill (Cont.)

			1	Total			Surface	1	Total			1	Total		
	-		1	Release	Distan	ce to	Area	1	Release	Dista	nce to	İ	Release	Distan	ice to
	IDLH	TLV	Rel.	kg or	(km) a	t 25C	Covered	Rel.	kg or	(km) :	at 50C	Re∣.	kg or	(km) at	100C
Chemical Name	mg/m3	mg/m3	Туре	kg/sec	IDLH	TLV	sq.m.	Type	kg/sec	IDLH	TLV	Type	kg/sec	IDLH	TLV
LPG (Butene)	44413	2338	I	90720	Ø.8	2.2	50265	I	4536Ø	Ø.6	1.7	I	22680	Ø.5	. 1.4
LPG (Butylene)	44413	2338	I	90720	0.8	2.2	50265	I	4536Ø	0.6	1.7	I	22680	Ø.5	1.4
LPG (Isobutane)	?	?	I	90720	-	-	-	I	4536Ø	-	-	I	2268Ø	-	-
LPG (Isobutylene)	?	?	I	90720	-	-	<u> </u>	I	4536Ø	-	-	I	2268Ø	-	-
LPG (Propane)	36742	1837	I	90720	Ø.9	2.3	63617	I	4536∅	Ø.7	1.9	I.	2268Ø	0.5	1.5
LPG (Propylene)	?	?	I	90720	_	-	_	I	4536Ø	-	-	I	22680	-	_
Maleic Anhydride	?	1	C	0.05	-	Ø.9	_	C	Ø.5	-	3.0	i C	5.6	-	20.7
Methyl Alcohol	33375	267	C	31.7	0.1	1.5	785	C	36.8	0.1	1.6	i I	2268Ø	0.6	2.8
Methyl Chloride	21Ø38	105	I	90720	1.0	6.7	78540	I	4536Ø	Ø.8	4.8	I	2268Ø	0.7	3.8
Methyl Ethyl Ketone	9014	601	[C	21.4	0.2	0.8	3142	C	51.9	Ø.3	1.3	I	22680	Ø.9	2.1
Methyl Methacrylate Monomer	16687	417	l C	9.0	0.1	0.6	785	C	28.5	0.2	1.1	I	22680	0.7	2.4
Motor Fuel Antiknock (8)			C	2.0	_	-	_	C	5.1			, c	32.9		
Tetraethyl Lead	40	0.1	C	0.006	<0.1	1.1	⟨785	C	0.04	0.1	2.7	C	0.6	Ø.5	22.2
Ethylene Dibromide	3131	157	C	0.2	⟨∅.1	0.2	⟨785	C	0.7	0.1	Ø.3	C	5.3	Ø.2	Ø.8
Ethylene Dichlorid	4123	41	[C	1.8	0.1	Ø.9	785	C	4.4	Ø.1	1.4	C	27.0	Ø.3	3.5
Nitric Acid, Fuming	263	5	C	11.7	0.9	8.2	63617	C	36.1	1.6	25.2	I	2268Ø	2.8	14.9
Octyl Alcohol	?	?	C	0.03	_	_	_	C	Ø.2	_	-	į c	5.6	_	_
Oleum (9)	80	1	C	10.5	1.6	39.8	201062	C	24.4	2.4	92.2	I	2268Ø	3.2	24.6
Petroleum Naphtha	41667	2083	C	1.3	⟨∅.1	0.1	⟨785	C	2.6	(Ø.1	Ø.2	i c	16.7	0.1	Ø.4
Phenol	392	20	C	0.1	⟨∅.1	Ø.3	⟨785	i c	Ø.9	0.2	0.9	i c	9.3	0.7	3.0
Phosphoric Acid	?	1	C	0.009	_	Ø.4	_	i c	Ø.0	_	_	C	0.0	_	_
Propylene Oxide	4837	48	I	90720	1.7	9.9	226980	I	45360	1.3	7.0	I	22680	1.1	5.0
Styrene Monomer, Inhibited	21698	868	C	1.3	⟨∅.1	0.1	⟨785	, C	5.7	(0.1	0.3	C	39.6	Ø.2	0.9
Sulfur Dioxide	267	5	I	90720	4.5	29.9	1590431	I	4536Ø	3.5	21.1	I	22680	2.8	14.9
Toluene	7678	384	j c	6.1	0.1	Ø.5	785	j c	23.0	Ø.2	1.1	i I	22680	0.9	2.9
Toluene Diisocyanate	73	Ø.1	jc	0.02	⟨∅,1	1.5	⟨785	j c	0.1	0.1	3.3	i c	1.2	0.6	31.7
Vinyl Acetate	?	36	C	32.6	_	4.1	_	C	77.5	-	8.0	I	2268Ø	-	5.7
Vinyl Chloride	?	13	I	90720	-	19.1	_	I	45360	-	13.5	I	2268Ø	_	9.5
Xylene	44238	442	i c	2.5	⟨∅.1	0.3	₹785	ìc	7.6	(0.1	Ø.6	ic	49.8	0.1	1.4

- Alcoholic beverage = 20% Ethanol/80% Water 1)
- 2) Formaldehyde = 37% formaldehyde/50% water/13% Methanol
- 3) Glycol ethers = Methyl Cellosolve
- 4) Hexamethylene Diamine Solution = 70% in Water
- 5) Hydrochloric acid = 20% HC1/80% water
- Hydrofluorosilicic acid = 25% 2HF-SiF4/75% water 6)
- Hydroiodic acid = 57% HI/43% water
- Motor Fuel Antiknock Compound = 50% Tetraethyl Lead, 30% Ethylene Dibromide, 20% Ethylene Dichloride 9) Oleum = 50% S03/50% H2S04

Note: I = instantaneous release (kg), I = continuous release (kg/sec).

298 Degrees K (7853.982 sq. meters) Temperature Spill diameter =

Wind speed =

Summary of Toxic Emiss	ion Conse	quence Calculations			1	Non-	nixed	Mixe	ed	Distar	nces for	Gas	
					[0)istano	e (m)	Distant	e (m)	J	Releases	5	Ï
ı	ASTM		ASTM		1	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	İ
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	1	#1	#2	#1	#2	Gas 1	Gas 2	Gas 3	Conclusions
Aceta I dehyde	5	Acetic Anhydride	107	Release of vinyl acetate	-	1100	< 100	900	200				 Mitigated
Acetaldehyde	5	Alkaline Liquid	10	Evaporation of residuals at 100 C	i	1100	Ø	700	Ø	I			Mitigated
Aceta I dehyde	5	Anhydrous Ammonia	10	Mitigate NH3 release	1	1100	4100	900	3900	l			Mitigated
Aceta I dehyde	5	Aniline Oil	7	Evaporation of residuals at 100 C		1100	100	700	700	í			Mitigated
Aceta i dehyde	5	Chlorine	104	Evaporation of residuals at 50 C **	- 1	1100	8000	900	7600	l			Mitigated
Aceta I dehyde	5	Dinitrotoluene, Liquid	27	Evaporation at 100 C; Explosion		1100	< 100	700	< 100	ļ			Mitigated
Aceta I dehyde	5	Ferric Chloride Solution	164	Evaporation of residuals at 50 C	-1	1100	Ø	900	Ø	Ι.			Mitigated
Aceta I dehyde	5	Hexamethylene Diamine Soln	7	Evaporation of residuals at 50 C	- 1	1100	〈 100	900	< 100	l			Mitigated
ceta i dehyde	5	Hydrogen Peroxide	194	Evaporation at 100 C; explosion **	1	1100	200	700	1300	· ·			Aggravated
ceta I dehyde	5	Pheno I	31	Evaporation of residuals at 100 C **	l	1100	< 100	700	700	l			Mitigated
lceta i dehyde	5	Phosphorus	105	Evaporation of residuals at 50 C		1100	Ø	900	Ø	1			Mitigated
lceta i dehyde	5	Potassium Hydroxide, Liq	10	Evaporation of residuals at 100 C	- 1	1100	Ø	700	Ø	l			Mitigated
lceta I dehyde	5	Sodium Hydroxide, Liq.	10	Evaporation of residuals at 100 C	l	1100	Ø	700	Ø	1			Mitigated
ceta I dehyde	5	Sodium Metal	21	Evaporation of residuals at 50 C **	-	1100	Ø	900	Ø	ĺ			Mitigated
Aceta I dehyde	5	Sulfur Dioxide	105	Unknown	- 1	1100	4500	?	?	1			Unknown
Aceta I dehyde	5	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	1	1100	< 100	700	600				Mitigated
Acetic Acid, Glacial 🦠	3	Acetaldehyde	5	Evaporation of residuals at 100 C	1	100	1100	1300	700				Aggravated
Acetic Acid, Glacial	3	Acetic Anhydride	107	Evaporation of residuals at 100 C	1	100	< 100	1300	400				Aggravated
Acetic Acid, Glacial	3	Acrylonitrile	26	Evaporation of residuals at 50 C	١	100	200	300	300)			Aggravated
Acetic Acid, Glacial	3	Alkaline Liquid	10	Evaporation of residuals at 100 C	-	100	Ø	1300	Ø				Aggravated
Acetic Acid, Glacial	3	Ammonium Nitrate Soln.	106	Evaporation of residuals at 50 C	1	100	Ø	300	Ø				Aggravated
Acetic Acid, Glacial	3	Anhydrous Ammonia	10	Mitigate release of NH3	1	100	4100	300	3900				Mitigated
Acetic Acid, Glacial	3	Aniline Oil	7	Evaporation of residuals at 50 C	- 1	100	100	900	200				Aggravated
Acetic Acid, Glacial	3	Butyraldehyde		Evaporation of residuals at 100 C	ļ	100	< 100	1300	< 100				Aggravated
Acetic Acid, Glacial	3	Chlorine	104	Evaporation of residuals at 100 C **	-	100	8000	1300	7600				Mitigated
cetic Acid, Glacial	3	Chloroprene	17	Evaporation of residuals at 100 C	ļ	100	800	1300	1700				Aggravated
cetic Acid, Glacial	3	Ethylene Oxide	34	Evaporation of residuals at 100 C	1	100	2500	1300	1600				Mitigated
cetic Acid, Glacial	3	Formaldehyde Solution		Evaporation of residuals at 100 C	1	100	100	1300	500				Aggravated
cetic Acid, Glacial	3	Hexamethylene Diamine Soln	7	Evaporation of residuals at 50 C	١	100	< 100	900	< 100				Aggravated
Acetic Acid, Glacial	3	Hydrogen Peroxide	104	Evaporation at 100 C; Explosion	1	100	200	•	1300				Aggravated
cetic Acid, Glacial	3	Potassium Hydroxide, Liq	10	Evaporation of residuals at 100 C	1	100	Ø	1300	Ø				Aggravated
cetic Acid, Glacial	3	Propylene Oxide	34	Evaporation of residuals at 50 C	1	100	1700	300	1300				Mitigated
cetic Acid, Glacial	3	Sodium Hydroxide, Liq.	10	Evaporation of residuals at 100 C	- 1	100	Ø	1300	Ø				Aggravated

•				•	Dista	nce (m)	Dis	tance (m)] R	eleases _.	1
	ASTM		ASTM		IDL	IDLH	10	LH IDLH	IDLH	IDLH IDLH	
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1 -	#2	 	1 #2	Gas 1	Gas 2 Gas 3	Conclusions
Acetic Acid, Glacial	3	Sodium Metal	21	Evaporation at 100 C; release H2 gas **	16	10 () 1	300 0	!		Aggravated
Acetic Acid, Glacial	3	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	16	00 < 100	1 1	300 600	1		Aggravated
Acetone	19	Chlorine	104	Mitigation of chlorine release **	16	10 800	3	600 7600	1		Mitigated
Acetone	19	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	16	10 201)	500 1300	1		Aggravated
Acetone	19	Sodium Metal	21	Releases H2 gas	16	10 (7	500 0	< 100		Aggravated
Acrylic Acid	3	Aceta I dehyde	5	Evaporation of residuals at 100 C	806	+ 1100	63	100 * 700	1		Mitigated
Acrylic Acid	3	Acetic Anhydride	107	Evaporation of residuals at 100 C	808	* < 100	63	100* 400	1		Aggravated
Acrylic Acid	3	Alkaline Liquid	16	Evaporation of residuals at 100 C	1 809	l + (63	100 + 0			Aggravated
Acrylic Acid	3	Ammonium Nitrate Soln.	108	Evaporation of residuals at 50 C	809	i* () 16	00+ 0	1		Aggravated
Acrylic Acid	3	Anhydrous Ammonia	10	Mitigate release of NH3	806	+ 4100	16	00* 3900	1		Mitigated
Acrylic Acid	3	Aniline Oil	7	Evaporation of residuals at 100 C	808	l* 100) 63	00* 700	1		Aggravated
Acrylic Acid	3	Chlorine	104	Evaporation of residuals at 100 C **	806	* 8000	63 63	00 + 7600	1		Mitigated
Acrylic Acid	3	Chloroprene	17	Evaporation of residuals at 100 C	806	# 8Ø	5 63	00* 1700	1		Aggravated
Acrylic Acid	3	Ethylene Oxide	34	Evaporation of residuals at 100 C	806	* 25Ø	J 63	00* 1600]		Mitigated
Acrylic Acid	3	Formaldehyde Solution	5	Evaporation of residuals at 100 C	806	+ 100	63	00 + 500	1		Aggravated
Acrylic Acid	3	Hexamethylene Diamine Soln	7	Evaporation of residuals at 100 C	806	* < 10°	63	00* < 100	1		Aggravated
Acrylic Acid	3	Hydrogen Peroxide	104	Evaporation at 100 C; Explosion	1 809	i* 206	63	00* 1300	1		Aggravated
Acrylic Acid	3	Potassium Hydroxide, Liq	10	Evaporation of residuals at 100 C	808	l +	63	00* 0	1		Aggravated
Acrylic Acid	3	Sodium Hydroxide, Liq.	10	Evaporation of residuals at 100 C	808	l# £	63	00* 0	•		Aggravated
Acrylic Acid	3	Sodium Metal	21	Evaporation at 100 C; release H2 gas **	1 800	i+ í	63	00+ 0	< 100		Aggravated
Acrylic Acid	3	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	808	+ < 100	63 63	00* 600	1		Aggravated
Acrylonitrile	26	Chlorine	104	Evaporation of residuals at 50 C **	26	10 800 <i>6</i>)	300 7600	1		Mitigated
Acrylonitrile	26	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	26	iØ (1	300 0			Aggravated
Acrylonitrile	26	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	20	0 200		900 1300	i		Aggravated
Acrylonitrile	28	Pheno I	31	Evaporation of residuals at 100 C	28	0 < 10		900 700	•		Aggravated
Acrylonitrile	28	Sulfur Dioxide		Unknown	20		•	? ?	1		Unknown
Alcoholic Beverage	4	Acetaldehyde	5	Evaporation at 50 C; releases acetal	< 10	0 1106	(100 900	,		Mitigated
Alcoholic Beverage	4	Butyraldehyde	5	Evaporation of residuals at 50 C	< 10	0 (100	(100 < 100	1		Same
Alcoholic Beverage	4	Chlorine	104	Evaporation at 50 C; explosion **	< 10	Ø 8000	1	100 7600	1		Mitigated
Alcoholic Beverage	4	Ethylene Oxide		Evaporation of residuals at 50 C	< 10		1 (1		Mitigated
Alcoholic Beverage	4	Hydrogen Peroxide		Evaporation at 50 C; explosion **	< 10		1		•		Aggravated
Alcoholic Beverage	4	Propylene Oxide		Evaporation of residuals at 50 C	< 10		1				Mitigated
Alcoholic Beverage	4	Sodium Metal	21	Release of hydrogen gas **	< 10	Ø 6	1 (100 0	1		Same

ASTW	ASTM	Summary of Toxic Emissi	ion Conse	equence Calculations			I	Non-		Mixe		Distan	ces for Gas	1
Cheaical Name #1 Group Cheaical Name #2 Group Toxic Emission Consequences #1 #2 #2 Gas 1 Gas 2 Gas 3 Conclusion Alcoholic Beverage A Toluene Dissocyanate 18 Evaporation of residuals at 188 C 186 186 688 Aggravat Alkaline Liquid 18 Acetic Anhydride 19 Evaporation of residuals at 188 C 8 188 8 688 Aggravat Alkaline Liquid 18 Acetic Anhydride 19 Evaporation of residuals at 188 C 8 188 8 688 8 188 Aggravat Alkaline Liquid 18 Acetic Ministrate Soln. 180 Color Consequences 8 8 8 1588 988 1588 988 Alkaline Liquid 18 Carbon Tetrachloride 17 Releases HELL and phospene 18 688 8 1588 988 1588 988 Alkaline Liquid 18 Chloroprene 17 Evaporation of residuals at 188 C 8 888 8 1578 988 Alkaline Liquid 18 Chloroprene 17 Evaporation of residuals at 188 C 8 888 8 1578 988 Alkaline Liquid 18 Chloroprene 17 Evaporation of residuals at 188 C 8 888 8 1578 988 Alkaline Liquid 18 Ethyl Acetate 13 Releases HEL and phospene 18 688 8 1578 988 Alkaline Liquid 18 Ethyl Acetate 13 Releases HEL and phospene 18 688 8 1578 988 Alkaline Liquid 18 Ethyl Acetate 13 Releases Alkaline Liquid 19 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 888 8 1778 Maleic Alkaline Liquid 19 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 888 8 1788 Alkaline Liquid 18 Methyl Ethyl Ketone 19 Evaporation of residuals at 188 C 8 888 8 188 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 8 8 188 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 8 8 188 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 188 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 188 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 188 Maleic Anhydride 187 Evaporation of residuals at	Chemical Name #1 Croup Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name #2 Group Chemical Name *2 Group Chemical Name *2 Group Chemical Name *2 Group Chemical Name *2 Group Chemical Name *2 Group Name *2						D	stand	e (m)	Distanc	e (m)	ł R	eleases	1
Alkaline Liquid Alkaline Liquid Action Action 19 Evaporation of residuals at 188 C	Alkaline Liquid	Chemical Name #1		Chemical Name #2		Toxic Emission Consequences	•			•		•		•
Alkaline Liquid Alkaline Liquid Action Action 19 Evaporation of residuals at 188 C	Alkaline Liquid 19	Alcoholic Reverage	4	Toluene Diisocyanate	18	Evaporation of residuals at 100 C		100	 / 100					-
Alkaline Liquid 18	Alkaline Liquid 18	_		•			- ¦ `		-	•		•		
Alkaline Liquid 18	Alkaline Liquid	•		· ·		•	1		•	•		•		
Alkaline Liquid 16 Ammonium Nitrate Soln. 166 Local consequences 8 8 8 8 1568 980 2188 Aggravat Alkaline Liquid 16 Chloroprene 17 Evaporation of residuals at 188 C 8 888 8 1568 980 2188 Aggravat Alkaline Liquid 18 Chloroprene 17 Evaporation of residuals at 188 C 8 888 8 1568 980 3888 Alkaline Liquid 18 Ethyl Acetate 13 Releases acetic acid and achanol 8 188 8 588 3888 Alkaline Liquid 18 Ethyl acene Bxide 34 Evaporation of residuals at 188 C 8 888 8 1588 988 1888 Alkaline Liquid 18 Maleic Anhydride 187 Evaporation of residuals at 188 C 8 2588 8 1688 1688 1688 1688 1688 1688 1688	Alkaline Liquid 18 Amenium Mitrate Soln. 186 Local consequences 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	•				•	1	_		•		•		
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Anhydrous Ammonia 10 Acrylonitrile 26 Mitigate release of NH3 4100 200 3900 300 Mitigate release of NH3 4100 600 3900 1100 Mitigate release of NH3 4100 600 3900 1100 Mitigate release of NH3 4100 600 3900 1100 Mitigate release of NH3 4100 800 4000 6900 7 2700 Mitigate release of NH3 4100 800 3900 2100 Mitigate release of NH3 4100 800 3900 2100 Mitigate release of NH3 4100 (100 3900 (100 Mitigate release of NH3 4100 (100 3900 200 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 4100 2500 3700 1600 Mitigate release of NH3 4100 2500 3700 2500 Mitigate release of NH3 4100 2500 3700 2500 Mitigate release of NH3 4100 2500 3700 2500 Mitigate release of NH3 4100 2500 3700 2500 Mitigate release o	Anhydrous Ammonia 10 Acrylonitrile 26 Mitigate release of NH3 4100 200 3900 300 Mitigate release of NH3 Anhydrous Ammonia 10 Carbon Tetrachloride 17 Mitigate release of NH3 4100 600 3900 1100 Mitigate release of NH3 Anhydrous Ammonia 10 Chloroprene 17 Mitigate release of NH3 4100 800 3900 2100 Mitigate release of NH3 Anhydrous Ammonia 10 Dinitrotoluene, Liquid 27 Mitigate release of NH3 4100 100 3900 100 Mitigate release of NH3 Anhydrous Ammonia 10 Ethyl Acetate 13 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 Anhydrous Ammonia 10 Ethylene Oxide 34 Mitigate release of NH3 ** 4100 2500 3700 1600 Mitigate release of NH3	-		· ·		_	•					•		-
Anhydrous Ammonia 16 Carbon Tetrachloride 17 Mitigate release of NH3 4100 600 3900 1100 Mitigate of NH3 4100 600 3900 1100 Mitigate of NH3 4100 8000 4000 6900 7 2700 Mitigate of NH3 4100 8000 3900 2100 Mitigate of NH3 4100 8000 3900 2100 Mitigate of NH3 4100 4000 8000 4000 <	Anhydrous Ammonia 16 Carbon Tetrachloride 17 Mitigate release of NH3 4100 600 3900 1100	=				_	•			•] , 1		-
Anhydrous Ammonia 10 Chlorine 104 Releases NCL3 and HCL gas 4100 8000 4000 6900 ? 2700 Mitigate of NH3 4100 800 3900 2100 Mitigate of NH3 4100 800 3900 2100 Mitigate of NH3 4100 3900 4000 3900 4000 Mitigate of NH3 4100 4	Anhydrous Ammonia 10 Chlorine 104 Releases NCL3 and HCL gas 4100 8000 4000 6900 ? 2700 Mitigate follows Ammonia 4100 8000 4000 6900 ? 2700 Mitigate follows Ammonia 4100 8000 4000 6900 ? 2700 Mitigate follows Ammonia 4100 8000 3900 2100 Mitigate follows Ammonia<	•		-		-	•] 		-
Anhydrous Ammonia 10 Chloroprene 17 Mitigate release of NH3 4100 800 3900 2100 2100 Mitigate release of NH3 Anhydrous Ammonia 10 Dinitrotoluene, Liquid 27 Mitigate release of NH3 4100 < 100 3900 < 100	Anhydrous Ammonia 10 Chloroprene 17 Mitigate release of NH3 4100 800 3900 2100 Mitigate	, <u> </u>											0700	
Anhydrous Ammonia 10 Dinitrotoluene, Liquid 27 Mitigate release of NH3 4100 < 100 3900 < 100 Mitigate	Anhydrous Ammonia 10 Dinitrotoluene, Liquid 27 Mitigate release of NH3 4100 < 100 3900 < 100 Mitigate					_	•			•		! 	2100	
Anhydrous Ammonia 10 Ethyl Acetate 13 Mitigate release of NH3 4100 100 3900 200 Mitigate of NH3 4100 2500 3700 1600 Mitigate of NH3 4100 2500 3700 1600 Mitigate of NH3 4100 2500 3700 3700 Mitigate of NH3 4100 2500 3700 3700 Mitigate of NH3 4100 3900 3700 3700 Mitigate of NH3 4100 3900 3700 3700	Anhydrous Ammonia 10 Ethyl Acetate 13 Mitigate release of NH3 4100 100 3900 200 Mitigate release of NH3 Anhydrous Ammonia 10 Ethylene Oxide 34 Mitigate release of NH3 ** 4100 2500 3700 1600 Mitigate release of NH3 **			· ·		-	•							
Anhydrous Ammonia 10 Ethylene Oxide 34 Mitigate release of NH3 ** 4100 2500 3700 1600 Mitigate college	Anhydrous Ammonia 10 Ethylene Oxide 34 Mitigate release of NH3 ** 4100 2500 3700 1600 Mitigate	i =		· · · · · · · · · · · · · · · · · · ·			•		•					
		-				_	•					•		-
Anhydrous Ammonia 10 Maleic Anhydride 107 Mitigate release of NH3 4100 900* 3900 3000*	Anhydrous Ammonia 10 Maleic Anhydride 107 Mitigate release of NH3 4100 900* 3900 3000*	· •				-	•							
		Anhydrous Ammonia	10	Maleic Anhydride	107	Mitigate release of NH3	ı	4100	900*	3900	3000*			Mitigated

Summary of Toxic Eniss	ion Conse	quence Calculations			 Non-	 mixed	 Mixe	 ed	 Distan	 ces for Gas	
•		, , , , , , , , , , , , , , , , , , , ,			Distan		Distance		•	eleases	l
	ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	i
Chemical Name #1	Group		Group	Toxic Emission Consequences	#1	#2	#1	#2	Gas 1	Gas 2 Gas 3	•
Anhydrous Ammonia	10	Methyl Chloride	17	Releases HCL and methyl amine gas	- 4100	1000	3500	 Ø	 4800	4800	 Aggravated
Anhydrous Ammonia	10	Methyl Ethyl Ketone	19	Mitigate release of NH3	4100	200	3900	300	Ì		Mitigated
Anhydrous Ammonia	10	Motor Fuel Antiknock Compo	24	Releases ethane gas	4100	100	3900	100	< 100		Mitigated
Anhydrous Ammonia	16	Propylene Oxide	34	Mitigate release of NH3 **	4100	1700	3700	1100	İ		Mitigated
Anhydrous Ammonia	16	Sodium Metal	21	Releases H2 gas; mitigate NH3 release	4100	Ø	3900	œ	ĺ		Mitigated
Anhydrous Ammonia	16	Styrene Monomer	16	Mitigate release of NH3	4100	< 100	3900	(100	Ì		Mitigated
Anhydrous Ammonia	18	Sulfur Dioxide	105	Mitigate release of SO2 and NH3	4100	4500	3700	3500	İ		Mitigated
Anhydrous Ammonia	16	Toluene Diisocyanate	18	Mitigate release of NH3	4100	< 100	3900	100	ì	-	Mitigated
Anhydrous Ammonia	16	Vinyl Acetate	13	Mitigate release of NH3	4100	4100+	3900	8000+	ĺ		Mitigated
Anhydrous Ammonia	16	Vinyl Chloride	17	Release acetylene gas	4100	19100+	3700	18100+]		Mitigated
Aniline Dil	7	Acetic Anhydride	107	Evaporation of residuals at 100 C	100	< 100	700	400	1		Aggravated
Aniline Oil	7	Acrylonitrile	26	Evaporation of residuals at 100 C	[100	200	700	900	1		Aggravated
Aniline Oil	7	Carbon Tetrachloride	17	Evaporation of residuals at 100 C	100	600	700	1500	1		Aggravated
Aniline Oil	7	Chlorine	184	Evaporation of residuals at 50 C **	100	8000	200	7600	1		Mitigated
Aniline Dil	7	Chloroprene	17	Evaporation of residuals at 100 C	100	800	700	1700	i		Aggravated
Aniline Oil	7	Ethylene Oxide		Evaporation of residuals at 100 C	100	2500	700	1600			Mitigated
Aniline Oil	7	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	100	Ø	200	Ø	Į.		Aggravated
Aniline Oil	7	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	100	200	700	1300	I		Aggravated
Aniline Oil	7	Maleic Anhydride	107	Evaporation of residuals at 100 C	100	900*	700	20700+	1		Aggravated
Aniline Dil	7	Methyl Chloride		Mitigate methyl chloride release	100	1000	200	1000	1		Same
Aniline Oil	7	Motor Fuel Antiknock Compo		Evaporation at 50 C; release ethane gas	100	100	200	100	l		Aggravated
Aniline Oil	7	Propylene Oxide	34	Evaporation of residuals at 100 C	100	1700	700	1100	1		Mitigatèd
Aniline Oil	7	Sodium Metal		Evaporation at 100 C; release H2 gas	100		700		< 100		Aggravated
Aniline Oil	7	Toluene Diisocyanate		Evaporation of residuals at 100 C	100	•	700	800	ĺ		Aggravated
Aniline Dil	7	Vinyl Chloride		Mitigate release of vinyl chloride	•	19100+	•	18100*	i		Mitigated
Benzene	16	Chlorine		Mitigation of chlorine release **	200		400	7600	1		Mitigated
Benzene	16	Hydrogen Peroxide		Evaporation at 100 C; explosion **	200		1000	1300	i		Aggravated
Butadiene	28	Chlorine	104		800		800	8000	•		Same
Butadiene	28	Ferric Chloride Solution		Mitigation of butadiene release	800	Ø	800	Ø	•		Same
Butadiene	28	Hydrogen Peroxide		Evaporation of residuals at 100 C **	800	200	800	400	•		Same
Butadiene	28	Pheno I		Mitigation of butadiene release	800		800	700	•		Same
Butane Butyl Acrylate	29	Chlorine	104	Fire Mitigation of chlorine release **	< 100		< 100 1500*	8000	1		Same Mitigated

Summary of Toxic Emiss	ion Conse	equence Calculations				mixed	Mixe		Distan	ces for G	ns (
				*	Distan	ce (m)	Distand	e (m)] F	eleases	1
	ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH I	DLH
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1 !	#2	#1	#2	Gas 1	Gas 2 Ga	3 Conclusion
Butyl Acrylate	13	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	700+	Ø	1500+	Ø			Aggravated
Butyl Acrylate	13	Hydrogen Peroxide	184	Evaporation at 100 C; explosion **	700+	200	3000*	1300	1		Aggravated
Butyl Acrylate	13	Sodium Metal	21	Evaporation of residuals at 100 C **	700+	Ø	3000*	Ø	i		Aggravated
Butyl Alcohol	4	Acetaldehyde	5	Evaporation of residuals at 100 C	< 100	1100	600	700	i		Mitigated
Butyl Alcohol	4	Butyraldehyde	5	Evaporation of residuals at 100 C	< 100	< 100	600	< 100	1		Same
Butyl Alcohol	4	Chlorine	164	Evaporation at 50 C; explosion **	< 100	8000	< 100	7800	I		Mitigated
Butyl Alcohol	4	Ethylene Oxide .	34	Evaporation of residuals at 50 C	< 100	2500	< 100	2000	t		Mitigated
Butyl Alcohol	4	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	< 100	200	600	1300	J		Aggravated
Butyl Alcohol	4	Propylene Oxide	34	Evaporation of residuals at 50 C	< 100	1700	< 100	1300	l		Mitigated
Butyl Alcohol	4	Sodium Metal	21	Release of hydrogen gas ++	< 100	Ø	500	Ø	1		Aggravated
Butyl Alcohol	4	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	< 100	< 100	500	600	I		Aggravated
Butyraldehyde	5	Acetic Anhydride	107	Release of vinyl acetate	< 100	< 100	< 100	200	ſ		Aggravated
Butyraldehyde	5	Alkaline Liquid	18	Evaporation of residuals at 100 C	< 100	Ø	< 100	0	1		Same
Butyraldehyde	5	Anhydrous Ammonia	10	Mitigate NH3 release	< 100	4100	< 100	3900	i		Mitigated
Butyraldehyde	5	Aniline Oil	7	Evaporation of residuals at 50 C	< 100	100	< 100	200	i		Aggravated
Butyraldehyde	5	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	< 100	7600	l		Mitigated
Butyraldehyde	8	Dinitrotoluene, Liquid	27	Evaporation at 100 C; Explosion	< 100	< 100	< 100	< 100 .	ì		Same
Butyraldehyde	5	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	< 100	Ø	(100	Ø	Ì		Same
Butyraldehyde	Б	Hexamethylene Diamine Soln	7	Evaporation of residuals at 50 C	< 100	< 100	< 100	< 100	İ		Same
Butyra i dehyde	5	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	< 100	200	< 100	1300	1		Aggravated
Butyraldehyde	5	Pheno I	31	Evaporation of residuals at 100 C **	< 100	(100	(100	700			Aggravated
Butyraldehyde	5	Potassium Hydroxide, Liq	10	Evaporation of residuals at 100 C	< 100	Ø	(100	Ø	l		Same
Butyraldehyde	5	Sodium Hydroxide, Liq.	10	Evaporation of residuals at 100 C	< 100	Ø	< 100	Ø	- 		Same
Butyraldehyde	5	Sodium Metal	21	Evaporation of residuals at 50 C **	< 100	Ø	(100	Ø			Same
Butyra I dehyde	5	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	< 100	< 100	< 100	600			Aggravated
Carbon Disulfide	20	Chlorine	104	Evaporation of residuals at 100 C	900	8000	900	8000			Mitigated
Carbon Disulfide	20	Ferric Chloride Solution	104	Mitigate release of CS2	900	Ø	2400	Ø			Aggravated
Carbon Disulfide	20	Hydrogen Peroxide	104	Releases SO2 gas; explosion **	900	200	2300	1300	3000		Aggravated
Carbon Disulfide	20	Propylene Oxide	34	Mitigate release of CS2	900	1700	2400	1300			Aggravated
Carbon Disulfide	20	Sodium Metal	21	Evaporation of residuals at 50 C	900	Ø	2000	Ø			Aggravated
arbon Tetrachloride	17	Carbon Disulfide	20	Evaporation of residuals at 50 C	600	900	1100	2000			Aggravated
arbon Tetrachloride	17	Sodium Metal	21	Evaporation at 100 C; Explosion	600	Ø	•	øj			Aggravated
Chlorine	104	Acetic Anhydride	107	Evaporation of residuals at 50 C **	8000	⟨ 100	7600	200			Mitigated

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Summary of Toxic Emissic	n Conse	quence Calculations			 Non-m	ixed	Mixe	 d	 Distan	ces for Gas	
					Distanc	e (m)	Distanc	e (m)	l R	eleases	
	ASTM	•	ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	
Chemical Name #1	Group	Chemical Name #2	Згоир	Toxic Emission Consequences	#1	#2	#1	#2	Gas 1	Gas 2 Gas 3	•
Chlorine	104	Ammonium Nitrate Soln.	106	Mitigation of chlorine release; explosion	8000	Ø	7600	Ø	<u>-</u> 		 Mitigated
Chlorine	104	Maleic Anhydride	107	Evaporation of residuals at 50 C **	8000	900+	7600	3000*	1		Mitigated
Chlorine	104	Phosphorus	105	Releases PCL3 gas	8000	Ø	7600	Ø	2200		Mitigated
Chloroprene	17	Chlorine	104	Mitigation of chlorine release **	800	8000	1700	7600	l		Mitigated
Chloroprene	17	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	800	Ø	1700	Ø	l		Aggravated
Chloroprene	17	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	800	200	1700	1300	ſ		Aggravated
Chloroprene	17	Motor Fuel Antiknock Compo	24	Evaporation of residuals at 50 C	800	100	1700	100	1		Aggravated
Chloroprene	17	Sodium Metal	21	Evaporation at 100 C; explosion **	800	Ø	1700	Ø			Aggravated
Coal Tar Distillate	101	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	100	7600	1		Witigated
Coal Tar Distillate	101	Hydrogen Peroxide	104	, ,	< 100	200	100	400	1		Aggravated
Combustible Liquid	101	Chlorine	104	Evaporation of residuals at 50 C **	〈 100	8000	,	7600	1		Mitigated
Combustible Liquid	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 100	200	1 5	400	l	,	Aggravated
Compound, Cleaning	101	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	?	7600	J		Mitigated
Compound, Cleaning	101	Hydrogen Peroxide	104		< 100	200	?	400			Aggravated
Crude Bil Petroleum	101	Chlorine	104	Evaporation of residuals at 50 C **	(100	8000	100	7600	l		Mitigated
Crude Dil Petroleum	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 100	200	100	400	Ι΄		Aggravated
Cyc Iohexane	29	Chlorine	164	Evaporation of residuals at 50 C **	100	8000	200	7600	i		Mitigated
Denatured Alcohol	4	Acetaldehyde		Evaporation at 100 C; releases acetal	〈 100	1100	< 100	700	?	·	Mitigated
Denatured Alcohol	4	Butyraldehyde	5	•	〈 100	< .100	< 100	< 100	l		Same
Denatured Alcohol	4	Chlorine	104		< 100		< 100	7600	l		Mitigated
Denatured Alcohol	4	Ethylene Oxide	34	•	〈 100		< 100	2000	1		Mitigated
Denatured Alcohol	4	Hydrogen Peroxide	104	• •	< 100		< 100	1300	l		Aggravated
Denatured Alcohol	4	Propylene Oxide	34	•	< 100	1700	< 100	1300	l		Mitigated
Denatured Alcohol	4	Sodium Metal	21	• •	< 100	Ø	< 100	Ø	l		Same
Denatured Alcohol	4	Toluene Diisocyanate	18	•	< 100		< 100	600	1		Aggravated
. Dinitrotoluene, Liquid	27	Acetic Anhydride	107				(100	400	l		Aggravated
Dinitrotoluene, Liquid	27	Chlorine	104	Evaporation at 50 C; explosion	< 100		< 100	7600			Mitigated
Dinitrotoluene, Liquid	27	•	104		< 100		< 100	1300			Aggravated
Dinitrotoluene, Liquid	27	•	107		< 100		< 100				Aggravated
Dinitrotoluene, Liquid	27	•	105		< 100		< 100	Ø			Same
Distillate Fuel Oil	101		104	- ·	< 100	8000	•	7600			Mitigated
Distillate Fuel Oil	101	, ,	104		(100	200	-	400	,		Aggravated
Ethyl Acetate	13	Chlorine	104	Mitigation of chlorine release **	100	8000	200	7600			Mitigated
				. 1.							
			•								5

Summary of Toxic Emission	Conse	equence Calculations			•	on-mi tance		Mixe Distanc		Distances for Ga Releases	s
	ASTM		ASTM		, IDI		IDLH	IDLH	IDLH	IDLH IDLH ID	_H i
Chemical Name #1	Group	Chemical Name ∦2	Group	Toxic Emission Consequences	#1	1	#2	#1	∦2 _	Gas 1 Gas 2 Gas	3 Conclusions
Ethyl Acetate	13	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	 1 1	100	200	 500	1300	 	 Aggravated
Ethyl Acetate	13	Sodium Metal	21	Evaporation of residuals at 100 C **	1	100	Ø	500	Ø]	Aggravated
Ethyl Acrylate	13	Chlorine	104	Mitigation of chlorine release **	1	100	8000	300	7600	l	Mitigated
Ethyl Acrylate	13	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	1	100	Ø	300	Ø	1	Aggravated
Ethyl Acrylate	13	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	1	100	200	900	1300		Aggravated
Ethyl Acrylate	13	Sodium Metal	21	Evaporation of residuals at 100 C **	1	100	Ø	900	Ø	1	Aggravated
Ethyl Alcohol	4	Acetaldehyde	5	Evaporation at 100 C; releases acetal	(1	100	1100	< 100	700	?	Mitigated
Ethyl Alcohol	4	Butyraldehyde	5	Evaporation of residuals at 100 C	(1	100	< 100	〈 100	(100	j	Same
Ethyl Alcohol	4	Chlorine	104	Evaporation at 50 C; explosion **	1 (1	100	8000	< 100	7600	İ	Mitigated
Ethyl Alcohol	4	Ethylene Oxide	34	Evaporation of residuals at 50 C	(1	100	2500	< 100	2000	1	Mitigated
Ethyl Alcohol	4	Hydrogen Peroxide	194	Evaporation at 100 C; explosion **	(1	100	200	< 100	1300	l	Aggravated
Ethyl Alcohol	4	Propylene Oxide	34	Evaporation of residuals at 50 C	(1	100	1700	⟨ 100	1300	ľ	Mitigated
Ethyl Alcohol	4	Sodium Metal	21	Release of hydrogen gas **	(1	00	Ø	< 100	Ø		Same
Ethyl Alcohol	4	Toluene Diisocyanate	18	Evaporation of residuals at 100 C **	(1	100	< 100	< 100	600	[Aggravated
Ethylene Oxide	34	Chlorine	104	Evaporation of residuals at 50 C **	25	90	8000	2000	7600	1	Mitigated
Ethylene Oxide	34	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	25	600	Ø	2000	0		Mitigated
Ethylene Oxide	34	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	25	90	200	1600	1300		Mitigated
Ethylene Oxide	34	Phosphorus	105	Evaporation of residuals at 100 C	25	90	Ø	1600	øj	,	Mitigated
Ethylene Oxide	34	Sulfur Dioxide	105	Unknown	25	90	4500	?	?		Unknown
Ferric Chloride Solution	104	Acetic Anhydride	107	Evaporation of residuals at 50 C]	Ø	(100	Ø	200		Aggravated
Ferric Chloride Solution	104	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	1	Ø	200	Ø	1300		Aggravated
Ferric Chloride Solution	184	Maleic Anhydride	107	Evaporation of residuals at 50 C	1	Ø	900+	Ø	3000+		Aggravated
Ferric Chloride Solution	104	Phosphorus	105	Local consequences	1	Ø	Ø	Ø	Ø		Same
Flammable Liquid	101	Chlorine	104	Evaporation of residuals at 50 C **	(1	.00	8000	?	7600		Mitigated
Flammable Liquid	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 1	00	200	?	400	!	Aggravated
Formaldehyde Solution	5	Acetic Anhydride	107	Release of acetic acid	1	ØØ <	(100	200	200	300	Aggravated
ormaldehyde Solution	5	Alkaline Liquid	10	Evaporation of residuals at 50 C	1	ØØ	ø	200	øį		Aggravated
ormaldehyde Solution	5	Anhydrous Ammonia	10	Mitigate NH3 release	1	ØØ	4100	200	3700	•	Mitigated
ormaldehyde Solution	Б	Aniline Oil	7	Evaporation of residuals at 100 C	1	00	100	500	700]		Aggravated
ormaldehyde Solution	5	Chlorine	104	Evaporation of residuals at 50 C **	1	00	8000	200	7600		Mitigated
ormaldehyde Solution	5	Ethylene Oxide	34	Evaporation of residuals at 50 C	1	ØØ	2500	200	2000		Mitigated
ormaldehyde Solution	5		104	Evaporation of residuals at 50 C	1	0 0	ø	200	øį		Aggravated
ormaldehyde Solution	5	Hexamethylene Diamine Soln	7	Evaporation of residuals at 100 C	•		100 j	500	(100		Aggravated

ASTM	Summary of Toxic Emission	Conse	quence Calculations			I No	n-mixed	Mix	 ed	 I Distan	nces for Gas	
ASTW Cheaical Name #1 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group Cheaical Name #2 Group G	•		•			•		•		•	•	1
Chesical Name #1		ASTM		ASTM		•	` '	•	٠, ٠	-		1
Formaldehyde Solution 5					Toxic Emission Consequences	•		•		•		Conclusion
Forsaldehyde Solution 5	Formaldehyde Solution	5	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	 1	 00 200		400	 		· Aggravated
Formaldehyde Solution 5	Formaldehyde Solution	5	Potassium Hydroxide, Liq	10	Evaporation of residuals at 50 C	1	99 Ø	200	Ø	l		Aggravated
Formaldehyde Solution 5	Formaldehyde Solution	5	Sodium Hydroxide, Liq.	10	Evaporation of residuals at 50 C	1	00 Ø	200	Ö	l		Aggravated
Forcal dehyde Solution 5 Tolluene Diisocyanate 18 Exporation of residuals at 186 C 186 186 186 7688 Mitigate Fuel Dii 181 Hydrogen Peroxide 184 Exporation at 58 C; explosion ** (186 286 188 486 Magravat Fuel, Aviation 181 Hydrogen Peroxide 184 Exporation at 58 C; explosion ** (186 286 188 486 Magravat Fuel, Aviation 181 Chlorine 184 Exporation at 58 C; explosion ** (186 286 188 486 Magravat Gasoline 181 Hydrogen Peroxide 184 Exporation at 58 C; explosion ** (186 286 188 488 Magravat Gasoline 181 Hydrogen Peroxide 184 Exporation of residuals at 58 C ** (188 286 188 488 Magravat Gasoline, nec 181 Hydrogen Peroxide 184 Exporation of residuals at 58 C ** (188 288 188 488 Magravat Gasoline, nec 181 Hydrogen Peroxide 184 Exporation of residuals at 58 C ** (188 288 188 488 Magravat Gasoline, nec 181 Hydrogen Peroxide 184 Exporation of residuals at 58 C ** (188 288 188 488 Magravat Glycol Ethers 14 Chlorine 184 Exporation of S8 C; explosion ** (188 288 188 488 Magravat Glycol Ethers 14 Chlorine 184 Exporation of residuals at 188 C (188 288 188 488 Magravat Glycol Ethers 14 Chlorine 184 Exporation of residuals at 188 C (188 288 188 488 Magravat Hexasethylene Dianine Sol 7 Acctro Anhydride 187 Exporation of residuals at 188 C (188 288 188 488 Magravat Hexasethylene Dianine Sol 7 Carbon Tetrachloride 187 Exporation of residuals at 188 C (188 288 188 188 Magravat Magravat Hexasethylene Dianine Sol 7 Chloropene 184 Exporation of residuals at 188 C (188 288 188 188 Magravat Magravat Mexasethylene Dianine Sol 7 Hydrogen Peroxide 184 Exporation of residuals at 188 C (188 288 188 188 188 Magravat Mexasethylene Dianine Sol 7 Hydrogen Peroxide 184 Exporation of residuals at 188 C (188 288 188 188 188 188 188 Magravat Me	Formaldehyde Solution	5	Sodium Metal	21	Release of hydrogen gas **	1	00 Ø	200	Ø	l		Aggravated
Fuel Dil 191	Formaldehyde Solution	5	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	1	00 < 100	200	600	Ì		Aggravated
Fuel Aviation 181 Hydrogen Peroxide 184 Evaporation at 58 C; explosion ** < 168 288 186 488	Fuel Dil .	101	Chlorine	104	Evaporation of residuals at 50 C ++	(1	00 8000	100	7600			Mitigated
Fuel, Aviation 181 Chlorine 184 Evaporation of residuals at 58 C ** C 188 8888 188 7888 Mitigate Fuel, Aviation 181 Hydrogen Peroxide 184 Evaporation at 58 C; explosion ** C 188 288 188 488 Mitigate Gasoline 181 Hydrogen Peroxide 184 Evaporation of residuals at 58 C ** C 188 8888 188 7688 Mitigate Gasoline, nec 181 Hydrogen Peroxide 184 Evaporation of residuals at 58 C ** C 188 8888 188 7688 Mitigate Gasoline, nec 181 Hydrogen Peroxide 184 Evaporation of residuals at 58 C ** C 188 8888 188 7688 Mitigate Gasoline, nec 181 Hydrogen Peroxide 184 Evaporation of residuals at 58 C ** C 188 8888 188 7688 Mitigate Gasoline, nec 181 Hydrogen Peroxide 184 Evaporation of residuals at 58 C ** C 188 8888 188 7688 Mitigation Gasoline, nec 184 Hydrogen Peroxide 184 Evaporation of residuals at 58 C ** C 188 8888 188 7688 Mitigation Gasoline, nec 184 Hydrogen Peroxide 184 Evaporation of Residuals at 58 C ** C 188 Research Mitigate Mitigate Gasoline, nec 184 Hydrogen Peroxide 184 Evaporation of Residuals at 188 C C 188 Research Mitigate M	Fuel Oil	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	(1	ØØ 200	100	400	İ		Aggravated
Fuel, Aviation	Fuel, Aviation	161	Chlorine	104	Evaporation of residuals at 50 C ++	< 1	00 8000	100	7600	ĺ		Mitigated
Gasoline	Fuel, Aviation	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	(1	00 20 0	100	400	l		Aggravated
Gasoline, nec 181 Chlorine 184 Evaporation of residuals at 58 C ** (188 8689 188 7688 Mitigate Gasoline, nec 181 Hydrogen Peroxide 184 Evaporation at 58 C c ** (188 8689 188 7688 Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 889 188 7888 Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 889 188 7888 Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 889 188 7888 Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 6889 188 7888 Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 6889 (188 188 1889 Maggravat Maggravat Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 6889 (188 1889 Maggravat Maggravat Maggravat Mitigate methylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 8898 (188 1889 Maggravat Maggravat Maggravat Maggravat Mexamethylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 8898 (188 1898 Maggravat Maggravat Maggravat Mexamethylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C (188 8898 (188 1898 Maggravat Maggra	Gasoline	101	Chlorine	104	Evaporation of residuals at 50 C **	(1	00 8000	100	7600			Mitigated
Gasoline, nec 181 Hydrogen Peroxide 184 Evaporation at 58 C; explosion ** < 108 208 108 466	Gasoline	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	(1	ØØ 20Ø	100	400]		Aggravated
Clycol Ethers	Gasoline, nec	101	Chlorine .	104	Evaporation of residuals at 50 C **	(1	00 8000	100	7600	l		Mitigated
Simple S	Gasoline, nec	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	(1	0 0 200	100	400	l		Aggravated
Hexamethylene Diamine Sol 7 Acetic Anhydride 187 Evaporation of residuals at 188 C	Glycol Ethers	14	Chlorine	104	Mitigation of chlorine release **	(1	00 8000	100	7600			Mitigated
Hexamethylene Diamine Sol 7 Acrylonitrile 26 Evaporation of residuals at 100 C 100 200 1500 900 Aggravat	Glycol Ethers	14	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	(1	00 Ø	100	Ø]		Aggravated
Rexamethylene Diamine Sol	Hexamethylene Diamine Sol	7	Acetic Anhydride	107	Evaporation of residuals at 100 C	(1	00 < 100	< 100	400	!		Aggravated
Hexamethylene Diamine Sol 7 Chlorine 184 Evaporation of residuals at 50 C ** < 100 8000 < 100 7500 Mitigate Hexamethylene Diamine Sol 7 Chloroprene 17 Evaporation of residuals at 100 C < 100 800 < 100 1700 Aggravat Hexamethylene Diamine Sol 7 Ethylene Dxide 34 Evaporation of residuals at 100 C < 100 2500 < 100 1600 Mitigate Hexamethylene Diamine Sol 7 Ferric Chloride Solution 104 Evaporation of residuals at 50 C < 100 2500 < 100 1300 Aggravat Hexamethylene Diamine Sol 7 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** < 100 200 < 100 200 < 100 200 < 100 200 Hexamethylene Diamine Sol 7 Maleic Anhydride 107 Evaporation of residuals at 100 C < 100 200 < 100 200 < 100 200 < 100 200 Hexamethylene Diamine Sol 7 Methyl Chloride 17 Mitigate methyl chloride release < 100 1000 < 100 200 < 100 200 < 100 200 < 100 200 Hexamethylene Diamine Sol 7 Motor Fuel Antiknock Compo 24 Evaporation of residuals at 100 C < 100 1000 < 100 1000 < 100 Hexamethylene Diamine Sol 7 Propylene Oxide 34 Evaporation of residuals at 100 C < 100 100 100 < 100 1100 Hexamethylene Diamine Sol 7 Sodium Metal 21 Evaporation of residuals at 100 C < 100 100 0 < 100 1100 Hexamethylene Diamine Sol 7 Toluene Diisocyanate 18 Evaporation of residuals at 100 C < 100 100 0 < 100 100 100 Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 100 0 < 100 100 100 Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 100 0 < 100 100 100 Hexamethylene Diamine Sol 7 Vinyl Chloride 5 Evaporation of residuals at 100 C < 100 100 0 < 100 100 100 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 100 0 < 100 100 0 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 100 0 < 100 100 0 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 100 0 < 100 0	Hexamethylene Diamine Sol	7	Acrylonitrile	26	Evaporation of residuals at 100 C	(1	00 200	< 100	900			Aggravated
Hexamethylene Diamine Sol	Hexamethylene Diamine Sol	7	Carbon Tetrachloride	17	Evaporation of residuals at 100 C	(1	00 600	< 100	1500	l		Aggravated
Hexamethylene Diamine Sol 7 Ethylene Oxide 34 Evaporation of residuals at 100 C < 100 2500 < 100 1600 Mitigate Hexamethylene Diamine Sol 7 Ferric Chloride Solution 104 Evaporation of residuals at 50 C < 100 0 < 100 0 Same Sa	Hexamethylene Diamine Sol	7	Chlorine ,	104	Evaporation of residuals at 50 C **	(1	00 8000	< 100	7600			Mitigated
Hexamethylene Diamine Sol 7 Ferric Chloride Solution 184 Evaporation of residuals at 50 C	-		Chloroprene	17		(1	00 800	< 100	1700	i		Aggravated
Hexamethylene Diamine Sol 7 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** < 100 200 < 100 1300 Aggravat Hexamethylene Diamine Sol 7 Maleic Anhydride 107 Evaporation of residuals at 100 C < 100 900* < 100 20700* Aggravat Hexamethylene Diamine Sol 7 Methyl Chloride 17 Mitigate methyl chloride release < 100 1000 < 100 1000 < 100 1000 Hexamethylene Diamine Sol 7 Motor Fuel Antiknock Compo 24 Evaporation of residuals at 50 C < 100 100 100 < 100 100 Hexamethylene Diamine Sol 7 Propylene Oxide 34 Evaporation of residuals at 100 C < 100 100 100 100 Hexamethylene Diamine Sol 7 Sodium Metal 21 Evaporation at 100 C; release H2 gas < 100 0 < 100 0 < 100 0 Hexamethylene Diamine Sol 7 Toluene Diisocyanate 18 Evaporation of residuals at 100 C < 100 100 < 100 600 Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 19100* < 100 18100* Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 19100* < 100 18100* Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 < 100 1100 < 100 1100 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 < 100 1100 < 100 1100 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 < 100 1100 < 100 1100 Hitigate Hydrochlo	Hexamethylene Diamine Sol	7	Ethylene Oxide	34	Evaporation of residuals at 100 C	(1	00 2500	< 100	1600	!		Mitigated
Hexamethylene Diamine Sol 7 Maleic Anhydride 107 Evaporation of residuals at 100 C < 100 900* < 100 20700* Aggravat Hexamethylene Diamine Sol 7 Methyl Chloride 17 Mitigate methyl chloride release < 100 1000 < 100 1000 < 100 1000 Hexamethylene Diamine Sol 7 Motor Fuel Antiknock Compo 24 Evaporation of residuals at 50 C < 100 100 < 100 100 < 100 100 Hexamethylene Diamine Sol 7 Propylene Oxide 34 Evaporation of residuals at 100 C < 100 1700 < 100 1100 Hexamethylene Diamine Sol 7 Sodium Metal 21 Evaporation at 100 C; release H2 gas < 100 0 < 100 0 < 100 0 Hexamethylene Diamine Sol 7 Toluene Diisocyanate 18 Evaporation of residuals at 100 C < 100 0 0 < 100 0 0 Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 19100* < 100 18100* Hexamethylene Diamine Sol 7 Vinyl Chloride 104 Evaporation of residuals at 50 C *** 200 8000 300 7600 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Hitigate Hydrochloric Acid 1 Acetaldehyde 1 Acetald			Ferric Chloride Solution	104		•		< 100	Ø			Same
Hexamethylene Diamine Sol 7 Methyl Chloride 17 Mitigate methyl chloride release < 100 1000 < 100 1000 Same Same Hexamethylene Diamine Sol 7 Motor Fuel Antiknock Compo 24 Evaporation of residuals at 50 C < 100 100 < 100 100 < 100 100 Same Same Hexamethylene Diamine Sol 7 Propylene Oxide 34 Evaporation of residuals at 100 C < 100 1700 < 100 1100 Mitigate Hexamethylene Diamine Sol 7 Sodium Metal 21 Evaporation at 100 C; release H2 gas < 100 0 < 100 0 < 100 0 Same Sa	i i		= =	104	,	(1	00 200	< 100	1300	l		Aggravated
Hexamethylene Diamine Sol 7 Motor Fuel Antiknock Compo 24 Evaporation of residuals at 50 C < 100 100 < 100 100 Mitigate Mitigate Hexamethylene Diamine Sol 7 Propylene Oxide 34 Evaporation of residuals at 100 C < 100 100 < 100 100 Mitigate Mitigate Hexamethylene Diamine Sol 7 Sodium Metal 21 Evaporation at 100 C; release H2 gas < 100 0 < 100 0 < 100 0 Same Same Hexamethylene Diamine Sol 7 Toluene Diisocyanate 18 Evaporation of residuals at 100 C < 100 < 100 < 100 600 Mitigate Mitigate Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 19100 < 100 18100 Mitigate Mitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Mitigate	Hexamethylene Diamine Sol	7			-			•	20700*	•		Aggravated
Hexamethylene Diamine Sol 7 Propylene Oxide 34 Evaporation of residuals at 100 C < 100	-		<u> </u>		- · · · · · · · · · · · · · · · · · · ·	•		•				•
Hexamethylene Diamine Sol 7 Sodium Metal 21 Evaporation at 100 C; release H2 gas < 100			•		•			< 100				
Hexamethylene Diamine Sol 7 Toluene Diisocyanate 18 Evaporation of residuals at 100 C < 100 < 100 < 100 600	•		· ·		·	•		< 100	1100			Mitigated
Hexamethylene Diamine Sol 7 Vinyl Chloride 17 Mitigate release of vinyl chloride < 100 19100* < 100 18100* Mitigate release of vinyl chloride Hexame 29 Chlorine 104 Evaporation of residuals at 50 C ** 200 8000 300 7600 300 7600 Mitigate Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700	-		· · · · · · · · · · · · · · · · · · ·					-				-
Hexane 29 Chlorine 104 Evaporation of residuals at 50 C ** 200 8000 300 7600 Mitigated Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Mitigated Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Evaporation of residuals at 100 C < 100 1100 500 700 Evaporation of residuals at 100 C < 100 1100 500 700 Evaporation of residuals at 100 C < 100 1100 500 700 Evaporation of residuals at 100 C < 100 1100 500 700 Evaporation of residuals at 100 C < 100 1100 500 700	=		•		- ·	•						Aggravated
Hydrochloric Acid 1 Acetaldehyde 5 Evaporation of residuals at 100 C < 100 1100 500 700 Mitigated	Hexamethylene Diamine Sol		-	17	· ·	•		(100				Mitigated
		29			· · · · · ·	•		•				Mitigated
11 1 1 1 1 1 1 1 1 1	=	1	<u> </u>		•			500	700			Mitigated
Hydrochloric Acid I Acetic Acid, Glacial 3 Evaporation of residuals at 50 C (100 100 100 300 Aggravato	Hydrochloric Acid	1	Acetic Acid, Glacial	3	Evaporation of residuals at 50 C	< 1	00 100	100	300			Aggravated

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Summary of Toxic Emiss	ion Conse	quence Calculations			-	Non-	nixed	Mix	ed	Distan	ces fo	r Gas	1
		*			D	istan	се (м)	Distan	ce (m)	F	elease	S	1
1	ASTM		ASTM		-	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	1
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	1	#1	#2	#1	#2	Gas 1	Gas 2	Gas 3	[Conclusion:
Hydrochloric Acid	1	Acetic Anhydride	107	Evaporation of residuals at 100 C **		< 100	< 100	500	400				- Aggravated
Hydrochloric Acid	1	Acetone	19	Evaporation of residuals at 50 C	1	(100	100	100	600	I			Aggravated
Hydrochloric Acid	1	Acrylic Acid	3	Evaporation of residuals at 50 C	-	(100	800+	100	1600*	f			Aggravated
Hydrochloric Acid	1	Acrylonitrile	26	Evaporation of residuals at 100 C	1	(100	200	500	900	1			Aggravated
Hydrochloric Acid	1	Alcoholic Beverage	4	Releases ethyl chloride gas	1	< 100	< 100	100	< 100	400			Aggravated
Hydrochloric Acid	1	Alkaline Liquid	10	Evaporation of residual HCL at 100 C	1	< 100	Ø	500	Ø	ŀ			Aggravated
Hydrochloric Acid	1	Anhydrous Ammonia	10	Mitigate NH3 release	1	< 100	4100	100	3700	l			Mitigated
Hydrochloric Acid	1	Aniline Oil	7	Evaporation of residuals at 50 C	i	< 100	100	100	200	ĺ			Aggravated
Hydrochloric Acid	1	Butadiene	28	Evaporation of residual HCL at 50 C	ĺ	< 100	800	100	800	1			Same
Hydrochloric Acid	1	Butyl Acrylate	13	Release acrylic acid and butanol	Ì	< 100	700+	500	3000+	6300+	< 100		Aggravated
Hydrochloric Acid	1	Buty! Alcohol	4	Evaporation of butyl chloride, alcohol	Ī	< 100	< 100	100	< 100	?			Aggravated
Hydrochloric Acid	1	Butyraldehyde	5	Evaporation of residuals at 50 C	-	< 100	< 100	100	(100	I			Aggravated
Hydrochloric Acid	1	Chlorine	104	Evaporation of residual HCL at 100 C	i	(100	8000	500	8000	< 100			Same
Hydrochloric Acid	1	Chloroprene	17	Evaporation of residuals at 100 C **		< 100	800	500	1700	Ì			Aggravated
Hydrochloric Acid	1	Coal Tar Distillate	101	Evaporation of residuals at 50 C	Ì	< 100	< 100	100	< 100	İ			Same
Hydrochloric Acid	1	Combustible Liquid	101	Evaporation of residuals at 50 C	i	< 100	?	100	?	İ			Same
Hydrochloric Acid	1	Compound, Cleaning	101	Evaporation of residuals at 50 C	i	(100	?	100	?	İ			Same
Hydrochloric Acid	1	Crude Oil Petroleum	101	Evaporation of residuals at 50 C	i	⟨ 100	< 100	100	< 100	i I			Same
Hydrochloric Acid	1	Denatured Alcohol	4	Releases ethyl chloride gas	i	⟨ 100	< 100	100	(100	400			Aggravated
Hydrochloric Acid	1	Distillate Fuel Oil	101	Evaporation of residuals at 50 C	i	⟨ 100	(100	1 100	(100	I			Same
Hydrochloric Acid	1	Ethyl Acetate	13	Release acetic acid and ethanol	i	(100	100	100	200	300	(100		Aggravated
Hydrochloric Acid	1	Ethyl Acrylate	13	Release acrylic acid and ethanol	i	(100	100	500	900	6300*	(100		Aggravated
Hydrochloric Acid	1	Ethyl Alcohol	4	Releases ethyl chloride gas	i.	(100	(100	100	(100	400	•		Aggravated
Hydrochloric Acid	1	Ethylene Oxide	34	Mitigate ethylene oxide release	Ė	(100	2500	500	2300	i			Mitigated
Hydrochloric Acid	1	Flammable Liquid	101	Evaporation of residuals at 50 C	-	(100	?	100	?	•			Same
Hydrochloric Acid	1	Formaldehyde Solution	5	Evaporation of residuals at 50 C		(100	100	100	200	I			Aggravated
Hydrochloric Acid	1	Fuel Oil	101	Evaporation of residuals at 50 C	i,	(100	⟨ 100	100	⟨ 100	İ			Same
lydrochloric Acid	1	Fuel, Aviation	101	Evaporation of residuals at 50 C	i.	(100	< 100	100	(100	I			Same
lydrochloric Acid	1	Gasoline	101	Evaporation of residuals at 50 C	i.	(. 100	< 100	100	⟨ 100]			Same
Hydrochloric Acid	1	Gasoline, nec	101	Evaporation of residuals at 50 C	į,	(100	< 100	100	< 100				Same
lydrochloric Acid	1	Glycol Ethers	14	Releases 1,2 dichloroethane	i,	(100	⟨ 100	100	100	300			Aggravated
lydrochloric Acid	1	Hexamethylene Diamine Soln		Evaporation of residuals at 50 C	•	(100	⟨ 100	100	(100				Aggravated
Hydrochloric Acid	1	Hydrogen Peroxide	104	Releases chlorine gas; explosion	•	(100	200	I Ø	1300	,			Aggravated

SUMMARY OF LOXIC EMISS	ion Conse	equence Calculations			•		ixed	Mix		•	ces for Gas	
¥	40711				•			Distan		•	Releases	İ
Cha-:a- N #1	ASTM		ASTM	T	•	DLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	•
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	 .1	#1	#2	#1	#2	Gas 1	Gas 2 Gas 3	Conclusion
Hydrochloric Acid	1	Isopropanol	4	Releases isopropyl chloride gas	(100	< 100	100	100	?		Aggravated
lydrochloric Acid	1	Maleic Anhydride	107	Evaporation of residuals at 100 C	(100	900*	500	20700+	j		Aggravated
Hydrochloric Acid	1	Methyl Alcohol	4	Releases methyl chloride gas	1 <	100	100	100	100	500		Aggravated
łydrochloric Acid	1	Methyl Ethyl Ketone	19	Evaporation of residuals at 50 C	1 (100	200	100	300	1		Aggravate
Hydrochloric Acid	1	Methyl Methacrylate	13	Release methacrylic acid and methanol	1 (100	100	500	700	?	600	Aggravated
Hydrochloric Acid	1	Motor Fuel Antiknock Compo	24	Boil-off HCL, release ethane gas **	1 <	100	100	100	100	ĺ		Same
Hydrochloric Acid	1	Nitric Acid, Fuming	2	Release of chlorine gas	1 <	100	900	100	1800	2700		Aggravated
lydrochloric Acid 🦈	1	Octyl Alcohol	4	Local consequences	(100	< 100	< 100	< 100	1		Same
lydrochloric Acid	1	Oil	101	Evaporation of residuals at 50 C	1	100	< 100	100	< 100	İ		Same
lydrochloric Acid	1	0 leum	2	Releases HCL gas, H2SO4 mist (from SO3)	1 <	100	1600	2500	2400	İ		Aggravated
Hydrochloric Acid	1	Petroleum Naphtha	101	Evaporation of residuals at 50 C	ik	100	< 100	100	< 100	i I		Same
łydrochloric Acid	1	Petroleum Residual Fuel	101	Evaporation of residuals at 50 C	ÌΚ	100	< 100	100	< 100	İ		Same
lydrochloric Acid	1	Petroleum, Refined	101	Evaporation of residuals at 50 C	1	100	< 100	100	< 100	İ		Same
lydrochloric Acid	1	Pheno I	31	Same as individual chemicals	İ	100	< 100	< 100	(100	İ		Same
Hydrochloric Acid	1	Phosphorus ·	105	Releases PH3 gas	14	100	Ø	100	Ø	800		Aggravated
Hydrochloric Acid	1	Potassium Hydroxide, Liq	10	Evaporation of residual HCL at 100 C	1 <	100	Ø	500	Ø	ĺ		Aggravate
Hydrochloric Acid	1	Propylene Oxide	34	Mitigate propylene oxide release	1 <	100	1700	500	1500	1	,	Mitigated
Hydrochloric Acid	1	Pulp Mill Liquid	101	Evaporation of residuals at 50 C	(100	?	100	?	1		Same
Hydrochloric Acid	1	Sodium Hydroxide, Liq.	10	Evaporation of residual HCL at 100 C	1 (100	, Ø	500	Ø	1	•	Aggravated
Hydrochloric Acid	1	Sodium Metal	21	Boil-off HCL, release H2 **	1 <	100	Ø	500	Ø	< 100		Aggravated
Hydrochloric Acid	1	Styrene Monomer	16	Evaporation of residuals at 100 C	1	100	< 100	500	200	l		Aggravated
Hydrochloric Acid	1	Sulfuric Acid	2	Displaces HCL from solution	(100	Ø	2500	Ø	I		Aggravated
Hydrochloric Acid	1	Sulfuric Acid, Spent	2	Displaces HCL from solution	(100	Ø	2500	Ø	1		Aggravated
łydrochloric Acid	1	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	1 <	100	< 100	100	600	1		Aggravated
Hydrochloric Acid	1	Vinyl Acetate	13	Release acetic acid and acetaldehyde	(100	4100+	500	5700+	300	900	Aggravated
lydrochloric Acid	1	Vinyl Chloride	17	Mitigate release of vinyl chloride	<	100	19100*	500	18100+			Aggravated
łydrocyanic Acid	1	Acetaldehyde	5	Mitigate release of HCN	1	88øø	1100	8400	700			Mitigated
lydrocyanic Acid	1	Acetic Anhydride	107	Release acetonitrile **	1	8800	< 100	8400	200	100		Mitigated
lydrocyanic Acid	1	Acetone	19	Mitigate HCN release	(8800	100	8400	600]		Mitigated
lydrocyanic Acid	1	Acrylonitrile	26	Mitigate HCN release	4	8800	200	8400	900			Mitigated
lydrocyanic Acid	1	Alcoholic Beverage	4	Evaporation of propionitrile	į (8800	< 100] 7600	< 100	· ?		Mitigated
lydrocyanic Acid	1	Alkaline Liquid	10	Mitigate HCN release	İ	8800	Ø	7600	Ø			Mitigated
lydrocyanic Acid	1	Ammonium Nitrate Soln.	106	Mitigate HCN release	i i	8800	Ø	8400	Ø			Mitigated

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Summary of Toxic Emiss	ion Conse	equence Calculations			1	Non-m		•	xed	•	ices fo		1
			40714		•	istanc		•	nce (m)	•	Release		1
Chemical Name #1	ASTM Group	Chemical Name #2	ASTM Group	Toxic Emission Consequences	1	IDLH #1	IDLH ∦2	IDL! #1	IDLH #2	IDLH		IDLH	 Conclusions
					 -	#±	#* 						
Hydrocyanic Acid	1	Anhydrous Ammonia	10	Mitigate HCN and NH3 releases		8800	4100	626	0 3400	İ			Mitigated
Hydrocyanic Acid	1	Bu ta diene	28	Mitigate HCN release	1	8800	800	849	Ø 800	1			Mitigated
Hydrocyanic Acid	1	Butyl Acrylate	13	Release acrylonitrile and butanol	1	8800	700+	846	0 1500*	300	< 100		Mitigated
Hydrocyanic Acid	1	Butyl Alcohol	4	Evaporation of pentanenitrile	i	8800	< 100	846	0 < 100] ?			 Mitigated
Hydrocyanic Acid	1	Butyraldehyde	5	Mitigate release of HCN	1	8800	< 100	846	Ø < 100	1			Mitigated
Hydrocyanic Acid	1	Carbon Tetrachloride	17	Mitigate HCN release	1	8800	600	848	Ø 1100	1			Mitigated
Hydrocyanic Acid	1	Chlorine	104	Releases CNCL and HCL gas	1	8800	8000	696	ø ø	9200	4300		Aggravated
Hydrocyanic Acid	1	Chloroprene	17	Mitigate HCN release	1	8800	800	846	0 2100	J			Mitigated
Hydrocyanic Acid	1	Denatured Alcohol	4	Evaporation of propionitrile		8800	〈 100	846	0 < 100	1 ?			Mitigated
Hydrocyanic Acid	1	Ethyl Acetate	13	Release acetonitrile and ethanol	1	8800	100] 848	Ø 200	100	< 100		Mitigated
Hydrocyanic Acid	1	Ethyl Acrylate	13	Release acryonitrile and ethanol	١	8800	100	848	0 300	300	< 100		Mitigated
Hydrocyanic Acid	1	Ethyl Alcohol	4	Evaporation of propionitrile	ı	8800	< 100	846	Ø < 100	?			Mitigated
Hydrocyanic Acid	1	Ethylene Oxide	34	Mitigate ethylene oxide and HCN releases	1	8800	2500	760	Ø 23ØØ	Ī			Mitigated
Hydrocyanic Acid	1	Ferric Chloride Solution	104	Mitigate release of HCN	ı	8800	Ø	769	ø ø	ì			Mitigated
Hydrocyanic Acid	1	Formaldehyde Solution	5	Mitigate release of HCN	l	8800	100	768	Ø 500	ĺ			Mitigated
Hydrocyanic Acid	1	Glycol Ethers	14	Mitigate HCN release	ı	8800	< 100	842	Ø 100]			Mitigated
Hydrocyanic Acid	1	Hydrogen Peroxide	104	Releases NOx gas; explosion	l	8800	200	620	Ø 1300	3700			Mitigated
Hydrocyanic Acid	1	Isopropanol	4	Evaporation of isopropyl cyanide	ļ	8800	< 100	840	0 100	?			Mitigated
Hydrocyanic Acid	1	Maleic Anhydride	107	Mitigate HCN release	ı	8800	900+	840	0 3000+	1			Mitigated
Hydrocyanic Acid	1	Methyl Alcohol	4	Evaporation of acetonitrile	1	8800	100	840	0 100	800			Mitigated
Hydrocyanic Acid	1	Methyl Chloride	17	Mitigate HCN, CH3CL; form CH3CN and HCL	1	8800	1000	760	0 800	1100	3800		Mitigated
Hydrocyanic Acid	1	Methyl Ethyl Ketone	19	Mitigate HCN release	ĺ	8800	200	840	0 300	i			Mitigated
Hydrocyanic Acid	1	Methyl Methacrylate	13	Release methacrylonitrile and methanol	ĺ	8800	100	840	0 200	?	< 100		Mitigated
Hydrocyanic Acid	1	Motor Fuel Antiknock Compo	24	Mitigate HCN release **	ĺ	8800	100	840	Ø 100	< 100			Mitigated
Hydrocyanic Acid	. 1	Nitric Acid, Fuming	2	Mitigate release of HCN	ĺ	8800	900	620	Ø 1600	Ì			Mitigated
Hydrocyanic Acid	1	Octyl Alcohol	4	Mitigate release of HCN	ı	8800	< 100	840	0 < 100	ĺ			Mitigated
Hydrocyanic Acid	1	Oleum	2	Releases NO2, CO, and SO2 gas	ĺ	8800	1600	860	ð Ø	2400	900	2600	Mitigated
Hydrocyanic Acid	1	Potassium Hydroxide, Liq	10	Mitigate HCN release	ı	8800	Ø	760	3 Ø	1			Mitigated
Hydrocyanic Acid	1	Propylene Oxide	34	Mitigate propylene oxide and HCN release	ĺ	8800	1700	760	1500	Ì			Mitigated
Hydrocyanic Acid	1	Sodium Hydroxide, Liq.		Mitigate HCN release		8800	Ø	76Ø	ð Ø	İ			Mitigated
Hydrocyanic Acid	1	Sodium Metal		Mitigate HCN release **	1	8800	Ø	840	9 9	< 100			Mitigated
Hydrocyanic Acid	1	Styrene Monomer		Evaporation of residuals at 100 C	1	8800	< 100	840	200				Mitigated

ASTN	Summary of Toxic Emission	n Conse	equence Calculations		·	Non-	ixed	 Mixe	 ed	 Distan	ces for Gas	1
ASTM	,		•			•		•	*	•		i
Cheaical Name #1 Crow Cheaical Name #2 Group Toxic Eaission Consequences #1 #2 #2 #2 Gas 1 Gas 2 Gas 3 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Concentration Cheaical Name #2 Cheaical Name #2 Cheaical Name *2		ASTM		ASTM				-		•		1
Hydrocyanic Acid	Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1	#2	•		:		Conclusion
Hydrocyanic Acid 1 Vinyl Actate 13 Release acctonitrile and acetaldehyde 8688 4188 8488 8088 188 988 1	Hydrocyanic Acid	1	Sulfuric Acid, Spent	2	Releases CO gas	- 8800	Ø	 7600	Ø	1100		- Mitigated
	Hydrocyanic Acid	1	Toluene Diisocyanate	18	Mitigate HCN release	8800	< 100	8400	600	i		Mitigated
	Hydrocyanic Acid	1	Vinyl Acetate	13	Release acetonitrile and acetaldehyde	8800	4100+	8400	8000*	100	900	Mitigated
Hydrofluorosilicic Acid 1	Hydrocyanic Acid	1	Vinyl Chloride	17	Mitigate HCN, vinyl chloride releases	8800	19100+	7600	16500+	ĺ		Mitigated
Hydrofluorosilicic Acid 1 Acetic Acid, Glacial 3 Evaporation of residuals at 50 C (100 1	Hydrofluorosilicic Acid	1	Acetaldehyde	5	Evaporation of residuals at 100 C	< 100	1100	800	700	İ		Mitigated
Hydrofluorosilicic Acid 1 Acetic Anhydride 197 Release acetyl fluoride +* (188 688 488 ? Aggr	Hydrofluorosilicic Acid	1	Acetic Acid, Glacial	3	Evaporation of residuals at 50 C	(100	100	100	300	}		Aggravated
Hydrofluorosilicic Acid 1 Acetone 19 Evaporation of residuals at 50 C C 100 100	Hydrofluorosilicic Acid	1	Acetic Anhydride	107	Release acetyl fluoride **	(100	< 100	800	400	?		Aggravated
Hydrofluorosilicic Acid 1	Hydrofluorosilicic Acid	1	Acetone	19	Evaporation of residuals at 50 C	< 100	100	100	600	j		Aggravated
Hydrofluorosilicic Acid 1 Acrylonitrile 26 Evaporation of residuals at 188 C < 188 288 988	Hydrofluorosilicic Acid	1	Acrylic Acid	3	Evaporation of residuals at 50 C	< 100	800+	100	1600+	Ì		Aggravated
Hydrofluorosilicic Acid 1	Hydrofluorosilicic Acid	1	Acrylonitrile	26	Evaporation of residuals at 100 C	< 100	200	800	900	-		Aggravated
Nydrofluorosilicic Acid 1	Hydrofluorosilicic Acid	1	Alcoholic Beverage	4	Releases ethyl fluoride gas	< 100	< 100	1 100	< 100] ?		Aggravated
Hydrofluorosilicic Acid 1	Hydrofluorosilicic Acid	1	Alkaline Liquid	10	Evaporation of residuals at 100 C	< 100	Ø	800	Ø	1		Aggravated
Hydrofluorosilicic Acid 1	Hydrofluorosilicic Acid	1	Ammonium Nitrate Soln.	106	Evaporation of residuals at 50 C	< 100	Ø	1 100	Ø	1		Aggravated
Hydrofluorosilicic Acid 1 Butaliene 28 Evaporation of residual 2HF-SiF4 at 50 C 100 800 100 800 130 800 800 130 800	Hydrofluorosilicic Acid	1	Anhydrous Ammonia	10	Mitigate NH3 release	< 100	4100] 100	3700	< 100		Mitigated
Hydrofluorosilicic Acid 1 Butyl Acrylate 13 Evaporation of residuals at 100 C < 100 700+ 800 3000+ Aggr Hydrofluorosilicic Acid 1 Butyl Alcohol 4 Evaporate butyl fluoride < 100 < 100 100 100 100	Hydrofluorosilicic Acid	1		7	-	-	100	100	200	l		Aggravated
Hydrofluorosilicic Acid 1 Butyl Alcohol 4 Evaporate butyl fluoride	•	1		28		-		100	800	1		Same
Hydrofluorosilicic Acid 1 Butyraldehyde 5 Evaporation of residuals at 188 C < 188 < 189 < 189 Aggr Hydrofluorosilicic Acid 1 Carbon Tetrachloride 17 Releases CL3CF gas < 188 688 188 < 189 < 189 Aggr Hydrofluorosilicic Acid 1 Chloroprene 17 Evaporation of residuals at 188 C < 188 888 888 1788 Aggr Hydrofluorosilicic Acid 1 Coal Tar Distillate 181 Evaporation of residuals at 188 C < 188 & 688 888 1788 Aggr Hydrofluorosilicic Acid 1 Combustible Liquid 181 Evaporation of residuals at 58 C < 188 & 7 < 188 & 7 < 188 & 7 < 188 & 7 < 188 & 7 < 188 & 7 < 188 & 7 < 188 & 7	-	1	-	13	•	•		800	3000+	1		Aggravated
Hydrofluorosilicic Acid 1	-	1	-	4	•	< 100	< 100	100	(100] ?		Aggravated
Hydrofluorosilicic Acid 1 Chloroprene 17 Evaporation of residuals at 100 C < 100 800 1700 1700 1800 1800 1700 1800 1800 1700 1800 1800 1700 1800 1800 1700 1800 1800 1700 18	•	1	•	5	•	•	•	•	-	i		Aggravated
Hydrofluorosilicic Acid 1 Coal Tar Distillate 101 Evaporation of residuals at 50 C < 100 < 100 < 100		1		17	-			•		•		Aggravated
Hydrofluorosilicic Acid 1 Combustible Liquid 101 Evaporation of residuals at 50 C < 100 ? < 100 ? Same Same Hydrofluorosilicic Acid 1 Compound, Cleaning 101 Evaporation of residuals at 50 C < 100 ? < 100 ? < 100 Same Sam	•	1	•			=		•	1700	1		Aggravated
Hydrofluorosilicic Acid 1 Compound, Cleaning 101 Evaporation of residuals at 50 C < 100 ? < 100 ? Same Same Hydrofluorosilicic Acid 1 Crude Oil Petroleum 101 Evaporation of residuals at 50 C < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 100 < 1	="	1			·			•	•			Same
Hydrofluorosilicic Acid 1 Crude Oil Petroleum 101 Evaporation of residuals at 50 C < 100 < 100 < 100 < 100	*	1	•		•	•		•		•		Same
Hydrofluorosilicic Acid 1 Denatured Alcohol 4 Releases ethyl fluoride gas < 100 < 100 100 < 100 ? Aggr Hydrofluorosilicic Acid 1 Distillate Fuel Oil 101 Evaporation of residuals at 50 C < 100 < 100 < 100 < 100 Same Hydrofluorosilicic Acid 1 Ethyl Acetate 13 Evaporation of residuals at 50 C < 100 100 100 200 Aggr Hydrofluorosilicic Acid 1 Ethyl Acrylate 13 Evaporation of residuals at 100 C < 100 100 800 900 Aggr Hydrofluorosilicic Acid 1 Ethyl Alcohol 4 Releases ethyl fluoride gas < 100 < 100 100 < 100 ? Aggr Hydrofluorosilicic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release < 100 2500 800 2300 Mitigate ethylene oxide release < 100 ? < 100 ? Same	•	1	• •		•	•		•		1		Same
Hydrofluorosilicic Acid 1 Distillate Fuel Dil 101 Evaporation of residuals at 50 C < 100 < 100 < 100 < 100 Same	9	1			•			-	•			Same
Hydrofluorosilicic Acid 1 Ethyl Acetate 13 Evaporation of residuals at 50 C < 100 100 100 200 Aggr Hydrofluorosilicic Acid 1 Ethyl Acrylate 13 Evaporation of residuals at 100 C < 100 100 800 900 Aggr Hydrofluorosilicic Acid 1 Ethyl Alcohol 4 Releases ethyl fluoride gas < 100 < 100 100 < 100 ? Aggr Hydrofluorosilicic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release < 100 2500 800 2300 Mitigate ethylene oxide release < 100 ? < 100 ? < 100 ?		1			•	-		-		? ·		Aggravated
Hydrofluorosilicic Acid 1 Ethyl Acrylate 13 Evaporation of residuals at 100 C < 100	•	1			•			•		!		Same
Hydrofluorosilicic Acid 1 Ethyl Alcohol 4 Releases ethyl fluoride gas < 100 < 100 100 < 100 ? Aggr Hydrofluorosilicic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release < 100 2500 800 2300 Mitigate ethylene oxide release < 100 ? < 100 ? Same	•	1	-		•			•		!		Aggravated
Hydrofluorosilicic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release < 100 2500 800 2300 Mitigate ethylene oxide release < 100 2500 800 2300 Same	•	1	= =	13		-		•				Aggravated
Hydrofluorosilicic Acid 1 Flammable Liquid 101 Evaporation of residuals at 50 C < 100 ? < 100 ?	•	1	*	4				•] 		Aggravated
	•	1	=		_ ·	-		•		!		Mitigated
Number transport to the series of the series	•	1				•		=				•
undictinotositicic vein I commune and an an an an an an an an an an an an an	Hydrofluorosilicic Acid	1	Formaldehyde Solution	. 5	Evaporation of residuals at 50 C	< 100	100	100	200	l		Aggravated
					125							
					•							

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Summary of Toxic Emission	n Conse	equence Calculations			No	n-mix	ed	Mix	ed	Distar	ces f	or Gas	İ
					Dist	ance	(m)	Distan	ce (m)	F	Releas	es	İ
l .	ASTM		ASTM		IDL	H I	DLH	IDLH	IDLH	IDLH	IDLH	IDLH	İ
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1		#2	#1	#2	Gas 1	Gas	2 Gas 3	Conclusion:
Hydrofluorosilicic Acid	1	Fuel Oil	101	Evaporation of residuals at 50 C	< 1	 00 (100	< 100	⟨ 100				- (
Hydrofluorosilicic Acid	, 1	Fuel, Aviation	101	Evaporation of residuals at 50 C	(1	ØØ (100	< 100	< 100	1			Same
Hydrofluorosilicic Acid	1	Gasoline	101	Evaporation of residuals at 50 C	(1	00 (100	< 100	(100	1			Same
Hydrofluorosilicic Acid	1	Gasoline, nec	101	Evaporation of residuals at 50 C	(1	00 (100	< 100	< 100	1			Same
Hydrofluorosilicic Acid	1	Glycol Ethers	14	Releases 1,2 difluoroethane	(1	00 (100	100	100	?			Aggravated
Hydrofluorosilicic Acid	1	Hexamethylene Diamine Soln	7	Evaporation of residuals at 50 C	(1	00 (100	100	< 100	1			Aggravated
Hydrofluorosilicic Acid	1	Hydrogen Peroxide	104	Releases HF from solution	(1	00	200	100	400	< 100			Aggravated
Hydrofluorosilicic Acid	1	Isopropanol	4	Evaporate isopropyl fluoride	(1	00 (100	100	100	?			Aggravated
Hydrofluorosilicic Acid	1	Maleic Anhydride	107	Evaporation of residuals at 100 C	(1	00	900*	800	20700+	1			Aggravated
Hydrofluorosilicic Acid	1	Methyl Alcohol	4	Releases methyl fluoride gas	1 (1	00	100	100	100	?			Same
Hydrofluorosilicic Acid	1	Methyl Chloride	17	Releases HCL, methyl fluoride gases	< 1	ØØ	1000	100	1000	2200		?	Aggravated
Hydrofluorosilicic Acid	1	Methyl Ethyl Ketone	19	Evaporation of residuals at 50 C	(1	00	200	100	300	i			Aggravated
Hydrofluorosilicic Acid	1	Methyl Methacrylate	13	Evaporation of residuals at 100 C	(1	ØØ	100	800	700	i			Aggravated
Hydrofluorosilicic Acid	1	Motor Fuel Antiknock Compo	24	Boil-off HF, release ethane gas **	(1	00	100	100	100	< 100			Same
Hydrofluorosilicic Acid	1	Nitric Acid, Fuming	2	Displaces HF from solution	< 1	00	900	3400	2800	i			Aggravated
Hydrofluorosilicic Acid	1	Octyl Alcohol	4	Local consequences	1 (1		100	< 100	(100	İ			Same
Hydrofluorosilicic Acid	1	Oil	101	Evaporation of residuals at 50 C	< 1	00 (100	< 100	< 100	i			Same
Hydrofluorosilicic Acid	1	Oleum	2	Releases HF gas, H2SO4 mist (from SO3)	1 (1		1600	3400	2400	i			Aggravated
Hydrofluorosilicic Acid	1	Petroleum Naphtha	101	Evaporation of residuals at 50 C	(1	8 8 (100	< 100	< 100	i			Same
Hydrofluorosilicic Acid	1	Petroleum Residual Fuel	101	Evaporation of residuals at 50 C	1 (1	00 (< 100	< 100	i			Same
Hydrofluorosilicic Acid	1	Petroleum, Refined	101	Evaporation of residuals at 50 C	1 (1	00 (100	< 100	(100	i			Same
Hydrofluorosilicic Acid	1	Pheno!	31	Evaporation of residuals at 50 C	(1		100	100	200	•			Aggravated
Hydrofluorosilicic Acid	1	Phosphorus	105	Evaporation of residuals at 50 C	1 (1	8Ø	ø	100	Ø	•			Aggravated
Hydrofluorositicic Acid	1	Potassium Hydroxide, Liq	10	Evaporation of residuals at 100 C	(1	80	ø	•	ø	i			Aggravated
Hydrofluorosilicic Acid	1	Propylene Oxide	34	Mitigate propylene oxide release	(1		1700	1 800	1500	•			Mitigated
Hydrofluorosilicic Acid	1	Pulp Mill Liquid	101	Evaporation of residuals at 50 C	(1		?		?	•			Same
Hydrofluorosilicic Acid	1	Sodium Hydroxide, Liq.	10	Evaporation of residuals at 100 C	1 (1		ø	. '	Ø	•			Aggravated
Hydrofluorosilicic Acid	1	Sodium Metal	21	Boil-off HF, release H2 **	1 (1		Ø	•		< 100			Aggravated
Hydrofluorosilicic Acid	1	Styrene Monomer	16	Evaporation of residuals at 100 C	1 (1		100	800	200			-	Aggravated
Hydrofluorosilicic Acid	1	Sulfuric Acid	2	Displaces HF from solution	(1	•	Ø		Ø	•			Aggravated
Hydrofluorosilicic Acid	1	Sulfuric Acid, Spent	2	None	(1			< 100	Ø	•			Same
Hydrofluorosilicic Açid	1	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	(1		100		600	I			Aggravated
Hydrofluorosilicic Acid	1	Vinyl Acetate	13	Evaporation of residuals at 100 C	\ 1	-		•	5700+	' 			Aggravated

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ASTM		ASTM		•		• •			•		1
Group	Chemical Name #2	Group	Toxic Emission Consequences		#1	#2	#1 #1	#2	•		•
1	Vinvl Chloride	17	Mitigate release of vinvl chloride	- 	 < 100	19100*		 7 18100±			- Mitigated
	•		•	' 	-		•		•		Mitigated
	•			1			•		•		Mitigated
	•		-	1			•		•		Mitigated
	-			l I		•	•		•		Mitigated
-			-	I I			•		•		Mitigated
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	_			1			•	-			Mitigated
	•		-	•			•		•		Mitigated
			_	i			•		•		Mitigated
	•		-	ı			•		•		Mitigated
1	Butadiene		_	ŀ			•		•		Mitigated
1			•	İ			•		•		Mitigated
1	•		-	i	8500		•		•	•	Mitigated
1	Butyraldehyde			i	8500	•	•	•	•		Mitigated
_	Carbon Tetrachloride		Releases CL3CF and HCl gas	i	8500	-		•	•	3100	Mitigated
1	Chloroprene		Mitigate HF release	i	8500		•		•		Mitigated
1	Coal Tar Distillate		_	i	8500	⟨ 100	8100		•		Mitigated
	Combustible Liquid		Mitigate HF release	i	8500	=	•	-	•		Mitigated
	Compound, Cleaning		Mitigate HF release	i	8500	?	•		•	*	Mitigated
	Crude Dil Petroleum		-	i	8500		:		•		Mitigated
1	Denatured Alcohol		Mitigate HF release; release C2H5F gas	i	8500	-	•	-			Mitigated
1	Distillate Fuel Oil		Mitigate HF release	•	8500	•	•	•	•		Mitigated
1	Ethyl Acetate		-	i	8500		•	-	•		Mitigated
	-			•			•		•		Mitigated
	•		_	-	8500						Mitigated
	· · · · · · · · · · · · · · · · · · ·			•	8500		٠	-	=		Mitigated
	-			•	8500				•		Mitigated
	· ·		-	•	8500		•		-		Mitigated
			- .	•	8500		-		-		Mitigated
			_	i	8500	-	•	-	•		Mitigated
	Gasoline		Mitigate HF release	i	8500		•	\ \ 100	•		Mitigated
	ASTM Group 1	Group Chemical Name #2	ASTM Group Chemical Name #2 Group 1 Vinyl Chloride 17 1 Acetaldehyde 5 1 Acetic Acid, Glacial 3 1 Acetic Anhydride 107 1 Acetone 19 1 Acrylic Acid 3 1 Acrylic Acid 3 1 Acrylonitrile 28 1 Alcoholic Beverage 4 1 Alkaline Liquid 10 1 Ammonium Nitrate Soln. 106 1 Anhydrous Ammonia 10 1 Aniline Oil 7 1 Butadiene 28 1 Butyl Acrylate 13 1 Butyl Alcohol 4 1 Butyraldehyde 5 1 Carbon Tetrachloride 17 1 Chloroprene 17 1 Coal Tar Distillate 101 1 Compound, Cleaning 101 1 Crude Oil Petroleum 101 1 Denatured Alcohol 4 1 Distillate Fuel Oil 101 1 Ethyl Acetate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acetate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acetate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Acrylate 13 1 Ethyl Alcohol 4 1 Flammable Liquid 101 1 Formaldehyde Solution 5 1 Fuel Oil 101 1 Fuel, Aviation 101	ASTM Group Chemical Name #2 Group Toxic Emission Consequences 1 Vinyl Chloride 1 Acetaldehyde 5 Mitigate release of vinyl chloride 1 Acetic Acid, Glacial 1 Acetic Anhydride 1 Acetone 1 9 Mitigate HF release 1 Acrylic Acid 1 Acrylonitrile 2 8 Mitigate HF release 1 Alcoholic Beverage 1 Alkaline Liquid 1 Amnonium Nitrate Soln. 1 Anhydrous Amnonia 1 Mitigate HF release 1 Butadiene 2 Mitigate HF release 1 Butyl Acrylate 1 Butyl Alcohol 1 Butyraldehyde 1 Carbon Tetrachloride 1 Carbon Tetrachloride 1 Compound, Cleaning 1 Crude 0il Petroleum 1 Denatured Alcohol 1 Distillate Fuel 0il 1 Ethyl Acetate 1 Ethyl Accylate 1 Butigate HF release 1 Bitigate HF release 1 Butigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Coal Tar Distillate 1 Compound, Cleaning 1 Mitigate HF release 1 Denatured Alcohol 2 Mitigate HF release 1 Denatured Alcohol 3 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Accohol 2 Mitigate HF release 1 Ethyl Alcohol 3 Mitigate HF release 1 Ethyl Alcohol 4 Mitigate HF release 1 Ethyl Alcohol 5 Mitigate HF release 1 Ethyl Alcohol 5 Mitigate HF release 1 Ethyl Alcohol 6 Mitigate HF release 1 Ethyl Alcohol 7 Mitigate HF release 1 Ethyl Alcohol 8 Mitigate HF release 1 Ethyl Alcohol 9 Mitigate HF release 1 Ethyl Alcohol 1 Mitigate HF release 1 Ethyl Alcohol 2 Mitigate HF release 3 Mitigate HF release 4 Mitigate HF release 5 Mitigate HF release 6 HF 7 Mitigate HF release 7 Mitigate HF release 8 Mitigate HF release 9 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Miti	ASTM Group Chemical Name #2 Group Toxic Emission Consequences 1 Vinyl Chloride 17 Mitigate release of vinyl chloride 1 Acetaldehyde 5 Mitigate release of HF 1 Acetic Acid, Glacial 3 Mitigate HF release 1 Acetic Anhydride 107 Release acetyl fluoride ** 1 Acetone 19 Mitigate HF release 1 Acrylonitrile 26 Mitigate HF release 1 Alcoholic Beverage 4 Mitigate HF release 1 Almonium Nitrate Soln. 106 Mitigate HF release 1 Anhydrous Ammonia 10 Mitigate HF release 1 Anhydrous Ammonia 10 Mitigate HF release 1 Anhydrous Ammonia 10 Mitigate HF release 1 Coal Tar Distillate 101 Mitigate HF release 1 Combustible Liquid 101 Mitigate HF release 1 Compound, Cleaning 101 Mitigate HF release 1 Compound, Cleaning 101 Mitigate HF release 1 Compound, Cleaning 101 Mitigate HF release 1 Distillate Fuel 0il 101 Mitigate HF release 1 Distillate Fuel 0il 101 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Ethyl Acetate 1 Mitigate HF release 1 Formaldehyde Solution 5 Mitigate ethylene oxide and HF releases 1 Formaldehyde Solution 5 Mitigate HF release 1 Mitigate HF release 1 Formaldehyde Solution 5 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF release 1 Mitigate HF rele	ASTM Group Chemical Name #2 Group Chemical Name #2 Group Toxic Emission Consequences #1 1 Vinyl Chloride 17 Mitigate release of vinyl chloride 1 Acetaldehyde 5 Mitigate release of HF 8566 1 Acetic Acid, Glacial 3 Mitigate HF release 8566 1 Acetic Acid, Glacial 3 Mitigate HF release 8566 1 Acetone 19 Mitigate HF release 8566 1 Acrylic Acid 3 Mitigate HF release 8566 1 Acrylic Acid 3 Mitigate HF release 8566 1 Acrylic Acid 3 Mitigate HF release 8566 1 Acrylonitrile 26 Mitigate HF release 8566 1 Alcoholic Beverage 4 Mitigate HF release 8566 1 Alkaline Liquid 10 Mitigate HF release 8566 1 Annonium Nitrate Soln 1 Mitigate HF release 1 Anhydrous Aamonia 10 Mitigate HF release 1 Anhydrous Aamonia 10 Mitigate HF release 1 Butyal Acrylate 1 Mitigate HF release 1 Butyal Acrylate 1 Mitigate HF release 1 Butyal Alcohol 4 Mitigate HF release 1 Butyal Alcohol 4 Mitigate HF release 1 Butyal Alcohol 5 Mitigate HF release 1 Butyal Alcohol 6 Mitigate HF release 1 Butyal Alcohol 7 Mitigate HF release 1 Butyal Alcohol 8 Mitigate HF release 1 Carbon Tetrachloride 1 Releases CL3CF and HCl gas 1 Compound, Cleaning 1 Mitigate HF release 1 Compound, Cleaning 1 Mitigate HF release 1 Compound, Cleaning 1 Mitigate HF release 1 Compound, Cleaning 1 Mitigate HF release 1 Compound, Cleaning 1 Mitigate HF release 1 Denatured Alcohol 4 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF release 1 Ethyl Acrylate 1 Mitigate HF releas	ASTM	ASTM	ASTM	ASTM Group Chemical Name #2 Group Toxic Emission Consequences #1 #2 #1 #2 Gas 1 1 Vinyl Chloride 17 Mitigate release of vinyl chloride (189 19100) 180 18180 788 1 Acetic Acid, Glacial 3 Mitigate HF release 8500 1100 8100 788 1 Acetic Acid, Glacial 3 Mitigate HF release 8500 1100 8100 788 1 Acetic Anhydride 167 Release acetyl fluoride ** 8500 100 8100 800 78 1 Acetone 19 Mitigate HF release 8500 100 8100 800 70 1 Acetone 19 Mitigate HF release 8500 100 8100 600 70 1 Acrylic Acid 3 Mitigate HF release 8500 100 8100 600 70 1 Acrylic Acid 3 Mitigate HF release 8500 100 8100 600 70 1 Acrylic Acid 3 Mitigate HF release 8500 100 8100 600 70 1 Acrylic Acid 3 Mitigate HF release 8500 200 8100 1000 70 1 Alcoholic Beverage 4 Mitigate HF release 8500 200 8100 600 70 1 Alkaline Liquid 18 Mitigate HF release 8500 0 100 70 70 1 Annonium Nitrate Soin 166 Mitigate HF release 8500 0 100 70 70 1 Anniline Oil 7 Mitigate HF release 8500 800 8100 800 800 1 Butyl Acrylate 13 Mitigate HF release 8500 180 8100 200 800 1 Butyraldehyde 5 Mitigate HF release 8500 180 8100 200 8	ASTW Chemical Name #2 Group Toxic Emission Consequences #1 #2 #2 Gas 1 Gas 2 Gas 3

Summary of Toxic Emiss	ion Conse	equence Calculations			•	mixed	Mix		•	ces for Gas	1
					Distan	ce (m)	Distan	ce (m)] F	Releases	1
	ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	•
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Enission Consequences	#1	#2	#1	#2	Gas 1	Gas 2 Gas 3	Conclusions
Hydrogen Fluoride	1	Gasoline, nec	101	Mitigațe HF release	8500	< 100	8100	(100			Mitigated
Hydrogen Fluoride	1	Glycol Ethers	14	Mitigate HF release	8500	< 100	8100	100	1		Mitigated
Hydrogen Fluoride	1	Hexamethylene Diamine Soln	ı 7	Mitigate HF release	8500	< 100	8100	< 100	1		Mitigated
Hydrogen Fluoride	1	Hydrogen Peroxide	104	Releases fluorine gas	8500	200	7400	400	2400		Mitigated
Hydrogen Fluoride	1 -	Isopropanol	4	Mitigate HF release; release C3H7F gas	8500	< 100	8100	100	?		Mitigated
Hydrogen Fluoride	1	Maleic Anhydride	107	Mitigate HF release	8500	900*	8100	3000+	1		Mitigated
Hydrogen Fluoride	1	Methyl Alcohol	4	Mitigate HF release; release CH3F gas	8500	100	8100	100] ?		Mitigated
Hydrogen Fluoride	1	Methyl Chloride	17	Releases HCL, methyl fluoride gases	8500	1000	7400	900	3000	?	Mitigated
Hydrogen Fluoride	1	Methyl Ethyl Ketone	19	Mitigate HF release	8500	200	8100	300	ľ		Mitigated
Hydrogen Fluoride	1	Methyl Methacrylate	13	Mitigate HF release	8500	100	8100	200	Ì		Mitigated
Hydrogen Fluoride	1	Motor Fuel Antiknock Compo	24	Mitigate HF release **	8500	100	8100	100	Ì		Mitigated
Hydrogen Fluoride	1	Octyl Alcohol	4	Mitigate HF release	8500	< 100	8100	< 100	< 100		Mitigated
Hydrogen Fluoride	1	Oil	101	Mitigate HF release	8500	< 100	8100	< 100	Ì		Mitigated
Hydrogen Fluoride	1	Petroleum Naphtha	161	Mitigate HF release	8500	< 100	8100	< 100	i ,		Mitigated
Hydrogen Fluoride	1	Petroleum Residual Fuel	101	Mitigate HF release	8500	(100	8100	⟨ 100	i		Mitigated
Hydrogen Fluoride	1	Petroleum, Refined	101	Mitigate HF release	8500	< 100	8100	(100	Í		Mitigated
Hydrogen Fluoride	1	Pheno I	31	Mitigate HF release	8500	(100	8100	200	i		Mitigated
Hydrogen Fluoride	1	Phosphorus	105	Releases PF3 and PH3 gas.	8500	ø	8100	ø	16700+	1700	Mitigated
Hydrogen Fluoride	1	Potassium Hydroxide, Liq	10	Mitigate HF release	8500	Ø	7400	ø	i		Mitigated
Hydrogen Fluoride	1	Propylene Oxide	34	Mitigate propylene oxide and HF releases	8500	1700	7400	1500	i		Mitigated
Hydrogen Fluoride	. 1	Pulp Mill Liquid	101	Mitigate HF release	8500	?	8100	?	i		Mitigated
Hydrogen Fluoride	1	Sodium Hydroxide, Liq.	10	Mitigate HF release	8500	ø	7400	ø	i		Mitigated
Hydrogen Fluoride	1	Sodium Metal	21	Mitigate HF release **	8500	Ø	8100	Ø	< 100		Mitigated
Hydrogen Fluoride	1	Styrene Monomer	16	Evaporation of residuals at 100 C	8500	< 100	8100	200	-		Mitigated
Hydrogen Fluoride	1	Sulfuric Acid, Spent	2	Dissolution of some HF	8500	Ø	8100	Ø	i		Mitigated
Hydrogen Fluoride	1	Toluene Diisocyanate	18	Mitigate HF release	8500	< 100	8100	600	i		Mitigated
Hydrogen Fluoride	1	Vinyl Acetate	13	Mitigate HF release	8500	4100+	8100	8000+	i		Mitigated
Hydrogen Fluoride	1	Vinyl Chloride	17	Mitigate HF, vinyl chloride releases	8500	19100+	•	16500+	i I		Mitigated
Hydrogen Peroxide	104	Acetic Anhydride	107	Evaporation at 100 C; explosion **	200	< 100	1300	400	•		Aggravated
Hydrogen Peroxide	104	Maleic Anhydride	107	Evaporation at 100 C; explosion **	200	900+	1300	20700+	•		Ággravated
Hydrogen Peroxide	104	Phosphorus	105	Evaporation at 50 C; explosion **	200	Ø	•	Ø	•		Aggravated
Hydroiodic Acid	1	Acetaldehyde	5	Evaporation of residuals at 100 C	ø	1100		700			Mitigated
Hydroiodic Acid	1	Acetic Acid, Glacial	3	Evaporation of residuals at 50 C	I Ø	100	•	300	•		Aggravated

Hydroiodic Acid							4			•		
ASTM	Summary of Toxic Emiss	ion Conse	equence Calculations			Non-m	ixed	Wixe	 ed	Dista	nces for Gas	1
Chemical Name #1 Group Chemical Name #2 Group Toxic Emission Consequences #1 #2 Gas 1 Gas 2 Gas 3 Conclusive Hydroiodic Acid 1 Acetic Anhydride 187 Evaporation of residuals at 188 C 8 8 488 Aggravative Hydroiodic Acid 1 Acetone 19 Evaporation of residuals at 58 C 8 888 8 688 Aggravative Hydroiodic Acid 1 Acrylonitrile 26 Evaporation of residuals at 58 C 8 888 8 688 Aggravative Hydroiodic Acid 1 Acrylonitrile 26 Evaporation of residuals at 188 C 8 288 8 688 Aggravative Hydroiodic Acid 1 Alkaline Liquid 18 Local Consequences 8 6 8 8 Same Hydroiodic Acid 1 Alkaline Liquid 18 Local Consequences 8 6 8 8 Same Hydroiodic Acid 1 Anhydrous Amenius 188						Distanc	e (m)	Distanc	e (m)	1 1	Releases	i
		ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	1
Hydroiedic Acid	Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	 #1	∦2	#1 	#2	Gas 1	Gas 2 Gas 3	Conclusion
Hydroiodic Acid	Hydroiodic Acid	1	Acetic Anhydride	107	Evaporation of residuals at 100 C **	Ø	< 100	0	400		J	Aggravated
Hydroiodic Acid	Hydroiodic Acid	1	Acetone	19	Evaporation of residuals at 50 C] Ø	100] Ø	600	Ì		Aggravated
Hydroiodic Acid 1 Acrylonitri 26 Evaporation of residuals at 188 C 8 288 8 988 Aggravat	Hydroiodic Acid	1	Acrylic Acid	3	Evaporation of residuals at 50 C	0	800*	0	1600+	ì		Aggravated
Hydroiodic Acid	Hydroiodic Acid	1	Acrylonitrile	26	Evaporation of residuals at 100 C	Ø	200	0	900	ì		Aggravated
Hydroiodic Acid	Hydroiodic Acid	1	Alcoholic Beverage	4	Ethyl iodide formed dissolves in water	1 6	< 100	Ø	< 100	i		
Hydroiodic Acid	Hydroiodic Acid	1	Alkaline Liquid	10	Local consequences		Ø			i ·		•
Hydroiodic Acid	Hydroiodic Acid	1	Ammonium Nitrate Soln.	106	Same as individual chemicals		. Ø	I Ø	ø	i		
Hydroiodic Acid 1	Hydroiodic Acid	1	Anhydrous Ammonia	10	Mitigate NH3 release	. 0	4100			•		•
Hydroiodic Acid 1	Hydroiodic Acid	1	Aniline Oil	7	Evaporation of residuals at 50 C	. 0	100		200	i	-	Aggravated
Hydroiodic Acid 1	Hydroiodic Acid	1	Butadiene	28	Evaporation of residual HI at 50 C	[0	800	. 0	800	į		
Hydroiodic Acid 1 Butyl Alcohol 4 Local consequences 6 < 100 8 < 100 150	Hydroiodic Acid	1	Butyl Acrylate	13	Release acrylic acid and butanol	0	700+		3000+	6300+	< 100	Aggravated
Hydroidic Acid	Hydroiodic Acid	1	Butyl Alcohol	4	Local consequences	0	< 100	0	< 100	Ì		Same
Hydroidic Acid 1 Chlorine 104 Releases HCL gas 6 8000 6 7600 2000 Mitigated Hydroidic Acid 1 Chloroprene 17 Evaporation of residuals at 100 C 6 800 8 1700 Aggravated Hydroidic Acid 1 Coal Tar Distillate 101 Evaporation of residuals at 50 C 6 800 6 1700 6 800 8 1700 Aggravated Hydroidic Acid 1 Combustible Liquid 101 Evaporation of residuals at 50 C 6 ? 4 100 ? Same Hydroidic Acid 1 Compound, Cleaning 101 Evaporation of residuals at 50 C 6 ? 4 100 ? Same Hydroidic Acid 1 Crude 01 Petroleum 101 Evaporation of residuals at 50 C 6 ? 4 100 7 Same Hydroidic Acid 1 Denatured Alcohol 4 Evaporate ethyl iodide 6 4 100 4 100 7 Aggravated Hydroidic Acid 1 Distillate Fuel 01 101 Evaporation of residuals at 50 C 6 4 100 6 100 7 Aggravated Hydroidic Acid 1 Ethyl Acctate 13 Release acctic acid and ethanol 6 100 6 200 300 4 100 Aggravated Hydroidic Acid 1 Ethyl Acrylate 13 Release accylic acid and ethanol 7 100 7 Aggravated Hydroidic Acid 1 Ethyl Alcohol 4 Evaporate ethyl iodide 7 100 7 4 100 7 4 100 8 100 900 6 300	Hydroiodic Acid	1	Butyraldehyde	5	Evaporation of residuals at 50 C	1 6	< 100	ļ Ø	< 100	Ì		Same
Hydroidic Acid 1 Chlorine 164 Releases HCL gas 6 8000 6 7600 2000 Mitigated Hydroidic Acid 1 Chloroprene 17 Evaporation of residuals at 100 C 6 800 8 1700 1000 1	Hydroiodic Acid	1	Carbon Tetrachloride	17	Evaporation of residuals at 100 C	0	600	1 0	1500	i		Aggravated
Hydroidic Acid 1 Chloroprene 17 Evaporation of residuals at 100 C 6 800 0 1700 Aggravature	Hydroiodic Acid	1	Chlorine	164	Releases HCL gas	6	8000	0	7600	2000	,	Mitigated
Hydroidic Acid 1 Coal Tar Distillate 101 Evaporation of residuals at 50 C 0 0 0 0 0 0 0 0 0	Hydroiodic Acid	1	Chloroprene	17	Evaporation of residuals at 100 C	j ø	800	Ø	1700	İ		Aggravated
Hydroiodic Acid 1 Compound, Cleaning 181 Evaporation of residuals at 58 C 8	Hydroiodic Acid	1	Coal Tar Distillate	101	Evaporation of residuals at 50 C	6	< 100	< 100	< 100	İ		
Hydroiodic Acid 1 Crude Oil Petroleum 101 Evaporation of residuals at 50 C 0 (100 (100 100	Hydroiodic Acid	1	Combustible Liquid	101	Evaporation of residuals at 50 C	Ø	?	< 100	?	1		Same
Hydroiodic Acid 1 Denatured Alcohol 4 Evaporate ethyl iodide Ø < 100 Ø < 100 ? Aggravate Hydroiodic Acid 1 Distillate Fuel Dil 101 Evaporation of residuals at 50 C Ø < 100 < 100 < 100 < 100 < 100 Same Hydroiodic Acid 1 Ethyl Acetate 13 Release acetic acid and ethanol Ø 100 Ø 900 6300 + < 100 Aggravate Hydroiodic Acid 1 Ethyl Alcohol 4 Evaporate ethyl iodide Ø < 100 Ø < 100 Ø < 100 ? Aggravate Hydroiodic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release Ø < 100 Ø < 100 ? Aggravate Hydroiodic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release Ø < 2500 Ø < 2300 Mitigate Hydroiodic Acid 1 Flammable Liquid 101 Evaporation of residuals at 50 C Ø ? < 100 ? Same Hydroiodic Acid 1 Fuel Dil 101 Evaporation of residuals at 50 C Ø < 100 < 100 < 100 Same Hydroiodic Acid 1 Fuel Dil 101 Evaporation of residuals at 50 C Ø < 100 < 100 < 100 Same Same Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C Ø < 100 < 100 < 100 Same Same Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C Ø < 100 < 100 < 100 Same Same Same Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C Ø < 100 < 100 < 100 Same Same Same Same Same Same Same Same	Hydroiodic Acid	1	Compound, Cleaning	101	Evaporation of residuals at 50 C) Ø	. ?	< 100	?	1		Same
Hydroiodic Acid 1 Distillate Fuel Dil 101 Evaporation of residuals at 50 C 0 < 100 < 100 < 100	Hydroiodic Acid	1	Crude Oil Petroleum	191	Evaporation of residuals at 50 C	0	< 100	< 100	< 100	1		Same
Hydroiodic Acid 1 Ethyl Acetate 13 Release acetic acid and ethanol Hydroiodic Acid 1 Ethyl Acrylate 13 Release acrylic acid and ethanol Hydroiodic Acid 1 Ethyl Alcohol 4 Evaporate ethyl iodide Hydroiodic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release Hydroiodic Acid 1 Flammable Liquid 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Fuel Oil 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Fuel, Aviation 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Fuel, Aviation 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 102 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec 103 Evaporation of residuals at 50 C Hydroiodic Acid 1 Gasoline, nec	Hydroiodic Acid	1	Denatured Alcohol	4	Evaporate ethyl iodide	0	< 100	Ø	< 100	?		 Aggravated
Hydroiodic Acid 1 Ethyl Acrylate 13 Release acrylic acid and ethano! Ø 100 Ø 900 6300* < 100 Aggravate Hydroiodic Acid 1 Ethyl Alcohol 4 Evaporate ethyl iodide Ø < 100 Ø < 100 ? Aggravate Hydroiodic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release Ø 2500 Ø 2300 Mitigate ethylene oxide release Ø 2500 Ø 2300 Mitigate ethylene oxide release Ø ? < 100 ? < 100 ? Same	Hydroiodic Acid	1	Distillate Fuel Dil	101	Evaporation of residuals at 50 C] Ø.	< 100	< 100	〈 100	Ì		Same
Hydroiodic Acid 1 Ethyl Alcohol 4 Evaporate ethyl iodide 0 < 100 0 < 100 ? Aggravate Hydroiodic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 0 2000 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500 Mitigate ethylene oxide release 0 2500	Hydroiodic Acid	1	Ethyl Acetate	13	Release acetic acid and ethanol	Ø	100	Ø	200	300	< 100	Aggravated
Hydroiodic Acid 1 Ethylene Oxide 34 Mitigate ethylene oxide release 0 2500 0 2300 Mitigate of Hydroiodic Acid 1 Flammable Liquid 101 Evaporation of residuals at 50 C 0 ? < 100 ?	Hydroiodic Acid	1	Ethyl Acrylate	13	Release acrylic acid and ethanol	0	100	1 0	900	6300+	< 100	Aggravated
Hydroiodic Acid 1 Flammable Liquid 101 Evaporation of residuals at 50 C 0 ? < 100	Hydroiodic Acid	1	Ethyl Alcohol	4	Evaporate ethyl iodide	1 6	< 100	0	< 100	?		Aggravated
Hydroiodic Acid 1 Flammable Liquid 101 Evaporation of residuals at 50 C 0 ? < 100	Hydroiodic Acid	1	Ethylene Oxide	34	Mitigate ethylene oxide release	0	2500	j ø	2300	Į		Mitigated
Hydroiodic Acid 1 Formaldehyde Solution 5 Evaporation of residuals at 50 C 0 100 0 200 Aggravate Hydroiodic Acid 1 Fuel Oil 101 Evaporation of residuals at 50 C 0 100 100 100 Same Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C 0 100 100 100 Same Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C 0 100 100 100 Same	Hydroiodic Acid	1	Flammable Liquid	161	Evaporation of residuals at 50 C	l Ø	?	< 100	?	l		_
Hydroiodic Acid 1 Fuel, Aviation 101 Evaporation of residuals at 50 C 0 100 100 Same Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C 0 100 100 Same Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C 0 100 100 Same	Hydroiodic Acid	1	Formaldehyde Solution	5	Evaporation of residuals at 50 C	Ø	100	Ø	200	1		Aggravated
Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C 0 < 100 < 100 < 100 100 Same Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C 0 < 100 < 100 < 100	Hydroiodic Acid	1	Fuel Dil	181	Evaporation of residuals at 50 C] 0	< 100	< 100	< 100	Ì		
Hydroiodic Acid 1 Gasoline 101 Evaporation of residuals at 50 C 0 < 100 < 100 < 100 100 Same Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C 0 < 100 < 100 < 100	•	1	Fuel, Aviation	101	Evaporation of residuals at 50 C	l Ø	〈 100	(100	< 100	ļ		Same
Hydroiodic Acid 1 Gasoline, nec 101 Evaporation of residuals at 50 C 0 < 100 < 100	=	1		101								-
		1	Gasoline, nec	101	•					•		
inguistable neta 1 difest control 14 chapotasion of residuals as the c 1 b (150) b 400) (nggi arabe	Hydroiodic Acid	1	Glycol Ethers		Evaporation of residuals at 100 C	•	< 100	•		•		Aggravated

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Summary of Toxic Emiss	ion Conse	equence Calculations			Non-	mixed	Mix	ed	Distar	nces for Gas	1
				•	Distan	ce (m)	Distan	ce (m)	[F	Releases	
	ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	l
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1	#2	 #1	#2 .	Gas 1	Gas 2 Gas 3	Conclusions
Hydroiodic Acid	1	Hexamethylene Diamine Sol	7	Evaporation of residuals at 50 C	0	< 100	Ø	< 100			Same
Hýdroiodic Acid	1	Hydrogen Peroxide	104	Evaporation of residuals at 50 C	Ø	200	Ø	400	1		Aggravated
Hydroiodic Acid	1	Isopropano l	4	Evaporate isopropyl iodide	Ø	(100	Ø	< 100	,		Aggravated
Hydroiodic Acid	1	Maleic Anhydride	107	Evaporation of residuals at 100 C	0	900*	0	20700+	1		Aggravated
Hydroiodic Acid	1	Methyl Alcohol	4	Releases methyl iodide gas	0	100	, ø	100	1200		Aggravated
Hydroiodic Acid	1	Methyl Chloride	17	Same as individual chemicals	Ø	1000	0	1000	1		Same
Hydroiodic Acid	1	Methyl Ethyl Ketone	19	Evaporation of residuals at 50 C	0	200	Ø	300	ł		Aggravated
Hydroiodic Acid	1	Methyl Methacrylate	13	Release methacrylic acid and methanol	0	100] Ø	700	?	600	Aggravated
Hýdroiodic Acid	1	Motor Fuel Antiknock Compe	24	Releases ethane gas **	0	100	Ø	100	< 100		Same
Hydroiodic Acid	1	Nitric Acid, Fuming	2	Evaporation of residuals at 50 C	Ø	900	< 100	2800	1		Aggravated
Hydroiodic Acid	1	Octyl Alcohol	4	Local consequences	Ø	< 100	0	< 100	ŀ		Same
Hydroiodic Acid	1	Oil	101	Evaporation of residuals at 50 C	0	< 100	< 100	< 100	1		Same
Hydroiodic Acid	1	01eum	2	Releases SO2 gas, H2SO4 mist	ļ Ø	1600	Ø	2400	1800		Aggravated
Hydroiodic Acid	1	Petroleum Naphtha	101	Evaporation of residuals at 50 C	0	< 100	< 100	< 100	8000+		Same
Hydroiodic Acid	1	Petroleum Residual Fuel	101	Evaporation of residuals at 50 C	1 0	< 100	< 100	< 100	ĺ		Same
Hydroiodic Acid	1	Petroleum, Refined	101	Evaporation of residuals at 50 C	0	< 100	< 100	< 100	1		Same
Hydroiodic Acid	1	Pheno I	31	Same as individual chemicals	j ø	< 100] Ø	< 100	ĺ		Same
Hydroiodic Acid	1	Phosphorus	105	Releases PH3 gas	0	Ø	6	Ø	1800		Aggravated
Hydroiodic Acid	1	Potassium Hydroxide, Liq	10	Local consequences	0	. 0	Ø	Ø	ĺ		Same
Hydroiodic Acid	1	Propylene Oxide	34	Mitigate propylene oxide release	Ø	1700	j ø	1500	t		Mitigated
H <u>y</u> droiodic Acid	1	Pulp Mill Liquid	101	Evaporation of residuals at 50 C	Ø	?	< 100	?	I		Same
Hydroiodic Acid	1	Sodium Hydroxide, Liq.	10	Local consequences	Ø	. 0	j ø	Ø	l		Same
Hydroiodic Acid	1	Sodium Metal	21	Releases H2 gas **	j ø	Ø	ļ Ø	Ø	< 100		Aggravated
Hydroiodic Acid	1	Styrene Monomer.	16	Evaporation of residuals at 100 C	(0	< 100	(0	200	}		Aggravated
Hydroiodic Acid	1	Sulfuric Acid	2	Releases SO2 gas	0	Ø	0	Ø	1800		Aggravated
Hydroiodic Acid	1	Sulfuric Acid, Spent	2	Releases SO2 gas	Ø	Ø	j Ø	Ø	1100		Aggravated
Hydroiodic Acid	1	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	0	< 100	Ø	699	i		Aggravated
Hydroiodic Acid	1	Vinyl Acetate	13	Release acetic acid and acetaldehyde	0	4100+	Ø	5700+	300		Aggravated
Hydroiodic Acid	1	Vinyl Chloride	17	Mitigate release of vinyl chloride	0	19100+	Ø	18100+			Mitigated
Isopropanol	4	Aceta I dehyde		Evaporation of residuals at 100 C	< 100	1100	500	700			Mitigated
Isopropanol	4	Butyraldehyde	5	Evaporation of residuals at 100 C	< 100	< 100	500	< 100			Same
Isopropanol	4	Chlorine		Evaporation at 50 C; explosion **	< 100	8000	100	7600	- 	i	Mitigated
[sopropanol	4	Ethylene Oxide		Evaporation of residuals at 50 C	(100	2500	100	2000			Mitigated

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Chemical Name #1 Group Chemical Name #2 Group Toxic Emission Consequences #1 #2 #1 #2 Gas 1 Gas 2 Gas 3 Conclusion Group Group Toxic Emission Consequences #1 #2 #1 #2 Gas 1 Gas 2 Gas 3 Conclusion Gas 2 Gas 3 Gas 2 Gas 3 Conclusion Gas 2 Gas 3 Gas 2 Gas	Summary of Toxic Emiss	ion Conse	equence Calculations			Non-		Mixe		-	nces for Gas	l
Chemical Name #1 Group Chemical Name #2 Group Toxic Emission Consequences #1 #2 #1 #2 Gas 1 Gas 2 Gas 3 Conclusion Consequences #1 #2 #2 #3 Gas 2 Gas 3 Conclusion Consequences #3 #3 #3 Gas 2 Gas 3 Conclusion Consequences #3 #3 #3 Gas 2 Gas 3 Conclusion Consequences #3 #3 #3 Gas 2 Gas 3 Conclusion Consequences #3 #3 Gas 2 Gas 3 Conclusion Consequences #3 #3 Gas 2 Gas 3	•		•			•	` '	•	• •	•	Releases	1
Isopropanol	Chemical Name #1		Chemical Name #2		Toxic Emission Consequences	•		•		•		•
Lapropanol	Isopropanol	4	Hydrogen Peroxide	184	Evaporation at 100 C: explosion **	 < 100	200	 500	1300	 		 Aggravated
Isopropanel	• •	4	• •			• •		•		•		Mitigated
Leg	• •	4	• •		-	•		•		•		Aggravated
LPG (Butene)	, ,	4		18		-		•		•		Aggravated
LPG (Butene)	· • •	28		104	•		•	•		•		Mitigated
LPG (Butene)		28	Chlorine			•		•		•		Mitigated
LPG (Isobutylene)	•					•		•		•		Mitigated
LPG (Isobutylane)	•		- "					•		•		Mitigated
LPG (Isobutylene) 28			•							•		Mitigated
LPG (Propane) 29 Chlorine 184 Fire 186 8888 186 7888 186								•		•		Mitigated
LPG (Propylene) 28 Chlorine 184 Fire (108 8000 (100 7600 Mitig LPG (Propylene) 28 Hydrogen Peroxide 184 Fire/explosion (100 200 (100 400 400 Image: Acctaldehyde 5 Exaporation of residuals at 100 C 100 100 600 400 100			• •		• •			•		•		Mitigated
LPG (Propylene) 28 Hydrogen Peroxide 184 Fire/explosion (100 200 100 400 100 400 100 400 100 400 100 400 100 400 100 400 100	· · · · ·					, ,		•		•		Mitigated
Methyl Alcohol 4 Acetaldehyde 5 Evaporation of residuals at 100 C 100 100 600 700 100 100 100 100 100 100 100 100 1			Hydrogen Peroxide	184		•				•		Aggravated
Methyl Alcohol 4 Butyraldehyde 5 Evaporation of residuals at 100 C 100 C 600 C 100 C 600 C 100 C 800 C 800 C 100 C <td></td> <td>4</td> <td></td> <td></td> <td>• •</td> <td></td> <td></td> <td>•</td> <td></td> <td>•</td> <td></td> <td>Mitigated</td>		4			• •			•		•		Mitigated
Methyl Alcohol 4 Chlorine 184 Evaporation at 58 C; explosion ** 180 8008 100 7600 Mitigg Methyl Alcohol 4 Ethylene Oxide 34 Evaporation of residuals at 50 C 100 2500 100 2000 Mitigg Methyl Alcohol 4 Ferric Chloride Solution 104 Release of formaldehyde 100 0 100 0 300 Maggrav Methyl Alcohol 4 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** 100 200 600 1300 Maggrav Methyl Alcohol 4 Propylene Oxide 34 Evaporation of residuals at 50 C 100 100 1300 Mitigg Methyl Alcohol 4 Toluene Diisocyanate 18 Evaporation of residuals at 100 C ** 100 600 60 600 Maggrav Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 600 600 600 Maggrav Methyl Chloride 17		4	· · · · · · · · · · · · · · · · · · ·	5				•		•		-
Methyl Alcohol 4 Ethylene Oxide 34 Evaporation of residuals at 50 C 100 2500 100 2000 Mitig Methyl Alcohol 4 Ferric Chloride Solution 104 Release of formaldehyde 100 0 100 0 300 Aggrav Methyl Alcohol 4 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** 100 200 600 1300 Maggrav Methyl Alcohol 4 Propylene Oxide 34 Evaporation of residuals at 50 C 100 100 100 1300 Maggrav Methyl Alcohol 4 Sodium Metal 21 Release of hydrogen gas ** 100 0 600 0 Aggrav Methyl Alcohol 4 Toluene Disocyanate 18 Evaporation of residuals at 100 C ** 100 0 600 600 600 Aggrav Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 0 1000 0 900 6900 900 900 900 900 900 900 900 900 900	-	4	•	104	•	•	•	•	-	•		Mitigated
Methyl Alcohol 4 Ferric Chloride Solution 184 Release of formaldehyde 100 0 100 0 300 Aggrav Methyl Alcohol 4 Hydrogen Peroxide 164 Evaporation at 100 C; explosion ** 160 200 600 1300 Aggrav Methyl Alcohol 4 Propylene Oxide 34 Evaporation of residuals at 50 C 160 1700 100 1300 Mitig Methyl Alcohol 4 Sodium Metal 21 Release of hydrogen gas ** 160 600 0 600 0 Aggrav Methyl Alcohol 4 Toluene Diisocyanate 18 Evaporation of residuals at 100 C ** 160 600 600 600 Aggrav Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 0 1000 0 1000 0 1000 0 1000 0 1000 0 1000 0 1000 0 1000 0 1000 0 1000	•	4		34		1 100		•		•		Mitigated
Methyl Alcohol 4 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** 100 200 600 1300 Aggraventary Methyl Alcohol 4 Propylene Dxide 34 Evaporation of residuals at 50 C 100 1700 100 1300 Mitig Methyl Alcohol 4 Sodium Metal 21 Release of hydrogen gas ** 100 0 600 0 Aggraventary Methyl Alcohol 4 Toluene Diisocyanate 18 Evaporation of residuals at 100 C ** 100 600 600 600 Aggraventary Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 800 900 6900 Mitig Methyl Chloride 17 Ferric Chloride Solution 104 Evaporation at 50 C; explosion ** 1000 400 100 400 Same Methyl Chloride 17 Sodium Metal 21 Evaporation at 50 C; explosion ** 1000 0 1000 0 1000 0	•	4	Ferric Chloride Solution	104	•	100	Ø	100	Ø	300		Aggravated
Methyl Alcohol 4 Propylene Oxide 34 Evaporation of residuals at 50 C 100 1700 100 1300 Mitigation Methyl Alcohol 4 Sodium Metal 21 Release of hydrogen gas ** 100 0 600 0 Aggravante Methyl Alcohol 4 Toluene Diisocyanate 18 Evaporation of residuals at 100 C ** 100 600 600 600 600 600 600 600 600 600 Aggravante Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 8000 900 6900 Methyl Chloride 17 Hydrogen Peroxide 104 Evaporation at 50 C; explosion ** 1000 200 1000 400 Methyl Chloride 17 Sodium Metal 21 Evaporation at 50 C; explosion ** 1000 200 1000 400 Methyl Chloride 1000 400 Methyl Chloride 1000 400 Methyl Chloride 1000 400 Methyl Chloride 1000 400 Methyl Chloride 10	•	4	Hydrogen Peroxide	104		100	200	600	1300	i		Aggravated
Methyl Alcohol 4 Sodium Metal 21 Release of hydrogen gas ** 100 0 600 0 Aggraven Methyl Alcohol 4 Toluene Diisocyanate 18 Evaporation of residuals at 100 C ** 100 0 600 600 Aggraven Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 8000 900 6900 Mitigation of chlorine release ** 1000 0 1000 0 1000 0 Same Same Methyl Chloride 17 Hydrogen Peroxide 104 Evaporation at 50 C; explosion ** 1000 0 1000 0 Same Methyl Ethyl Ketone 19 Chlorine 104 Mitigation of chlorine release ** 200 8000 300 7600 Mitigation of chlorine release ** 200 8000 300 7600 Mitigation of chlorine release ** 200 8000 300 7600 Mitigation of chlorine release ** 200 8000 300 7600 Mitigation of chlorine release ** 200 8000 300 <t< td=""><td>Methyl Alcohol</td><td>4</td><td>Propylene Oxide</td><td>34</td><td>· · · · · · · · · · · · · · · · · · ·</td><td>100</td><td>1700</td><td>100</td><td>1300</td><td>i</td><td></td><td>Mitigated</td></t<>	Methyl Alcohol	4	Propylene Oxide	34	· · · · · · · · · · · · · · · · · · ·	100	1700	100	1300	i		Mitigated
Methyl Alcohol 4 Toluene Diisocyanate 18 Evaporation of residuals at 100 C ** 100 (100 600	-	4	• •	21	-	100	Ø	600		i		Aggravated
Methyl Chloride 17 Chlorine 104 Mitigation of chlorine release ** 1000 8000 900 6900 Mitigation of chlorine release ** 1000 8000 900 6900 Mitigation of chlorine release ** 1000 8000 900 6900 Mitigation of chloride 17 Ferric Chloride Solution 104 Evaporation at 50 C; explosion ** 1000 200 1000 400 Same Methyl Chloride 17 Sodium Metal 21 Evaporation at 50 C; explosion 1000 0 1000 0 Same Methyl Ethyl Ketone 19 Chlorine 104 Mitigation of chlorine release ** 200 8000 300 7600 Mitigation Methyl Ethyl Ketone 19 Ferric Chloride Solution 104 Evaporation of residuals at 50 C 200 0 300 0 Mitigation 104 Evaporation at 100 C; explosion ** 200 200 900 1300 Maggray Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Mitigation 104 Evaporation at 100 C; explosion ** 200 0 0 900 0 Maggray Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Maggray Maggray Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Maggray Maggray Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Maggray Maggray Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Maggray Maggray Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Maggray Maggr	Methyl Alcohol	4	Toluene Diisocyanate	18	Evaporation of residuals at 100 C **	100	< 100	600	600	İ		Aggravated
Methyl Chloride 17 Ferric Chloride Solution 104 Evaporation of residuals at 50 C 1000 0 1000 </td <td>Methyl Chloride</td> <td>. 17</td> <td>Chlorine</td> <td>164</td> <td>Mitigation of chlorine release **</td> <td>1000</td> <td>8000</td> <td>900</td> <td>6900</td> <td>i</td> <td></td> <td>Mitigated</td>	Methyl Chloride	. 17	Chlorine	164	Mitigation of chlorine release **	1000	8000	900	6900	i		Mitigated
Methyl Chloride 17 Sodium Metal 21 Evaporation at 50 C; explosion 1000 0 10000 0 10000 0 1000 0 1000 0 1000 0 1000 0 10000 0 1000 0 1000 0 1000 0 1000 0 1000 0 1000 0	Methyl Chloride	17	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	1000	Ø	1000	Ø	i		-
Methyl Chloride 17 Sodium Metal 21 Evaporation at 50 C; explosion 1000 0 1000 0 Same Methyl Ethyl Ketone 19 Chlorine 104 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation Methyl Ethyl Ketone 19 Ferric Chloride Solution 104 Evaporation of residuals at 50 C 200 0 300 0 Aggrave Methyl Ethyl Ketone 19 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** 200 200 900 1300 Aggrave Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Aggrave	Methyl Chloride	17	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	1000	200	1000	400	Ī		Same
Methyl Ethyl Ketone 19 Chlorine 104 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 8000 300 7800 Mitigation of chlorine release ** 200 800 300 7800 Mitigation of chlorine release ** 200 800 300 7800 Mitigation of chlorine release ** 200 800 300 7800 Mitigation of chlorine release ** 200 800 300 7800 Mitigation of chlorine release ** 200 800 300 7800 Mitigation of chlorine release ** 200 800 300 7800 Mitigation of chlorine release ** Mitigation of chlorine release ** 200 800 300 300 Mitigation of chlorine release ** Mitigation of chlorine release ** 200 800 300 300 Mitigation of chlorine release ** Mitigation of chlorine release ** 200 800 300 300 Mitigation of chlorine release ** Mitigation of chlorine release ** 200 800 300 300 Mitigation of chlorine release ** Mitigation of chlorine release ** 200 800 300 300 Mi	Methyl Chloride	17	Sodium Metal	21	Evaporation at 50 C; explosion	1000	Ø	1000	Ø	İ		-
Methyl Ethyl Ketone 19 Ferric Chloride Solution 104 Evaporation of residuals at 50 C 200 0 300 0 Aggrav Methyl Ethyl Ketone 19 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** 200 200 900 1300 Aggrav Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Aggrav	Methyl Ethyl Ketone	19	Chlorine	104	Mitigation of chlorine release **	200	8000	300		İ		Mitigated
Methyl Ethyl Ketone 19 Hydrogen Peroxide 104 Evaporation at 100 C; explosion ** 200 200 900 1300 Aggrave Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 00 900 00 Aggrave	-	19	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	200	Ø	300	Ø	Ì		Aggravated
Methyl Ethyl Ketone 19 Sodium Metal 21 Releases H2 gas 200 0 900 0 Aggrav	Methyl Ethyl Ketone	19	Hydrogen Peroxide	164	Evaporation at 100 C; explosion **	200	200	900	1300	ĺ		Aggravated
	• •	19		21	Releases H2 gas	200	Ø	900	Ø	İ		Aggravated
		13	Chlorine	104	Mitigation of chlorine release **	100	8000	200	7600	Ī		Mitigated
Methyl Methacrylate 13 Ferric Chloride Solution 104 Evaporation of residuals at 50 C 100 0 200 0 Aggrav		13	Ferric Chloride Solution	101	-	100	Ø	200	Ø	i ·		Aggravated
	Methyl Methacrylate	13		104	•	100	200	700	1300	İ		Aggravated

Summary of Toxic Emission	Conse	equence Calculations			Non-i		Mixe		•	ces for Ga	s
•	1CTU		ACTU		Distance		Distanc		•	Releases	. !
Chemical Name #1	ASTM Group	Chemical Name #2	ASTM Group	Toxic Emission Consequences	IDLH #1	IDLH #2	IDLH #1	IDLH #2	IDLH	IDLH ID	_H 3 Conclusions
					-	-π		#Z			
Methyl Methacrylate	13	Sodium Metal	21	Evaporation of residuals at 100 C **	100	Ø	700	Ø	İ		Aggravated
Motor Fuel Antiknock Comp	24	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	100	200	500	1300	[Aggravated
Nitric Acid, Funing	2	Acetaldehyde	5	Evaporation at 100 C; release of NO **	900	1100	2800	700	2000		Aggravated
Nitric Acid, Funing	2	Acetic Acid, Glacial	3	Evaporation of residuals at 50 C	900	100	1600	300	1		Aggravated
Nitric Acid, Fuming	2	Acetic Anhydride	107	Evaporation of residuals at 100 C **	900	< 100	2800	200	1		Aggravated
Nitric Acid, Fuming	2	Acetone	19	Evaporation of residuals at 100 C **	900	100	2800	500	1		Aggravated
Nitric Acid, Fuming	2	Acrylic Acid	3	Evaporation of residuals at 50 C	900	800*	1600	1600+]		Aggravated
Nitric Acid, Fuming	2	Acrylonitrile	26	Evaporation of residuals at 100 C **	900	200	2800	900	1		Aggravated
Nitric Acid, Fuming	2	Alcoholic Beverage	4	Evaporation of residuals at 100 C **	900	< 100	2800	< 100	1	< 100	Aggravated
Nitric Acid, Fuming	2	Alkaline Liquid	10	Evaporation of residuals at 100 C	900	Ø	2800	· ø	l		Aggravated
Nitric Acid, Fuming	2	Ammonium Nitrate Soln.	106	Heat of dissolution of HNO3	900	Ø	1600	Ø	1		Aggravated
Nitric Acid, Funing	2	Anhydrous Ammonia	10	Mitigate release of NH3	900	4100	1600	3200	i		Mitigated
Nitric Acid, Fuming	2	Aniline Oil	7	Evaporation of residuals at 100 C **	900	100	2800	700	į		Aggravated
Nitric Acid, Fuming	2	Benzene	16	Evaporation of residuals at 100 C **	900	200	2800	1000	i		Aggravated
Nitric Acid, Fuming	2	Butadiene	28	Evaporation of residuals at 50 C **	900	800	1 1600	600	i		Aggravated
Nitric Acid, Funing	2	Butyl Acrylate	13	Evaporation of residuals at 100 C	I 900	700+	2800	3000+	i		Aggravated
Nitric Acid, Funing	2	Butyl Alcohol	4	Evaporation of residuals at 100 C **	I 900	< 100	2800	600	i		Aggravated
Nitric Acid, Fuming	2	Butyraldehyde	5	Evaporation at 100 C; release of NO **	900	(100	2800	< 100	2400		Aggravated
Nitric Acid, Fuming	2	Carbon Disulfide	20	Evaporation of residuals at 50 C	1 900	900	1 1600	2000	, <u>1,55</u>		Aggravated
Nitric Acid, Fuming	2	Carbon Tetrachloride		Unknown	900	600	. ?	?	, 		Unknown
Nitric Acid, Fuming	2	Chloroprene	17	Evaporation of residuals at 100 C **	1 900	800	1 2800	1700	I		Aggravated
Nitric Acid, Fuming	2	Coal Tar Distillate	101	Evaporation of residuals at 50 C	1 900	< 100	1 1600	< 100			Aggravated
Nitric Acid, Fuming	2	Combustible Liquid	101	Evaporation of residuals at 50 C	900	?	•	?	I	•	Aggravated
Nitric Acid, Funing	2	Compound, Cleaning	181	Evaporation of residuals at 50 C	900	?	1600	?	1		Aggravated
Nitric Acid, Fuming	2	Crude Oil Petroleum	101	Evaporation of residuals at 50 C	1 900	(100	1 1600	(100	•		Aggravated
Nitric Acid, Fuming	2	Denatured Alcohol	4	Evaporation of residuals at 100 C **	1 900	(100	2800	(100	! !		Aggravated
Nitric Acid, Fuming	2	Dinitrotoluene, Liquid	27	Evaporation of residuals at 50 C	1 900	Ø	•	0	! 		Aggravated
Nitric Acid, Fuming	2	Distillate Fuel Gil	101	Evaporation of residuals at 50 C	1 900	(100	•	(100	l .		Aggravated
Nitric Acid, Fuming	2	Ethyl Acetate	13	Evaporation of residuals at 50 C	900	100	1600	200			Aggravated
Nitric Acid, Funing	2	Ethyl Acrylate		Evaporation of residuals at 100 C	1 900	100	1 2800	900]		Aggravated
Nitric Acid, Fuming	2	Ethyl Alcohol		Evaporation of residuals at 100 C **	900	(100	2800	< 100			Aggravated
	2	Ethylene Oxide	34	Evaporation of residuals at 100 C	1 900	2500	1 2800	1600	i !		Aggravated Aggravated
Nitric Acid, Funing	_			•	900		•	_	 / 100		
Nitric Acid, Fuming	2	Flammable Liquid	101	Evaporation of residuals at 50 C	ไ ลดด	?	1600	ŗ	< 100		Aggravated

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ummary or it	DXIC ENISS	ion Conse	quence Calculations	*		Non	-mixed	Mixe		Distan	ces for	Gas	
			•	*		Dista	nce (m)	Distanc	e (m)	R	eleases		!
		ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	i
Chemical	Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1	. #2	#1	#2	Gas 1	Gas 2 (Gas 3	Conclusio
itric Acid,	Fuming	2	Formaldehyde Solution	5	Releases formic acid and NO **	90	0 100	2800	500	2300	2000		Aggravate
itric Acid,	Fuming	2	Fuel Oil	101	Evaporation of residuals at 50 C	90	0 (100	1600	< 100	l			Aggravate
itric Acid,	Fuming	2	Fuel, Aviation	101	Evaporation of residuals at 50 C	90	0 < 100	1600	(100	ł			Aggravate
itric Acid,	Fuming	2	Gasoline	101	Evaporation of residuals at 50 C	90	0 < 100	1600	〈 100	l			Aggravate
itric Acid,	Fuming	2	Gasoline, nec	101	Evaporation of residuals at 50 C	90	0 (100	1600	< 100	1			Aggravate
itric Acid,	Fuming	2	Glycol Ethers	14	Evaporation of residuals at 100 C **	90	0 < 100	2800	400	1			Aggravate
itric Acid,	Funing	2	Hexamethylene Diamine Soln	7	Evaporation of residuals at 100 C **] 90	0 (100	2800	< 100				Aggravate
itric Acid,	Funing	2	Hydrogen Peroxide	104	Evaporation of residuals at 50 C **	90	200	1600	400				Aggravate
itric Acid,	Fuming	2	Isopropanol	4	Evaporation of residuals at 100 C **	90	0 (100	2800	500	j			Aggravate
itric Acid,	Funing	2	LPG	28	Evaporation of residuals at 50 C ++	90	ø 6øø	1600	700	İ			Aggravate
itric Acid,	Fuming	2	LPG (Butene)	28	Evaporation of residuals at 50 C ++	90	800	1600	600	l			Aggravate
itric Acid,	Funing	2	LPG (Isobutylene)	28	Evaporation of residuals at 50 C **	90	Ø ?	1600	?	İ			Aggravate
itric Acid,	Funing	2	LPG (Propylene)	28	Evaporation of residuals at 50 C **	90	ø ?	1600	?	İ			Aggravate
itric Acid,	Fuming	2	Maleic Anhydride	107	Evaporation of residuals at 100 C **	90	900*	2800	3000*	İ			Aggravat
itric Acid,		2	Methyl Alcohol	4	Evaporation of residuals at 100 C **	90	7 100	2800	600	İ			Aggravate
itric Acid,		2	Methyl Chloride	17	Releases HCL, NO and CO gas **	90	1000	1 1600	1000] 2200	2500	900	Aggravate
itric Acid,		2	Methyl Ethyl Ketone	19	Evaporation of residuals at 100 C **	90	7 200	2800	900	į			Aggravate
itric Acid,		2	Methyl Methacrylate	13	Evaporation of residuals at 100 C	90	100	2800	700	İ			Aggravate
itric Acid,		2	Motor Fuel Antiknock Compo	24	Releases ethane gas **.	90	7 100	2800	500	(100			Aggravate
itric Acid,		2	Octyl Alcohol	4	Evaporation of residuals at 50 C	90	0 (100	1600	< 100	i			Aggravate
itric Acid,	=	2	Oil	101	Evaporation of residuals at 50 C	90	0 (100	1600	⟨ 100	i .			Aggravate
itric Acid,	=	2	Oleum	2	Evaporation of residuals at 50 C	90	1600	1600	2400	İ			Aggravate
itric Acid,		2	Petroleum Naphtha	101	Evaporation of residuals at 50 C	90	7 (100	1600	< 100	I			Aggravate
itric Acid,		2	Petroleum Residual Fuel	101	Evaporation of residuals at 50 C	90	7 (100	1600	< 100				Aggravate
itric Acid,		2	Petroleum, Refined	101	Evaporation of residuals at 50 C	900	7 (100	1600	< 100	İ			Aggravate
itric Acid,		2	Phenol	31	Evaporation of residuals at 100 C **	900	(100	2800	700	8000*			Aggravate
itric Acid,		2	Phosphorus	1Ø5	Evaporation of residuals at 100 C **	900	•	2800	ø				Aggravate
itric Acid,		2	Potassium Hydroxide, Liq	10	Evaporation of residuals at 100 C	1 900		2800	Ø				Aggravate
itric Acid,		2	Propylene Oxide	34	Evaporation of residuals at 100 C	I 900		2800	1500				Aggravate
itric Acid,		2	Pulp Mill Liquid	101	Evaporation of residuals at 50 C	901		•	. ?	, 			Aggravate
itric Acid,	_	2	Sodium Hydroxide, Liq.		Evaporation of residuals at 100 C	1 906		•	. ø				Aggravate
itric Acid,	_	2	Sodium Metal	21	Releases hydrogen gas **.	900		2800	_	< 100			Aggravate
itric Acid,		2	Styrene Monomer	16	Evaporation of residuals at 100 C **	900		•	200				Aggravate
IVITE ACIU,	ı amınğ	•	OATIONO MONOREI	-	Eraporation of feetabate at 100 C 44	, 301	, , 100	1 2000	200	l,			Luggi arabe

Summary of Toxic Emissi	on Conse	quence Calculations			Non-	iixed	Mix	ed	Distan	ces for	Gas	1
		•			Distanc	e (m)	Distan	ce (m)	R	eleases		1
	ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	I
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1	#2	#1 	#2	Gas 1	Gas 2 0	as 3	Conclusions
Nitric Acid, Fuming	2	Sulfur Dioxide	105	Releases NO and SO3 gases	900	4500	2800	4300	1800	3600		 Mitigated
Nitric Acid, Fuming	2	Toluene	16	Evaporation of residuals at 100 C **	900	100	2800	900	1			Aggravated
Nitric Acid, Fuming	2	Toluene Diisocyanate	18	Releases NO2 gas **	900	< 100	2800	600	900			Aggravated
Nitric Acid, Fuming	2	Vinyl Acetate	13	Evaporation of residuals at 50 C	900	4100+	1600	5700+	l			Aggravated
Nitric Acid, Fuming	2	Vinyl Chloride	17	Evaporation of residuals at 50 C **	900	19100+	1600	13500+	< 100			Aggravated
Nitric Acid, Fuming	2	Xy I ene	16	Evaporation of residuals at 100 C **	900	< 100	2800	100	1			Aggravated
Octyl Alcohol	4	Acetaldehyde	5	Evaporation of residuals at 100 C	< 100	1100	< 100	700	?			Mitigated
Octyl Alcohol	4	Butyra I dehyde	5	Evaporation of residuals at 100 C	< 100	< 100	< 100	< 100	j			Same
Octyl Alcohol	4	Chlorine	104	Evaporation at 50 C; explosion **	< 100	8000	< 100	7600	1			Mitigated
Octyl Alcohol	- 4	Ethylene Oxide	34	Evaporation of residuals at 50 C	< 100	2500	< 100	2000	1			Mitigated
Octyl Alcohol	4	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	< 100	200	< 100	1300	1			Aggravated
Octyl Alcohol	4	Propylene Oxide	34	Evaporation of residuals at 50 C	< 100	1700	< 100	1300	l			Mitigated
Octyl Alcohol	4	Sodium Metal	21	Release of hydrogen gas **	< 100	Ø	< 100	Ø	(100			Same
Octyl Alcohol	4	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	< 100	< 100	< 100	600	į			Aggravated
Dil	101	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	1 100	7600	1			Mitigated
Oil	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 100	200	1 100	400	1			Aggravated
0 leum	2	Acetaldehyde	5	Evaporation at 100 C. S02 gas **	1600	1100	3200	700	I 2600			Aggravated
Oleum	2	Acetic Acid, Glacial	3	Evaporation at 100 C, SO2 gas **	1 1600	100	3200	1300	! 2600			Aggravated
0 leum	2	Acetic Anhydride	107	Evaporation at 100 C, SD2 gas **	1600	⟨. 100	3200	400	I 2600			Aggravated
Oleum	2	Acetone	19	Evaporation at 100 C, SO2 gas **	1600	100	3200	500	I 2600			Aggravated
Oleum	2	Acrylic Acid	3	Evaporation at 100 C, SO2 gas **	1600	800+	3200	6300+	I 2600			Aggravated
0 leum	2	Acrylonitrile	26	Releases NO2, SO2 and CO gas **	1 1600	200	3200	900	I 800	1500	500	Aggravated
0 leus	2	Alcoholic Beverage	4	Releases acetaldehyde and SO2 gas **	1600	⟨ 100	3200	< 100	900	2600		Aggravated
Oleum	2	Alkaline Liquid	10	Release residual S03	1600	ø	3200	· ø	İ			Aggravated
Oleum	2	Ammonium Nitrate Soln.	106	Explosion	1600	ø	3200	Ø				Aggravated
Oleum	2	Anhydrous Ammonia	10	Releases SO2 gas	1 1600	4100	3200	3700	2600			Mitigated
O l eum	2	Aniline Oil	7	Releases NOx, SO2 and CO gases **	1 1600	100	3200	700	700	1500	500	Aggravated
Oleum	2	Benzene	16	Evaporation at 100 C, SO2 gas **	1600	200	1 3200	1000	2600			Aggravated
Dieum	2	Butadiene	28	Releases SO2 gas **	1 1800	800	2400	800 1	2600			Aggravated
Dieum	2	Butyl Acrylate	13	Evaporation at 100 C, SB2 gas **	1 1800	700+	3200	3000+	2600			Aggravated
Dieum	2	Butyl Alcohol	4	Releases butyraldehyde and SO2 gas **	1600	(100	1 3200	800	< 100	2600		Aggravated
leu∎	2	Butyraldehyde	5	Evaporation at 100 C, SO2 gas **	1 1600	< 100	1 3200	(100	2600			Aggravated
Dieum	2	Carbon Tetrachloride		Releases COCL2 and HCL gas	1 1600	600	3200	1500	2700	1000		Aggravated

Summary of Toxic Emiss		440			1	Non-m		Mix		•	ces for	U45	1
	ASTM		ASTM		- 1	istanc ייחו		Distan		•	Releases IDLH	. 101 11	
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	1	IDLH #1	IDLH #2	IDLH #1	IDLH ∦2	IDLH Gas 1			 Conclusions
Oleum	2	Chloroprene	17	Releases HCL, SO2 and CO gas **	- 	1600	800		2600	900	1500	 500	- Aggravated
Oleum	2	Coal Tar Distillate	101	Evaporation of residuals at 50 C	İ	1600	< 100	2400	< 100	Í			Aggravated
Oleum	2	Combustible Liquid	101	Evaporation of residuals at 50 C	Ì	1600	< 100	2400	?	ĺ			Aggravated
Oleum	2	Compound, Cleaning	101	Evaporation of residuals at 50 C	١	1600	< 100	2400	?	ĺ			Aggravated
Oleum	2	Crude Dil Petroleum	161	Evaporation of residuals at 50 C	Ì	1600	< 100	2400	< 100	İ			Aggravated
Oleum	2	Denatured Alcohol	4	Releases acetaldehyde and SO2 gas ++	i	1600	< 100	3200	< 100	900	2600	•	Aggravated
0 leus	2	Dinitrotoluene, Liquid	27	Evaporation at 100 C; explosion **	Ì	1600	Ø	3200	< 100	ĺ			Aggravated
Oleum	2	Distillate Fuel Oil	101	Evaporation of residuals at 50 C	Ī	1600	< 100	2400	< 100	ĺ			Aggravated
0 leus	2	Ethyl Acetate	13	Evaporation at 100 C, SO2 gas **	Ì	1600	100	3200	500	2600			Aggravated
Dieum	2	Ethyl Acrylate	13	Evaporation at 100 C, SO2 gas **	l	1600	100	3200	900	2600			Aggravated
0 leus	2	Ethyl Alcohol	4	Releases acetaldehyde and SO2 gas **	1	1600	< 100	3200	< 100	900	2600		Aggravated
0 leu n	2	Ethylene Oxide	34	Evaporation at 100 C, SO2 gas **	-	1600	2500	3200	2200	2600			Aggravated
Oleum	2	Flammable Liquid	101	Evaporation of residuals at 50 C	-	1600	〈 100	2400	?	l			Aggravated
0 leum	2	Formaldehyde Solution	5	Evaporation at 100 C, SD2 gas	-1	1600	100	2400	500	2600			Aggravated
Oleum	2	Fuel Oil	101	Evaporation of residuals at 50 C	- 1	1600	〈 100	2400	< 100	l			Aggravated
0 leum	2 ~	Fuel, Aviation	101	Evaporation of residuals at 50 C	1	1600	< 100	2400	< 100	l			Aggravated
0 i eu n	2	Gasoline	101	Evaporation of residuals at 50 C	-	1600	< 100	2400	< 100	l			Aggravated
0 i eum	2	Gasoline, nec	101	Evaporation of residuals at 50 C		1600	〈 100	2400	< 100	1			Aggravated
Oleum	2	Glycol Ethers	14	Evaporation at 100 C, SQ2 gas **	1	1600	⟨ 100	3200	400	2600			Aggravated
Oleum	2	Hexamethylene Diamine Solr	1 7	Evaporation at 100 C, SD2 gas	1	1600	〈 100	2400	< 100	2600			Aggravated
Oleum	2	Hydrogen Peroxide	104	Evaporation at 100 C; Explosion	- [1600	200	3200	1300	l			Aggravated
Oléum	2	Isopropanol	4	Evaporation at 100 C, SO2 gas **	1	1600	〈 100	3200	500	2600			Aggravated
0 leum	2	LPG	28	Releases S02 gas **	1	1600	600	2400	700	2600			Aggravated
Oleum	2	LPG (Butene)	28	Releases SO2 gas **	1	1600	800	2400	500	2600			Aggravated
0 leum	2	LPG (Isobutylene)	28	Releases S02 gas **	1	1600	< 100	2400	< 100	2600			Aggravated
0 i eum	2	LPG (Propylene)	28	Releases S02 gas **	1	1600	< 100	2400	< 100	2800			Aggravated
0 leum	2	Maleic Anhydride	107	Evaporation at 100 C, SO2 gas	1	1600	900+	3200	20700+	2600			Aggravated
Oleum	2	Methyl Alcohol	4	Releases formaldehyde and SO2 gas **	١	1600	< 100	3200	600	2600	2600		Aggravated
0 leum	2	Methyl Chloride	17	Releases HCL, S02 and C0 gas **	1	1600	1000	2400	1000	1200	1500	500	Aggravated
0 leum	2	Methyl Ethyl Ketone	19	Evaporation at 100 C, SO2 gas **	-	1600	200	3200	900	2600			Aggravated
0 leum	2	Methyl Methacrylate	13	Evaporation at 100 C, SO2 gas **	l	1600	100	3200	700	2800			Aggravated
0 leum	2	Motor Fuel Antiknock Compo	24	Releases S02 and ethane gas	1	1600	100	3200	100	2600	< 100		Aggravated
0 leum	2	Octyl Alcohol	4	Evaporation at 100 C, SO2 gas **	1	1600	< 100	3200	< 100	2600			Aggravated
•													•
				•									

Summary of Toxic Emissic	n Conse	equence Calculations			Non-	∎ixed	Mixe	ed	•	ces for		1
ı		•			Distan	ce (m)	Distand	ce (m)	F	le l eases	;	1
	ASTM		ASTM	•	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH	•
Chemical Name #1	Group	Chemical Name #2	Gгоир	Toxic Emission Consequences	# 1	#2	#1	#2	Gas 1	Gas 2	Gas 3	[Conclusions
Oleum	.2	Oil	101	Evaporation of residuals at 50 C	1600	< 100	2400	< 100				Aggravated
0 leum	2	Petroleum Naphtha	101	Evaporation of residuals at 50 C	1600	< 100	2400	< 100	ł			Aggravated
0 leum	2	Petroleum Residual Fuel	101	Evaporation of residuals at 50 C	1600	〈 100	2400	< 100	1			Aggravated
O leum	2	Petroleum, Refined	101	Evaporation of residuals at 50 C	1600	< 100	2400	< 100	1			Aggravated
0 leum	2	Pheno I	31	Evaporation at 100 C, SO2 gas	1600	< 100	3200	700	2600			Aggravated
Oleum	2	Phosphoru s	105	Releases H3P04 mist and S02 gas **	1600	Ø	3200	Ø	11700+	1500		Aggravated
O I eum	2	Potassium Hydroxide, Liq	10	Release residual SO3	1600	Ø	3200	. 8	1			Aggravated
0 leum	2	Propylene Oxide	34	Evaporation at 100 C, SD2 gas **	1600	1700	3200	1500	2600			Aggravated
O l eum	2	Pulp Mill Liquid	101	Evaporation of residuals at 50 C	1600	< 100	2400	?	1	•		Aggravated
0 leum	2	Sodium Hydroxide, Liq.	10	Release residual SO3	1600	Ø	3200	Ø	1			Aggravated
0 leum	2	Sodium Metal	21	Releases SO2 and hydrogen gas **	1600	Ø	2400	Ø	2600	(100		Aggravated
Dieum	2	Styrene Monomer	16	Evaporation at 100 C, SO2 gas **	1600	< 100	3200	200	2600			Aggravated
Dieum	2	Toluene	16	Evaporation at 100 C, SO2 gas **	1600	100	3200	900	2600			Aggravated
0 leum	2	Toluene Diisocyanate	18	Releases SO2, CO and NO2 gas	1600	〈 100	3200	600	1500	500	800	Aggravated
0 l eum	2	Vinyl Acetate	13	Evaporation at 100 C, SO2 gas **	1600	4100*	3200	5700*	2600			Aggravated
01'eum	2	Vinyl Chloride	17	Releases HCL, SO2 and CO gas **	1600	19100*	2400	19000*	1100	1500	500	Aggravated
0 l'eum	2	Xy lene	16	Evaporation at 100 C, SO2 gas **	1600	< 100	3200	100	2600			Aggravated
Petroleum Naphtha	101	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	100	7600	1			Mitigated
Petroleum Naphtha	181	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 100	200] 100	400	1			Aggravated
Petroleum Residual Fuel	101	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	100	7600	J			 Mitigated
Petroleum Residual Fuel	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 100	200	100	400	1			Aggravated
Petroleum, Refined	101	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	100	7600	1			Mitigated
Petroleum, Refined	161	Hydrogen Peroxide	164	Evaporation at 50 C; explosion **	< 100	200	100	400	1			Aggravated
Pheno I	31	Chlorine	104	Evaporation of residuals at 50 C **	< 100	8000	200	7600	1			Mitigated
Pheno I	31	Ethylene Oxide	34	Evaporation of residuals at 100 C **	< 100	2500	700	1600	l			Mitigated
Pheno I	31	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	< 100	200	700	1300	1			Aggravated
Pheno I	31	Phosphorus	105	Evaporation of residuals at 50 C	< 100	Ø	200	Ø	1			Aggravated
Pheno I	31	Propylene Oxide	34	Evaporation of residuals at 100 C **	< 100		700	1100	ŀ			Mitigated
Phosphoric Acid	1	Acetaldehyde	5	Evaporation of residuals at 100 C	įø	1100	jø	700	1			Mitigated
Phosphoric Acid	1	Acetic Acid, Glacial	3	Evaporation of residuals at 50 C	j ø	100	1 0	300	İ			Aggravated
Phosphoric Acid	1	Acetic Anhydride	107	Evaporation of residuals at 100 C **		(100	. 0	400	1			Aggravated
Phosphoric Acid	1	Acetone	19	Evaporation of residuals at 50 C	j ø	100	Ø	600	Ī			Aggravated
Phosphoric Acid	1	Acrylic Acid	3	Evaporation of residuals at 50 C	. 0	800+	j ø	1600+	İ			Aggravated

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Summary of Toxic Emissi	ion Conse	quence Calculations		· 	Non-i	 nixed	 Mixe	. 	 Distan	 ces for G	 as
					Distanc	ce (m)	Distanc	e (m)	! R	eleases	i
	ASTM		ASTM		IDLH	IDLH	IDLH	IDLH	IDLH	IDLH I	DLH I
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	j #1	#2	#1	#2	Gas 1	-	s 3 Conclusion
Phosphoric Acid	1	Acrylonitrile	26	Evaporation of residuals at 100 C	 Ø	200		900	 		 Aggravated
Phosphoric Acid	1	Alcoholic Beverage	4	Local consequences	0	(100	Ø	< 100	1		Same
Phosphoric Acid	1	Alkaline Liquid	10	Local consequences	0	Ø	. 0	Ø	i I		Same
Phosphoric Acid	1	Ammonium Nitrate Soln.	106	Same as individual chemicals	0	Ø	ı ø	Ø	I		Same
Phosphoric Acid	1	Anhydrous Ammonia	10	Mitigate NH3 release	Ø	4100	I Ø	3700	I		Mitigated
Phosphoric Acid	1	Aniline Oil	7	Evaporation of residuals at 50 C	j ø	100	i ø	200	•		Aggravated
Phosphoric Acid	. 1	Butadiene	28	Evaporation of residual H3PO4 at 50 C	0	800	. 0	800	•		Same
Phosphoric Acid	1	Butyl Acrylate	13	Local consequences	0	700+	Ø	700+	•		Same
Phosphoric Acid	1	Butyl Alcohol	4	Local consequences	. 0	< 100	. ø	< 100	i i		Same
Phosphoric Acid	1	Butyraldehyde	5	Evaporation of residuals at 50 C	Ø	⟨ 100	Ø	(100	i		Same
Phosphoric Acid	1	Chlorine	104	Same as individual chemicals	į ø	8000	. 0	8000	İ		Same
Phosphoric Acid	1	Chloroprene	17	Evaporation of residuals at 100 C	0	800] Ø	1700	İ		Aggravated
Phosphoric Acid	1	Coal Tar Distillate	101	Evaporation of residuals at 50 C] 0	< 100	< 100	< 100	i		Same
Phosphoric Acid	1	Combustible Liquid	101	Evaporation of residuals at 50 C		?	< 100	?	I		Same
Phosphoric Acid	1	Compound, Cleaning	101	Evaporation of residuals at 50 C	į ø	?	< 100	?	i I		Same
Phosphoric Acid	1	Crude Dil Petroleum	101	Evaporation of residuals at 50 C	j. ø	(100	(100	< 100			Same
Phosphoric Acid	1	Denatured Alcohol	4	Local consequences	į ø	(100	. 0	(100	i		Same
Phosphoric Acid	1	Distillate Fuel Oil	101	Evaporation of residuals at 50 C	0	< 100	< 100	(100	•		Same
Phosphoric Acid	1	Ethyl Acetate	13	Local consequences	j .	100	0	100	i		Same
Phosphoric Acid	1	Ethyl Acrylate	13	Local consequences	J Ø	100	. 0	100	[Same
Phosphoric Acid	1	Ethyl Alcohol	4	Local consequences	j ø	< 100	Ø	< 100	i		Same
Phosphoric Acid	1	Ethylene Oxide	34	Mitigate ethylene oxide release	0	2500	j ø	2300	i İ		Mitigated
Phosphoric Acid	1	Flammable Liquid	101	Evaporation of residuals at 50 C	. 0	?	(100	?	•		Same
Phosphoric Acid	1	Formaldehyde Solution	5	Evaporation of residuals at 50 C	į ø	100	. 6	200	j		Aggravated
Phosphoric Acid	1	Fuel Oil	101	Evaporation of residuals at 50 C	0	(100	< 100	(100	I		Same
Phosphoric Acid	1	Fuel, Aviation	101	Evaporation of residuals at 50 C	I Ø	(100	< 100	(100	İ		Same
Phosphoric Acid	1	Gasoline	101	Evaporation of residuals at 50 C	. 0	< 100	(100	(100	ĺ		Same
Phosphoric Acid	1	Gasoline, nec	101	Evaporation of residuals at 50 C	I Ø		(100	(100	ĺ		Same
Phosphoric Acid	1	Glycol Ethers	14	Local consequences	0	⟨ 100	•	(100			Same
Phosphoric Acid	1	Hexamethylene Diamine Solr		Evaporation of residuals at 50 C	i ø	< 100	•	< 100	•		Same
Phosphoric Acid	1	Hydrogen Peroxide	104	Evaporation of residuals at 50 C	1 0	200	•	400	, 		Aggravated
Phosphoric Acid	1	Isopropanol	4	Local consequences	1 0	< 100		⟨ 100	, 		ISame
Phosphoric Acid	1	Maleic Anhydride	107	Evaporation of residuals at 100 C	1 0	900+	•	20700*	•		Aggravated

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Summary of Toxic Emissio	n Conse	quence Calculations			•	nixed	Mix		•	ces for Gas	1
		-		•	Distan	ce (m)	Distan	ice (m)	F	Releases	1
	ASTM		ASTM	,	IDLH	IDLH	IDLH	IDLH	IDLH	IDLH IDLH	•
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences] #1	#2	#1	#2		Gas 2 Gas 3	Conclusion
Phosphoric Acid	1	Methyl Alcohol	4	Local consequences	1. 0	100	0	100	1		Same
Phosphoric Acid	1	Methyl Ethyl Ketone	19	Evaporation of residuals at 50 C	j Ø	200	0	300	.1		Aggravated
Phosphoric Acid	1	Methyl Methacrylate	13	Local consequences	Ø	100	0	100	1		Same
Phosphoric Acid	1	Motor Fuel Antiknock Compo	24	Releases ethane gas **	0	100	0	100	< 100		Same
Phosphoric Acid	1	Octyl Alcohol	4	Local consequences	Ø	< 100	0	(100	1		Same
Phosphoric Acid	1	0i1	101	Evaporation of residuals at 50 C	Ø	< 100	< 100	〈 100			Same
Phosphoric Acid	1	Petroleum Naphtha	101	Evaporation of residuals at 50 C	0	(100	< 100	< 100	1		Same
Phosphoric Acid	1	Petroleum Residual Fuel	101	Evaporation of residuals at 50 C	0	< 100	< 100	< 100	1		Same
Phosphoric Acid	1	Petroleum, Refined	101	Evaporation of residuals at 50 C	0	< 100	< 100	< 100	1		Same
Phosphoric Acid	1	Pheno I	31	Same as individual chemicals	Ø	(100] 0	< 100	1		Same
Phosphoric Acid	1	Potassium Hydroxide, Liq	10	Local consequences	l Ø	0	0	Ø	1		Same
Phosphoric Acid	1	Propylene Oxide	34	Mitigate propylene oxide release	Ø	1700	0	1500	1		Mitigated
Phosphoric Acid	1	Pulp Mill Liquid	101	Evaporation of residuals at 50 C	0	< 100	< 100	?	1		Same
Phosphoric Acid	1	Sodium Hydroxide, Liq.	10	Local consequences	0	Ø	0	Ø	1		Same
Phosphoric Acid	1	Sodium Metal	21	Releases H2 gas **	0	Ø	0	Ø	< 100		Aggravated
Phosphoric Acid	1	Styrene Monomer	16	Evaporation of residuals at 100 C	Ø	< 100	1 0	200	1		Aggravated
Phosphoric Acid	1	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	B	< 100	Ø	800	1		Aggravated
Phosphoric Acid	1	Vinyl Acetate	13	Local consequences	0	4100+	Ø	4100+	1		Same
Phosphoric Acid	1	Vinyl Chloride	17	Mitigate release of vinyl chloride	į Ø	19100+	0	18100+	ł		Mitigated
Phosphorus	105	Acetic Anhydride	107	Unknown	0	< 100	?	?	1		Unknown
Phosphorus	105	Annonium Nitrate Soln.	106	Explosion	0	Ø	1 0	. 0	1		Same
Phosphorus	105	Maleic Anhydride	107	Evaporation of residuals at 100 C] Ø	900*	[Ø	20700*	1		Aggravated
Potassium Hydroxide, Liq	10	Acetic Anhydride	107	Evaporation of residuals at 100 C	Ø	〈 100	j ø	400	1		Aggravated
Potassium Hydroxide, Liq	10	Acetone	19	Evaporation of residuals at 50 C	0	100	0	600	1		Aggravated
Potassium Hydroxide, Liq		Acrylonitrile	26	Evaporation of residuals at 100 C] 0	200	Ø	900	I	-	Aggravated
Potassium Hydroxide, Liq	10	Ammonium Nitrate Soln.	106	Local consequences	0	Ø	Ø	Ø	ļ		Same
Potassium Hydroxide, Liq	10	Carbon Tetrachloride	17	Releases HCL and phosgene	0	600	Ø	1500	900	2100	Aggravated
Potassium Hydroxide, Liq	10	Chloroprene	17	Evaporation of residuals at 100 C	0	800	Ø	1700	i		Aggravated
Potassium Hydroxide, Liq		Dinitrotoluene, Liquid	27	Evaporation at 100 C; Explosion	į ø	< 100	0	< 100	(100		Same
Potassium Hydroxide, Liq		Ethyl Acetate	13	Releases acetic acid and ethanol	į ø	100	Ø	500	300	< 100	Aggravated
Potassium Hydroxide, Liq		Ethylene Oxide	34	Evaporation of residuals at 100 C **	Ø	2500	0	1600	l		Mitigated
Potassium Hydroxide, Liq		Maleic Anhydride	107	Evaporation of residuals at 100 C	įø	900+	j ø	20700+	1		Aggravated
otassium Hydroxide, Liq		Methyl Chloride	17	Mitigate CH3CL; release methanol	i ø	1000	I Ø	1000	100		Same

Summary of Toxic Emission	CONSE	dance celculations		•	Non-m		Mixe		•	ces for Gas	1
ı	ASTM		ASTM		Distanc	` '	Distance	• •	•	eleases	
	поир Эгоир	Chemical Name #2		Tania Fairaira Casassas	IDLH		IDLH		•	IDLH IDLH	•
CHEMICAL NAME #1		Chemical Name #2	Group	Toxic Emission Consequences	#1 -	#2 	#1 	#2	Gas 1 	Gas 2 Gas 3	Conclusions
Potassium Hydroxide, Liq	10	Methyl Ethyl Ketone	19	Evaporation of residuals at 50 C	, 0	200	, 0	300	! 		Aggravated
Potassium Hydroxide, Liq	10	Phosphorus	105	Releases PH3 gas	1 0	Ø	Ø	Ø	800		Aggravated
Potassium Hydroxide, Liq	10	Propylene Oxide	34	Evaporation of residuals at 100 C **	Ø	1700	Ø	1100	l		Mitigated
Potassium Hydroxide, Liq	10	Sodium Metal	21	Releases H2 gas	Ø	Ø	j ø	Ø	(100		Same
Potassium Hydroxide, Liq	10	Styrene Monomer	16	Evaporation of residuals at 100 C	Ø	< 100	0	200	ì		Aggravated
Potassium Hydroxide, Liq	10	Sulfur Dioxide	105	Mitigate release of SO2	0	4500	0	4100		•	Mitigated
Potassium Hydroxide, Liq	10	Toluene Diisocyanate	18	Evaporation of residuals at 100 C] Ø	< 100	8	600			Aggravated
Potassium Hydroxide, Liq	10	Viny! Acetate	13	Releases acetic acid and acetaldehyde	Ø	4100+	1 0	5700+	300	900	Aggravated
Potassium Hydroxide, Liq	10	Vinyl Chloride	17	Release acetylene gas	Ø	19100+	j Ø	18100+]		Mitigated
Propylene Oxide	34	Chlorine	164	Evaporation of residuals at 50 C **	1700	8000	1300	7600	1		Mitigated
Propylene Oxide	34	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	1700	Ø	1300	Ø			Mitigated
Propylene Oxide	34	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	1700	200	[1100	1300			Mitigated
Propylene Oxide	34	Phosphorus	105	Evaporation of residuals at 100 C	1700	Ø	1100	ß	1		Mitigated
Pulp Mill Liquid	101	Chlorine	184	Evaporation of residuals at 50 C **	(100	8000	?	7600	1		Mitigated
Pulp Mill Liquid	101	Hydrogen Peroxide	104	Evaporation at 50 C; explosion **	< 100	200	?	400			Aggravated
Sodium Hydroxide, Liq.	10	Acetic Anhydride	107	Evaporation of residuals at 100 C	ļ Ø	< 100	0	400	1		Aggravated
Sodium Hydroxide, Liq.	10	Acetone	19	Evaporation of residuals at 50 C	Ø	100	j ø	600	İ		Aggravated
Sodium Hydroxide, Liq.	10	Acrylonitrile	26	Evaporation of residuals at 100 C	Į Ø	200	Ø	900			Aggravated
Sodium Hydroxide, Liq. 🧢	10	Ammonium Nitrate Soln.	108	Local consequences	Ø	. Ø	9	Ø			Same
Sodium Hydroxide, Liq.	10	Carbon Tetrachloride	17	Releases HCL and phosgene	Ø	600	j Ø	1500	900	2100	Aggravated
Sodium Hydroxide, Liq.	10	Chloroprene	17	Evaporation of residuals at 100 C	1 0	800	ļØ	1700			Aggravated
Sodium Hydroxide, Liq.	10	Dinitrotoluene, Liquid	27	Evaporation at 100 C; Explosion	1 0	< 100	0	< 100	l .		Same
Sodium Hydroxide, Liq.	10	Ethyl Acetate	13	Releases acetic acid and ethanol	j ø	100	j ø	500	300	< 100	Aggravated
Sodium Hydroxide, Liq.	10	Ethylene Oxide	34	Evaporation of residuals at 100 C **	0	2500	Ø	1600			Mitigated
Sodium Hydroxide, Liq.	10	Maleic Anhydride	107	Evaporation of residuals at 100 C	ļ Ø	900+	0	20700+			Aggravated
Sodium Hydroxide, Liq.	10	Methyl Chloride	17	Mitigate CH3CL; release methanol	l Ø	1000	Ø	1000	100		Same
Sodium Hydroxide, Liq.	10	Methyl Ethyl Ketone	19	Evaporation of residuals at 50 C	ļ Ø	200	0	300			Aggravated
Sodium Hydroxide, Liq.	10	Phosphorus	1Ø5	Releases PH3 gas	0	Ø	Ø	Ø	800		Aggravated
Sodium Hydroxide, Liq.	10	Propylene Oxide	34	Evaporation of residuals at 100 C **	j ø	1700	0	1100			Mitigated
Sodium Hydroxide, Liq.	10	Styrene Monomer	16	Evaporation of residuals at 100 C	0	< 100	0	200			Aggravated
Sodium Hydroxide, Liq.	10	Sulfur Dioxide	105	Mitigate release of SO2	0	4500	Ø	4100			Mitigated
Sodium Hydroxide, Liq.	10	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	Ø	(100	jø	800	ı		Aggravated
odium Hydroxide, Liq.	10	Vinyl Acetate		Releases acetic acid and acetaldehyde	. 6	4100+	Ø	5700+	300	900	Aggravated
				· .							

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Summary of Toxic Emissic	on Conse	quence Calculations		•	•	aixed	Mix		•	ces for	Gas	ļ.
	ASTM	•	ASTM	,	Distand	ce (m) IDLH	Distan	ce (m) IDLH	F	Releases IDLH	IDLH	
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1	#2	#1	#2	•			 Conclusions
Sodium Hydroxide, Liq.	10	Vinyl Chloride	17	Release acetylene gas	- Ø	19100*	 Ø	18100+				- Mitigated
Sodium Metal	21	Acetic Anhydride	107	Evaporation of residuals at 100 C	0	< 100	jø	400	I			Aggravated
Sodium Metal	21	Acrylonitrile	26	Evaporation of residuals at 100 C **	0	200	j ø	900	ĺ			Aggravated
Sodium Metal	21	Ammonium Nitrate Soln.	106	Releases hydrogen gas; explosion **	0	Ø	1 0	Ø	< 100			Same
Sodium Metal	21	Butadiene	28	Mitigate release of butadiene] 0	800	9	800	i			Same
Sodium Metal	21	Chlorine	104	Fire	6	8000	0	8000	i			Sane
Sodium Metal	21	Dinitrotoluene, Liquid	27	Evaporation at 100 C; Explosion] Ø	< 100	1 0	(100	1			Same
Sodium Metal	21	Ethylene Oxide	34	Mitigate release of ethylene oxide	1 8	2500	0	1600	1			Mitigated
Sodium Metal	21	Ferric Chloride Solution	104	Releases hydrogen gas **	0	Ø	0	Ø	< 100			Same
Sodium Metal	21	Hydrogen Peroxide	164	Releases hydrogen gas **	0	200	0	1300	< 100			Aggravated
Sodium Metal	21	Maleic Anhydride	107	Evaporation of residuals at 100 C	[0	900+	0	20700*				Aggravated
odium Metal	21	Phosphorus	105	Fire; releases P205 mist	0	Ø	0	Ø	?			Same
Sodium Metal	21	Propylene Oxide	34	Mitigate release of propylene oxide	j ø	1700	0	1100	I			Mitigated
Sodium Metal	21	Pulp Mill Liquid	101	Releases hydrogen gas	0	< 100	Ø	?	< 100			Same
Sodium Metal	21	Sulfur Dioxide	105	Fire	[0	4500	0	4500	1			Same
Styrene Monomer	16	Chlorine	184	Mitigation of chlorine release **	< 100	8000	< 100	7600	ĺ			Mitigated
Styrene Monomer	16	Ferric Chloride Solution	104	Evaporation of residuals at 50 C	< 100	Ø	< 100	Ø	i			Same
Styrene Monomer	16	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	(100	200	200	1300	1			Aggravated
Styrene Monomer	16	Sodium Metal	21	Evaporation of residuals at 50 C	< 100	. Ø	< 100	ø	İ			Same
Sulfur Dioxide	105	Acetic Anhydride	107	Unknown	4500	< 100	?	?	1			JUnknown
Sulfuric Acid	2	Aceta I dehyde	5	Evaporation of residuals at 100 C **	0	1100	j ø	700	ĺ			Mitigated
Sulfuric Acid	2	Acetic Acid, Glacial	3	Evaporation of residuals at 100 C **	. 0	100	1 0	1300	İ			Aggravated
Sulfuric Acid	2	Acetic Anhydride	107	Evaporation of residuals at 100 C **	jø	< 100	Ø	400	Ì			Aggravated
Sulfuric Acid	2	Acetone	19	Evaporation of residuals at 100 C	0	100	1 0	500	i			Aggravated
Sulfuric Acid	2	Acrylic Acid	3	Evaporation of residuals at 100 C	Ø	800+	1 0	6300+	l			Aggravated
Sulfuric Acid	2	Acrylonitrile	26	Releases NO2, SO2 and CO gas **	0	200	jø	900	600	1100	400	Aggravated
ulfuric Acid	2	Alcoholic Beverage	4	Releases diethyl ether **	1 6	< 100	0	< 100	(100			Same
ulfuric Acid	2	Alkaline Liquid	10	Local consequences	0	Ø	Ø	Ø	l			Same
ulfuric Acid	2	Ammonium Nitrate Soln.	106	Explosion	j ø	Ø	j ø	Ø	l			Same
Sulfuric Acid	2	Anhydrous Ammonia	10	Mitigate release of NH3	Ø	4100	Ø	3700				Mitigated
ulfuric Acid	2	Aniline Oil	7	Releases NOx, SO2 and CO gases **	1 0	100	Ø	700	500	1100	400	Aggravated
ulfuric Acid	2	Benzene	16	Evaporation of residuals at 100 C **	. 6	200	j 6	1000	I			Aggravated
Sulfuric Acid	2	Butadiene		Mitigate release of butadiene **	. 0	800	1 6	800	•			Same

Summary of Toxic Emissi	on Conse	quence Calculations			•	nixed	Mixe		•	ices for		1
					Distan	٠,	Distanc	• •	•	Releases		!
Chemical Name #1	ASTM Group		ASTM Group	Toxic Emission Consequences	IDLH #1	IDLH #2	IDLH #1	IDLH ∦2	IDLH Gas 1			 Conclusion
Sulfuric Acid	2	Butyl Acrylate	13	Evaporation of residuals at 100 C **	Ø	700+	Ø	3000*				Aggravat
Sulfuric Acid	2	Butyl Alcohol	4	Releases dibutyl ether **	0	< 100	0	600	< 100			Aggravat
Sulfuric Acid	2	Butyraldehyde	5	Evaporation of residuals at 100 C **	0	< 100	į Ø	< 100	1			Same
Sulfuric Acid	2	Carbon Tetrachloride	17	Releases COCL2 and HCL gas	Ø	600	j Ø	1500	2700	1900		[Aggravat
Sulfuric Acid	2	Chloroprene	17	Releases HCL, SO2 and CO gas **	0	800	Ø	2600	700	1100	400	Aggravat
Sulfuric Acid	2	Coal Tar Distillate	101	Same as individual chemicals	0	< 100	1 0	< 100]			Same
Sulfuric Acid	2	Combustible Liquid	101	Same as individual chemicals	0	< 100	0	?	1			Same
Sulfuric Acid	2	Compound, Cleaning	101	Same as individual chemicals	Ø	< 100	j ø	?	ı			Same
Sulfuric Acid	2	Crude Gil Petroleum	101	Same as individual chemicals	Ø	< 100	Ø	< 100	1			Same
Sulfuric Acid	2	Denatured Alcohol	4	Releases diethyl ether **	0	(100	0	< 100	< 100			Same
Sulfuric Acid	2	Dinitrotoluene, Liquid	27	Evaporation at 100 C; explosion **	0	Ø	0	< 100	ĺ			Same
Sulfuric Acid	2	Distillate Fuel Oil	101	Same as individual chemicals	j ø	< 100	0	〈 100	,			Same
Sulfuric Acid	2	Ethyl Acetate	13	Evaporation of residuals at 50 C	Ø	100		200	i			Aggravat
Sulfuric Acid	2	Ethyl Acrylate	13	Evaporation of residuals at 100 C	. 0	100	Ø	900	i			Aggravat
Sulfuric Acid	2	Ethyl Alcohol	4	Releases diethyl ether **		< 100	. 0	< 100	< 100			Same
Sulfuric Acid	2	Ethylene Oxide	34	Evaporation of residuals at 100 C **	j ø	2500		2200	i I			Mitigate
Sulfuric Acid	2	Flammable Liquid	101	Same as individual chemicals	jø	< 100	Ø	?	i i			Same
Sulfuric Acid	2	Formaldehyde Solution	5	Evaporation of residuals at 100 C	. 0	100	•	500	•			Aggravat
Sulfuric Acid	2	Fuel Dil	101	Same as individual chemicals	. 0	(.100	ø	< 100	i			Same
Sulfuric Acid	2	Fuel, Aviation	101	Same as individual chemicals		(100	•	< 100	•			Same
Sulfuric Acid	2	Gasoline	101	Same as individual chemicals	•	(100	•	< 100	•			Same
Sulfuric Acid	2	Gasoline, nec	101	Same as individual chemicals	•	.⟨ 100	•	< 100	•			Same
Sulfuric Acid	ź	Glycol Ethers		Evaporation of residuals at 100 C **	ĺ	⟨ 100	• •	•	< 100			Aggravat
Sulfuric Acid	2	Hexamethylene Diamine Soln		Evaporation of residuals at 100 C	Ø	< 100	•		• •			Same
Sulfuric Acid	2	Hydrogen Peroxide	104	Evaporation at 100 C; Explosion	1 0	200	•	1300	' I			Aggravat
Sulfuric Acid	2	Isopropanol	4	Evaporation of residuals at 100 C **	1 8	(100		500	! 			Aggravat
Sulfuric Acid	2	Maleic Anhydride	107	Evaporation of residuals at 100 C	1 6	900+	•	20700*	•			Aggravat
Sulfuric Acid	2	Methyl Alcohol	4	Releases dimethyl ether **	1 0	⟨ 100	•		 < 100			Aggravat
Sulfuric Acid	2	Methyl Chloride	17	Releases HCL, SO2 and CO gas **	1 0	1000	•	1000	•	1100	100	Aggravat
Sulfuric Acid	2	Methyl Ethyl Ketone	19	Evaporation of residuals at 100 C	1 8	200	•	900		1100	700	Aggravat
Sulfuric Acid	2	Methyl Methacrylate	13	Evaporation of residuals at 100 C **	1 0	100	•	700				
Sulfuric Acid	2	Motor Fuel Antiknock Compo		Releases ethane gas	8	100			 < 100			Aggravat
	_	•		Evaporation of residuals at 100 C **	1 0	(100						Same
Sulfuric Acid	2	Octyl Alcohol	4	CASHOLSON OL LESIGNS ST TAN (**	ן ט	100	ן ש	< 100	ממנ) ן			Same

iummary of loxic Eniss	ion Conse	equence Calculations			Non	-mixed	•	ixed		Distan	ces for G	as .	1
					Dista	nce (m)	Dist			l R	eleases		
	ASTM		ASTM	,	IDLH	IDLH	IDLI	н :	IDLH	IDLH	IDLH I	DLH	!
Chemical Name #1	Group	Chemical Name #2	Group	Toxic Emission Consequences	#1	. ∦2	#1		#2	Gas 1	Gas 2 Ga	s 3	Conclusion
Sulfuric Acid	2	oil	101	Same as individual chemicals	, 	Ø < 10	- : 8	ø	(100	i			Same
Gulfuric Acid	2	Petroleum Naphtha	101	Same as individual chemicals	1	0 < 10	3	ø	(100	l			Same
iulfuric Acid	2	Petroleum Residual Fuel	101	Same as individual chemicals	1	0 (10	7]	Ø	(100	i			Same
Sulfuric Acid	2	Petroleum, Refined	101	Same as individual chemicals	1	Ø < 10	ð)	Ø	(100	1			Same
Sulfuric Acid	2	Pheno I	31	Evaporation of residuals at 100 C	1	0 < 10	9	Ø	700				Aggravated
Sulfuric Acid	2	Phosphoru s	105	Releases H3P04 mist **	1	Ø	ð	0	. 0	l			Aggravated
Sulfuric Acid	2	Potassium Hydroxide, Liq	10	Local consequences	1	Ø .	ð <u>1</u>	Ø	Ø	l			Same
Sulfuric Acid	2	Propylene Oxide	34	Evaporation of residuals at 100 C **	!	Ø 17Ø	ð [Ø	1500	l			Mitigated
Sulfuric Acid	2	Pulp Mill Liquid	161	Same as individual chemicals	1	Ø < 10	3	Ø	?	[Same
Sulfuric Acid	2	Sodium Hydroxide, Liq.	10	Local consequences	ŀ	Ø	3	Ø	Ø	[Same
iulfuric Acid	2	Sodium Metal	21	Releases SO2 and hydrogen gas **	Ì	Ø	3	Ø	Ø	1500	< 100		Aggravated
Sulfuric Acid	. 2	Styrene Monomer	16	Evaporation of residuals at 100 C **	i	0 (10	3]	Ø	200	1			Aggravated
ulfuric Acid	2	Toluene	16	Evaporation of residuals at 100 C **	1.	Ø 10	3 j	Ø	900	< 100			Aggravated
ulfuric Acid	2	Toluene Diisocyanate	18	Evaporation of residuals at 50 C	ĺ	Ø < 10	7	Ø	600				Aggravated
ulfuric Acid	2	Vinyl Acetate	13	Evaporation of residuals at 100 C **	İ	0 4100	, j	Ø	5700+				Aggravated
Sulfuric Acid	2	Vinyl Chloride	17	Releases HCL, S02 and C0 gas **	Ì	Ø 1910Ø	ı İ	Ø 19	9000+	800	1100		Aggravated
Sulfuric Acid	2	Xylene	16	Evaporation of residuals at 100 C **	i	Ø (10	7	Ø	100				Aggravated
ulfuric Acid, Spent	2	Acetaldehyde	5	Evaporation of residuals at 100 C	i	Ø 110)	Ø	700	Ì			Witigated
ulfuric Acid, Spent	2	Acetic Acid, Glacial	3	Evaporation of residuals at 100 C	ì	Ø , 9	5	Ø	1300]			Aggravated
ulfuric Acid, Spent	. 2	Acetic Anhydride	107	Evaporation of residuals at 100 C	i	0 < 10	5 j	Ø	400	ĺ			Aggravated
Sulfuric Acid, Spent	2	. Acetone	19	Evaporation of residuals at 50 C	i	Ø 10	5 j	Ø	600				Aggravated
Sulfuric Acid, Spent	2	Acrylic Acid	3	Evaporation of residuals at 100 C	i	Ø 8ØØ:	, i	ø e	300*	ĺ			Aggravated
Sulfuric Acid, Spent	2	Acrylonitrile	26	Releases NO2 gas **	i	Ø 201	ı i	ø	900	1000			Aggravated
Sulfuric Acid, Spent	2	Alcoholic Beverage	4	Releases diethyl ether	i	Ø < 100	•	ø	100	< 100			Same
Sulfuric Acid, Spent	2	Alkaline Liquid	10	Local consequences	i	•	; ;	Ø	Ø	, ,	•		Same
ulfuric Acid, Spent	2	Ammonium Nitrate Soln.	106	Local consequences	i		i	Ø	Ø				Same
ulfuric Acid, Spent	2	Anhydrous Ammonia		Mitigate release of NH3	i	Ø 41Ø	•	Ø	3700				Mitigated
ulfuric Acid, Spent	2	Aniline Oil	7	Evaporation of residuals at 100 C	•	0 106		Ø	700				Aggravated
ulfuric Acid, Spent	. 2	Benzene	16	Evaporation of residuals at 50 C	•	Ø 200	•	Ø	400				Aggravated
ulfuric Acid, Spent	2	Butadiene		Mitigate release of butadiene	•	Ø 8Ø6	•	Ø	800				Same
ulfuric Acid, Spent	2	Butyl Acrylate	13	Evaporation of residuals at 50 C	•	0 700:	•	ø 1	500+				Aggravated
ulfuric Acid, Spent	2	Butyl Alcohol		Releases dibutyl ether **	•	0 (10)	•	g ·		< 100			Aggravated
ulfuric Acid, Spent	2	Butyraldehyde		Evaporation of residuals at 100 C		0 (100	•	-	100				Same

Summary of Toxic Emissi	on Conse	quence Calculations			•	mixed	Mix		Distances for Gas	!
					Distan		Distan		Releases	Ī
Chemical Name #1	ASTM Group		ASTM Group	Toxic Emission Consequences	IDLH #1	IDLH #2	IDLH #1	IDLH #2	IDLH IDLH IDLH Gas 1 Gas 2 Gas 3	•
 Sulfuric Acid, Spent	2	Carbon Tetrachloride	 17	Releases HCL gas	 0 1	600		1500		 Aggravated
Sulfuric Acid, Spent	. 2	Chloroprene		Evaporation of residuals at 100 C	1 8	800	•		1	Aggravated
Sulfuric Acid, Spent	2	Coal Tar Distillate		Same as individual chemicals	1 0	1 < 100			, 	Same
Sulfuric Acid, Spent	2	Combustible Liquid		Same as individual chemicals	1 6	1 (100		?	, 1	Same
Sulfuric Acid, Spent	2	Compound, Cleaning		Same as individual chemicals		1 (100		?	•	Same
Sulfuric Acid, Spent	2	Crude Oil Petroleum		Same as individual chemicals		1 (100			i	Same
Sulfuric Acid, Spent	2	Denatured Alcohol		Releases diethyl ether **	0	(100	•	-	 < 100	Same
Sulfuric Acid, Spent	2	Dinitrotoluene, Liquid		Evaporation at 100 C; explosion **	j 0) 0		-	, , 	Same
Sulfuric Acid, Spent	2	Distillate Fuel Dil		Same as individual chemicals.	0			•	İ	Same
Sulfuric Acid, Spent	2	Ethyl Acetate	13	Evaporation of residuals at 50 C		100	. 6	266		Aggravated
Sulfuric Acid, Spent	2	Ethyl Acrylate	13	Evaporation of residuals at 50 C	0	100	0	300	İ	Aggravated
Sulfuric Acid, Spent	2	Ethyl Alcohol	4	Releases diethyl ether **	jø	(100	0	< 100	< 100	Same
Sulfuric Acid, Spent	2	Ethylene Oxide	34	Evaporation of residuals at 100 C **	0	2500	9	2200		Mitigated
Sulfuric Acid, Spent	2	Flammable Liquid	101	Same as individual chemicals	0	< 100] Ø	?	l	Same
Sulfuric Acid, Spent	2	Formaldehyde Solution	5	Evaporation of residuals at 50°C	0	100	[Ø	200	 	Aggravated
Sulfuric Acid, Spent	2	Fuel Dil	101	Same as individual chemicals	0	< 100 ,	8	< 100	1	Same
Sulfuric Acid, Spent	2	Fuel, Aviation	161	Same as individual chemicals	ļ Ø	(100	Ø	< 100	1	Same
Sulfuric Acid, Spent	2	Gasoline	161	Same as individual chemicals	0	(100	.0	< 100	< 100	Same
Sulfuric Acid, Spent	2	Gasoline, nec		Same as individual chemicals	0	< 100	0		< 100	Same
Sulfuric Acid, Spent	2	Glycol Ethers		Evaporation of residuals at 50 C] 0	< 100	. 0	100	< 100 ·	Aggravated
Sulfuric Acid, Spent	2	Hexamethylene Diamine Soln		Evaporation of residuals at 100 C	Ø	< 100	0	•	l	Same
Sulfuric Acid, Spent	2	Isopropanol	4	Evaporation of residuals at 100 C **	0	< 100	•	500	l	Aggravated
Sulfuric Acid, Spent	2	Maleic Anhydride	107	Evaporation of residuals at 100 C	0	900+	•	20700+	•	Aggravated
Sulfuric Acid, Spent	2	Methyl Alcohol		Releases dimethyl ether **	0	< 100	•		< 100	Aggravated
Sulfuric Acid, Spent	2	Methyl Chloride		Releases S02 gas **	0		•		1100	Aggravated
Sulfuric Acid, Spent	2	Methyl Ethyl Ketone		Evaporation of residuals at 50 C	Ø	200		.300		Aggravated
Sulfuric Acid, Spent	2	Methyl Methacrylate		Evaporation of residuals at 50 C	0	100	•	200	•	Aggravated
Sulfuric Acid, Spent	2	Motor Fuel Antiknock Compo		Releases ethane gas	0	100	•		< 100	Same
Sulfuric Acid, Spent	2	Octyl Alcohol	4	Evaporation of residuals at 100 C **	0	,	•	< 100	•	Same
Sulfuric Acid, Spent	2	Oil		Same as individual chemicals	0	•	-	< 100	•	Same
Sulfuric Acid, Spent	2	Petroleum Naphtha		Same as individual chemicals	0		•	< 100	• •	Same
Sulfuric Acid, Spent	2	Petroleum Residual Fuel		Same as individual chemicals	Ø	•	•	< 100	•	Same
Sulfuric Acid, Spent	2	Petroleum, Refined	101	Same as individual chemicals	0	< 100	l Ø	< 100	(100	Same

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											•
1											
Summary of Toxic Emissi	ion Conse	quence Calculations			Nor	ixed	 Mix	ed	l Dista	nces for Gas	 1
-		•		•	•	nce (m)	Distan		•	Releases	i
	ASTM		ASTM		j IDLI		IDLH	IDLH	IDLH	IDLH IDL	ні
Chemical Name #1	Group	Chemical Name ∦2	Group	Toxic Emission Consequences	#1	#2	#1	#2	Gas 1		3 Conclusions
Sulfuric Acid, Spent	2	Phenoi	31	Evaporation of residuals at 50 C	 	0 < 100		200	 		 Aggravated
Sulfuric Acid, Spent	2	Phosphorus		Same as individual chemicals **	i	0 0	•				Same
Sulfuric Acid, Spent	2	Potassium Hydroxide, Liq	10	Local consequences	i	ø · ø	0	Ø	İ		Same
Sulfuric Acid, Spent	2	Propylene Oxide	34	Evaporation of residuals at 100 C **	j	0 1700	0	1500	İ		Mitigated
Sulfuric Acid, Spent	2	Pulp Will Liquid	101	Same as individual chemicals	1	0 < 100	0	?	1		Same
Sulfuric Acid, Spent	2	Sodium Hydroxide, Liq.	10	Local consequences	1	Ø Ø	0	Ø	1		Same
Sulfuric Acid, Spent	2	Sodium Metal	21	Releases hydrogen gas **	1	ø ø	0	Ø	< 100		Same
iulfuric Acid, Spent	2	Styrene Monomer	16	Evaporation of residuals at 50 C	1	0 < 100) 9	< 100	ì		Same
ulfuric Acid, Spent	2	Toluene	16	Evaporation of residuals at 50 C	1	0 100	jø	200	< 100		Aggravated
ulfuric Acid, Spent	2	Toluene Diisocyanate	18	Evaporation of residuals at 100 C	1	Ø < 100	0	800	İ		Aggravated
ulfuric Acid, Spent	2	Vinyl Acetate	13	Evaporation of residuals at 50 C	1	Ø 4100*	0	8000+	ĺ		Aggravated
ulfuric Acid, Spent	2	Vinyl Chloride	17	Mitigate release of vinyl chloride	Ì	Ø 191ØØ*	. 6	18199+	Ì		Mitigated
ulfuric Acid, Spent	2	Xylane	16	Evaporation of residuals at 50 C	Ì	Ø < 100	. 0	< 100	Ī		Same
o l uene	16	Chlorine	104	Mitigation of chlorine release **	1 10	0 8000	200	7600	6900+		Mitigated
o luene	16	Hydrogen Peroxide	164	Evaporation at 100 C; explosion **	10	0 200	900	1300	i		Aggravated
Toluene Diisocyanate	18	Ammonium Nitrate Soln.	106	Evaporation of residuals at 50 C	< 10	ø ø	100	Ø	į		Aggravated
oluene Diisocyanate	18	Chlorine	184	Mitigation of chlorine release **	< 10	0 8000	100	7600	ĺ		Mitigated
Coluene Diisocyanate	18	Ethylene Oxide	34	Evaporation of residuals at 100 C	< 10	Ø 25 00	600	1600	i		Mitigated
oluene Diisocyanate	18	Hydrogen Peroxide	104	Evaporation at 100 C; explosion **	< 10	Ø 200	600	1300	İ		Aggravated
oluene Diisocyanate	18	Phosphorus	105	Evaporation of residuals at 50 C	< 10	ø ø	100	Ø	i		Aggravated
Toluene Diisocyanate	18	Sodium Metal	21	Evaporation at 50 C; release H2 gas	< 10		100	Ø	< 100		Aggravated
Toluene Diisocyanate	18	Sulfur Dioxide	105	Unknown	< 10		; ?	?	İ		Unknown
inyl Acetate	13	Chlorine	104	Mitigation of chlorine release **	4100	* 8000	8000+	7600	i		Mitigated
/inyl Acetate	13	Ferric Chloride Solution		Evaporation of residuals at 50 C	4100	* Ø	8000+	ø	i		Aggravated
inyl Acetate	13	Hydrogen Peroxide		Evaporation at 100 C; explosion **	4100		5700+	1300	Ī		Aggravated
inyl Acetate	13	Sodium Metal		Evaporation of residuals at 100 C **	4100	* Ø	5700+	Ø	i		Aggravated
inyl Chloride	17	Chlorine		Mitigation of chlorine release **	19100		16500+	6900	•		Mitigated
inyl Chloride	17	Ferric Chloride Solution		Evaporation of residuals at 100 C	19100		18100+	Ø			Mitigated
inyl Chloride	17	Hydrogen Peroxide		Evaporation at 50 C; explosion **	19100		18100+	400	•		Aggravated
inyl Chloride	17	Motor Fuel Antiknock Compo		Evaporation of residuals at 50 C	19100		18100+	100	•		Mitigated
inyl Chloride	17	Sodium Metal		Evaporation at 50 C; explosion	19100		18100+	Ø	•		Mitigated
y lene	16	Chlorine		Mitigation of chlorine release **	< 10		(100	7600	I		Mitigated
y lene	. 16	Hydrogen Peroxide		Evaporation at 100 C; explosion **	\ < 10		•	1300	1		Aggravated

* = Distance to the TLV (Threshold Limiting Value) reported because of unavailability of the IDLH value.

** = Assumes that the mixture and/or vapors do not ignite.

IDLH = Immediately Dangerous to Life and Health concentration.

Mitigated = Consequences of mixed chemicals are less than the consequences of the unmixed chemicals.

Aggravated = Consequences of mixed chemicals are greater than the consequences of the unmixed chemicals.

Same = Consequences of mixed chemicals are equivalent to the consequences of the unmixed chemicals.

? = Unknown consequences.

Note: All distances in meters from source of spill.

Table I3. Continuous Release Fireballs

											Fire				ance 1				
			Release	3						Dist	Ball					Dista		Area	Net
		Temp	•				Combust	•		to LFL				Pain	Leth	Pain	Leth	Covered	Area
Case	Chemical Name	K 	kg/sec	LFL.	Conc	Wt.	kJ/kg	(m)	kg	•	*	Fract	sec			П	M	sq.⊞.	sq.m.
25c	Acetic Acid	298	2.6	Ø.Ø54	Ø.Ø95	60.1	13118	1.73	71	19	22	Ø.2	3	15	4	26	15	167	
50c	Acetic Acid	298	9.0	Ø.Ø54	0.095	60.1	13110	2.84	791	26	47	Ø.2	7	47	12	70	35	988	821
25c	Acetic Anhydride	298	1.1	0.027	0.050	102	16420	1.49	3Ø	8	16	0.2	3	12	3	20	11	95	
5Øc	Acetic Anhydride	298	5.6	0.027	0.050	102	16429	2.55	4Ø8	21	38	Ø.2	6	39	10	58	29	649	554
100c	Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	11091	74	112	Ø.2	16	178	46	234	102	8145	8050
25c	Acetone	298	48.4	0.021	0.050	58.1	28500	8.40	19344	122	134	Ø.2	18	303	78	370	146	16644	
25c	Acrylic Acid	298	1.1	0.024	0.065	72.1	18840	1.62	39	11	18	Ø.2	3	14	3	23	12	118	
50c	Acrylic Acid	298	4.0	0.024	0.065	72.1	18840	2.49	315	23	35	Ø.2	5	37	9	54	27	559	441
25c	Acrylonitrile	298	13.2	0.030	Ø.Ø53	53.1	33200	4.67	2215	49	86	9.4	10	178	43	203	76	4573	
50c	Acrylonitrile	298	41.9	0.030	0.053	53.1	33200	5.95	14060	98	121	Ø.4	17	399	103	460	164	21000	16427
25c	Alcoholic Bev.	298	Ø.1	0.033	0.065	46.1	2688Ø	0.81	1	3	. 5	Ø.2	1	3	1	6	3	9	
50c	Alcoholic Bev.	298	Ø.3	0.033	0.065	46.1	2688Ø	1.17	5	5	9	₿.2	2	7	2	11	6	31	22
100c	Alcoholic Bev.	298	0.8	0.033	0.065	46.1	2688Ø	1.61	26	9	15	Ø.2	2	14	3	22	11	97	88
25c	Aniline	298	. Ø.2	0.013	0.026	93.1	34830	1.10	3	5	8	Ø.4	1	8	2	12	6	28	
50c	Aniline	298	Ø.6	0.013	0.026	93.1	34830	1.57	19	9	14	Ø.4	2	19	5	26	12	107	79
100c	Aniline	298	-9.4	0.013	0.026	93.1	34830	3.90	1532	48	59	Ø.4	9	147	37	176	67	3480	3452
25c	Benzene	298	17.6	0.013	0.027	78.1	40600	5.08	4645	77	84	Ø.4	12	264	68	307	110	9474	
50c	Benzene	298	56.3	0.013	0.027	78.1	40600	7.42	29850	155	155	Ø.4	21	625	162	702	240	45097	35623
25c	Butyl Acrylate	298	1.3	0.014	0.023	128	32240	1.85	51	11	19	Ø.2	3	21	5	30	15	171	
50c	Butyl Acrylate	298	6.1	0.014	0.023	128	32240	3.08	606	29	43	Ø.2	6	65	16	87	38	1136	965
100c	Butyl Acrylate	298	26.4	0.014	0.023	128	32240	5.00	6313	79	93	Ø. <u>2</u>	13	192	49	239	96	7217	7046
25c	Butyl alcohol	298	Ø.9	0.014	0.034	74.1	33100	1.84	39	13	18	8.2	3	19	5	27	13	142	
50c	Butyl alcohol	298	7.4	0.014	0.034	74.1	33100	3.69	1146	45	53	Ø.2	8	88	22	115	49	1891	1749
25c	Butyraldehyde	298	31.8	0.025	0.037	72.1	3538Ø	5.39	8481	78	103	Ø.2	14	231	59	282	111	9607	
25c	Carbon disulfide	298	68.6	0.013	0.065	76.1	14210	6.92	41592	177	173	Ø.4	23	431	112	517	198	3Ø928	
25c	Chloroprene	298	47.6	0.040	0.050	88.5	30000	5.07	10786	s 66	111	Ø.4	15	336	86	391	142	15824	
25c	Coal Tar Distill	298	Ø.7	0.013	0.030	100	40570	1.58	23	16	15	Ø.4	2	23	6	30	13	133	
5Øc	Coa! Tar Distill	298	2.7	0,013	0.030	100	40570	2.47	199	22	3Ø	0.4	5	52	15	77	30	730	597
100c	Coal Tar Distill	298	13.1	0.013	Ø.030	100	40570	4.16	2497	56	69	0.4	10	198	51	233	85	5668	5535
25c	Crude Oil	298	0.7	0.060	0.023	100	43500	1.28	9	4	11	Ø.4	2	15	4	21	9	67	
50c	Crude Oil	298	2.3	0.060	0.023	100	43500	1.90	62	8	21	0.4	3	37	9	47	19	297	230
100c	Crude Oil	298	13.4	0.060	0.023	100	43500	3.40	1034	23	52	0.4	8	137	35	162	60	2864	2797
25c	Cyclohexane	298	24.0	0.013	0.023	84.2	43459	5.61	7193	88	97	0.4	14	335	86	384	135	14235	
5Øc	Cyclohexane	298	61.8	0.013	Ø.Ø23	84.2	43459	7.66	32671	155	160	Ø.4	22	674	175	754	255	51002	36767
25c	Ethyl acetate	298	20.1	0.022	0.040	88.1	23510	4.37	3897	57	80	Ø.2	11	131	34	171	73	4224	
50c	Ethyl acetate	298	67.1	0.022	0.040	88.1	23510	8.50	26816	117	150	Ø.2	20	320	83	395	158	19547	15323
25c	Ethyl acrylate	298	7.7	0.018	0.034	100	27630	3.24	877	33	49	Ø.2	. 7	71	18	98	42	1418	
50c	Ethyl acrylate	298	28.5	Ø.Ø18	9.934	199	27630	5.00	7119	73	97	Ø.2	14	188	48	236	97	7347	5929
25c	Ethyl Alcohol	298	8.7	0.033	0.065	46.1	2688Ø	3.54	1180	40	54	0.2	8	81	20	108	47	1764	
5Øc	Ethyl Alcohol	298	26.1	0.033	0.065	46.1	26880	5.09	6845	77	96	0.2	13	182	47	230	95	7024	5260
25c	Formaldehyde	298	0.04	0.070	Ø.173	30.0	19035	0.52	Ø	1	. 3	0.2	! 1	1	Ø	3	2	3	
5Øc	Formaldehyde	298	. Ø.2	0.070	0.173	30.0	19035	Ø.88	2	3	7	Ø.2	! 1	4	1	7	4	15	13
100c	Formaldehyde	298	1.5	0.070	Ø.173	30.0	19035	1.72	58	11	20	0.2	. 3	17	4	27	14	150	157
25c	Fuel Oil	298	Ø.9	0.013	Ø.·023	100	43100	1.80	34	11	. 17	Ø.4	3	28	7	37	15	187	
50c	Fuel Oil	298	2.7	0.013	0.023	100	43100	2.58	199	22	30	Ø.4	- 5	64	16	79	31	753	565
100c	Fuel Oil	298	17.1	0.013	Ø.Ø23	100	43100	4.75	3824	85	79	Ø.4	11	249	64	289	103	8365	8178

Table I3. (Cont.) Continuous Release Fireballs

	20, (00,,01)	• • • • • • • • • • • • • • • • • • • •								,	Fire			Dista	ance î	Tota I			
		Amb.	Release	•			Heat.	Cloud	Vanor	Dist						Dista	nce	Area	Net
		Temp	Rate		Stoic	Mole	Combust		•	to LFL		Rad	Dur			Pain			Area
Case	Chemical Name	K	kg/sec	LFL.			kJ/kg	(m)	kg			Fract			#		E	sq.m.	sq.m.
25c	Fuel, Aviation	298	Ø.7	0.007	Ø.Ø23	100	43100	1.83	33	14	17	8.4	3	28	7	36	15	183	
5Øc	Fuel, Aviation	298	2.1	Ø.007	0.023	100	43100	2.63	193	27	30	0.4	5	63	16	78	31	734	552
100c	Fuel, Aviation	298	13.2	0.007	0.023	100	43100	4.83	3664	81	78	Ø.4	11	244	62	283	101	8073	7891
25c	Glycol Ethers	298	2.2	0.025	0.050	78.1	22000	2.09	114	15	25	Ø.4	4	35	9	48	21	356	
50c	Glycol Ethers	298			0.050		22000	3.11	797	31	47	0.4	7	86	22	110	45	1522	1266
100c	Glycol Ethers	298	58.6	0.025	0.050	76.1	22000	6.17	21834	109	140	0.4	19	398	103	468	173	23504	23147
25c	Hexane	298	32.1	0.012	0.022	86.2	44765	6.28	12014	110	115	0.4	16	431	111	488	169	22317	
50c	Hexane	298	93.8	Ø.Ø12	Ø.Ø22	86.2	44765	8.95	66803	208	202	Ø.4	27	953	249	1053	349	95880	73563
25c	Hexamethy lened ia	298	0.04	0.007	0.021	116	28400	Ø.69	ø	2	4	0.4	1	3	1	4	2	5	
50c	Hexamethy lenedia	298	Ø.2	Ø.007	0.021	116	28400	1.17	4	6	8	Ø.4	1	9	2	13	6	31	27
100c	Hexamethy lened ia	298	1.5	Ø.ØØ7	0.021	116	28400	2.28	103	20	24	0.4	4	38	. 9	5Ø	22	367	362
25c	Isopropanol	298	6.0	Ø.Ø23	0.044	60.1	30150	3.24	690	34	45	Ø.2	7	67	17	89	39	1221	
50c	Isopropanol	298	28.3	Ø.Ø23	8.844	60.1	30150	5.41	8252	85	102	Ø.2	14	210	54	261	105	8641	7420
25c	Maleic Anhydride	298	Ø.05	Ø. Ø14	Ø.Ø65	98.1	14200	Ø.58	Ø	2	4	Ø.4	1	2	Ø	4	2	4	•
50c	Maleic Anhydride		Ø.5	Ø. Ø14	Ø.Ø65	98.1	14200	1.24	13	8	12	0.4	2	10	3	16	9	59	55
100c	Maleic Anhydride	298	5.6	Ø.Ø14	0.065	98.1	14200	2.75	620	32	44	0.4	6	62	16	83	37	1097	1092
25c	Methyl alcohol	298	31.7	Ø.060	Ø.122	32.0	19580	5.00	8125	75	101	Ø.2	14	168	43	219	94	6914"	
50c	Methyl alcohol	298	36.8	0.060	Ø.122	32:0	19580	5.25	10315	82	109	0.2	15	188	48	242	1Ø3	8342	1428
25c	Methylethylketon	298	21.4	0.018	0.037	72.1	31360	5.00	5481	75	89	0.2	13	177	45	222	90	6355	
5øc	Methylethylketon	298	51.9	Ø.Ø18	Ø. Ø37	72.1	31360	6.69	22617	127	141	Ø.2	19	342	88	412	159	19905	13550
25c	Methylmethacryla		9.0	Ø.Ø21	0.034	100	26400	3.33	1026	33	-51	Ø.2	8	75	19	101	45	1570	
50c	Methylmethacryla	298	28.5	Ø.Ø21	0.034	100	26400	4.87	6490	67	94	Ø.2	13	176	45	223	92	6674	5103
25c	Petroleum Naphth	298	1.3	Ø.008	0.030	100	42400	2.10	83	19	23	0.4	4	42	10	53	. 22	371	
50c	Petroleum Naphth	298	2.6	0.008	0.030	100	42400	2.64	251	28	32	Ø.4	5	70	18	86	34	898	527
100c	Petroleum Naphth	298	16.7	Ø.Ø08	0.030	100	42400	4.88	4927	88	86	0.4	12	278	71	321	114	10221	985Ø
25c	Pheno I	298	Ø.1	0.017	Ø.Ø29	94.1	31170	∅.82	1	3	5	0.2	1	3	1	6	3	9	
50c	Phenol	298	0.9	0.017	0.029	94.1	31170	1.69	30	10	16	Ø.2	3	16	4	24	12	115	106
100c	Pheno I	298	9.3	0.017	0.029	94.1	31170	3.65	1274	40	55	0.2	8	90	23	118	50	1999	1991
25c	Styrene	298	1.3	Ø.Ø11	0.021	104	41960	2.10	67	15	21	0.4	3	38	9	48	20	312	
50c	Styrene	298	5.7	0.011	0.021	104	41960	3.42	712	37	46	0.4	7	113	29	136	51	2070	1758
100c	Styrene	298	39.6	Ø.Ø11	0.021	194	41960	6.48	15825	117	126	0.4	17	474	122	537	185	26970	26658
25c	Toluene	298	6.1	0.040	0.023	92.1	40550	2.89	393	19	38	0.4	6	84	21	1Ø3	40	1258	
5Øc	Toluene	298	23.0	Ø.040	0.023	92.1	40550	4.48	3289	42	75	0.4	11	225	58	263	95	7113	5856
25c	Toluene Diisocya	298	0.02	0.009	0.022	174	23900	Ø.46	. Ø	1	2	0.4	Ø	1	Ø	2	1	2	
50c	Toluene Diisocya	298	0.1	0.009	0.022	174	23900	Ø.72	1	2	5	0.4	1	3	1	6	3	. 8	6
100c	Toluene Diisocya	298	1.2	0.009	Ø. Ø22	174	23900	1.77	49	12	19	0.4	3	25	6	34	16	191	190
25c	Viny! acetate	298	32.6	Ø.Ø26	0.044	86.1	22690	4.94	7749	70	100	0.2	14	177	46	227	95	7138	
50c	Vinyl acetate	298	77.5	0.026	0.044	86.1	22690	6.57	30973	117	157	Ø.2	21	336	87	414	166	21542	14405
25c	Xylene	298	2.5	0.011	0.020	106	40841	2.60	188	22	3Ø	Ø.4	5	60	. 15	75	3Ø	699	
50c	Xylene	298	7.8	0.011	0.020	106	40841	3.78	1114	43	53	0.4	8	137	35	163	61	2936	2237
100c	Xylene	298	49.8	Ø.Ø11	0.020	106	40841	6.99	22550	132	141	0.4	19	551	143	621	213	35713	35015

Table I3. (Cont.) Puff Release Fireballs

1001	e 13. (Conc.) Fu	, ,,,	, ,							Fire			Dista	ance .	Total			
			Fuel				Heat	Cloud	Vapor	Ball			From	Ball	Dista	nce	Area	Net
		Temp	Weight		Stoic	Mole	Combust	Rad.	> LFL	diam	Rad	Dur	Pain	Leth	Pain	Leth	Covered	Агеа
Case	Chemical Name	K	(kg)	LFL	Conc	Wt.	kJ/kg	(m)	kg		Fract	sec		R	R		sq.∎.	sq.∎.
	Acetaldehyde	298	90720	0.040	Ø.Ø77	44.1	2464Ø	71.69	83149	217	Ø.2	29	553	145	661	253	50198	5Ø198
5Øc	Acetaidehyde	298	45360	0.040	0.077	44.1	24640	57.03	41575	173	Ø.2	23	461	184	488	191	28558	Mitigated
100c	Acetaldehyde	298	2268Ø	0.040	0.077	44.1	24840	45.37	20787	138	Ø.2	19	291	75	36Ø	144	16326	Mitigated
100c	Acetic Acid	298	2268Ø	0.054	Ø.Ø95	60.1	13110	37.70	19830	136	0.2	19	208	54	276	122	11600	11433
50c	Acetone	298	45360	0.021	0.050	58.1	28500	62.05	44126	176	0.2	24	444	115	532	203	32509	15865
100c	Acetone	298	22680	0.021	0.050	58.1	28500	49.36	22063	140	0.2	19	322	83	392	154	18511	1867
100c	Acrylic Acid	298	22680	0.024	0.065	72.1	18840	43.19	21903	140	Ø.2	19	261	68	331	138	14857	14739
100c	Acrylonitrile	298	22680	0.030	0.053	53.1	33200	47.51	21460	139	0.4	19	485	126	555	195	299ø2	8902
	Ammonia	298	90720	0.150	0.218	17.0	18610	66.52	61229	196	0.4	26	590	154	688	252	49807	49807
50c	Ammonia	298	4536Ø	0.150	Ø.218	17.0	18610	52.92	30615	156	Ø.4	21	428	111	506	189	28115	Mitigated
100c	Ammonia	298	2268Ø	Ø.15Ø	Ø.218	17.0	18610	42.10	15307	125	_ Ø.4	17	311	80	373	142	15946	Mitigated
100c	Benzene	298	22680	0.013	0.027	78.1	40600	53.73	22507	141	0.4	19	549	142	619	213	35515	26041
	Butadiene	298	90720	0.020	0.037	54.1	44600	84.96	88617	221	Ø.4	29	1084	283	1194	394	121871	121871
50°C	Butadiene	298	4536Ø	0.020	0.037	54.1	44600	67.59	44308	176	Ø.4	24	786	205	875	293	67312	Mitigated
100c	Butadiene	298	2268Ø	0.020	0.037	54.1	44600	53.77	22154	141	8.4	19	571	148	641	218	37331	Mitigated
	Butane	298	90720	0.018	0.031	58.1	45760	86.73	89075	221	0.4	29	1100	288	1211	398	124699	124699
100c	Butyl alcohol	298	22680	0.014	0.034	74.1	33100	52.13	22466	141	0.2	19	350	91	420	161	20404	20262
5Øc	Butyraidehyde	298	45360	0.025	0.037	72.1	3538Ø	59.18	43675	175	0.2	24	492	128	580	215	36547	26940
100c	Butyraldehyde	298	22680	0.025	0.037	72.1	35380	47.Ø8	21837	140	0.2	19	357	92	427	162	20703	11096
50c	Carbon disulfide	298	45360	0.013	0.065	76.1	14216	58.96	45014	177	0.4	24	447	116	536	205	32988	2060
100c	Carbon disulfide	298	2268Ø	0.013	0.065	76.1	14210	46.90	225Ø7	141	0.4	19	325	84	395	155	18783	Mitigated
	Chloroprene	298	45360	0.040	0.050	88.5	30000	48.73	41575	173	0.4	23	626	163	713	249	48769	32945
100c	Chloroprene	298	2268Ø	0.040	0.050	88.5	30000	38.77	20787	138	0.4	19	455	118	523	186	27295	11471
100c	Cyclohexane	298	2268Ø	0.013	Ø.Ø23	84.2	43459	53.80	22495	141	0.4	19	567	147	638	218	37160	22925
100c	Ethyl acetate	298	2268Ø	0.022	0.040	88.1	23510	44.39	22Ø32	140	0.2	19	292	76	362	148	16693	12469
100c	Ethyl acrylate	298	22680	0.018	0.034	100	27630	45.29	22269	141	Ø.2	19	318	82	389	153	18340	16922
100c	Ethyl alcohol	298	2268Ø	0.033	0.065	46.1	26880	47.49	21283	139	Ø.2	19	3Ø8	8Ø	377	149	17422	15658
	Ethylene oxide	298	90720	0.036	0.077	44.1	27650	72.96	84282	217	0.2	29	589	154	698	263	54260	54260
	Ethylene oxide	298	45360	0.036	0.077	44.1		58.04			0.2	23	428	111	514	198	30780	Mitigated
100c	Ethylene oxide	298	2268Ø	0.036	0.077	44.1	27650	46.18	21071	138	Ø.2	19	310	8Ø	380	149	17545	Mitigated
100c	Hexane	298	22680	0.012	0.022	86.2	44765	54.75	22544	141	0.4	19	576	149	547	220	37982	15665
	Hydrogen cyanide	298	90720	0.056	Ø.143	27.0	24530	72.01	78799	213	0.4	28	761	199	867	3Ø5	73186	73186
	Hydrogen cyanide		45360	0.056	Ø.143	27.0	24530	57.28	39400	179	Ø.4	23	552	144	637	228		Mitigated
100c	Hydrogen cyanide	298	22680	0.058	0.143	27.0	24530	45.57	19700	135	0.4	18	401	104	469	171	23050	Mitigated
100c	Isopropanoi	298	2268Ø	0.023	0.044	60.1		49.16			Ø.2	19	331	86	461	156	19022	17801
	LPG-Butene	298			0.034		45330	88.33	89493	222	0.4		1097	287	1208	398	124352	
5Øc	LPG-Butene	298	45360	0.016	0.034	56.1	45330	70.27	44747	177	Ø.4	24	796	207	885	296	68656	Mitigated
100c	LPG-Butene	298			0.034		45330	55.90	22373	141	Ø.4			150	648	220	38061	Mitigated
	LPG-Isobutane	298	90720	0.018	0.031	58.1	45660	86.73	89075	221	Ø.4	29	1099	287	1210	398	124502	124502
	LPG-Isobutane	298			0.031			69.00			0.4			208	886	296	68730	Mitigated
100c	LPG-Isobutane	298			0.031			54.89								220		Mitigated
	LPG-Isobutylene	298			0.147			67.96					1091		1202	396	123261	
	LPG-Isobutylene	298			Ø.147			54.06								294		Mitigated
100c	LPG-Isobuty lene	298			8.147			43.01								219		Mitigated
	LPG-Propane	298			0.040			88.81					1103		1214	399	125027	
	LPG-Propane	298			0.040			70.56								296		Mitigated
188c	LPG-Propane	298	22680	Ø.Ø21	9.949	44.1	45360	56.21	22094	148	9.4	19	581	150	651	221	38238	Mitigated

Table I3. (Cont.) Puff Release Fireballs

										Fire			Dist	ance	Total			
		Amb.	Fuel				Heat	Cloud	Vapor	Ball			From	Ball	Dist	ance	Area	Net
		Temp	Weight		Stoic	Mole	Combust	Rad.	> LFL	diam	Rad	Dur	Pain	Leth	Pain	Leth	Covered	Area
Case	Chemical Name	K	(kg)	LFL	Conc	Wt.	kJ/kg	(m)	kg		Fract	sec			R	3	sq.m.	sq.m.
	LPG-Propylene	298	90720	Ø.111	0.044	42.1	45790	67.38	67112	202	Ø.4	27	965	252	1066	353	97827	97827
50c	LPG-Propylene	298	45360	Ø.111	0.044	42.1	4579Ø	53.60	33556	161	Ø.4	22	701	182	781	262	54090	Mitigated
100c	LPG-Propylene	298	22686	Ø.111	8.844	42.1	45790	42.64	16778	128	Ø.4	18	509	131	573	196	30032	Mitigated
100c	Methyl alcohol	298	22680	0.060	0.122	32.0	1958Ø	43.72	19444	135	Ø.2	18	252	65	319	132	13775	6861
	Methyl chloride	298	90720	0.107	Ø.122	50.5	12680	54.02	878Ø9	293	0.4	27	510	133	612	235	43200	43200
5Øc	Methyl chloride	298	45360	0.107	₿.122	50.5	1268Ø	42.97	33905	161	8.4	22	370	96	451	177	24595	Mitigated
100c	Methyl chloride	298	2268Ø	0.107	Ø.122	50.5	12680	34.19	16952	129	. 0.4	18	269	69	333	134	14071	Mitigated
100c	Methy lethy lketor	1 298	22680	Ø.Ø18	0.037	72.1	31360	49.76	22269	141	0.2	19	339	88	410	158	19656	13301
100c	Methy Imethacry I	. 298	2268Ø	Ø.Ø21	0.034	100	26400	44.15	22094	140	Ø.2	19	310	8Ø	38Ø	150	17784	16214
	Propylene Oxide	298	90720	0.021	0.05	58.1	30230	78.30	88375	221	Ø.2	29	630	165	740	275	59482	59482
50c	Propylene Oxide	298	4536Ø	Ø.Ø21	0.05	58.1	30230	62.29	44188	176	- Ø.2	24	457	119	545	207	33660	Mitigated
100c	Propylene Oxide	298	22680	0.021	0.05	58.1	30230	49.56	22094	148	Ø.2	19	332	86	4Ø2	156	19138	Mitigated
100c	Toluene	298	22680	0.040	0.023	92.1	40550	43.56	20787	138	0.4	19	528	137	597	206	33183	31952
100c	Vinyl acetate	298	22680	0.026	0.044	86.1	22690	42.80	21770	140	Ø.2	19	286	74	355	144	16234	9096
	Vinyl chloride	298	90720	0.036	0.077	63.5	18490	64.69	84282	217	0.4	29	682	178	790	287	64672	64672
5Øc	Vinyl chloride	298	45360	0.036	0.077	63.5	1849Ø	51.46	42141	173	8.4	23	495	129	581	215	36433	Mitigated
100c	Vinyl chloride	298	22680	0.036	0.077	63.5	1849Ø	40.94	21071	138	Ø.4	19	359	93	428	162	20621	Mitigated

Table I3. (Cont.) Fire Balls

14016	e 13. (Cont.) Fir	C 54	Fuel							Fire			Dist:	ance 1	Total			Non	•
		Amb.	Weight				Heat	Cloud	Vapor						Dista	nce	Area	Mixed	Net
			kg/sec		Stoic	Mole	Combust		> LFL		Rad	Dur					Covered	Агеа	Area
Case	Chemical Name	K	or kg*	LFL	Conc	Wt.	kJ/kg	(m)	kg		Fract			M	п	M	sq.m.	sq.m.	sq.m.
5Øc	Acetic Acid	298	9	Ø.Ø54	Ø.Ø95	60.1	13110	2.84	791	47	Ø.2	7	47	12	 7ø	35	988	167	821
w/	Acrylonitrile	298	41.9	0.030	Ø.Ø53	53.1	33200	5.97	14991	124	Ø.4	17	411	106	473	168	22154	4748	17414
	Aniline	298	Ø.6	0.013	0.026	93.1	3483Ø	1.58	810	48	Ø.3	7	95	24	118	48	1787	195	1592 ~
w/	Ammonia*	298	45360	Ø.15Ø	Ø.218	17	18610	52.93	31405	157	Ø.4	21	433	112	512	191	28710	49974	Mitigated
w/	HexaCH3 Diamine	298	0.05	0.007	0.021	116	28400	0.74	791	47	Ø.3	7	84	21	108	45	1589	172	1417
. w/	Propylene Oxide*	298	45360	0.021	0.050	58.1	30230	62.23	44979	177	0.2	24	461	120	550	209	34152	59649	Mitigated.
100c	Acetic Acid*	298	22680	0.054	0.095	60.1	13110	37.68	19830	136	0.2	19	208	54	276	122	11600	167	11433
w/	Acetaldehyde∗	298	22680	0.040	0.077	44.1	24640	45.40	40618	171	Ø.2	23	397	103	483	189	28025	50365	Mitigated
w/	Butyraldehyde*	298	22680	0.025	0.037	72.1	3538Ø	47.08	41668	173	0.2	23	481	125	588	212	35159	9774	25385
w/	Formaldehyde	298	1.5	0.070	Ø.173	3Ø	19035	1.72	19889	136	Ø.2	19	251	65	319	133	13831	170	13661
w/	Chloroprene*	298	2268Ø	0.040	0.050	88.5	30000	38.74	40618	171	- Ø.3	23	537	140	622	225	39824	15991	23833
w/	TDI	298	1.2	0.009	0.022	174	23900	1.76	19879	136	0.3	19	344	89	412	157	19319	169	19150
w/	Sodium (H2)*	298	3944	0.040	Ø.295	2	119780	56.68	23445	143	0.2	19	679	176	750	247	48077	167	47910
w/	Ethylene oxide*	298	2268Ø	0.036	0.077	44.1	27650	46.20	40901	172	0.2	23	422	110	5Ø8	196	30041	54427	Mitigated
w/	Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	30927	157	0.2	21	286	74	364	153	18266	262	18004
5Øc	Acrylic Acid	298	4.0	0.024	0.065	72.1	18840	2.49	. 315	35	Ø.2	5	37	9	54	27	560	118	442
w/	Acrylonitrile	298	41.9	0.030	0.053	53.1	33200	5.97	14515	122	Ø.4	17	4Ø5	1Ø5	466	166	21566	4691	16875
100c	Acrylic Acid*	298	22680	0.024	0.065	72.1	18840	43.19	21903	140	0.2	19	261	68	331	138	14857	118	14739
w/	Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	33000	160	Ø.2	22	294	76	374	156	19231	213	19018
w/	Aniline	298	9.4	0.013	0.026	93.1	34830	3.91	23436	143	0.3	19	448	116	520	188	27662	146	27516
w/	TDI	298	1.2	0.009	0.022	174	23900	1.76	21952	140	0.3	19	360	93	430	163	20948	120	20828
w/	Formaldehyde	298	1.5	0.070	Ø.173	3Ø	19035	1.72	21962	140	0.2	19	263	58	333	138	14964	121	14843
w/	Sodium (H2)*	298	3944	0.040	Ø.295	2	119780	56.68	25518	147	0.2	20	706	183	78Ø	257	51698	118	51580
w/	Butyraldehyde*	298	22680	0.025	0.037	72.1	3538Ø	47.08	43741	176	0.2	24	492	128	580	216	36592	9725	26867
w/	HexaCH3 Diamine	298	0.6	0.007	0.021	116	28400	1.68	21927	140	Ø.3	19	393	102	463	172	23138	123	23015
w/	Chioroprene*	298	22680	0.040	0.050	88.5	30000	38.74	42891	174	0.3	23	549	143	636	230	41506	15942	25564
5 0 c	Acetaldehyde*	298	45360	0.040	0.077	44.1	24640	57.07	41575	173	Ø.2	23	401	104	488	191	28558	50198	Mitigated
w /	Ammonia*	298	4536Ø	Ø.15Ø	Ø.218	17	18610	52.93	72189	207	Ø.3	27	551	144	655	247	48050	50198	Mitigated
w/	Acetic Anhydride	298	5.6	0.027	0.050	102	16420	2.54	41983	173	0.2	23	329	86	416	172	23285	50198	Mitigated
•	HCN*	298	45360	0.056	0.143	27	24530	57.31	80974	215	Ø.3	28	667	174	775	282	62362	73186	Mitigated
w /	Sodium (H2)*	298	3944	0.040	Ø.295	2	11978Ø	56.68	45190	177	0.2	24	920	239	1008	328	84507	5Ø198	34309
100c	Acetaldehyde*	298	22680	0.040	0.077	44.1	24640	45.40	20787	138	Ø.2	19	291	75	360	144	16326	5Ø198	Mitigated
w/	Butyl Alcohol∗	298	22680	0.014	0.034	74.1	33100	52.06	43253	175	0.2	23	474	123	561	211	34861	5Ø198	Mitigated
w/	Isopropanol*	298	22680	0.023	0.044	60.1	30150	49.25	42756	174	0.2	23	450	117	537	204	32715	50198	Mitigated
w/	Methyl Alcohol∗	298	22680	0.060	Ø.122	32	19580	43.75	40232	171	0.2	23	352	92	438	177	24608	50198	Mitigated
	Aniline	298			0.026		34830	3.91	22320	141	0.3	19	438	113	5ø9	184	26563	5Ø198	Mitigated
w /	TDI	298	1.2	0.009	0.022	174	23900	1.76	20836	138	Ø.3	19	352	91	421	160	20075	5Ø198	Mitigated
5Øc	Butyraldehyde*	298	45360	0.025	0.037	72.1	35380	59.18	43675	175	Ø.2	24	492	128	58Ø	216	36547	9607	26940
w/	Acetic Anhydride	298	5.6	0.027	0.050	102	16420	2.54	44083	176	0.2	24	337	88	425	176	24208	9702	14506
w /	HexaCH3 Diamine	298	0.05	0.007	0.021	116	28400	Ø.74	43675	175	Ø.3	24	540	140	628	228	40889	9612	31277
	Butyraldehyde*	298	22680	0.025	0.037	72.1			21837		Ø.2	19	357	92	427	162	20703	9607	11096
	Butyl Alcohol*	298			0.034		33100	52.06	44303	176			479			213	35554	9749	
	Isopropanol∗	298			0.044		30150	49.25	43806	176	0.2	24	455	118	543	206	33370	10828	22542 1
	Ethyl Alcohol*	298			0.065		26880	47.46	43121	175	Ø.2	23			514	198	30869	11371	
	Methy! Alcohol*	298	22680	0.060	0.122	32			41281							179	25122	16521	8601
	Aniline	298			0.026				23370							187		9635	
	TDI	298			0.022		23900		21886							163		9609	
w/	Pheno l	298	9.3	0.017	0.029	94.1	31170	3.65	23111	142	0.2	19	344	89	415	160	20190	9616	10574

Table I3. (Cont.) Fire Balls Fuel Fire Distance Total Non Amb. Weight Heat Cloud Vapor Ball From Ball Distance Mixed Net. Temp kg/sec Stoic Mole Combust Rad. > LFL diam Rad Dur Pain Leth Pain Leth Covered Area Case Chemical Name K or kg* kJ/kg LFL Conc Wt. (m) kg Fract sec SQ.M. sq.m. ---- -----50c Formaldehyde 0.2 0.070 0.173 Ø 88 0 2 w/ Acetic Anhydride 298 5.8 0.027 0.050 2.54 0.2 100c Formaldehyde 1 5 0 070 0 173 1.72 Ø.2 w/ Aniline 9.4 0.013 0.026 93.1 3483Ø Ø.3 3.91 w/ HexaCH3 Diamine 0.6 0.007 0.021 1.68 9 3 w/ TDI 1.2 0.009 0.022 1.78 0.3 5Øc HCN∗ 45360 Ø.056 Ø.143 24530 57.31 39400 0.4 73186 Mitigated w/ Acetone* 45360 0.021 0.050 58.1 28500 62 23 83587 Ø 3 73186 Mitigated w/ Acrylonitrile 41.9 0.030 0.053 53.1 5.97 53600 0 4 73186 Mitigated w/ Vinyl acetate 77.5 0.026 0.044 86.1 6.58 70372 0.3 73186 Mitigated w/ Ethyl acetate 67.1 0.022 0.040 88.1 6.50 66216 0.3 73186 Mitigated w/ Isopropanol 28.3 0.023 0.044 60.1 5.42 47652 0.3 73186 Mitigated w/ Methyl alcohol 36.8 0.060 0.122 5.25 49714 0.3 73186 Mitigated w/ Butyl Acrylate 6.1 0.014 0.023 3.08 40006 73186 Mitigated 0.3 w/ Butyl alcohol 7.4 0.014 0.034 74.1 3 68 40546 Ø 3 73186 Mitigated 100c HCN* 22680 0.056 0.143 24530 45.59 19700 0.4 73186 Mitigated w/ Styrene 39.6 0.011 0.021 6.45 35525 0.4 73186 Mitigated Ammonia* 45360 Ø.150 Ø.218 18610 52.93 30615 0.4 5Ø6 49807 Mitigated w/ MEK 51.9 0.018 0.037 72.1 6.68 53231 0.3 56162 Mitigated w/ Vinyl acetate 77.5 0.026 0.044 86.1 6.58 61587 0.3 49807 Mitigated w/ Ethyl acetate 67.1 0.022 0.040 88.1 6.50 57431 49807 Mitigated w/ Styrene 5.7 0.011 0.021 104 3.40 31326 0.4 48Ø45 49807 Mitigated w/ Acetic Anhydride 298 5.6 0.027 0.050 102 2.54 31023 0.4 49807 Mitigated 50c Ethylene oxide* 45360 0.036 0.077 44.1 27650 58.07 42141 0.2 54260 Mitigated w/ Isopropano! 28.3 0.023 0.044 60.1 5.42 50393 9 2 54260 Mitigated w/ Methyl alcohol 36.8 0.060 0.122 5.25 52456 0.2 54260 Mitigated w/ Butyl alcohol 7.4 0.014 0.034 74.1 3.68 43287 0.2 54260 Mitigated 100c Ethylene oxide* 22680 0.036 0.077 44.1 27650 46.20 21071 Ø.2 54260 Mitigated w/ Aniline 9.4 0.013 0.026 93.1 34830 3.91 22603 Ø.3 54260 Mitigated w/ Phenol 9.3 0.017 0.029 94.1 31170 3.65 22345 0.2 54260 Mitigated 50c Propylene oxide* 298 45360 0.021 0.050 58.1 30230 62.23 44188 0.2 59482 Mitigated w/ Isopropanol 28.3 0.023 0.044 60.1 5.42 52440 0.2 60703 Mitigated w/ Methyl alcohol 36.8 0.060 0.122 1958@ 5.25 54503 0.2 33674 Mitigated w/ Butyl alcohol 7.4 0.014 0.034 74.1 3.68 45334 0.2 59624 Mitigated 100c Propylene oxide* 22680 0.021 0.050 58.1 30230 49.51 22094 0.2 33674 Mitigated 33674 Mitigated w/ Phenol 9.3 0.017 0.029 94.1 3.65 23368 0.2 w/ Aniline 9.4 0.013 0.026 93.1 3.91 23626 Ø.3 33674 Mitigated 50c Aniline Ø.6 Ø.013 Ø.026 93.1 1.58 0.4 w/ Methyl chloride* 298 4536Ø Ø.1Ø7 Ø.122 5Ø.5 12680 43.00 33923 0.4 43200 Mitigated w/ Vinyl chloride* 45360 0.036 0.077 63.5 18490 51.49 42160 0.4 64672 Mitigated 100c Aniline 9.4 0.013 0.026 93.1 34830 3.91 0.4 w/ Chloroprene* 30000 38 74 22320 9 4 22680 0.040 0.050 88.5 w/ Acrylonitrile* 22680 0.030 0.053 53.1 33200 47.62 23027 9.4 w/ Sodium (H2)* 3944 Ø.040 Ø.295 11978Ø 56.68 0.3 w/ Acetic Anhydride 298 44.1 0.027 0.050 102 16420 5.03 12629 0.2 14200 2.75 2152 w/ Maleic Anhydride 298 5.6 Ø.014 Ø.065 98.1 Ø.2 q

Table I3. (Cont.) Fire Balls

			Fuel							Fire			Dist	ance `	Tota l			Non	
	,	mb.	Weight					Cloud	•					Ball			Area	Mixed	Net
	Ţ	•	kg/sec				Combust	Rad.) LFL	diam	Rad	Dur	Pain	Leth	Pain	Leth	Covered	Årea	Area
Case	Chemical Name	K	or kg*	LFL	Conc	Wt.	kJ/kg	(m)	kg		Fract	sec		п	A	R	sq.m.	sq.m.	sq.∎.
100c	HexaCH3 Diamine	298	0.6	0.007	0.021	116	28400	1.68	24	15	Ø.4	2	19	5	27	12	118	5	113
w/	Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	11121	112	0.2	16	178	46	234	102	8162	100	8062
w/	Maleic Anhydride	298	5.6	0.014	0.065	98.1	14200	2.75	644	44	Ø.2	7	44	11	66	33	87Ø	9	861
w/	Chloroprene*	298	2258Ø	0.040	0.050	88.5	30000	38.74	20811	138	Ø.4	19	455	118	524	187	27321	15829	11492
w/	Acrylonitrile*	298	2268Ø	0.030	0.053	53.1	33200	47.62	21518	139	0.4	19	486	126	555	195	2997Ø	4578	25392
50c	Sodium (H2)*	298	3944	0.040	Ø.295	2	11978Ø	56.68	3615	78	Ø.2	11	286	73	325	112	9835	Ø	9835 +
w/	Carbon disulfide	298	45360	0.013	0.065	76.1	14210	59.00	48628	182	0.4	24	463	121	554	212	35136	30928	4208
w/	HCN*	298	45360	Ø.Ø56	0.143	27	24530	57.31	43015	175	Ø.4	23	575	150	663	237	44090	73186	Mitigated
w/	Styrene	298	5.7	0.011	0.021	104	41960	3.40	4327	82	₫.3	12	225	58	266	99	767 ø	312	7358
100c	Sodium (H2)∗	298	3944	0.040	0.295	2	11978Ø	56.68	3615	78	Ø.2	11	286	73	325	112	9835	0	9835
w/	Acetone*	298	22680	0.021	0.050	58.1	28500	49.51	25709	148	· Ø.2	28	346	9Ø	419	163	20951	16644	4307
w/	Ethyl Acrylate	298	22680	0.018	0.034	100	2763Ø	45.24	25884	148	0.2	20	341	88	415	162	20712	1418	19294
w/	Methylmethacryla	298	22680	0.021	0.034	100	26400	44.10	25709	148	Ø.2	20	333	86	406	160	20097	1570	18527
w/	Acetic Anhydride	298	44.1	Ø.Ø27	0.050	102	16420	5.03	14712	123	Ø.2	17	203	52	264	114	10164	95	10069
w/	Acrylonitrile*	298	22680	0.030	0.053	53.1	33200	47.62	25109	146	Ø.4	20	522	135	595	208	34111	4573	29538
w/	Aniline	298	9.4	0.013	0.026	93.1	34830	3.91	5147	87	Ø.3	12	222	57	266	101	7943	28	7915
w/	Butyl Acrylate	298	26.4	0.014	0.023	128	32246	4.99	9934	198	0.2	15	237	61	291	115	10386	171	10215
	Butyl Alcohol*	298	22580	Ø. Ø14	0.034	74.1	33100	52.06	26Ø81	148	Ø.2	20	375	97	449	171	23042	142	22900
w/	Ethyl Acetate*	298	2268Ø	0.022	0.040	88.1	2351Ø	44.43	25647	147	Ø.2	20	314	81	387	155	18859	4224	14635
w/	Ethyl Alcohol*	298	22680	0.033	0.065	46.1	2688Ø	47.46	24898	146	0.2	20	331	86	404	159	19775	1764	18011
w/	Formaldehyde	298	1.5	0.070	Ø.173	30	19035	1.72	3673	78	Ø.2	11	115	29	154	68	3673	3	3670
w/	HexaCH3 Diamine	298	Ø.6	0.007	0.021	116	28400	1.68	3639	78	Ø.3	11	171	44	210	83	5362	5	5357
•	Isopropanol*	298	22680	0.023	0.044	60.1	30150	49.25	25583	147	Ø.2			92		166	21526	1221	20305
		298	22680	Ø.Ø6Ø	Ø.122	32	1958Ø	43.75	23059	142	Ø.2	19	272	71				6914	8861
	MEK*	298			0.037	72.1		49.70			0.2		364			168	22214	6355	15859
٠.	Vinyl Acetate*	298			0.044			42.87			6.2		307	79			18361	7138	11223
•	Pheno I	298	9.3	0.017	0.029	94.1		3.65			Ø.2		90	23		50	1999	9	1990
w/	Butadiene*	298			0.037			53.70			0.4		586				39143		Mitigated
-	Acrylonitrile*	298			0.053			47.62			Ø.3							4582	41571
100c		298			0.022			1.76	49		Ø.4							2	189
	Ethyl Alcohol*	298	-	0.033	0.065	46.1		47.46			Ø.2							1766	15688
•	Butyl Alcohol*	298			0.034			52.06				19					20440	144	20296
•	Isopropanol*	298	22680	0.023	0.044	60.1	30150	49.25	22817	149	0.2	19	331	86	491	156		1223	17833
	Methyl Alcohol*	298			Ø.122			43.75			Ø.2							6916	6886
	Aniline	298			0.026			3.91			0.4							30	3540
	HexaCH3 Diamine	298			0.021			1.68	73									7	272
	Alcoholic Bev.	298			Ø.Ø65			1.17	5		Ø. 2				11			9	22
	Acetaldehyde*	298			Ø.077			57.07			Ø. 2							_	Mitigated
	Butyraldehyde*	298			0.017			59.18			Ø.2							9616	26934
	Ethylene oxide*	298			0.037			58.97			Ø.2								Mitigated
	Propylene oxide*				0.050			62.23			Ø. 2								Mitigated
	Sodium (H2)*	298																	_
	, ,				Ø.295	2 48 1		•			Ø.2							9	9839 88
	Alcoholic Bev.	298			0.065			1.61										. 9	
₩/	TDI	298	1.2	9 פטע. ש	0.022	1/4	23900	1.76	75	22	Ø.3	3	26	6	37	17	237	11	226

			Fuel	*						Fire			Dista					Non	
			Weight					Cloud	•						Dista		Area	Mixed	Net
	6 1		kg/sec*				Combust											Area	Area.
Case	Chemical Name	K	or kg	LFL	Conc	Wt.	kJ/kg	(m)	kg		Fract	sec				M	sq.m.	sq.m.	sq.m.
1	Hydrogen	298	3944	Ø. Ø4Ø	Ø.295	2.0	11978Ø	56.68	3615	78	Ø.2	11	286	73	325	112	9835	Ø	9835
2	Methyl amine	298	55738	0.049	0.085	31.1	31400										,		4
•	Ammonia	298	6Ø154	Ø.15Ø	Ø.218	17.0	1861ø												
•	Average	298	115892	Ø.Ø89	Ø.143	21.7	24761	77.77	90972	223	Ø.4	29	817	214	929	325	83Ø88	93007	Mitigate
3	Ethane	298	17040	0.030	0.057	30.1	47510												
	Ammonia	298	81064	Ø.15Ø	0.218	17.0	18610												
	Average	298	98184	Ø.1Ø5	Ø.187	18.4	2363Ø	73.72	73714	208	Ø.4	28	724	189	828	293	87545	49807	17738
4	Ethane	298	14940	Ø.Ø3Ø	0.057	30.1	47510	49.45	14159	121	0.4	17	479	124	540	184	26670	Ø	26670
5	Ethane	298	34080	0.030	0.057	30.1	4751Ø												
•	HCN	298	60048	0.056	0.143	27.0	24530												
•	Average	298	94128	0.043	0.094	28.0	3285€	80.66	85399	218	· Ø.4	29	914	239	1023	348	95238	73186	22052
6	Hydrogen	298	534	0.040	Ø.295	2.0	119780												
•	Ammonia	298	81648	0.150	0.218	17.0	18610												
•	Average	298	82182	Ø.131	0.221	16.2	19267	66.76	57891	192	Ø.4	26	585	152	681	249	48566	498Ø7	Mitigate
7	Vinyl chloride	298	45360	0.036	0.077	63.5	18490	51.49	42141										
	Acetylene	298	3715	0.025	0.077	26.0	48240	32.15	45718	178	0.4	24	830	216	919	3Ø5	73062	64672	8390
8	Ammonia	298	45360	Ø.15Ø	0.218	17.0	18610	52.93	76333	211	Ø.4	28	653	171	759	276	598Ø1	114479	Mitigate
9	HexaCH3 Diamine	298	0.2	Ø.007	0.021	116	28400	1.17	45722	178	0.4	24	637	166	726	255	50971	64677	Mitigate
19	Aniline*	298	0.6	Ø.Ø13	0.026	93.1	34830	1.58	45737	178	0.4	24	705	184	794	273	58360	64700	Mitigate
11	Ethyl ether	298	14578	0.018	0.034	74.1	33840	43.17	14314	122	0.3	17	352	91	413	152	18073		
•	Ethyl Alcohol	298	9072	0.033	0.065	46.1	26880	35.10	22827	142	Ø.3	19	389	101	460	172	23153	1764	21389
12	Ethyl ether	298	72970	Ø.Ø18	0.034	74.1	33840	73.44	71647	206	Ø.3	27	741	193	844	297	69080		
1	Ethyl Alcohol	298	22680	Ø.Ø33	0.065	46.1	26880	47.49	9293Ø	225	Ø.3	3Ø	745	195	857	307	74Ø68	1764	72304
13	Butene	298	6856	0.016	0.034	56.1	4533Ø	37.61	6763	95	0.3	13	288	74	336	122	11604		
•	Butyl Alcohol	298	22680	0.014	0.034	74.1	- 33100	52.06	29229	154	Ø.3	21	484	126	561	202	32202	142	32060
14	Propene	298	31700	Ø.111	0.044	42.1	45790	47.72	23451	143	Ø.3	19	514	133	586	205	32925	•	
•	Isopropanol	298	22680	Ø.Ø23	0.044	60.1	30150	49.25	45419	178	Ø.3	24	566	147	655	236	43836	1221	42615
15	Methyl ether	298	65205	0,034	0.065	46.1	28710	66.96	60986	196	0.3	26	633	165	731	263	54337		
•	Methyl Alcohol	298	22680	Ø.060	Ø.122	32.0	1958Ø	43.75	80430	214	Ø.3	28	594	155	702	262	54103	6914	47189
16	Ethyl ether	298	36485	0.018	0.034	74.1	33840	58.43	35823	164	Ø.3	22	538	140	629	222	38673		
•	Ethyl Alcohol	298	9072	Ø.Ø33	0.065	46.1	2688Ø	35.10	44337	176	Ø.3	24	529	138	617	226	40030	1764	38266
17	Ethy! ether	298	7289	Ø.Ø18	0.034	74.1	3384Ø	34.34	7157	97	0.3	14	255	66	3Ø4	114	10227		
3	Ethy! Alcohol	298	2268Ø	Ø.Ø33	0.065	46.1	2688Ø	47.49	28440	152	Ø.3	21	431.	112	507	188	27742	1764	25978
18	Methyl ether	298	32603	0.034	0.065	46.1	28710	53.27	3Ø493	156	Ø.3	21	46Ø	119	538	197	30568	-	
•	Methyl Alcohol	298	22680	0.060	0.122	32.0	19580	43.75	49938	183	Ø.3	24	477	124	569	216	36590	6914	29676
19	Chloroethane	298	24410	Ø.Ø62	Ø.Ø65	99.0	7700	34.07	20793	138	Ø.4	19	23Ø	60	299	128	12950		
1	Ethyl Alcohol	298	9072	Ø.Ø33	0.065	46.1	2688Ø	35.10	29306	154	0.3	21	437	113	514	190	28436	1764	26672
20	Chloroethane	298	32117	Ø. Ø62	0.065	99.0	7700	37.30	27358	151	0.4	20	261	68	337	143	16075		
•	Ethyl Alcohol	298	4536Ø	Ø.033	0.065	46.1	26880	59.70	69924	205	Ø.3	27	653	170	755	273	58431	1764	56667
	Chloromethane	298	25114					35.38		133	Ø.3	18	244	63	311	130	13206		
•	Methyl alcohol*	298	36.8	0.060	0.122	32		5.25		154	Ø.3	21	371	96	448	173	23547	6914	16633
	Isopropylchlor.	298	19536					41.54		133	Ø.3	18	342	88	408	155	18801		
	Isopropanol*	298			0.044			5.42		150	Ø.3	20	444	115	519	190	28376	1221	27155
	Phosphine	298			0.095			17.03	495	41	0.3	6	72	18	92	38	1162	Ø	1162
	Phosphine	298			0.095		32000		556	42	Ø.3	6	76	19	97	40	1273	Ø	1273

Table I3. (Cont.)

Case 1 = Sodium metal / Alkaline Liquid, Ammonium Nitrate Solution, Ferric Chloride Solution, Hydrochloric Acid,
Hydrofluorosilicic Acid, Hydrogen Peroxide, Hydroiodic Acid, Phosphoric Acid, Potassium Hydroxide Liq.,
Pulp Mill Liquid, Sulfuric Acid Spent, Toluene Diisocyanate

Case 2 = Anhydrous Ammonia / Methyl Chloride

Case 3 = Anhydrous Ammonia / Motor Fuel Antiknock Compound

Case 4 = Motor Fuel Antiknock Compound / Hydrochloric Acid, Hydroiodic Acid, Hydrofluorosilicic Acid, Phosphoric Acid,
Sulfuric Acid Spent, Aniline Dil

Case 5 = Motor Fuel Antiknock Compound / Hydrocyanic Acid 1

Case 6 = Anhydrous Ammonia / Sodium Metal

Case 7 = Vinyl Chloride / Alkaline Liquid, Potassium Hydroxide Liq., Sodium Hydroxide Liq.

Case 8 = Vinyl Chloride / Anhydrous Ammonia

Case 9 = Vinyl Chloride / Hexamethylene Diamine Solution

Case 10 = Vinyl Chloride / Aniline Oil

Case 11 = Alcoholic Beverage / Nitric Acid Funning, Oleum, Sulfuric Acid

Case 12 = Ethyl Alcohol or Denatured Alcohol / Nitric Acid Fuming, Gleum, Sulfuric Acid

Case 13 = Butyl Alcohol / Nitric Acid Fuming, Oleum

Case 14 = Isopropanol / Nitric Acid Fuming, Oleum

Case 15 = Methyl Alcohol / Nitric Acid Fuming, Oleum, Sulfuric Acid

Case 16 = Sulfuric Acid Spent / Ethyl Alcohol, Denatured Alcohol

Case 17 = Sulfuric Acid Spent / Alcoholic Beverage

Case 18 = Sulfuric Acid Spent / Methyl Alcohol

Case 19 = Hydrochloric Acid / Alcoholic Beverage

Case 20 = Hydrochloric Acid / Ethyl Alcohol, Denatured Alcohol

Case 21 = Hydrochloric Acid / Methyl Alcohol

Case 22 = Hydrochloric Acid / Isopropano!

Case 23 = Phosphorus / Alkaline Liquid, Potassium Hydroxide Liq., Sodium Hydroxide Liq.

Case 24 = Phosphorus / Hydrochloric Acid, Hydroiodic Acid

Puff Release Unconfined Vapor Cloud Explosions

	•										Maximum	
		Amb.	Fue I				Heat	Cloud	Virt.	Vapor	TNT	
		Темр	Weight		Stoic	Mole	${\sf Comb.}$	Rad.	Dist.) LFL	Equiv	Expl.
#	Chemical Name	K	(kg)	LFL	Conc	Wt.	kJ/kg	(m)	m 	kg	kg TNT	Eff.
	Acetaldehyde	298	90720	0.040	0.077	44.1	24640	71.69	75.30	83149	489442	0.309
50c	Acetaldehyde	298	45360	0.040	0.077	44.1	24640	57.Ø3	53.24	41575	244721	0.161
100c	Aceta I dehyde	298	22680	0.040	0.077	44.1	24640	45.37	37.65	20787	122360	0.068
100c	Acetic Acid	298	2268Ø	0.054	0.095	60.1	13110	37.68	37.65	19830	62106	0.023
50c	Acetone	298	45360	0.021	0.050	58.1	28500	62.23	53.24	44188	300848	0.200
100c	Acetone	298	2268Ø	0.021	0.050	58.1	28500	49.51	37.65	22094	150424	0.090
100c	Acrylic Acid	298	2268Ø	0.024	0.065	72.1	18840	43.19	37.65	21903	98581	0.050
100c	Acrylonitrile	298	2268Ø	0.030	0.053	53.1	33200	47.62	37.65	21495	170478	0.106
	Ammonia	298	90720	0.150	Ø.218	17.0	18610	66.52	75.30	61229	272210	Ø.18Ø
50c	Ammonia	298	45360	0.150	0.218	17.0	18610	52.92	53.24	30615	136105	0.079
100c	Ammonia	298	2268Ø	0.150	0.218	17.0	18610	42.10	37.65	15307	68Ø53	0.027
100c	Benzene	298	22680	0.013	0.027	78.1	40600	53.80	37.65	22507	218293	0.142
	Butadiene	298	90720	0.020	0.037	54.1	44600	84.96	75.30	88617	944173	Ø.484
50c	Butadiene	298	4536Ø	0.020	0.037	54.1	44600	87.59	53,24	44308	472086	0.300
100c	Butadiene	· 298	2268Ø	0.020	0.037	54.1	44600	53.77	37.65	22154	236043	Ø.155
	Butane	298	90720	0.018	0.031	58.1	45760	86.73	75.30	89075	973738	0.493
100c	Butyl Alcohol	298	2268Ø	0.014	0.034	74.1	33100	52.06	37.65	22466	177644	0.112
50c	Butyraldehyde	298	45360	0.025	0.037	72.1	35380	59.18	53.24	43675	369137	0.243
100c	Butyraldehyde	298	2268Ø	0.025	0.037	72.1	35380	47.08	37.65	21837	184568	0.117
50c	Carbon Disulfide	298	45360	0.013	0.065	76.1	14210	59.00	53.24	45014	152805	0.092
100c	Carbon Disulfide	298	2288Ø	0.013	0.065	76.1	14210	46.94	37.65	22507	76403	0.033
50c	Chloroprene	298	45360	0.040	0.050	88.5	30000	48.70	53.24	41575	297956	Ø.198
100c	Chloroprene	298	2268Ø	0.040	0.050	88.5	30000	38.74	37.65	20787	148978	0.089
100c	Cyclohexane	298	22680	0.013	0.023	84.2	43459	53.89	37.65	22507	233665	0.153
100c	Ethyl Acetate	298	22680	0.022	0.040	88.1	23510	44.43	37.65	22032	123738	0.069
100c	Ethyl Acrylate	298	2288Ø	0.018	0.034	100	27630	45.24	37.65	22269	146986	0.088
100c	Ethyl Alcohol	298	22680	0.033	0.065	46.1	26880	47.49	37.65	21283	136669	0.080
	Ethylene oxide	298	90720	0.038	0.077	44.1	27650	72.96	75.3Ø	84282	556713	0.341
50c	Ethylene oxide	298	45380	Ø. Ø36	0.077	44.1	27650	58.04	53.24	42141	278357	Ø.185
100c	Ethylene oxide	298	22680	0.036	0.077	44.1	27650	46.18	37.65	21071	139178	Ø.Ø82
	Hexane	298	2268Ø	0.012	Ø.Ø22	86.2	44765	54.58	37.65	22544	241082	Ø.159

						TNO	TNO		07/28/88
						Dist.	Dist.		
TNT						Heavy	Glass	Area	Net
Equiv	Expl.	Dist.	Pso	Pref	Dur	Damage	Damage	Covered	Area
kg TNT	Prob	m	psi	psi	sec	m	m	sq.m.	sq.m.
151096	0.485	371	2.000	4.212	0.42	177	884	108103	108103
39468	0.354	232	2.000	4.213	0.27	140	702	42273	Mitigated
8359	0.238	132	2.007	4.227	Ø.16	111	557	13685	Mitigated
1451	Ø.231	57	2.006	4.227	0.09	89	444	2552	2552
60092	Ø.365	275	1.999	4.211	Ø.33	150	.752	59396	42854
13592	0.247	162	2.005	4.222	0.20	119	597	20612	3870
4912	0.246	105	1.995	4.202	0.14	104	518	8659	8659
18067	Ø.243	182	2.003	4.218	0.20	124	622	26016	25862
49130	Ø.426	232	1.996	4.203	0.30	145	727	42273	42273
10768	0.300	132	1.998	4.204	Ø.18	115	577	13685	Mitigated
1884	Ø.194	65	2.013	4.242	0.10	92	458	3318	Mitigated
31026	0.250	226	1.996	4.203	Ø.25	135	676	4Ø115	37909
456991	Ø.497	570	2.000	4.213	0.60	220	1101	255178	255176
141552	0.365	383	2.003	4.219	0.40	175	874	115209	Mitigated
36600	Ø.247	241	1.995	4.201	Ø.28	139	893	45617	Mitigated
479655	Ø.498	581	1.998	4.208	0.61	222	1112	265120	265120
19808	0.250	189	2.001	4.214	0.23	126	631	28Ø55	28055
89555	Ø.363	322	1.997	4.206	0.33	161	8Ø5	81433	75758
21560	0.245	196	1.994	4.200	0.21	128	639	30172	24497
14092	Ø.368	149	2.002	4.217	0.24	120	600	17437	2043
2543	0.250	77	1.993	4.196	Ø.14	95	476	4657	Mitigated
58945	0.354	273	1.999	4.209	0.25	150	749	58535	52173
13292	Ø.238	161	1.997	4.207	Ø.15	119	595	2Ø358	13996
35830	0.250	239	1.995	4.202	Ø.26	138	691	44863	38359
8587	Ø.247	134	1.999	4.209	0.16	112	559	14103	13876
12885	0.248	159	1.997	4.205	0.18	118	592	19856	19856
10873	0.241	148	2.000	4.211	0.18	116	578	17203	17203
189943	Ø.487	407	1.996	4.203	₿.45	185	923	130100	130100
514Ø8	0.358	258	2.003	4.219	0.29	147	733	52279	Mitigated
11347	0.240	151	1.994	4.199	Ø.18	116	581	17908	Mitigated
38253	0.250	245	1.996	4.204	0.27	140	698	47144	29473

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Puff Release Unconfined Vapor Cloud Explosions

					•						Maximum	
		Amb.	Fuel				Heat	Cloud	Virt.	Vapor	TNT	
		Temp	Weight		Stoic	Mole				•	Equiv	Expl.
#	Chemical Name	K	(kg)			Wt.			n	kg	kg TNT	-
	Hydrogen Cyanide	298	90720	0.056	Ø.143	27.0	24530	72.01	75.30	78799	461765	Ø.294
50c	Hydrogen Cyanide	298	45360	0.058	0.143	27.0	24530	57.28	53.24	39400	230882	0.151
100c	Hydrogen Cyanide	298	2268Ø	0.056	0.143	27.0	24530	45.57	37.65	19700	115441	0.063
100c	Isopropanol	298	2268Ø	0.023	0.044	60.1	30150	49.25	37.65	21968	158228	Ø.098
	LPG-Butene	298	90720	0.016	0.034	56.1	45330	88.33	75.30	89493	969121	0.491
5Øc	LPG-Butene	298	4536Ø	0.016	0.034	56.1	45330	70.27	53.24	44747	484560	Ø.3Ø8
100c	LPG-Butene	298	2268Ø	0.016	0.034	56.1	45330	55.90	37.65	22373	242280	0.160
;	LPG-Isobutane	298	90720	Ø.Ø18	0.031	58.1	45660	86.73	75.30	89Ø75	971610	Ø.492
}	LPG-Isobutylene	298	90720	0.018	0.147	58.1	45030	87.98	75.30	89075	958204	0.488
į	LPG-Propane	298	90720	0.021	0.040	44.1	46360	88.81	75.30	88375	978760	Ø.494
İ	LPG-Propylene	298	90720	0.111	0.044	42.1	45790	67.38	75.30	67112	734129	Ø.415
løøc	Methyl Alcohol	298	2268Ø	0.060	0.122	32.∅	19580	43.75	37.65	19444	90950	0.044
	Methyl chloride	298	90720	0.107	0.122	50.5	1268Ø	54.02	75.30	678Ø9	205404	Ø.133
50c	Methyl chloride	298	45360	Ø.107	0.122	50.5	12680	42.97	53.24	33905	102702	0.053
100c	Methyl chloride	298	22680	0.107	0.122	50.5	12680	34.19	37.65	16952	51351	0.017
100c	Methy lethy lketon	298	2268Ø	0.018	0.037	72.1	31360	49.70	37.65	22269	166829	0.103
100c	Methy lmethacry la	298	2268Ø	0.021	0.034	100	26400	44.10	37.65	22094	139340	0.082
	Propylene Oxide	298	90720	0.021	0.050	58.1	30230	78.30	75.30	88375	638221	Ø.377
5Øc	Propylene Oxide	298	45360	0.021	0.050	58.1	30230	62.29	53.24	44188	319110	0.212
100c	Propylene Oxide	298	2268Ø	0.021	0.050	58.1	30230	49.58	37.85	22094	159555	0.097
- 1	Toluene	298	2268Ø	0.040	0.023	92.1	40550	43.46	37.65	20787	201368	0.130
ØØc	Vinyl Acetate	298	22680	0.026	0.044	86.1	22690	42.87	37.65	21770	118004	0.065
,	Vinyl chloride	298	90720	0.036	0.077	63.5	18490	64.69	75.30	84282	372283	Ø.244
БØс	Vinyl chloride	298	45380	Ø.Ø36	0.077	63.5	18490	51.46	53.24	42141	186142	Ø.118
	Vinvl chloride	298	2268#	0.038	0.077	63.5	18490	46.94	37.65	21071	93071	Ø. Ø46

						TNO	TNO		Ø7/28/88
						Dist.	Dist.		
TNT						Heavy	Glass	Area	Net
Equiv	Expl.	Dist.	Pso	Pref	Dur	Damage	Damage	Covered	Area
kg TNT	Prob		ps i	psi	sec		m	sq.m.	sq.m.
135976	0.474	356	1.997	4.206	0.41	173	867	99538	99538
34939	0.344	221	1.996	4.204	Ø.26	138	688	38360	Mitigated
7260	0.230					109	546	12272	Mitigated
15262	0.246	170	2.003	4.218	Ø.20	121	607	22698	22698
476099	0.499	579	2.000	4.212	Ø.62	222	1110	263298	263298
148391	0.367	39Ø	2.002	4.217	Ø.42	176	881	119459	Mitigated
38651	0.249	246	1.996	4.204	0.27	140	700	47529	Mitigated
478015	0.498	580	1.999	4.211	Ø.61	222		2842Ø8	264208
467715	Ø.498	575	2.000	4.213	0.48	221	1106	259672	259672
483530	0.496	583	1.997	4.206	Ø.63	223	1114	266948	266948
304364	0.443	489	1.998	4.204	Ø.41	.202	1012	1878Ø5	1878Ø5
4004	Ø.228	98	1.987	4.183	0.13	101	505	7238	6163
27229	0.445	177	1.997	4.206	Ø.23	132	662	24606	246Ø6
5443	Ø.318	94	2.003	4.218	Ø.13	105	525	6940	Mitigated
849	0.208	42	1.988	4.187	0.07	83	417	1385	Mitigated
17208	0.248	179	1.995	4.202	0.21	124	618	25165	23503
11378	0.247	151	1.997	4.205	Ø.17	116	582	17908	17908
240553	0.496	446	1.999	4.210	0.50	193	968	156228	156228
6753Ø	0.365	288	2.000	4.211	0.33	153	767	65144	Mitigated
15556	0.247	172	1.993	4.197	0.20	122	609	23235	Mitigated
26085	0.238	211	1.998	4.208	Ø.19	132	658	34987	34967
7658	0.245	128	1.991	4.193	0.15	11,0	55Ø	12868	11347
91017	0.487	302	1.997	4.205	Ø.36	161	8Ø7	71631	71631
21967	0.356	181	2.005	4.224	0.22	128	641	25730	Mitigated
4246	Ø.24Ø	98	2.001	4.214	Ø.13	102	5Ø9	7543	Mitigated

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c Acetic c Acetic c Acetic c Acrylic c Acrylin c Acrylon c Aniline c Aniline c Benzene c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Carbon c Coal Ta	cal Name c Acid c Acid c Anhydride c Anhydride ne ic Acid ic Acid onitrile ne ne ne	Temp K 298 298 298 298	2.0 9.0 1.1 5.6 44.1 46.4 1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 56.3 6.1 26.4	LFL 0.054 0.054 0.027 0.027 0.027 0.021 0.024 0.024 0.030 0.013 0.013 0.013 0.014 0.014	Conc 0.095 0.095 0.050 0.050 0.050 0.065 0.065 0.065 0.065 0.028 0.027 0.027 0.023 0.023	Wt. 60.1 102 102 58.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1 128	Combust kJ/kg 13110 13110 16420 16420 16420 28500 18840 33200 33200 34830 34830 44830	Rad. (m) 1.73 2.84 1.49 2.55 5.03 6.40 1.82 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	Dist. m 2.1 7.0 1.4 5.1 26.3 34.8 1.6 4.4 11.8 29.6 0.4 1.1 9.8 17.0 43.2 1.8	kg 71 791 30 408 11091 19344 39 315 2215 14060 3 19 1532 4645 29850		Effc Ø.012 Ø.076 Ø.002 Ø.060 Ø.001 Ø.013	523 9963 28 6672 9	Ø. 153 Ø. 227 Ø. Ø33 Ø. 182 Ø. Ø21 Ø. Ø72	41 146 14 128 8 53	psi 2.008 1.992 1.956 2.002	4.115 4.216 4.022 4.197	\$ec 0.07 0.18 0.02 0.14	79 114 58 108	Dist Glass Damage m 395 571 292 540 262 399	154 12868 5ø 22ø6	Net Area sq.m. 1320 16742 154 12714 50 2206
Acetic C Acetic C Acrylic C Acrylic C Acrylin C Aniline C Aniline C Benzene C Butyl C Butyl C Butyl C Butyl C Butyl C Butyl C Carbon C Chlorop C Coal Ta C Coal Ta C Coal Ta	cal Name c Acid c Acid c Anhydride c Anhydride c Anhydride ic Acid ic Acid onitrile onitrile ne ne ne Acrylate Acrylate alcohol	X 298 298 298 298 298 298 298 298 298 298	kg/sec 2.0 9.0 1.1 5.6 44.1 46.4 1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 58.3 1.3 6.1 26.4	0.054 0.054 0.027 0.027 0.021 0.024 0.024 0.030 0.013 0.013 0.013 0.013	Conc 0.095 0.095 0.050 0.050 0.050 0.065 0.065 0.065 0.065 0.028 0.027 0.027 0.023 0.023	Wt. 60.1 102 102 58.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1 128	kJ/kg 13110 13110 16420 16420 16420 28500 18840 33200 34830 34830 40600 40600	(m) 1.73 2.84 1.49 2.55 5.03 6.40 1.62 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	2.1 7.0 1.4 5.1 26.3 34.8 1.8 4.4 11.8 29.6 0.4 1.1 9.8 17.0 43.2 1.8	kg 71 791 30 408 11091 19344 39 315 2215 14060 3 19 1532 4645 29850	43504 131701 17567 111514 12750 45048	Effc Ø.012 Ø.076 Ø.002 Ø.060 Ø.001 Ø.013	523 9963 28 6672 9	Prob Ø.153 Ø.227 Ø.033 Ø.182 Ø.021 Ø.072	41 146 14 128 8 53	psi 2.008 1.992 1.956 2.002 1.914 1.993	psi 4.230 4.195 4.115 4.216 4.022 4.197	\$ec 0.07 0.18 0.02 0.14	79 114 58 108	Damage m 395 571 292 540	Covered sq.m 1320 16742 154 12868 50 2208	1320 16742 154 12714 50 2206
Acetic C Acetic C Acetic C Acetic C Acetic C Acetic C Acetic C Acetic C Acrylic C Acrylic C Acrylin C Aniline C Aniline C Benzene C Butyl C Butyl C Butyl C Butyl C Butyl C Butyl C Carbon C Chlorop C Coal Ta C Coal Ta C Coal Ta	c Acid c Acid c Anhydride c Anhydride c Anhydride ne ic Acid ic Acid onitrile ne ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298 298	2.0 9.0 1.1 5.6 44.1 46.4 1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 56.3 6.1 26.4	0.054 0.054 0.027 0.027 0.021 0.024 0.024 0.030 0.013 0.013 0.013 0.013	0.095 0.050 0.050 0.050 0.050 0.065 0.065 0.063 0.063 0.026 0.027 0.027 0.023	60.1 60.1 102 102 58.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1 128	13110 13110 16420 16420 28500 18840 33200 34830 34830 34830 40600 40600	1.73 2.84 1.49 2.55 5.03 6.40 1.82 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	2.1 7.0 1.4 5.1 28.3 34.8 1.8 4.4 11.8 29.6 0.4 1.1 9.8 17.0 43.2 1.8	71 791 30 408 11091 19344 39 315 2215 14060 3 19 1532 4645 29850	43504 131701 17567 111514 12750 45048	9.012 0.076 0.002 0.060 0.001 0.013	523 ,9963 28 6672 9	Ø. 153 Ø. 227 Ø. Ø33 Ø. 182 Ø. Ø21 Ø. Ø72	41 146 14 128 8 53	2.008 1.992 1.956 2.002 1.914 1.993	4.230 4.195 4.115 4.216 4.022 4.197	6.07 6.18 6.02 6.14 6.02 6.02	58 1ø8 52 8ø	571 292 540 262	1328 16742 154 12868 58 2286	1320 16742 154 12714 50 2206
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c Acetic c Acetic c Acetic c Acrylic c Acrylic c Acrylin c Acrylin c Aniline c Aniline c Aniline c Benzene c Benzene c Butyl / d Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Carbon c Chlorop c Coal Ta	c Anhydride c Anhydride ne ic Acid ic Acid onitrile onitrile ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298 298	5.6 44.1 46.4 1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 58.3 1.3 6.1 26.4	0.027 0.027 0.021 0.024 0.024 6.030 0.013 0.013 0.013 0.014 0.014	0.050 0.050 0.050 0.065 0.065 0.053 0.028 0.028 0.027 0.027 0.027	102 102 58.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1	16420 16420 28500 18840 18840 33200 34830 34830 40600 40600	2.55 5.03 6.40 1.62 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	5.1 26.3 34.8 1.8 4.4 11.8 29.8 Ø.4 1.1 9.8 17.0 43.2 1.8	408 11091 19344 39 315 2215 14060 3 19 1532 4645 29850	131701 17567 111514 12750 45048	0.976 0.902 0.969 0.901 0.913	,9963 28 6672 9	Ø.227Ø.933Ø.182Ø.021Ø.972	146 14 128 8 53	1.956 2.002 1.914 1.993	4.115 4.216 4.022 4.197	Ø.18 Ø.02 Ø.14 Ø.02 Ø.67	58 1ø8 52 8ø	571 292 540 262	16742 154 12868 50 2206	16742 154 12714 50 2206
### Acetic Acetone Acrylic Acrylic Acryline Acryline Aniline Aniline Benzene Butyl Butyl Butyl Butyl Butyl Butyl C Butyl C Butyl C Butyl C C Butyl C C C C C C C C C C C C C C C C C C C	c Anhydride ne ic Acid ic Acid onitrile ne ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298 298	44.1 46.4 1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 56.3 1.3 6.1 26.4	0.027 0.021 0.024 0.030 0.030 0.013 0.013 0.013 0.013 0.014	0.050 0.055 0.065 0.065 0.053 0.028 0.028 0.027 0.027 0.027	102 58.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1	16420 28500 18840 18840 33200 34830 34830 34830 40600 40600	5.03 6.40 1.82 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	26.3 34.8 1.6 4.4 11.8 29.6 0.4 1.1 9.8 17.0 43.2 1.8	11091 19344 39 315 2215 14060 3 19 1532 4645 29850	131701 17567 111514 12750 45048	0.976 0.902 0.969 0.901 0.913	,9963 28 6672 9	Ø.227Ø.933Ø.182Ø.021Ø.972	146 14 128 8 53	1.956 2.002 1.914 1.993	4.115 4.216 4.022 4.197	Ø.18 Ø.02 Ø.14 Ø.02 Ø.67	58 1ø8 52 8ø	571 292 540 262	16742 154 12868 50 2206	16742 154 12714 50 2206
c Acetone c Acrylic c Acrylic c Acrylon c Acrylon c Aniline c Aniline c Benzene c Benzene c Butyl d Butyl d Butyl c Butyl c Butyl c Carbon c Chlorop c Coal Ta	ne ic Acid ic Acid onitrile onitrile ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298 298	46.4 1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 56.3 1.3 6.1 26.4	0.021 0.024 0.024 6.030 0.030 0.013 0.013 0.013 0.014 0.014	Ø.050 Ø.065 Ø.065 Ø.053 Ø.026 Ø.026 Ø.027 Ø.027 Ø.027	58.1 72.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1	28500 18840 18840 33200 34830 34830 34830 40600 40600	6.40 1.82 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	34.8 1.6 4.4 11.8 29.8 0.4 1.1 9.8 17.0 43.2 1.8	19344 39 315 2215 14060 3 19 1532 4645 29850	131701 17567 111514 12750 45048	0.976 0.902 0.969 0.901 0.913	,9963 28 6672 9	Ø.227Ø.933Ø.182Ø.021Ø.972	146 14 128 8 53	1.956 2.002 1.914 1.993	4.115 4.216 4.022 4.197	Ø.18 Ø.02 Ø.14 Ø.02 Ø.67	58 1ø8 52 8ø	571 292 540 262	16742 154 12868 50 2206	16742 154 12714 50 2206
Acrylic Acrylic Acrylic Acrylon Acrylon Aniline Aniline Aniline Benzene Benzene Butyl Butyl Butyl Butyl C Butyl C Butyl C Butyl C C C C C C C C C C C C C C C C C C C	ic Acid ic Acid onitrile onitrile ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298 298	1.082 3.995 13.2 41.9 0.2 0.6 9.4 17.6 56.3 1.3 6.1 26.4	0.024 0.024 0.030 0.030 0.013 0.013 0.013 0.014 0.014	0.065 0.065 0.053 0.053 0.028 0.028 0.026 0.027 0.027 0.023	72.1 72.1 53.1 53.1 93.1 93.1 78.1 78.1 128	1884Ø 1884Ø 3320Ø 3320Ø 3483Ø 3483Ø 4060Ø 4060Ø 3224Ø	1.62 2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	1.8 4.4 11.8 29.6 0.4 1.1 9.8 17.0 43.2	39 315 2215 14060 3 19 1532 4845 29850	17567 111514 12750 45048	0.902 0.969 0.901 0.901	28 6672 9 58ø	Ø.033 Ø.182 Ø.021 Ø.072	14 128 8 53	1.956 2.002 1.914 1.993	4.115 4.216 4.022 4.197	Ø. Ø2 Ø. 14 Ø. Ø2 Ø. Ø7	58 1Ø8 52 8Ø	292 540 -	154 12868 5ø 22ø6	154 12714 50 2206
c Acrylic c Acrylor c Acrylor c Aniline c Aniline c Aniline c Benzene c Benzene c Butyl c Butyl c Butyl c Butyl c Butyl c Carbon c Chlorop c Coal Ta c Coal Ta	ic Acid onitrile onitrile ne ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298 298	3.995 13.2 41.9 0.2 0.6 9.4 17.6 56.3 1.3 6.1 26.4	0.024 6.030 9.030 0.013 0.013 0.013 0.014 0.014	0.065 0.053 0.053 0.028 0.028 0.028 0.027 0.027 0.023 0.023	72.1 53.1 53.1 93.1 93.1 78.1 78.1 128	1884Ø 3320Ø 3320Ø 3483Ø 3483Ø 4060Ø 4224Ø	2.49 4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	4.4 11.8 29.8 Ø.4 1.1 9.8 17.0 43.2	315 2215 14060 3 19 1532 4845 29850	111514 1275 0 45 0 48	0.060 0.001 0.013	6672 9 58ø	Ø.182 Ø.021 Ø.072	128 8 53	2.002 1.914 1.993	4.216 4.022 4.197	Ø.14 Ø.02 Ø.67	1Ø8 52 8Ø	54 0 	12868 5ø 22ø6	12714 50 2206
c Acrylon c Acrylon c Aniline c Aniline c Aniline c Benzene c Butyl c Butyl c Butyl c Butyl c Butyl c Butyl c Carbon c Chlorop c Coal Ta c Coal Ta	onitrile onitrile ne ne ne ne Acrylate Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298 298	13.2 41.9 0.2 0.6 9.4 17.6 56.3 1.3 6.1 26.4	6.030 8.030 0.013 0.013 0.013 0.013 0.014	 Ø. Ø53 Ø. Ø28 Ø. Ø26 Ø. Ø27 Ø. Ø27 Ø. Ø23 Ø. Ø23 	53.1 53.1 93.1 93.1 78.1 78.1 128	33200 33200 34830 34830 34830 40600 40600 32240	4.07 5.95 1.10 1.57 3.90 5.06 7.42 1.85	11.8 29.8 Ø.4 1.1 9.8 17.0 43.2	2215 14060 3 19 1532 4845 29850	111514 1275 0 45 0 48	0.060 0.001 0.013	6672 9 58ø	Ø.182 Ø.021 Ø.072	128 8 53	2.002 1.914 1.993	4.216 4.022 4.197	Ø.14 Ø.02 Ø.67	1Ø8 52 8Ø	54 0 	12868 5ø 22ø6	12714 50 2206
c Acrylon c Aniline c Aniline d Aniline d Benzene c Benzene c Butyl d Butyl d Butyl c Butyl d Butyl c Butyl c Butyl c Butyl d C Butyl d C Butyl d C Butyl d C Butyl d C Butyl d C Butyl d C Butyl d C Butyl d C Carbon c Coal Ta d C Coal Ta	onitrile ne ne ne ne ne Acrylate Acrylate alcohol	298 298 298 298 298 298 298 298	41.9 Ø.2 Ø.6 9.4 17.6 56.3 1.3 6.1 26.4	8.030 6.013 6.013 6.013 6.013 6.013 6.014	9.953 9.926 9.926 9.027 9.027 9.923 9.923	53.1 93.1 93.1 93.1 78.1 78.1	33200 34830 34830 34830 40600 40600 32240	5.95 1.10 1.57 3.90 5.06 7.42 1.85	29.8 Ø.4 1.1 9.8 17.0 43.2 1.8	14060 3 19 1532 4845 29850	111514 1275 0 45 0 48	0.060 0.001 0.013	6672 9 58ø	Ø.182 Ø.021 Ø.072	128 8 53	2.002 1.914 1.993	4.216 4.022 4.197	Ø.14 Ø.02 Ø.67	1Ø8 52 8Ø	54 0 	12868 5ø 22ø6	12714 50 2206
c Aniline c Aniline c Benzene c Benzene c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Carbon c Chlorop c Coal Ta c Coal Ta	ne ne ne ne Acrylate Acrylate Acrylate	298 298 298 298 298 298 298 298	9.2 9.4 17.6 56.3 1.3 6.1 26.4	0.013 0.013 0.013 0.013 0.013 0.014	0.028 0.026 0.026 0.027 0.027 0.023	93.1 93.1 93.1 78.1 78.1 128	34830 34830 34830 40800 40800 32240	1.10 1.57 3.90 5.06 7.42 1.85	0.4 1.1 9.8 17.0 43.2 1.8	3 19 1532 4845 2985Ø	1275 0 45 0 48	Ø.001 Ø.013	9 58Ø	0.021 0.072	8 53	1.914 1.993	4.022 4.197	0.02 0.67	52 8ø	262	5Ø 22Ø8	50 2206
c Aniline c Benzene c Benzene c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Carbon c Chlorop c Coal Ta dc Coal Ta	ne ne ne Acrylate Acrylate Acrylate alcohol	298 298 298 298 298 298 298	Ø.6 9.4 17.6 56.3 1.3 6.1 26.4	0.013 0.013 0.013 0.013 0.014 0.014	8.026 0.026 0.027 0.027 0.023 0.023	93.1 93.1 78.1 78.1 128	3483Ø 3483Ø 4Ø6ØØ 4Ø6ØØ 3224Ø	1.57 3.90 5.06 7.42 1.85	1.1 9.8 17.0 43.2 1.8	19 1532 4645 2985Ø	45048	Ø.Ø13	58Ø	0.072	53	1.993	4.197	0.07	80		2206	2206
Bc Aniline c Benzene c Benzene c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Butyl / c Carbon c Chlorop c Coal Ta c Coal Ta	ne ne Acrylate Acrylate Acrylate acrylate	298 298 298 298 298 298	9.4 17.6 58.3 1.3 6.1 26.4	0.013 0.013 0.013 0.014 0.014	0.026 0.027 0.027 0.023 0.023	93.1 78.1 78.1 128	34830 40600 40600 32240	3.90 5.06 7.42 1.85	9.8 17.0 43.2 1.8	1532 4845 2985Ø	45048	Ø.Ø13	58Ø	0.072	53	1.993	4.197	0.07	80		2206	2206
Benzence Benzence Butyl / Butyl / Butyl a Butyl a Butyl a C Butyl a	ne Acrylate Acrylate Acrylate alcohol	298 298 298 298 298	17.6 58.3 1.3 6.1 26.4	0.013 0.013 0.014 0.014	0.027 0.027 0.023 0.023	78.1 78.1 128	40600 40600 32240	5.06 7.42 1.85	17.0 43.2 1.8	4845 2985Ø	45048	Ø.Ø13	58Ø	0.072	53	1.993	4.197	0.07	80		2206	2206
c Butyl / c Butyl / dc Butyl / c Butyl a c Butyl a c Butyral c Carbon c Chlorop c Coal Ta dc Coal Ta	Acrylate Acrylate Acrylate alcohol	298 298 298	1.3 6.1 26.4	0.014 0.014	Ø.023 Ø.023	128	32240	1.85	1.8	29850												
c Butyl / Øc Butyl / c Butyl a c Butyl a c Butyral c Carbon c Chlorop c Coal Ta c Coal Ta	Acrylate Acrylate alcohol	298 298	6.1 26.4	0.014	0.023				1.8										148	742	80263	58057
Oc Butyl a c Butyl a c Butyl a c Butyral c Carbon c Chlorop c Coal Ta c Coal Ta	Acrylate alcohol	298	26.4			128	32240	2 40								*						
c Butyl a c Butyl a c Butyral c Carbon c Chiorop c Coal Ta c Coal Ta	alcohol	,		0.014				3.08	6.2	606												
c Butyl a c Butyral c Carbon c Chlorop c Coal Ta c Coal Ta		298	a o		0.023	128	32240	5.00	19.9	6313	48622	0.015	725	0.096	56	1.977	4.162	0.07	82	410	2463	2463
c Butyral c Carbon c Chlorop c Coal Ta c Coal Ta Øc Coal Ta	a I coho I		₽.9	0.014	0.034	74.1	33100	1.84	1.8	39												
c Carbon c Chlorop c Coal Ta c Coal Ta Øc Coal Ta		298	7.4	0.014	0.034	74.1	33100	3.69	8.5	1146	9064	0.000										
c Chlorop c Coal Ta c Coal Ta Øc Coal Ta	a I dehyde	298	31.8	0.025	0.037	72.1	3538ø	5.39	23.0	8481	71680	0.030	2145	0.124	85	2.001	4.214	0.10	93	466	56 7 5	5675
c Coal Ta c Coal Ta Øc Coal Ta	n disulfide	298	68.6	0.013	Ø.Ø65	76.1	14210	6.92			141189		11734	0.354	140	1.991	4.192	0.23	117	584	15394	15394
c Coal Ta Øc Coal Ta	•	298			0.050		30000				77299	0.034	2623	0.150	90	1.989	4.189	0.09	96	478	6362	6362
Øc Coal Ta	Tar Distill				0.030	*	40570		1.2													
	Tar Distill				0.030		40570		3.1		0445-	# ##C	•			0.555	4				.=.	
r (rude (0.030		40570	4.16			24197	0.003		0.038	. 24	2.003	4.220	0.03	65	325	452	452
		298			Ø.Ø23		43500	1.28	0.8	9			* ;	ŧ.								
c Crude 0		298			0.023			1.90	2.0		10740	a aac										
Øc Crude 0 c Cyclohe		298			0.023						10746		9202	a 140	01	1 007	4 oar	a 11	ΩF	479	eca.	0004
c Cyclone c Cyclohe		298 208			Ø. Ø23					-	7468Ø 339193		76Ø91	Ø. 108					95 157	473	6504 75477	8504 80072
c Cyclone c Ethyla		298 298			0.023 0.040						21888			0.061					157 63	783 314	75477 227	68973 227
c Ethyla		298			0.040						150610								119	597	20106	19879
C CUIVI A	aceuate	290	07.1	0.022	0.010	00.1		0.50	10.9	20010	130010	0.031	13031	6,210	100	1,900	4.100	0.19		160	20100	1907
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Cont	inuaua Dalamaa Um	£:	inad Vanan Cl		_ ! : _														~		
Cont	inuous Release Un	conti	ined vapor Ci	lona Ex	piosic	ons				14								TNO	TNO		07/28/88
	•	Amb.	Release			Uaat	لمستداع	W: _+	V	Max TNT	U	THE						Dist	Dist		
į.			Rate	Ctain	Mala				•			TNT	-	D:-4	ο.	n (•	-	Glass		Net
Cooo		•				Combust												_	Damage	Covered	
Case	Chemical Name	K	kg/sec LFL	Conc	WC.	k1/kg	(#)	m	kg	kg INI	Effe	kg TNT	Prol) In	psi	psi	sec	m	M	sq.m.	sq.m.
25c	Ethyl acrylate	298	7.7 0.018	0.034	100	27630	3 24	7 4	877												
		298				27630				46989	0.014	656	0.107	52	1 989	4.188	a a7	81	405	2124	2124
i i	Ethyl Alcohol	298				26880			1185		0.000		0.10	02	1.505	7.100	0.01	01	400	2127	2127
	Ethyl Alcohol					26880				43820			0.103	. 4A	1 979	4.150	a a7	79	396	1810	1810
	Formaldehyde	298-				19035			0024	.5020	J. U.Z	. 000	U , 10t	. 40	1.012	7.100	0.01	13	330	1016	1010
,	Formaldehyde	298	0.2 0.070			19035			2												
	Formaldehyde	298	1.5 0.070			19035			58												
	Fuel Oil	298	Ø.9 Ø.013			43100			34												
1	Fuel Oil	298	2.7 0.013			43100			199												
		298	17.1 0.013			43100			3824	39372	0.010	388	0.060	4R	1 983	4.175	g ar	76	382	1662	1662
		298	0.7 0.007			43100		1.4	33	20012		300	J. JUL	70		7.173	J. 20	70	J02	1002	1002
		298	2.1 0.007			43100		3.5	193												
	*	298	13.2 0.607			43100				37724	а аво	341	ø ø57	43	ว สวส	4.256	a ae	75	376	1452	1452
		298	2.2 0.025			22000		2.7		51721		0.12	0.001	10		1.200	0.00	,,,	. 0,0	1402	1702
		298	7.4 0.025			22000			797												
		298	58.6 0.025							114749	Ø Ø89	7155	Ø 945	195	1 991	4 102	Ø 16	109	545	12272	12272
	•	298	32.1 0.012							128475						4.190		113	566	17671	
	Hexane	298	93.8 Ø.012									290888						201	1003		
i	Hexamethy lened ia		0.01 0.007			28400		0.0	0		J. 101	20000	₩. TT&	491	1.550		J. UZ	201	1203	103349	111013
	Hexamethy lenedia		Ø.1 Ø.007			28400		0.2	Ø							•					
	Hexamethy lenedia		Ø.8 Ø.007			28400		1.2	24												
		298	6.0 0.023			30150		8.6	890												
		298	28.3 Ø.023			30150				59439	g . Ø22	1284	Ø. 191	69	1.980	4.187	g gq	88	438	3739	3739
	Maleic Anhydride		0.05 0.014			14200		0.1	Ø		J. VLL			-			5		400	0109	5133
	Maleic Anhydride		Ø.5 Ø.014			14200			13												
	Maleic Anhydride		5.6 0.014			14200								•							
	Methyl alcohol		31.7 0.060			19580				38002	Ø . ØØ9	34A	Ø.119	37	1.971	4.149	0.06	75	377	1075	1075
	Methyl alcohol		36.8 0.060			19580							0.145			4.154		82	408	1963	
	Methy lethy lketon		21.4 Ø.Ø18			31360							0.084			4.134		77	387	1662	
	Methy lethy lketon		51.9 Ø.Ø18			31360												124	621	26016	
	Methylmethacryla		9.0 0.021			26400				6473		-,013	01	101				467	021	20010	27307
	Methylmethacryla		28.5 0.021			26400						438	g goa	44	1 979	4.151	g gs	77	387	1521	1521
			20.0 0.522	5.551		20 120			0100	10001	0.011	150	2.550	• • •	1.0.2		D . D G	• •	001	1021	1021
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ntinuous Release Un	iconf i	ned Vap	or Cl	oud Exp	losio	ns				Max								TNO Dist	TNO Dist		07/28/88	
	Amb.	Release	:			Heat	Cloud	Virt.	Vapor	TNT	Max	TNT						Heavy	Glass	Area	Net	
•	Temp	Rate		Stoic	Mole	Combust	Rad.	Dist.	> LFL	Equiv	Expl	Equi v	Expl	Dist	Pso	Pref	Dur	Damage	Damage	Covered	Årea	
e Chemical Name	K	kg/sec	LFL	Conc	Wt.	kJ/kg	(m)	n	kg	kg TNT	Effc	kg TNT	Prob	П	psi	ps i	sec	m	M	sq.m.	sq.m.	
Petroleum Naphth	298	1.3	0.008	0.030	100	42400	2.10	2.3	83													
Petroleum Naphth	298	2.6	0.008	0.030	100	42400	2.64	4.0	251												•	
ic Petroleum Naphth	298	16.7	0.008	0.030	100	42400	4.88	17.5	4927	49905	0.016	782	0.076	6Ø	1.986	4.182	0.07	83	413	2827	2827	
Pheno I	298	0.1	0.017	0.029	94.1	31170	Ø.82	0.2	1													
Pheno I	298	0.9	Ø. Ø17	₿.₿29	94.1	31170	1.69	1.4	30													
c Phenol	298	9.3	0.017	0.029	94.1	31170	3.85	8.9	1274	9488	0.000											
Styrene	298	1.3	0.011	0.021	104	41960	2.10	2.0	67													
Styrene	298	5.7	0.011	0.021	104	41960	3.42	6.7	712													
c Styrene	298	39.6	0.011	₿.021	164	41960	6.48	31.4	15825	158627	0.097	15350	0.198	179	1.969	4.144	Ø.19	121	607	25165	25165	
Toluene	298	6.1	0.040	0.023	92.1	40550	2.89	5.0	393													
Toluene	298	23.0	Ø.040	0.023	92.1	40550	4.48	14.3	3289	3186Ø	0.006	202	0.051	35	1.988	4.186	Ø.Ø4	71	358	962	962	
Toluene Diisocya	298	0.02	0.009	0.022	174	23900	0.48	0.1	Ø		1											
Toluene Diisocya	298	0.1	0.009	0.022	174	23900	0.72	0.2	1													•
c Toluene Diisocya	298	1.2	0.009	0.022	174	23900	1.77	1.7	49													
: Vinyl acetate	298	32.6	Ø. Ø26	0.044	86.1	22690	4.94	22.0	7749	42003	0.011	471	Ø.115	43	2.009	4.232	Ø.Ø6	78	390	1452	1452	
: Vinyl acetate	298	77.5	0.026	0.044	86.1	22690	6.57	44.0	30973	167886	0.104	17455	0.302	174	1.991	4.193	0.20	124	619	23779	22327	
: Xylene	298	2.5	0.011	0.020	108	40841	2.60	3.4	188													
: Xylene	298	7.6	0.011	0.020	106	40841	3.76	8.3	1114	10869	0.000											
ic Xylene .	298	49.B	Ø.Ø11	0.020	106	40841	6.99	37.5	22550	220007	0.143	31547	Ø.25Ø	228	1.991	4.193	0.25	135	677	4Ø828	40828	

Section 2

															•				TNO	TNO			
			Release	9							Max								Dist.	Dist.			
		Àmb.	Rate							Vapor	TNT	Max	TNT						Heavy	Glass	Area	Unmixed	l Net
1		•	kg/sec				Combust	Rad.	Dist.		-	•	Equiv	•		Pso	Pref	Dur	Damage	Damage	Covered	Агеа	Area
Case	Chemical Name	K	or kg*	LFL	Conc	Wt.	kJ/kg	(m)	m 	kg	kg TNT	Effc	kg. TNT	Prob	m 	psi	ps i	sec	m 	m	sq.m.	sq.m.	· sq.m.
50c	Acetic Acid	298	9.0	0.054	0.095	60.1	13110	2.84	7.0	791	2477	0.000	Ø										
w/	Acrylonitrile	298	41.9	0.030	0.053	53.1	33200	5.95	30.5	14060	113991	0.062	7040	0.190	130	2.002	4.217	0.15	108	540	13273	154	13119
. w/	Aniline	298	0.6	0.013	0.026	93.1	34830	1.57	7.1	19	2633	0.000	Ø										
w/	Ammonia+	298	45360	0.150	0.218	17.0	18610	52.93	53.7	30615	138582	0.081	11234	0.304	134	1.999	4.210	0.18	115	577	14103	42273	Mitigated
w/	HexaCH3 Diamine	298	0.05	0.007	0.021	116	28400	0.74	7.0	Ø	2480	0.000	, Ø										
w/	Propylene Oxide+	298			0.050		30230	62.29	53.7	44188	321587	0.213	68565	∅.368	289	2.002	4.217	Ø.33	153	767	65597	156228	Mitigated
100c	Acetic Acid∗	298	22680	0.054	0.095	60.1	13110	37.68	37.6	19830	62106	0.023	1451								V		
w/	Aceta I dehyde+	. 298	2268Ø	0.040	0.077	44.1	24640	45.40	53.2	20787	184466	0.117	21533	0.350	180	1.998	4.208	Ø.19	111	557	25447	108103	Mitigated
w/	Butyra dehyde+	298	2268Ø	Ø.Ø25	0.037	72.1	3538Ø	47.14	53.2	21837	246674	Ø.163	40124	0.354	234	1.998	4.203	0.23	128	639	43005	5675	37330
w/	Formal dehyde	298	1.5	0.070	Ø.173	30.0	19035	1.72	37.7	58	62372	0.024	1468	0.231	58	1.986	4.180	0.01	14	72	2642	Ø	2642
w/	Chloroprene*	298	22680	0.040	0.050	88.5	30000	38.77	53.2	20787	211084	Ø.137	28876	0.350	204	1.998	4.207	Ø.18	119	595	32685	6362	26323
w/	TDI	298	1.2	0.009	0.022	174	23900	1.77	37.7	49	62384	0.024	1469	Ø.231	58	1.986	4.182	0.02	15	73	2642	Ø	2642
,w/	Sodium (H2)*	298	3944	0.040	0.295	2.0	11978Ø	56.68	40.8	3615	185544	0.102	16911	0.256	174	2.004	4.220	Ø.23	105	527	23779	Ø	23779
w/	Ethylene oxide+	298	2268Ø	0.036	0.077	44.1	. 27650	46.20	53.2	21071	201284	₿.129	26062	0.351	195	2.002	4.217	0.21	116	581	29865	130100	Mitigated
w/	Acetic Anhydride	298	44.1	Ø.Ø27	0.050	102	16420	5.03	45.9	11091	105610	0.055	5834	0.302	105	1.998	4.208	0.11	79	395	8659	Ø	8659
50c	Acrylic Ácid	298	4	0.024	0.065	72.1	1884Ø	2.49	4.4	315	1418	0.000	Ø										
w/	Acrylonitrile	298	41.9	0.030	0.053	53.1	33200	5.95	30.0	14060	112932	0.061	6881	0.185	130	1.989	4.187	0.14	108	540	13273	154	13119
100c	Acrylic Acid+	298	2268Ø	0.024	0.065	72.1	18840	43.15	37.6	21903	98581	0.050	4912				•						
w/	Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	45.9	11091	142084	Ø.Ø84	11909	0.313	146	1.991	4.192	Ø.12	79	395	16742	Ø	16742
w/	Aniline	298	9.4	0.013	0.026	93.1	34830	3.90	38.9	1532	111331	0.060	6645	Ø.258	119	1.993	4.196	0.07	52	262	11122	Ø	11122
w/	TDI	298	1.2	0.009	0.022	174	23900	1.77	37.7	49	98859	0.050	4947	0.246	105	2.002	4.216	0.02	15	73	8659	Ø	8659
w/	Formal dehyde	298	1.5	0.070	0.173	30.0	19035	1.72	37.7	58	98846	Ø.050	4945	0.246	105	2.001	4.215	0.02	14	72	8659	Ø	8659
w/	Sodium (H2)∗	298	3944	0.040	0.295	2.8	119780	58.88	40.8	3815	202018	Ø.13Ø	26268	0.270	208	2.003	4.220	0.25	105	527	33979	Ø	33979
w/	Butyraldehyde*	298	22680	0.025	0.037	72.1	3538Ø	47.14	53.2	21837	283149	Ø.188	53213	0.363	262	1.999	4.210	0.24	128	639	53913	5675	48238
w/	HexaCH3 Diamine	298	Ø.6	0.007	0.021	116	28400	1.69	37.7	24	98743	0.050	4932	0.246	105	1.999	4.210	0.02	12	61	8659	Ø	8659
w/	Chloroprene*	298	22680	0.040	0.050	88.5	30000	38.77	53.2	20787	247559	Ø.163	40423	0.359	234	2.003	4.219	Ø.19	119	595	43005	6362	38643
50c	Aceta I dehyde*	298	45360	0.040	0.077	44.1	24840	57.07	53.2	41575	244721	Ø.161	39466										
w/	Ammonia+	298	4536Ø	0.150	Ø.218	17.0	18610	52.93	75.3	30815	380828	0.250	95029	Ø.457	307	2.000	4.213	Ø.26	115	577	74023	150378	Mitigated
w/	Acetic Anhydride	298	5.8	Ø.027	0.050	102	16420	2.55	53.5	408	246322	Ø.162	40005	Ø.355	233	2.001	4.214	0.05	26	131	42638	108103	Mitigated
1	HCN+	298	45360	a ase	Ø 143	97 A	24530	57.31	75.3	30400	475603	a 200	143460	a 19a	383	2 002	4.217	ø 33	138	688	103401	207641	Mitigated
w/ .		230	,0000	.	D. 170	21.0	27000			00700	710000	D. JUL	TAGAGE	0,700	203	2.002	7.211	0 .00	100	000	IUJTJI	LUIUTI	

																			TNO	TNO			
			Release	3 .							Max								Dist.	Dist.			
		Amb.	Rate				Heat	Cloud	Virt.	Vapor	TNT	Max	TNT						Heavy	Glass	Area	Unmixed	Net
		Temp	kg/sec	* *	Stoic	Mole	Combust	Rad.	Dist.	> LFL	Equiv	Expl	Equiv	Expl	Dist	Pso	Pref	Dur	Damage	Damage	Covered	Area	Агеа
Case	Chemical Name	K	or kg*	LFL	Conc	Wt.	kJ/kg	(m)	, M	kg	kg TNT	Effc	kg TNT	Prob	m	p s i	psi	sec	m	m	sq.m.	sq.m,	sq.m.
100c	Aceta I dehyde*	298	22680	0.040	0.077	44.1	24640	45,40	37.6	20787	122360	Ø.Ø68	8359										
w/	Butyl Alcohol*	298	2268Ø	0.014	0.034	74.1	33100	52.08	53.2	22466	300005	0.199	59757	Ø.361	274	2.003	4.219	Ø.27	126	631	58965	108103	Mitigated
w/	Isopropanol*	298	2268Ø	0.023	8.044	60.1	30150	49.25	53.2	21968	28Ø588	0.186	52246	Ø.359	260	2.000	4.212	0.25	121	607	53093	108103	Mitigated
w/	Methyl Alcohol*	298	2268Ø	0.080	0.122	32.0	1958Ø	43.75	53.2	19444	213311	∅.138	29533	0.348	206	1.997	4.206	Ø.19	101	505	33329	109178	Mitigated
w/	Aniline :	298	9.4	0.013	0.028	93.1	34830	3.90	38.9	1532	135110	0.078	10584	0.249	145	2.001	4.214	0.08	52	262	16513	108103	Mitigated
w/	TDI	298	1.2	Ø.Ø09	Ø.Ø22	174	23900	1.77	37.7	49	122639	0.069	8405	Ø.238	133	1.994	4.199	0.02	15	73	13893	108103	Mitigated
5Øc	Butyraldehyde*	298	45360	Ø.Ø25	0.037	72.1	3538Ø	59.25	53.2	43675	369137	0.243	89555									,	
w/	Acetic Anhydride	298	5.6	0.027	6.050	102	18420	2.55	53.5	408	370738	0.244	90298	Ø.364	322	2.003	4.219	0.08	26	131	81433	5675	75758
w/	Aniline	298	Ø.6	0.013	0.026	93.1	3483Ø	1.57	53.3	19	369293	0.243	89628	0.363	322	1.998	4.207	0.03	12	60	81433	5675	75758
w/	HexaCH3 Diamine	298	0.05	0.007	0.021	116	28400	Ø.74	53.2	Ø	369140	0.243	89557	Ø.363	322	1.997	4.206	Ø.01	3	16	81433	5675	75758
100c	Butyra dehyde+	298	22680	0.025	6.037	72.1	35380	47.14	37.6	21837	184568	0.117	21560										
w/	Butyl Alcohol*	298	2268Ø	0.014	Ø.034	74.1	33100	52.08	53.2	22466	362213	∅.238	8637Ø	Ø.365	317	2.001	4.214	0.29	126	631	78924	5675	73249
w/	Isopropanol*	298	22680	0.023	0.044	60.1	30150	49.25	53.2	21968	342796	0.227	77667	0.363	304	2.002	4.217	Ø.27	121	607	72583	5675	669Ø8
w/	Methyl Alcohol*	298	2268Ø	0.060	Ø.122	32.0	1958Ø	43.75	53.2	19444	275518	Ø.183	50351	0.352	256	2.001	4.215	0.21	101	505	51472	6750	44722
w/	TDI	298	1.2	0.009	Ø.Ø22	174	23900	1.77	37.7	49	184847	0.117	21632	Ø.245	198	1.997	4.206	Ø.Ø3	15	73	30172	5675	24497
w/	Pheno I	298	9.3	Ø.Ø17	0.029	94.1	31170	3.65	38.7	1274	194056	0.124	24Ø68	0.254	203	2.003	4.218	0.08	48	238	32365	5675	2669Ø
w/	Ethyl Alcohol*	298	2268Ø	0.033	0.065	46.1	2688Ø	47.46	53.2	21283	321237	0.213	68418	0.360	289	2.004	4.221	0,25	116	578	65597	5675	59922
5Øc	Formaldehyde	298	0.2	0.070	0.173	30.0	19035	Ø.88	Ø.4	2	11	0.000	Ø				•						
w/	Acetic Anhydride	298	5.6	0.027	0.050	102	16420	2.55	5.1	498	1612	0.000	Ø										
100c	Formal dehyde	298	1.5	0.070	0.173	30.0	19035	1.72	1.9	58	266	0.000	Ø								•		
w/	Aniline	298	9.4	Ø.Ø13	Ø.026	93.1	3483Ø	3.90	10.0	1532	13016	0.001	9	0.022	8	1.957	4.117	0.02	52	262	. 50	Ø	5Ø
w/	HexaCH3 Diamine	298	0.6	0.007	0.021	116	28400	1.69	2.3	24	427	0.000	Ø										
5Øc	HCN+	298	45360	0.058	Ø.143	27.0	24530	57.31	53.2	39400	230882	0.151	34939										
w/	Acetone*	298	4536Ø	Ø. Ø21	0.050	58.1	28500	62.23	75.3	44188	531731	Ø.329	175186	Ø.486	393	2.003	4.220	Ø.38	150	752	121304	116280	5024
w/	Acrylonitrile	298	41.9	0.030	0.053	53.1	33200	5.95	60.9	14060	342396	0.226	77491	Ø.4ØØ	296	2.002	4.217	Ø.22	1Ø8	540	68813	99692	Mitigated
w/	Vinyl acetate	298	77.5	0.028	0.044	86.1	22690	8.57	69.1	30973	398769	0.260	103657	0.452	325	1.996	4.204	Ø.27	124	619	82958	100990	Mitigated
w/	Ethyl acetate	298	67.1	0.022	0.040	88.1	23510	6.50	67.2	26816	381492	0.250	95345	0.441	316	1.997	4.206	0.27	119	597	78427	99785	Mitigated
w/	Isopropanol	298	28.3	0.023	0.044	60.1	30150	5.41	57.9	8252	290321	Ø.193	55960	Ø.379	263	1.996	4.204	Ø.18	88	438	54325	99538	Mitigated
w/	Methyl alcohol	298	36.8	0.060	0.122	32.0	1958Ø	5.25	59.Ø	10315	279130	Ø.185	51698	Ø.387	253	2.001	4:215	Ø.17	82	408	50273	100613	Mitigated
•	Butyl Acrylate	298	6.1	0.014	0.023	128	32240	3.08	53.6	608	235547	Ø.155	36439	0.347	224	2.002	4.216	0.07	37	187	39408		Mitigated
	Butyl alcohol	298	7.4	0.014	0.034	74.1	33100	3.69	53.9	1146	239946	0.158	37878	Ø.349	227	2.005	4.223	0.10	47	234	40471		Mitigated

																			TNO	TNO			
			Releas	е							Max								Dist.	Dist.			
		Amb.	Rate				Heat	Cloud	Virt.	Vapor	TNT	Max	TNT						Heavy	Glass	Area	Unmixed	Net
		-	kg/sec				Combust		Dist.	> LFL	Equi v	Expl	Equiv	Expl	Dist	Pso	Pref	Dur	Damage	Damage	Covered	Хгеа	Área
Ca	se Chemical Name	K	or kg*	LFL	Conc	Wt.	kJ/kg	(w)	m	kg	kg TNT	Effc	kg TNT	Prob	m	psi	ps i	sec	m	m	sq.m.	sq.m.	sq.m.
	c HCN+	298			0.143						115441												
	/ Styrene	298			0.021								49814	0.326	259	2.002	4.217	0.24	121	607	52685	99538	Mitigated
	Ammonia≠	298			0.218						136105			4		4 005	4 000		404				10555
	/ MEK	298			0.037						305540						4.200		124	621		43935	12055
	/ Vinyl acetate	298			0.044			6.57			•		61348						124	619		43725	9777
	/ Ethyl acetate	298			0.040						286715						4.205		119	597		42500	6981
	/ Styrene	298			0.021			3.42			143241								43	216			Mitigated
	/ Acetic Anhydride				0.050			2.55					11068	0.302	133	2.004	4.221	Ø.04	26	131	13893	422/3	Mitigated
	Ethylene oxide+				0.077	–		58.07			278357					0 440	4 010	<i>a</i> 10				100100	
	/ Isopropanol	298			0.044		30150						75482						88	438			Mitigated
	/ Methyl alcohol	298			0.122		1958Ø	5.25					70878						82	408			Mitigated
	/ Butyl alcohol	298			0.034			3.69					54843	0.361	265	1.995	4.200	W.10	47	234	55155	130100 1	ditigated
	c Ethylene oxide*				0.077			46.20			139178			a 05a	400		4 100	a ao	50	000	00007	100100	
	/ Aniline	298			0.026			3.90					13907				4.199		52	262			Witigated
	/ Phenol	298			0.029			3.65					13228	0.249	160	1.992	4.195	0.01	48	238	20106	130100 1	Mitigated
	Propylene oxide				0.050			62.29			319110			a 007			4 010	<i>a</i> 0 <i>a</i>		400	01018	450000	
	/ Isopropanol	298			0.044			5.41					93953						88	438			ditigated
	/ Methyl alcohol	298			0.122		19580						88733						82	408			Mitigated
	/ Butyl alcohol	298			0.034		-	3.69					71345	0.370	294	1.997	4.205	Ø.11	47	234	67887	156228 1	ditigated
	c Propylene oxide				0.050			49.56			159555			a 050	100	1 007	4 000	a ao	40	000	05447	150000	***
	/ Phenol	298			0.029			3.65					17727				4.208		48	238			litigated
	/ Aniline	298			0.026			3.90			172305			0.257	183	1.990	4.204	60.0	52	262	26392	156228 1	ditigated
	Aniline	298			0.026			1.57	1.1	19		0.000	5400			0 000	4 000	a 12	105	For	0044	04000	
	/ Methyl chloride				0.122			43.00			102858			0.318			4.228		105	525			litigated
	/ Vinyl chloride*				0.077			51.49			186298			Ø.350	182	1,994	4.200	0.22	128	641	26016	/1631 N	litigated
	c Aniline	298			0.026		34830				12750		9		470		4 104	a 10		F05	00590		47144
	/ Chloroprene≠	298			0.050			38.77			161728						4.194		119	595	23506	6362	17144
	/ Acrylonitrile*	298			0.053			47.62			183228						4.214		124	622	29255	154	29101
	/ Sodium (H2)*	298					119780				116187			0.080	145	1.443	4.198	w.20	105	527	16513	0	16513
	/ Maleic Anhydride				0.065		14200	2.75			14854		15			4 007	4 100	a ac	70	005	0764	_	0794
٧	/ Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	28.1	11091	56254	w.w20	1101	Ø.169	59	1.980	4.169	80.0	79	395	2734	0	2734

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			Releas								Max								TNO Dist.	TNO Dist.			
		≜ mh	Rate	5			Heat	Cloud	Viet	Vapor		Max	TNT							Glass	A	11 1	I N-1
			kg/sec		Staic	Wala	Combust			•				Eval	Diet	Dea	Deaf	Du-	-		Covered	Unmixed	
Case	Chemical Name							_	DISU.		•	•	kg TNT			psi	psi	sec	•	vamage m	sq.m.	sq.n.	Area sq.m.
										•	_					•			IA		5q.m.	5q.m.	sq.m.
100c	HexaCH3 Diamine	298	. Ø.6	0.007	Ø.021	116	28400	1.69	1.2	24	162	0.000	ø										
w/	Acetic Anhydride	298	44.1	0.027	0.050	102	16420	5.03	26.4	11091	43665	0.012	529	Ø.153	42	1.974	4.155	0.07	79	395	1385	Ø	1385
w/	Chloroprene*	298	2268Ø	0.040	0.050	88.5	30000	38.77	37.7	20787	149140	Ø.089	13326	Ø.238	161	2.000	4.211	Ø.15	119	595	20358	6362	13996
w/	Acrylonitrile*	298	22680	0.030	0.053	53.1	33200	47.62	37.7	21495	170640	0.106	18105	0.243	182	2.005	4.222	0.20	124	622	26016	154	25862
w/	Maleic Anhydride	298	5.6	0.014	0.085	98.1	14200	2.75	6.3	620	2266	0.000	Ø			•					•		
50c	Sodium (H2)*	298	3944	0.040	Ø.295	2.0	11978Ø	58.68	15.7	3615	103438	0.054	5540	0.056	133	1.991	4.193	Ø.19	105	527	13893		
w/	HCN+	298	45360	0.056	0.143	27.0	24530	57.31	55.5	39400	334320	Ø.221	73977	Ø.36Ø	296	2.002	4.216	0.30	138	688	68813	99538	Mitigated
w/	Styrene	298	5.7	Ø.Ø11	0.021	104	41980	3.42	17.1	712	110574	0.059	6534	0.087	140	1.992	4.195	Ø.Ø6	43	216	15394	Ø	15394
100c	Sodium (H2)*	298	3944	0.040	Ø.295	2.0	119780	56.68	15.7	3615	103438	0.054	5540										
w/	Acetone*	298	22680	0.021	0.050	58.1	28500	49.51	40.8	22094	253862	Ø.168	42583	Ø.271	252	1.998	4.208	Ø.24	119	597	49878	16742	33134
`w/	Acrylonitrile*	298	2268Ø	0.030	0.053	53,1	33200	47.51	40.8	21460	273638	0.181	49655	Ø.267	267	2.002	4.218	Ø.24	124	622	5599Ø	154	55838
w/	Aniline	298	9.4	0.013	0.026	93.1	3483Ø	3.90	18.5	1532	116187	0.063	7375	0.080	145	1.993	4.196	Ø.Ø8	52	262	16513	Ø	18513
w/	Butyl Acrylate	298	26.4	0.014	0.023	128	32240	5.00	25.3	6313	152060	0.092	13935	0.140	176	2.004	4.222	∅.12	82	410	24328	8	24328
w/	Butyl Alcohol*	298	22680	0.014	0.034	74.1	33100	52.06	40.8	22466	281Ø82	0.187	52431	0.273	273	1.998	4.208	0.27	126	631	58535	Ø	58535
w/	Ethyl Acetate*	298	2268Ø	0.022	0.040	88.1	23510	44.43	40.8	22032	227176	0.149	33766	Ø.271	230	2.000	4.213	Ø.21	112	559	41548	227	41321
w/	Ethyl Alcohol*	298	2268Ø	0.033	0.065	46.1	26880	47.46	40.8	21283	240106	Ø.158	37931	0.266	241	1.997	4.206	0.23	116	578	45617	Ø	45617
w/	Formal dehyde	298	1.5	0.070	Ø.173	30.0	19ø35	1.72	15.8	58	103703	0.054	5576	0.057	133	1.995	4.202	0.02	14	72	13893	6	13893
w/	HexaCH3 Diamine	298	Ø.6	0.007	0.021	116	28400	1.69	15.7	24	103599	0.054	5562	Ø.Ø57	133	1.994	4.199	0.02	12	61	13893	Ø	13893
w/	Isopropanol*	298	2268Ø	0.023	0.044	60.1	30150	49.25	40.8	21968	261666	Ø.173	45320	0.270	258	1.999	4.211	0.25	121	607	52279	Ø	52279
w/	Methyl Alcohol*	298	2288Ø	0.080	0.122	32.0	1958Ø	43.75	40.8	19444	194388	0.124	24158	0.254	202	1.993	4.197	Ø.19	101	505	32047	1075	30972
w/	Vinyl Acetate∗	298	2268Ø	0.026	0.044	86.1	22690	42.87	40.8	21770	221442	0.144	31986	0.269	225	2.002	4.217	0.20	110	550	39761	11347	28414
· w/	MEK	298	2268Ø	Ø.Ø18	0.037	72.1	31360	49.70	40.8	22269	270267	0.179	48419	0.272	265	1.996	4.204	0.25	124	618	55155	1662	53493
w/	Ethyl Acrylate*	298	22680	0.018	0.034	100	2763Ø	45.24	40.8	22269	250424	Ø.165	41399	0.272	249	2.001	4.214	0.22	118	592	48695	Ø	48695
w/	Methy Inethacry I	298	22680	Ø.Ø21	0.034	100	26400	44.10	40.8	22094	242778	0.160	38816	Ø.271	243	1.999	4.210	Ø.21	116	582	46377	Ø	46377
w/	Chloroprene*	298	22680	0.040	0.050	88.5	30000	38.77	40.8	20787	252415	Ø.167	42083	Ø.263	251	1.996	4.204	Ø.19	119	595	49481	6362	43119
100c	Pheno I	298	9.3	Ø.017	Ø.029	94.1	31170	3.65	8.9	1274	9488	0.000	3										
w/	Butadiene*	298	22680	0.020	0.037	54.1	44600	53.70	38.7	22154	245531	Ø.162	39738	Ø.258	247	2.002	4.217	Ø.26	139	693	47916	255176	Mitigated
w/	Acrylonitrile*	298	22680	0.030	0.053	53.1	33200	47.51	38.7	21460	179688	0.113	20318	Ø.251	190	1.999	4.211	0.20	124	622	28353	154	28199
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1			Releas	е							Max								Dist.	Dist.			
		Amb.	Rate				Heat	Cloud	Virt.	Vapor	TNT	Max	TNT				`		Heavy	Glass	Area	Unmixed	Net
ı		Temp	kg/sec		Stoic	Mole	Combust	Rad.	Dist.	> LFL	Equiv	Expl	Equiv	Expl	Dist	Pso	Pref	Dur	Damage	Damage	Covered	Area	Årea
	Chemical Name		•		Conc	Wt.	kJ/kg	(m)	m	kg	kg TNT	Effc	kg TNT	Ргоь	m					m	sq.m.		sq.m.
100c	TDI	298			0.022							0.000											
ŵ/	Ethyl Alcohol*	298	22680	0.033	0.065	46.1	2688Ø	47.46	37.7	21283	136947	0.080	10925	0.242	148	2.004	4.220	Ø.18	116	578	17203	Ø	17203
₩ /	Butyl Alcohol*	298	2268Ø	0.014	0.034	74.1	33100	52.06	37.7	22468	177923	0.112	19877	0.250	189	2.004	4.220	0.23	126	631	28055	Ø	28055
w/	HexaCH3 Diamine	298	0.8	0.007	0.021	116	28400	1.69	2.1	24	440	0.000	Ø										
w/	Isopropanol*	298	2268Ø	0.023	0.044	80.1	30150	4925	37.7	21968	1585Ø6	0.097	15324	0.246	170	2.008	4.226	0.20	121	607	22698	Ø	22698
· /	Methyl Alcohol*	298	22680	9.060	Ø.122	32.0	19580	43.75	37.7	19444	91228	0.044	4035	Ø.228	96	1.993	4.198	0.13	101	505	7238	1075	6163
w/	Aniline	298	9.4	0.013	0.026	93.1	34830	3.90	9.9	1532	13028	0.001	9	0.022	8	1'. 965	4.135	0.02	52	262	50	Ø	5Ø
50c	Alcoholic Bev.	298	0.3	0.033	0.065	46.1	26880	1.17	0.1	5	35	0.000	Ø	_									
/	Acetaldehyde	298	4536Ø	0.040	0.077	44.1	24640	57.07	53.2	41575	244755	0.161	39478	0.354	232	2.000	4.213	Ø.27	140	702	42273	108103 1	litigated
* /	Butyraldehyde+	298	45360	0.025	0.037	72.1	3538Ø	59.25	53.2	43675	369171	Ø.243	89571	Ø.363	322	1.997	4.206	Ø.33	161	805	81433	5675	75758
w/	Ethylene oxide+	298	4536Ø	Ø. Ø36	0.077	44.1	27650	58.07	53.2	42141	278391	Ø.185	51421	0.356	258	2.003	4.219	Ø.29	147	733	52279	130100 N	litigated
w/	Propylene oxides	298	4536Ø	0.021	0.050	58.1	30230	62.29	53.2	44188	319145	0.212	87544	Ø.365	288	2.000	4.212	Ø.33	1.53	767			litigated
	Sodium (H2)*	298		0.040	Ø.295	2.0	119780	56.68	15.7	3615	103472	0.054	5545	0.056	133	1.992	4.194	Ø.19	105	527	13893		13893
100c	Alcoholic Bev.	298	0.8	0.033	Ø.Ø65	48.1	2688Ø	1.61	0.2	26	166	0.000	ø										
w/	TDI	298	1.2	0.009	0.022	174	23900	1.77	2.2	49	445	0.000	ø										
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•												-					-	_			Area
																			54.a.	5q.m.	sq.m.
298	3944	0.040	Ø.295	2.0	11978Ø	56.68	15.7	3615	103438	0.054	5540	0.056	133	1.991	4.193	Ø.19	105	527	13893	Ø	13893
298	55738	0.049	0.085	31.1	31400																
298	60154	Ø.15	0.218	17	18610																
298 °	115892	0.089	0.143	21.7	24761	77.77	85.1	90972	538118	0.332	178922	0.502	387	2.000	4.213	Ø.45	183	913	117628	87816	30012
298	17040	0.03	0.057	30.1	47510																
298	81Ø64	₿.15	0.218	17	18610																
298	981#4	0.105	0.167	18.4	2363Ø	73.72	78.3	73714	416114	0.270	112249	0.461	326	1.999	4.210	Ø.39	168	838	83469	42273	41196
298	14940	0.03	0.057	30.1	47510	49.45	30.6	14159	160703	0.098	15811	0.183	180	1.998	4.204	0.20	122	610	25447	Ø	25447
298	34080	0.03	0.057	30.1	47510															_	
298	60048	0.056	Ø.143	27	2453∅				,												
298	94128	0.043	0.094	28	3285Ø	80.66	76.7	85399	670179	Ø.39Ø	261340	0.490	459	2.000	4.212	Ø.51	196	982	165468	99538	6593Ø
298	534	0.04	Ø.295	2	119780																
298	81648	Ø.15	∅.218	17	18610																
298	82182	Ø.131	0.221	16.2	19267	66.76	71.7	57891	266455	Ø.177	47Ø35	0.415	231	1.998	4.207	0.30	144	722	41910	42273	Mitigated
298	45360	Ø.Ø36	0.077	63.5	18490	51.49	53.2	42141	186142	0.118	21967	0.356	181	2.005	4.224	Ø.22	128	641	25730		
298	3715	0.025	0.077	26.0	48240	32.15	55.4	3577	227363	Ø.149	33825	Ø.371	216	1.996	4.202	Ø.15	78	388	36644	71631	Mitigated
298	45360	0.150	Ø.218	17.0	18610	52.93	76.8	30615	363468	Ø.239	86944	0.468	294	2.003	4.219	Ø.26	115	577	67887	136572	Mitigated
298	0.2	0.007	0.021	116	28400	1.18	55.4	4	227391	0.149	33834	Ø.371	216	1.998	4.203	0.02	7	34	36844	71631	Mitigated
298	€ Ø.6	0.013	0.026	93.1	3483Ø	1.58	55.4	19	227519	Ø.149	33874	Ø.371	216	1.997	4.205	0.02	12	60	36644	71631	Mitigated
298	14578	0.018	0.034	74.1	33840	43.03	30.2	14314	115713												_
298	9072	0.033	Ø.065	46.1	2688Ø	35.07	38.4	8513	170380	0.108	18Ø43	0.124	181	2.004	4.222	Ø.15	85	426	25730	Ø	25730
298	72970	Ø. Ø18	0.034	74.1	33840	73.22	67.5	71647	579199												
298	2268Ø	Ø. Ø33	0.065	46.1	26880	47.46	77.3	21283	715867	0.408	291892	Ø.241	479	1.997	4.206	Ø.31	116	578	180203	Ø	180203
298	6856	Ø.Ø16	0.034	56.1	45330	37.67	20.7	6763	73240												
298	2268Ø	0.014	0.034	74.1	33100	52.08	43.0	22466	25Ø884	Ø.166	41557	0.250	247	2.003	4.218	Ø.26	126	631	47916	ø.	47916
298	31700	Ø.111	0.644	42.1	45790	47.63						**									
298					30150	49.25	58.3	21988	414752	Ø. 269	111566	Ø.246	345	2.000	4.213	Ø. 28	121	607	93482	Ø	93482
298													_			-				-	_
										Ø.319	162222	Ø.228	383	1.999	4.210	Ø. 26	101	505	115209	1075	114134
298											-										
298									-	0 227	78314	Ø 124	305	2 441	4 214	Ø. 19	85	426	73Ø62	ø	73Ø62
	Amb. Temp K 298 298 298 298 298 298 298 298 298 298	Release Amb. Rate Temp kg/sece K or kg	Release Amb. Rate Temp kg/sec* K or kg LFL	Release Amb. Rate Temp kg/sec+ Stoic K or kg LFL Conc	Release Amb. Rate Temp kg/sec* Stoic Mole K or kg LFL Conc Wt. 298 3944 0.040 0.295 2.0 298 55738 0.049 0.085 31.1 298 60154 0.15 0.218 17 298 17640 0.03 0.057 30.1 298 81064 0.15 0.218 17 298 14940 0.03 0.057 30.1 298 34080 0.03 0.057 30.1 298 34080 0.03 0.057 30.1 298 34080 0.03 0.057 30.1 298 81048 0.056 0.143 27 298 94128 0.043 0.094 28 298 534 0.04 0.295 2 298 81648 0.15 0.218 17 298 82182 0.131 0.221 16.2 298 45360 0.036 0.077 63.5 298 3715 0.025 0.077 26.0 298 45360 0.160 0.218 17.0 4 298 0.2 0.007 0.021 116 298 0.6 0.013 0.026 93.1 298 14578 0.018 0.034 74.1 298 9072 0.033 0.065 46.1 298 22680 0.034 0.034 74.1 298 22680 0.014 0.034 74.1 298 22680 0.023 0.044 60.1 298 22680 0.023 0.044 60.1 298 22680 0.023 0.044 60.1 298 22680 0.023 0.044 60.1 298 22680 0.023 0.044 60.1	Release Amb. Rate Temp kg/sec* K or kg LFL Conc Wt. kJ/kg	Release Amb. Rate Temp kg/sec* Stoic Mole Combust Rad. K or kg LFL Conc Wt. kJ/kg (m)	Release Amb. Rate Temp kg/sec* K or kg LFL Conc Wt. kJ/kg (m) m 298 3944 0.040 0.295 2.0 119780 56.68 15.7 298 55738 0.049 0.085 31.1 31400 298 60154 0.15 0.218 17 18610 298 115892 0.089 0.143 21.7 24761 77.77 85.1 298 17040 0.03 0.057 30.1 47510 298 81064 0.15 0.218 17 18610 298 98104 0.105 0.187 18.4 23630 73.72 78.3 298 14940 0.03 0.057 30.1 47510 298 34080 0.03 0.057 30.1 47510 298 84084 0.056 0.143 27 24530 298 94128 0.043 0.094 28 32850 80.66 76.7 298 8534 0.04 0.295 2 119780 298 81648 0.15 0.218 17 18610 298 82182 0.131 0.221 16.2 19267 66.76 71.7 298 45360 0.036 0.077 63.5 18490 51.49 53.2 298 3715 0.025 0.077 26.0 48240 32.15 55.4 298 45360 0.150 0.218 17.0 18610 52.93 76.8 298 0.2 0.007 0.021 116 28400 1.18 55.4 298 0.6 0.013 0.026 93.1 34830 1.58 55.4 298 0.72 0.033 0.065 46.1 26800 35.07 38.4 298 9072 0.033 0.065 46.1 26800 47.46 77.3 298 22680 0.034 0.044 42.1 33840 73.22 67.5 298 22680 0.014 0.034 74.1 33840 73.22 67.5 298 22680 0.014 0.034 74.1 33840 73.22 67.5 298 22680 0.014 0.034 74.1 33840 73.22 67.5 298 22680 0.014 0.034 74.1 33100 52.06 43.0 298 31700 0.111 0.044 42.1 45790 47.63 44.5 298 22680 0.023 0.044 60.1 30150 49.25 68.3 298 22680 0.023 0.044 60.1 30150 49.25 68.3 298 22680 0.034 0.065 46.1 26700 66.91 63.8 298 22680 0.080 0.122 32.0 19580 43.75 74.1 298 36485 0.018 0.034 74.1 33840 58.25 47.8	Release Amb. 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Vapor TNT Max TNT Temp kg/sec	Release	Release	Release	Release Roless R

1																			TNO	TNO			
			Release	•							Max								Dist.	Dist.			
		Amb.	Rate							Vapor		Max	TNT						Heavy	Glass	Area	Unmixed	Net
		Temp	kg/sec	*	Stoic	Mole	Combust	Rạd.	Dist.) LFL	Equiv	Expl	Equiv	Expl	Dist	Pso	Pref	Dur	Damage	Damage	Covered	Area	Area
Case	Chemical Name	K	or kg	LFL'	Conc	Wt.	kJ/kg	(m)		kg	kg TNT	Effc	kg TNT	Prob	m	ps i	psi	sec	m	m	sq.m.	sq.m.	sq.m.
17	Ethyl ether	298	· 7289	0.018	Ø. Ø34	74.1	3384Ø	34.23	21.3	7157	57856												
	Ethyl Alcohol	298	2268Ø	0.033	0.065	46.1	2688Ø	47.48	43.3	21283	194525	0.124	24195	0.241	199	2.001	4.214	Ø.21	116	578	31103	ø	31103
18	Methyl ether	298	32603	0.034	0.065	46.1	28710	53.23	45.1	30493	209141												•
	Methyl Alcohol	298	22680	Ø.060	Ø.122	32.0	19580	43.75	58,.8	19444	300091	Ø.199	59791	Ø.228	269	1.999	4.209	0.22	101	505	56832	1075	55757
19	Chloroethane	298	24410	0.082	0.065	99.0	7700	34.06	39.1	20793	38247												
.•	Ethyl Alcohol	.298	9072	0.033	Ø.Ø65	46.1	2688Ø	35.07	45.7	8513	92915	0.046	4228	0.124	90	1.994	4.200	0.11	85	426	6362	Ø	6362
20	Chloroethane	298	32117	Ø.062	0.065	99.0	7700	37.29	44.8	27358	50323												
	Ethyl Alcohol	298	4538Ø	Ø.Ø33	0.065	46.1	2688Ø	59.65	69.6	42567	323661	Ø.215	69435	0.358	275	1.998	4.208	Ø.32	146	728	59398	Ø	59396
21	Chloromethane	298	25114	Ø.107	0.122	50.5	12680	35.38	39.8	18772	56862												
•	Methyl alcohol*	298	36.8	0.080	0.122	32.0	1958Ø	5.25	47.1	10315	105109	0.055	5766	0.145	103	2.004	4.220	Ø.12	82	408	8332	1075	7257
22	Isopropylchior.	298	19538	0.028	0.044	78.6	25000	41.47	34.9	18635	111292												
•	Isopropanol*	298	28.3	Ø.Ø23	0.044	60.1	30150	5.41	41.7	8252	170730	0.106	18127	0.121	179	1.992	4.195	Ø.15	88	438	25165	Ø	25165
23	Phosphine	298	497	0.01	0.095	34	32000	17.02	5.6	495	3787	0.000											
24	Phosphine	298	558	0.01	0.095	34	32000	17.69	5.9	556	4251	0.000											

Case 1 = Sodium metal / Alkaline Liquid, Ammonium Nitrate Solution, Ferric Chloride Solution, Hydrochloric Acid, Hydrofluorosilicic Acid, Hydrogen Peroxide, Hydroiodic Acid, Phosphoric Acid, Potassium Hydroxide Liq., Pulp Mill Liquid, Sulfuric Acid Spent, Toluene Diisocyanate

Case 2 = Anhydrous Ammonia / Methyl Chloride

Case 3 = Anhydrous Ammonia / Motor Fuel Antiknock Compound

Case 4 = Motor Fuel Antiknock Compound / Hydrochloric Acid, Hydroiodic Acid, Hydrofluorosilicic Acid, Phosphoric Acid, Sulfuric Acid Spent, Aniline Dil

Case 5 = Motor Fuel Antiknock Compound / Hydrocyanic Acid

Case 6 = Anhydrous Ammonia / Sodium Metal

Case 7 = Vinyl Chloride / Alkaline Liquid, Potassium Hydroxide Liq., Sodium Hydroxide Liq.

Case 8 = Vinyl Chloride / Anhydrous Ammonia

Case 9 = Vinyl Chloride / Hexamethylene Diamine Solution

Case 10 = Vinyl Chloride / Aniline Oil

Case 11 = Alcoholic Beverage / Nitric Acid Fuming, Dleum, Sulfuric Acid

Case 12 = Ethyl Alcohol or Denatured Alcohol / Nitric Acid Fuming, Oleum, Sulfuric Acid

Case 13 = Butyl Alcohol / Nitric Acid Fuming, Oleum

Case 14 = Isopropanol / Nitric Acid Fuming, Oleum

Case 15 = Methyl Alcohol / Nitric Acid Funing, Dleum, Sulfuric Acid

Case 18 = Sulfuric Acid, Spent / Ethyl Alcohol, Denatured Alcohol

Case 17 = Sulfuric Acid, Spent / Alcoholic Beverage

Case 18 = Sulfuric Acid, Spent / Methyl Alcohol

Case 19 = Hydrochloric Acid / Alcoholic Beverage

Case 20 = Hydrochloric Acid / Ethyl Alcohol, Denatured Alcohol

Case 21 = Hydrochloric Acid / Methyl Alcohol

Case 22 = Hydrochloric Acid / Isopropanol

Case 23 = Phosphorus / Alkaline Liquid, Potassium Hydroxide Liq., Sodium Hydroxide Liq.

Case 24 = Phosphorus / Hydrochloric Acid, Hydroiodic Acid



Table I5. Explosions

						•	Distance	to (m)		
						Side-on	Side-on	Refl.	Refl.	
		Heat of			TNT	Glass	Severe			Area
		Decomp.	Weight		Equiv	Damage	Damage	Damage	Damage	Covered
Case	Chemical Name	kcal/mol	16	Wt.	IB TNT	.15 psig	2 psig	.15 psig	2 psig	sq.m.
1	Polyethylidene Peroxide	97.2	2000	60.0	3234.4	811	122	1623	243	46534
2	Hydrogen Peroxide	23.1	1900	34.0	1292.0	598	90	1195	179	25239
3	Hydrogen Peroxide	23.1	95Ø	34.Ø	646.0	474	71	949	142	15900
4	Ethyl Hypochlorite	89.3	1000	80.5	1107.8	568	85	1135	178	22780
5	Ethyl Hypochlorite	89.3	2000	80.5	2215.6	715	107	1430	215	36161
6	Dinitrotoluene	157.3	2000	183.Ø	1717.6	657	99	1314	197	30516
7	Ammonium Nitrate	88.1	2000	8Ø.Ø	2199.2	713	107	1427	214	35982
8	Butyl Hypochlorite	79.5	2000	108.5	1463.2	623	93	1246	187	27423
9	Isopropyl Hypochlorite	103.2	2000	94.5	2181.8	712	197	1423	213	35792
10	Methy! Hypochlorite	94.2	2000	66.5	2831.0	776	116	1552	233	42580
11	Octyl Hypochlorite	59.7	2000	164.5	724.8	493	74	986	148	17168
12	Nitrogen trichloride	54.972	2000	120.4	912.5	532	80	1064	160	20017
13	DNT and H202		2000		1849	673	101	1347	202	32053
14	Acetyl Nitrate	130.08	2000	105	2475.5	742	111	1484	223	38936
15	Ethyl Nitrate	74.28	2000	91	1631.1	646	97	1292	194	29482
16	Ethyl Nitrate	74.28	1000	91	815.5	513	77	1025	154	18573
17	Butyl Nitrate	161.6	2000	119	2713.6	765	115	1530	23Ø	41394
18	Enol Nitrate	139.85	2000	89	3139.9	803	121	1607	241	45623
19	Isopropyl Nitrate	165.17	2000	105	3143.3	804	121	1607	241	45656
20	Methyl Nitrate	134.63	2000	77	3493.8	833	125	1665	25Ø	48990
21	Octyl Nitrate	142.17	2000	175	1623.4	645	97	1290	193	29389
22	Nitrobenzene	186.05	2000	123	3022.5	793	119	1587	238	44479
23	Maleyl Dinitrate	148.09	2000	206	1436.5	619	93	1238	186	27088
24	3-nitroacrylonitrile	67.74	2999	98	1381.2	611	92	1222	183	26389

Table I5. (Cont.)

Case	1st Chem	nical Name	2nd Chemical	Name
1	Acetaldeh	nvde	Hydrogen Perox	cide
	Hydrogen		Acetic Acid	••
u	Hydrogen		Acetic Anhydr	ide
II	Hydrogen	Peroxide	Acetone	
ti	Hydrogen	Peroxide	Acrylic Acid	
п	Hydrogen		Acrylonitrile	
. 11	Hydrogen-		Aniline Oil	
11	Hydrogen		Benzene	
11	Hydrogen		Butyl Acrylate	9
П		Peroxide	Butyl Alcohol	
н	Hydrogen	Peroxide	Butyraldehyde	
. 11		Peroxide	Carbon Disulf	ide
u		Peroxide	Chloroprene	-
п		Peroxide	Denatured Alco	ohol
н	Hydrogen	Peroxide	Ethyl Acetate	
11	Hydrogen	Peroxide	Ethyl Acrylate	9
11	Hydrogen	Peroxide	Ethyl Alcohol	•
11	Hydrogen	Peroxide	Ethylene Oxide	•
11	Hydrogen	Peroxide	Isopropanol	
n	Hydrogen	Peroxide	Maleic Anhydr	ide
П	Hydrogen	Peroxide	Methyl Alcoho	
11	Hydrogen		Methyl Ethyl I	(etone
ff	Hydrogen	Peroxide	Methyl Methaci	rylate
11	Hydrogen	Peroxide	Motor Fuel And	tiknock Comp
"	Hydrogen	Peroxide	Nitric Acid, A	-uming
11	Hydrogen	Peroxide	Octyl Alcohol	
11	Hydrogen	Peroxide	01eum	•
11	Hydrogen		Phenol	
H	Hydrogen		Phosphorus	
	Hydrogen		Propylene Oxid	ie ·
11	Hydrogen	Peroxide	Sodium Metal	
11	Hydrogen	Peroxide	Styrene Monome	er
	Hydrogen		Toluene	
	Hydrogen		Toluene Diiso	cyanate
	Hydrogen		Vinyl Acetate	
П	Hydrogen		Vinyl Chloride	9
11	y a og o	Peroxide	Xylene	
3	Hydrogen	Peroxide	Alcoholic Beve	erage
11	Hydrogen	Peroxide	Butadiene	
11	Hydrogen	Peroxide	Coal Tar Dist	
	Hydrogen	Peroxide	Combustible L	
	Hydrogen	Peroxide	Compound, Clea	aning
	Hydrogen		Crude Oil	1 0 1 3
11			Distillate Fue	
. "	Hydrogen		Ferric Chloric	
"		Peroxide	Flammable Liqu	
	Hydrogen		Formaldehyde S)UIII.
11	nyurogen	Peroxide	Fuel Oil	•
ít 		Peroxide	Fuel, Aviation Gasoline	1
	nyurogen	Peroxide	3011116	

Case	1st Chemical Name	2nd Chemical Name
11	Hydrogen Peroxide	Gasoline, nec
11	Hydrogen Peroxide	Hexamethylene Diamine Sol
11	Hydrogen Peroxide	Hydrochloric Acid
ıı	Hydrogen Peroxide	Hydrocyanic Acid
11	Hydrogen Peroxide	LPG-Butene
11	Hydrogen Peroxide	LPG-Butylene
11	Hydrogen Peroxide	LPG-Isobutylene
11	Hydrogen Peroxide	LPG-Propylene
II	Hydrogen Peroxide	Methyl Chloride
11	Hydrogen Peroxide	0il
u	Hydrogen Peroxide	Petroleum Naphtha
11	Hydrogen Peroxide	Petroleum Residual Fuel
п	Hydrogen Peroxide	Petroleum, Partially Refi
	Hydrogen Peroxide	Pulp Mill Liquid
4	Chlorine	Alcoholic Beverage
	Chlorine	Denatured Alcohol
	Chlorine	Ethyl Alcohol
	Dinitrotoluene	Acetaldehyde
	Dinitrotoluene	Alkaline Liquid
п	Dinitrotoluene	Butyraldehyde
11	Dinitrotoluene	Chlorine
п	Dinitrotoluene -	Oleum
п	Dinitrotoluene	Phosphorus
11	Dinitrotoluene	Potassium Hydroxide
11	Dinitrotoluene	Sodium Hydroxide
H	Dinitrotoluene	Sodium Metal
If	Dinitrotoluene	Sulfuric Acid
11	Dinitrotoluene	Sulfuric Acid, Spent
7	Ammonium Nitrate Soln.	Chlorine
11	Ammonium Nitrate Soln.	Oleum
18	Ammonium Nitrate Soln.	Phosphorus
ñ	Ammonium Nitrate Soln.	Sodium Metal
13	Ammonium Nitrate Soln.	Sulfuric Acid
8	Butyl Alcohol	Chlorine
	Isopropanol	Chlorine
	Methyl Alcohol	Chlorine
11	Octyl Alcohol	Chlorine
	Anhydrous Ammonia	Chlorine Chlorine
	Dinitrotoluene	Hydrogen Peroxide
	Nitric Acid	Acetic Acid
	Nitric Acid	Acetic Anhydride
	Nitric Acid	Denatured Alcohol
11	Nitric Acid	Ethyl Alcohol
11	Nitric Acid	Glycol Ethers
	Nitric Acid	Alcoholic Beverage
	Nitric Acid	Butyl Alcohol
18	Nitric Acid	Acetaldehyde
	Nitric Acid	Acetone
10	Nitric Acid	Butyraldehyde
19	Nitric Acid	Isopropanol

Table I5. (Cont.)

Case 1st Chemical Name	2nd Chemical Name
20 Nitric Acid	Methyl Alcohol
21 Nitric Acid	Octyl Alcohol
22 Nitric Acid	Benzene
23 Nitric Acid	Maleic Anhydride
24 Nitric Acid	Acrylonitrile

· Table I6. Tank Car Explosions during Pool Fires

		•													Tank		Side on	Side on	Refl.	Refl.	
	Crit	SLT	Boil	Flash	Spec.		Lig	Vap		Flsh	Vapor	Liquid	Press			lbs of	Glass				λгеа
	Temp	Temp	.Temp	Fract	Heat		•	-				•					Damage				Covered
Chemical Name	C	С			Ratio	Ratio		0 bp	C		#/cuft	#/cuft	psia	psia	psig	Equiv	n	n	m	л	sq.m.
Acetic Acid	322	253	118	0.63							1.5913	83.27	305		240	18	143	22	287	43	1453
Acetone	238	174	56	0.53	1.12	0.76	49.30	0.130	150	0.39	1.6783	62.99	26Ø	175	240	582	458	69	916	137	14838
Acrylonitrile	263	218	77	Ø.12	1.15	0.62	49.92	Ø.113	150	Ø.25	0.3737	55.98	330	109	240	683	483	72	966	145	16506
Ammonia	132	83	-33	0.45	1.31	0.53	42.43	0.045	67	0.34	0.8293	56.10	640	500	500	2392	734	110	1467	220	38054
Aniline	. 426	262	184	Ø.33	1.08	1.09	63.65	0.005	150	-0.2	0.0127	69.61	80								
Benzene	289	225	80	0.66	1.10	1.00	46.11	Ø.178	150	0.31	3.0475	60.54	310	82	240	98	253	38	5Ø8	76	4531
Butadiene	152	104	-4	0.59	1.12			Ø.156		Ø.54	2.6402	50.74	275	275	750	785	508	76	1012	152	18109
Butane	152	105	-1	Ø.83	1.11	0.91	38.19	0.159	105	0.58	2.7016	47.48	245	245	750	677	482	72	963	144	16398
Carbon Dioxide	31	-6	-79	0.38	1.30	0.47	73.63	0.154	-6	0.34	2.1046	94.60	375	375	240	1188	581	87	1162	174	23873
Carbon Disulfide	279	168	48	0.37	1.21	0.66	80.50	0.181	15Ø	Ø.29	1.0717	95.48	255	193	24Ø	983	546	82	1091	164	21039
Chlorine	144	. 93	-35	0.50	1.35	0.60	97.34	0.220	93	0.45	4.4782	129.93	500	500	75 ø	1324	6Ø2	90	1205	181	25649
Cyclohexane	28Ø	220	81	Ø.66	1.08	0.97	48.05	0.170	150	0.32	3.0209	63.30	265	- 8Ø	240	103	257	39	514	77	4665
Ethyl Alcohol	243	189	78	0.46	1.13	0.61	48.67	0.092	150	Ø.27	1.3428	62.93	330	132	240	342	384	. 58	768	115	18414
Ethylene Oxide	196	140	10	Ø.51	1.21	0.68	55.54	0.116	140	Ø.47	2.0684	73.19	435	435	500	1787	666	100	1332	200	31334
Hexane	234	184	69	0.74	1.08	1.24	40.56	0.190	150	0.52	3.3357	53.38	205	116	240	217	33Ø	49	659	99	7677
Hydrogen Cyanide	184	128	26	0.30	1.30	0.37	43.06	0.070	128	0.27	0.8438	54.66	270	270	1250	899	530	79	1059	159	19820
Hydrogen Fluoride	188	132	20	0.69	1.11	1.15	60.53	0.050	132	0.64	0.6983	77.93	350	35∅	1000	6728	1036	155	2071	311	75829
Isopropanol	235	200	82	Ø.55	1.10	0.60	48.67	Ø.128	150	0.27	4.4127	68.33	355	122	240	130	278	42	556	83	5461
LPG-Butene	146	98	-8,	0.61	1.10	0.88	37.44	∅.158	98	Ø.56	2.5125	48.81	250	250	750	752	499	75	998	150	17587
LPG-Isobutane	135	88	-12	Ø.62	1.09	0.91	34.94	Ø.166	88	Ø.57	2.5623	45.42	23Ø	230	75Ø	643	474	71	947	142	15855
LPG-Isobutylene	145	98	-7		1.10		36.82	Ø.162	96	0.00	2.4517	47.77		Ø	750						
LPG-Propane	97	53	-42	Ø.57	1.13	Ø.79	36.19	0.150	53	0.52	2.3716	47.15	27Ø	270	850	749	498	75	997	149	17553
LPG-Propylene	92	52	-48	Ø.56	1.15	0.71	38.08	Ø.138	52	0.51	2.7902	50.78	315	315	85Ø	794	508	76	1016	152	18240
Methyl Alcohol	240	186	65	0.35	1.20	0.40	48.80	0.073	150	Ø.21	1.2271	61.33	425	195	240	437	416	62	832	125	12247
Sulfur Dioxide	158	5Ø	-10	Ø.22	1.29	0.58	91.10	0.190	5Ø	Ø.21	Ø.5486	100.85	115	115	500	63Ø	470	71	941	141	15633
Vinyl Acetate	252	189	73	0.63	1.10	1.01	58.03	0.180	150	Ø.4Ø	2.2065	73.79	230	143	240	425	413	62	825	124	12031
Xy lene	346	235	139	0.40	1.07	0.87	53.66	0.150	150	0.05	Ø.6693	61.93	110	22	240	5	94	14	188	28	625

Table I6. Tank Car Explosions during Pool Fires

																	Side	Side				
															Tank		on	on	Refl.	Refl.		
	Crit	SLT	Boil	Flash	Spec.		Liq	Vap		Flsh	Vapor	Liquid	Press		Burst	lbs of	Glass	Severe	Glass	Severe	Агеа	
	Temp	Temp	.Temp	Fract	Heat	Vapor	Dens	Dens	Temp	p.Frct	Dens.	Dens.	0 SLT	Pres	Press	TNT	Damage	Damage	Damage	Damage	Covered	
Chemical Name	С	С	С	0 SLT	Ratio	Ratio	0 bp	Q bp	C	i	#/cuft	#/cuft	psia	psia	psig	Equiv	M	n	m	m	sq.m.	
Acetaldehyde	188		20.4		1.182		48.5	0.117	 7						240	80	236	35	472	71	3944	
Acrylic Acid	342		141		1.121										240	89	245	37	491	74	4256	
Butyl Acrylate	327		149		1.08										240	97	252	38	505	76	4500	
Chloroprene					1.12										240	90	246	37	491	74	4262	
Ethyl Acrylate	279		99.6		1.08			Ø. 222	2						240	97	252	38	505	76	4500	
Methyl Methacryla	294		101		1.059			0.19)						24Ø	102	256	38	512	77	4637	
Propylene Oxide	209		34.3		1.133		50.5	8.14	ŀ						240	87	243	37	487	73	4191	
Styrene	373		145		1.074			0.18	3						240	98	253	38	5Ø7	76	4536	
Vinyl Chloride	158		13.8		1.186			0.47	7.						75Ø	305	369	55	738	111	9633	
Tetraethyl Lead					1.161										750	321	376	56	752	113	9984	
Ammonium Nitrate		/														24978	1604	241	3208	481	1818Ø8	
Dinitrotoluene											~					83626	2399	360	4798	720	406882	
Tetraethyl Lead						~						•				12.4	127	19	254	38	1140	
Hydrogen Peroxide	:															67341	2232	335	4484	67Ø	352176	

 Volume of Tank car =
 20000 gallons
 Z (0.15 psig) = 180

 Liquid fraction 6 Rupture =
 0.4
 Z (2 psig) =
 27

1 + 1

APPENDIX J

CONSEQUENCE AND RISK RANKINGS OF INCOMPATIBLE CHEMICAL COMBINATIONS

APPENDIX J

Consequence and Risk Rankings of Incompatible Chemical Combinations

Appendix J contains the following Tables:

- Table J1 Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Pages J2-J30);
- Table J2 Summary of Consequences (Net Area Above the Lethal Limit) from Spills of Single Chemicals (Pages J31-J33);
- Table J3 Consequence Rankings by Binary Combinations (Sort by Area Above Lethal Limit) (Pages J34-J47);
- Table J4 Consequence Rankings Based on ASTM Groups (Sort by Total Area Above the Lethal Limit) (Pages J48-J50)
- Table J5 Risk Rankings by Binary Combinations (Pages J51-J64);
- Table J6 Risk by Specific Chemical (Pages J65-J66);
- Table J7 Risk by ASTM Group (Page J67)

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations

ASTM #	Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
5	Aceta I dehyde	27	Dinitrotoluene, Liquid	Explosions	30516
5	Acetaldehyde	104	Hydrogen Peroxide	Explosions	46534
5	Acetaldehyde	104	Hydrogen Peroxide	Toxic Emission	37699
.5	Acetaldehyde	21	Sodium Metal	Fire Ball	34309
3	Acetic Acid, Glacial	5	Acetaldehyde	Toxic Emission	37699
3	Acetic Acid, Glacial	107	Acetic Anhydride	Fire Ball	18004
3	Acetic Acid, Glacial	167	Acetic Anhydride	Toxic Emission	131947
3	Acetic Acid, Glacial	107	Acetic Anhydride	Vapor Cloud Explosion	8659
3	Acetic Acid, Glacial	26	Acrylonitrile	Fire Ball	17414
3	Acetic Acid, Glacial	26	Acrylonitrile	Toxic Emission	3927
3	Acetic Acid, Glacial	26	Acrylonitrile	Vapor Cloud Explosion	13119
3	Acetic Acid, Glacial	10	Alkaline Liquid	Fire Ball	11433
3	Acetic Acid, Glacial	10	Alkaline Liquid	Toxic Emission	131947
3	Acetic Acid, Glacial	10	Alkaline Liquid	Vapor Cloud Explosion	2552
3	Acetic Acid, Glacial	106	Ammonium Nitrate Soln.	Fire Ball	821
3	Acetic Acid, Glacial	106	Ammonium Nitrate Soln.	Toxic Emission	6283
3	Acetic Acid, Glacial	7	Aniline Oil, Liquid	Fire Ball	1592
3	Acetic Acid, Glacial	7	Aniline Oil, Liquid	Toxic Emission	62832
3	Acetic Acid, Glacial	5	Butyraldehyde	Fire Ball	25385
3	Acetic Acid, Glacial	5	Butyraldehyde	Toxic Emission	131947
3	Acetic Acid, Glacial	5	Butyraidehyde	Vapor Cloud Explosion	37330
3	Acetic Acid, Glacial	17	Chloroprene, Inhibited	Fire Ball	23833
3	Acetic Acid, Glacial	17	Chloroprene, Inhibited	Toxic Emission	176715
3	Acetic Acid, Glacial	17	Chloroprene, Inhibited	Vapor Cloud Explosion	26323
3	Acetic Acid, Glacial	5	Formaldehyde Solution	Fire Ball	13661
3	Acetic Acid, Glacial	5	Formaldehyde Solution	Toxic Emission	131947
3	Acetic Acid, Glacial	5	Formaldehyde Solution	Vapor Cloud Explosion	2642
3	Acetic Acid, Glacial	-7	Hexamethylene Diamine Soln.	Fire Ball	1417
3	Acetic Acid, Glacial	7	Hexamethylene Diamine Soln.	Toxic Emission	62832
3	Acetic Acid, Glacial	104	Hydrogen Peroxide	Explosions	25239
3	Acetic Acid, Glacial	104	Hydrogen Peroxide	Fire Ball	11433
3	Acetic Acid, Glacial	104	Hydrogen Peroxide	Toxic Emission	129591
3	Acetic Acid, Glacial	194	Hydrogen Peroxide	Vapor Cloud Explosion	2552
3	Acetic Acid, Glacial	10	Potassium Hydroxide, Liq	Fire Ball	11433
3	Acetic Acid, Glacial	10	Potassium Hydroxide, Liq	Toxic Emission	131947
3	Acetic Acid, Glacial	10	Potassium Hydroxide, Liq	Vapor Cloud Explosion	2552
3	Acetic Acid, Glacial	10	Sodium Hydroxide, Liq.	Fire Ball	11433
3	Acetic Acid, Glacial	10	Sodium Hydroxide, Liq.	Toxic Emission	131947
3	Acetic Acid, Glacial	10	Sodium Hydroxide, Liq.	Vapor Cloud Explosion	2552
3	Acetic Acid, Glacial	21	Sodium Metal	Fire Ball	47910
3	Acetic Acid, Glacial	21	Sodium Metal	Toxic Emission	131947
3	Acetic Acid, Glacial	21	Sodium Metal	Vapor Cloud Explosion	23779
3	Acetic Acid, Glacial	18	Toluene Diisocyanate	Fire Ball	19150
3	Acetic Acid, Glacial	18	Toluene Diisocyanate	Toxic Emission	131947
3	Acetic Acid, Glacial	18	Toluene Diisocyanate	Vapor Cloud Explosion	2642

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
19	Acetone	104	Hydrogen Peroxide	Explosions	25239
19	Acetone	184	Hydrogen Peroxide	Fire Ball	1867
19	Acetone	104	Hydrogen Peroxide	Toxic Emission	12959
19	Acetone	104	Hydrogen Peroxide	Vapor Cloud Explosion	387
19	Acetone	21	Sodium Metal	Fire Ball	430
19	Acetone	21	Sodium Metal	Toxic Emission	1885
19	Acetone	21	Sodium Metal	Vapor Cloud Explosion	3313
3	Acrylic Acid	107	Acetic Anhydride	Fire Ball	1901
3	Acrylic Acid	107	Acetic Anhydride	Toxic Emission	1256
3	Acrylic Acid	107	Acetic Anhydride	Vapor Cloud Explosion	1674
3	Acrylic Acid	26	Acrylonitrile	Fire Ball	1687
3	Acrylic Acid	26	Acrylonitrile	Toxic Emission	392
3	Acrylic Acid	26	Acrylonitrile	Vapor Cloud Explosion	1311
3	Acrylic Acid	10	Alkaline Liquid	Fire Ball	1473
. 3	Acrylic Acid	10	Alkaline Liquid	Toxic Emission	٠ .
3	Acrylic Acid	10	Alkaline Liquid	Vapor Cloud Explosion	865
3	Acrylic Acid	106	Ammonium Nitrate Soln.	Fire Ball	44
3	Acrylic Acid	106	Ammonium Nitrate Soln.	Toxic Emission	Į.
3	Acrylic Acid	7	Aniline Oil, Liquid	Fire Ball	2751
3	Acrylic Acid	7	Aniline Oil, Liquid	Toxic Emission	3769
3	Acrylic Acid	7	Aniline Oil, Liquid	Vapor Cloud Explosion	1112
3	Acrylic Acid	5	Butyraldehyde	Fire Ball	2686
3	Acrylic Acid	5	Butyraldehyde	Toxic Emission	ł
3	Acrylic Acid 🕝	. 5	Butyraldehyde	Vapor Cloud Explosion	4823
3	Acrylic Acid	17	Chloroprene, Inhibited	Fire Ball	2556
3	Acrylic Acid	17	Chloroprene, Inhibited	Toxic Emission	17671
3	Acrylic Acid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	3664
3	Acrylic Acid	5	Formaldehyde Solution	Fire Ball	1484
3	Acrylic Acid	5	Formaldehyde Solution	Toxic Emission	1885
3	Acrylic Acid	5	Formaldehyde Solution	Vapor Cloud Explosion	865
3	Acrylic Acid	7	Hexamethylene Diamine Soln.	Fire Ball	23Ø1
3	Acrylic Acid	7	Hexamethylene Diamine Soln.	Toxic Emission	
3	Acrylic Acid	7	Hexamethylene Diamine Soln.	Vapor Cloud Explosion	865
3	Acrylic Acid	104	Hydrogen Peroxide	Explosions	2523
3	Acrylic Acid	104	Hydrogen Peroxide	Fire Ball	1473
3	Acrylic Acid	104	Hydrogen Peroxide	Toxic Emission	12959
3	Acrylic Acid	104	Hydrogen Peroxide	Vapor Cloud Explosion	865
3	Acrylic Acid	10	Potassium Hydroxide, Liq	Fire Ball	1473
3	Acrylic Acid	10	Potassium Hydroxide, Liq	Toxic Emission	
3	Acrylic Acid	10	Potassium Hydroxide, Liq	Vapor Cloud Explosion	865
3	Acrylic Acid	10	Sodium Hydroxide, Liq.	Fire Ball	1473
· 3	Acrylic Acid	10	Sodium Hydroxide, Liq.	Toxic Emission	
3	Acrylic Acid	10	Sodium Hydroxide, Liq.	Vapor Cloud Explosion	865
3	411- 4-11	21	Sodium Metal	Fire Ball	5158
3	Acrylic Acid	21	Sodium Metal	Toxic Emission	ì

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM	,	ASTM			Area Covered
#	Chemical Name	#	Chemical Name	Consequence Type	sq.m.
3	Acrylic Acid	21	Sodium Metal	Vapor Cloud Explosion	33979
3	Acrylic Acid	18	Toluene Diisocyanate	Fire Ball	20828
3	Acrylic Acid	18	Toluene Diisocyanate	Toxic Emission	28274
3	Acrylic Acid	18	Toluene Diisocyanate	Vapor Cloud Explosion	8659
26	Acrylonitrile	104	Ferric Chloride Solution	Fire Ball	16427
26	Acrylonitrile	104	Ferric Chloride Solution	Toxic Emission	3927
26	Acrylonitrile	164	Ferric Chloride Solution	Vapor Cloud Explosion	12714
26	Acrylonitrile .	104	Hydrogen Peroxide	Explosions	25239
26	Acrylonitrile	164	Hydrogen Peroxide	Fire Ball	8902
26	Acrylonitrile	104	Hydrogen Peroxide	Toxic Emission	129591
26	Acrylonitrile	104	Hydrogen Peroxide	Vapor Cloud Explosion	25862
26	Acrylonitrile	31	Pheno I	Fire Ball	41571
26	Acrylonitrile .	31	Phenol	Toxic Emission	60476
26	Acrylonitrile	31	Phenol	Vapor Cloud Explosion	28199
4	Alcoholic Beverage	5	Butyraldehyde	Fire Ball	26934
4	Alcoholic Beverage	5	Butyraldehyde	Vapor Cloud Explosion	75758
4	Alcoholic Beverage	104	Chlorine	Explosions	22780
4	Alcoholic Beverage	- 104	Hydrogen Peroxide	Explosions	15900
4	Alcoholic Beverage	104	Hydrogen Peroxide	Toxic Emission	9425
4	Alcoholic Beverage	21	Sodium Metal	Fire Ball	9839
4	Alcoholic Beverage	21	Sodium Metal	Vapor Cloud Explosion	13893
4	Alcoholic Beverage	18	Toluene Diisocyanate	Fire Ball	226
4	Alcoholic Beverage	18	Toluene Diisocyanate	Toxic Emission	28274
10	Alkaline Liquid	107	Acetic Anhydride	Fire Ball	8050
10	Alkaline Liquid	107	Acetic Anhydride	Toxic Emission	12568
10	Alkaline Liquid	107	Acetic Anhydride	Vapor Cloud Explosion	1320
10	Alkaline Liquid	19	Acetone	Fire Ball	15865
10	Alkaline Liquid	19	Acetone	·Toxic Emission	27489
10	Alkaline Liquid	19	Acetone	Vapor Cloud Explosion	42654
10	Alkaline Liquid	26	Acrylonitrile	Fire Ball	8902
10	Alkaline Liquid	26	Acrylonitrile	Toxic Emission	60476
10	Alkaline Liquid	17	Carbon Tetrachloride	Toxic Emission	318086
10	Alkaline Liquid	17	Chloroprene, Inhibited	Fire Ball	11471
10	Alkaline Liquid	17	Chioroprene, Inhibited	Toxic Emission	176715
10	Alkaline Liquid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
10	Alkaline Liquid	27	Dinitrotoluene, Liquid	Explosions	3Ø516
10	Alkaline Liquid	13	Ethyl Acetate	Fire Ball	12469
10	Alkaline Liquid	13	Ethyl Acetate	Toxic Emission	18850
10	Alkaline Liquid	13	Ethyl Acetate	Vapor Cloud Explosion	13876
10	Alkaline Liquid	107	Maleic Anhydride	Fire Bali	1092
10	Alkaline Liquid	107.	Maleic Anhydride	Toxic Emission	1
10	Alkaline Liquid	19	Methyl Ethyl Ketone	Fire Ball	13556
10	Alkaline Liquid	19	Methyl Ethyl Ketone	Toxic Emission	3927
10	Alkaline Liquid	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	24354
10	Alkaline Liquid	105	Phosphorus	Fire Ball	1162

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
10	Alkaline Liquid	105	Phosphorus	Toxic Emission	50265
10	Alkaline Liquid	21	Sodium Metal	Fire Ball	9835
10	Alkaline Liquid	21	Sodium Metal	Vapor Cloud Explosion	13893
10	Alkaline Liquid	16	Styrene Monomer	Fire Ball	26658
10	Alkaline Liquid	16	Styrene Monomer	Toxic Emission	3142
10	Alkaline Liquid	16	Styrene Monomer	Vapor Cloud Explosion	25165
10	Alkaline Liquid	18	Toluene Diisocyanate	Fire Ball	190
10	Alkaline Liquid	18	Toluene Diisocyanate	Toxic Emission	28274
10	Alkaline Liquid	13	Viny! Acetate	Fire Ball	9096
10	Alkaline Liquid	13	Vinyl Acetate	Toxic Emission	63617
10	Alkaline Liquid	13	Vinyl Acetate	Vapor Cloud Explosion	11347
10	Alkaline Liquid	17	Vinyl Chloride	Fire Ball	8390
106	Ammonium Nitrate Soln.	107	Acetic Anhydride	Fire Ball	554
106	Ammonium Nitrate Soln.	107	Acetic Anhydride	Toxic Emission	3142
10	Anhydrous Ammonia	104	Chlorine	Explosions	20017
10	Anhydrous Ammonia	13	Ethyl Acetate	Vapor Cloud Explosion	6981
10	Anhydrous Ammonia	17	Methyl Chloride	Toxic Emission	489303
10	Anhydrous Ammonia	17	Methyl Chloride	Vapor Cloud Explosion	30012
10	Anhydrous Ammonia	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	12059
10	Anhydrous Ammonia	24	Motor Fuel Antiknock Compound	Fire Ball	17738
10	Anhydrous Ammonia	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	41196
10	Anhydrous Ammonia	13	Vinyl Acetate	Vapor Cloud Explosion	9777
7	Aniline Oil, Liquid	107	Acetic Anhydride	Fire Ball	8894
7	Aniline Oil, Liquid	107	Acetic Anhydride	Toxic Emission	37699
7	Aniline Oil, Liquid	107	Acetic Anhydride	Vapor Cloud Explosion	2734
7	Aniline Oil, Liquid	26	Acrylonitrile	Fire Ball	27126
7	Aniline Oil, Liquid	26	Acrylonitrile	Toxic Emission	60476
7	Aniline Oil, Liquid	26	Acrylonitrile	Vapor Cloud Explosion	29101
7	Aniline Oil, Liquid] 17	Carbon Tetrachloride	Toxic Emission	148446
7	Aniline Cil, Liquid	17	Chloroprene, Inhibited	 Fire Ball	1311
7	Aniline Oil, Liquid	17	Chloroprene, Inhibited	Toxic Emission	17671
7	Aniline Oil, Liquid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	1714
7	Aniline Oil, Liquid	104	Ferric Chloride Solution	Toxic Emission	2356
7	Aniline Dil, Liquid	104	Hydrogen Peroxide	Explosions	25239
7	Aniline Oil, Liquid	184	Hydrogen Peroxide	Fire Ball	3452
7	Aniline Oil, Liquid	104	Hydrogen Peroxide	Toxic Emission	129591
7	Aniline Oil, Liquid	107	Maleic Anhydride	Fire Ball	2133
7	Aniline Oil, Liquid	107	Maleic Anhydride	Toxic Emission	37699
7	Aniline Oil, Liquid	24	Motor Fuel Antiknock Compound	Fire Ball	26679
7	Aniline Oil, Liquid	24	Motor Fuel Antiknock Compound	Toxic Emission	2356
7	Aniline Oil, Liquid	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	25447
7	Aniline Oil, Liquid	21	Sodium Metal	Fire Ball	17466
7	Aniline Oil, Liquid	21	Sodium Metal	Toxic Emission	37699
7	Aniline Oil, Liquid	21	Sodium Metal	Vapor Cloud Explosion	16513
7	Aniline Bil, Liquid	18	Toluene Diisocyanate	Fire Ball	3546

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
7	Aniline Oil, Liquid	18	Toluene Diisocyanate	Toxic Emission	37699
16	Benzene	164	Hydrogen Peroxide	Explosions	25239
16	Benzene	104	Hydrogen Peroxide	Fire Ball	26041
16	Benzene	104	Hydrogen Peroxide	Toxic Emission	129591
16	Benzene	104	Hydrogen Peroxide	Vapor Cloud Explosion	37909
28	Butadiene	104	Hydrogen Peroxide	Explosions	15900
13	Butyl Acrylate	104	Ferric Chloride Solution	Fire Ball	965
13	Butyl Acrylate	104	Ferric Chloride Solution	Toxic Emission] ?
13	Butyl Acrylate	194	Hydrogen Peroxide	Explosions	25239
13	Butyl Acrylate	104	Hydrogen Peroxide	Fire Ball	7046
13	Butyl Acrylate	104	Hydrogen Peroxide	Toxic Emission	129591
13	Butyl Acrylate	104	Hydrogen Peroxide	Vapor Cloud Explosion	2463
13	Butyl Acrylate	21	Sodium Metal	Fire Ball	10215
13	Butyl Acrylate	21	Sodium Metal	Toxic Emission	1
13	Butyl Acrylate	21	Sodium Metal	Vapor Cloud Explosion	2432
4	Butyl Alcohol	5	Butyraldehyde	Fire Ball	2580
4	Butyl Alcohol	5	Butyraldehyde	Vapor Cloud Explosion	7324
4	Butyl Alcohol	194	Chiorine	Explosions	2742
4	Butyl Alcohol	104	Hydrogen Peroxide	Explosions	2523
4	Butyl Alcohol	104	Hydrogen Peroxide	Fire Ball	2026
4	Butyl Alcohol	104	Hydrogen Peroxide	Toxic Emission	12959
4	Butyl Alcohol	104	Hydrogen Peroxide	Vapor Cloud Explosion	2805
4	Butyl Alcohol	21	Sodium Metal	Fire Ball	2290
4	Butyl Alcohol	21	Sodium Metal	Toxic Emission	1963
4	Butyl Alcohol	21	Sodium Metal	Vapor Cloud Explosion	5853
4	Butyl Alcohol	18	Toluene Diisocyanate	Fire Ball	2029
4	Butyl Alcohol	18	Toluene Diisocyanate	Toxic Ensission	2827
4	Butyl Alcohol	18	Toluene Diisocyanate	Vapor Cloud Explosion	2805
5	Butyraldehyde	107	Acetic Anhydride	Fire Ball	1450
5	Butyraldehyde	107	Acetic Anhydride	Toxic Enission	314
5	Butyraldehyde	107	Acetic Anhydride	Vapor Cloud Explosion	7575
5	Butyraldehyde	10	Alkaline Liquid	Fire Ball	1109
5	Butyraldehyde	10	Alkaline Liquid	Vapor Cloud Explosion	2449
5	Butyraidehyde	7	Aniline Oil, Liquid	Fire Ball	1796
5	Butyraldehyde	7	Aniline Oil, Liquid	Toxic Emission	235
5	Butyraldehyde	7	Aniline Oil, Liquid	Vapor Cloud Explosion	7575
5	Butyraldehyde	27	Dinitrotoluene, Liquid	Explosions	3051
5	Butyraldehyde	104	Ferric Chloride Solution	Fire Ball	2694
5	Butyraldehyde	104	Ferric Chloride Solution	Vapor Cloud Explosion	2449
5	Butyraldehyde	7	Hexamethylene Diamine Soln.	Fire Ball	3127
5	Butyraidehyde	7	Hexamethylene Diamine Soln.	Vapor Cloud Explosion	7575
5	Butyraldehyde	104	Hydrogen Peroxide	Explosions	2523
5	Butyraldehyde	104	Hydrogen Peroxide	Fire Ball	1109
5	Butyraldehyde	104	Hydrogen Peroxide	Toxic Emission	12959
5	Butyraldehyde	104	Hydrogen Peroxide	Vapor Cloud Explosion	7575

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM		ASTM			Area Covered
#	Chemical Name	#	Chemical Name	Consequence Type	sq.m.
5	Butyraldehyde	31	Phenoi	Fire Ball	10574
5	Butyraldehyde	31	Phenol	Toxic Emission	3848
5	Butyraldehyde	31	Pheno I	Vapor Cloud Explosion	2669
5	Butyraldehyde	10	Potassium Hydroxide, Liq	Fire Ball	1109
5	Butyraldehyde	10	Potassium Hydroxide, Liq	Vapor Cloud Explosion	2449
5	Butyraldehyde	10	Sodium Hydroxide, Liq.	Fire Ball	1109
5	Butyraldehyde	10	Sodium Hydroxide, Liq.	Vapor Cloud Explosion	2449
5	Butyraldehyde	21	Sodium Metal	Fire Ball	2694
5	Butyraldehyde	21	Sodium Metal	Vapor Cloud Explosion	7575
5	Butyraldehyde	18	Toluene Diisocyanate	Fire Ball	1128
5	Butyraldehyde	18	Toluene Diisocyanate	Toxic Emission	2827
5	Butyraldehyde	18	Toluene Diisocyanate	Vapor Cloud Explosion	2449
20	Carbon Disulfide	104	Ferric Chloride Solution	Toxic Emission	38877
20	Carbon Disulfide	104	Hydrogen Peroxide	Explosions	2523
20	Carbon Disulfide	104	Hydrogen Peroxide	Toxic Emission	64324
20	Carbon Disulfide	34	Propylene Oxide	Toxic Emission	22540
20	Carbon Disulfide	21	Sodium Metal	Fire Ball	206
20	Carbon Disulfide	21	Sodium Metal	Toxic Emission	25054
17	Carbon Tetrachloride	20	Carbon Disulfide	Toxic Emission	25054
17	Carbon Tetrachloride	21	Sodium Metal	Toxic Emission	14844
104	Chlorine	1Ø6	Ammonium Nitrate Soln.	Explosions	3598
17	Chloroprene, Inhibited	104	Ferric Chloride Solution	Fire Ball	3294
17	Chloroprene, Inhibited	104	Ferric Chloride Solution	Toxic Emission	17671
17	Chloroprene, Inhibited	104	Ferric Chloride Solution	Vapor Cloud Explosion	5217
17	Chloroprene, Inhibited	104	Hydrogen Peroxide	Explosions	2523
17	Chloroprene, Inhibited	164	Hydrogen Peroxide	Fire Ball	1147
17	Chloroprene, Inhibited	104	Hydrogen Peroxide	Toxic Emission	17671
17	Chloroprene, Inhibited	104	Hydrogen Peroxide	Vapor Cloud Explosion	1399
17	Chloroprene, Inhibited	24	Motor Fuel Antiknock Compound	Fire Ball	3294
17	Chloroprene, Inhibited	24	Motor Fuel Antiknock Compound	Toxic Emission	17671
17	Chloroprene, Inhibited	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	5217
17	Chloroprene, Inhibited	21	Sodium Metal	Fire Ball	1147
17	Chloroprene, Inhibited	21	Sodium Metal	Toxic Emission	17671
17	Chloroprene, Inhibited	21	Sodium Metal	Vapor Cloud Explosion	4311
101	Coal Tar Distillate	104	Hydrogen Peroxide	Explosions	1590
101	Coal Tar Distillate	104	Hydrogen Peroxide	Fire Ball	59
101	Coal Tar Distillate	104	Hydrogen Peroxide	Toxic Emission	942
1Ø1	Combustible Liquid	104	Hydrogen Peroxide	Explosions	1590
101	Combustible Liquid	104	Hydrogen Peroxide	Toxic Emission	942
101	Compound, Cleaning.	104	Hydrogen Peroxide	Explosions	1590
101	Compound, Cleaning	104	Hydrogen Peroxide	Toxic Emission	942
101	Crude Dil Petroleum	104	Hydrogen Peroxide	Explosions	1590
101	Crude Oil Petroleum	104	Hydrogen Peroxide	Fire Ball	23
101	Crude Oil Petroleum	104	Hydrogen Peroxide	Toxic Emission	942
4	Denatured Alcohol	5	Butyraldehyde	Fire Ball	1949

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
4	Denatured Alcohol	5	Butyraldehyde	Vapor Cloud Explosion	59922
4	Denatured Alcohol	104	Chlorine	Explosions	36161
4	Denatured Alcohol	164	Hydrogen Peroxide	Explosions	25239
4	Denatured Alcohol	104	Hydrogen Peroxide	Fire Ball	15658
4	Denatured Alcohol	184	Hydrogen Peroxide	Toxic Emission	129591
4	Denatured Alcohol	104	Hydrogen Peroxide	Vapor Cloud Explosion	17203
4	Denatured Alcohol	21	Sodium Metal	Fire Ball	18Ø11
4	Denatured Alcohol	21	Sodium Metal	Vapor Cloud Explosion	45617
4	Denatured Alcohol	18	Toluene Diisocyanate	Fire Ball	15688
4	Denatured Alcohol	18	Toluene Diisocyanate	Toxic Emission	28274
4	Denatured Alcohol	18	Toluene Diisocyanate	Vapor Cloud Explosion	17203
27	Dinitrotoluene, Liquid	107	Acetic Anhydride	Fire Ball	8Ø5Ø
27	Dinitrotoluene, Liquid	197	Acetic Anhydride	Toxic Emission	12566
27	Dinitrotoluene, Liquid	107	Acetic Anhydride	Vapor Cloud Explosion	1320
27	Dinitrotoluene, Liquid	104	Chlorine	Explosions	30516
27	Dinitrotoluene, Liquid	104	Hydrogen Peroxide	Explosions	32Ø53
27	Dinitrotoluene, Liquid	164	Hydrogen Peroxide	Toxic Emission	129591
27	Dinitrotoluene, Liquid	197	Maleic Anhydride	Fire Ball	1092
27	Dinitrotoluene, Liquid	107	Maleic Anhydride	Toxic Emission	?
27	Dinitrotoluene, Liquid	105	Phosphorus	Explosions	30516
101	Distillate Fuel Oil	104	Hydrogen Peroxide	Explosions	15900
101	Distillate Fuel Oil	104	Hydrogen Peroxide	Toxic Emission	9425
13	Ethyl Acetate	104	Hydrogen Peroxide	Explosions	25239
13	Ethyl Acetate	194	Hydrogen Peroxide	Fire Ball	12469
13	Ethyl Acetate	104	Hydrogen Peroxide	Toxic Emission	129591
13	Ethyl Acetate	104	Hydrogen Peroxide	Vapor Cloud Explosion	13876
13	Ethyl Acetate	21	Sodium Metal	Fire Ball	14635
13	Ethyl Acetate	21	Sodium Metal	Toxic Emission	18850
13	Ethyl Acetate	21	Sodium Metal	Vapor Cloud Explosion	41321
13	Ethyl Acrylate	104	Ferric Chloride Solution	Fire Ball	5929
13	Ethyl Acrylate	184	Ferric Chloride Solution	Toxic Emission	6283
13	Ethyl Acrylate	104	Hydrogen Peroxide	Explosions	25239
13	Ethyl Acrylate	104	Hydrogen Peroxide	Fire Ball	16922
13	Ethyl Acrylate	104	Hydrogen Peroxide	Toxic Emission	129591
13	Ethyl Acrylate	104	Hydrogen Peroxide	Vapor Cloud Explosion	19856
13	Ethyl Acrylate	21	Sodium Metal	Fire Ball	19294
13	Ethyl Acrylate	21	Sodium Metal	Toxic Emission	62832
13	Ethyl Acrylate	21	Sodium Metal	Vapor Cloud Explosion	48695
4	Ethyl Alcohol	5	Butyraldehyde	Fire Ball	19498
4	Ethyl Alcohol	5	Butyraldehyde	Vapor Cloud Explosion	59922
4	Ethyl Alcohol	104	Chlorine	Explosions	36161
4	Ethyl Alcohol	184	Hydrogen Peroxide	Explosions	25239
4	Ethyl Alcohol	104	Hydrogen Peroxide	Fire Ball	15658
4	Ethyl Alcohol	104	Hydrogen Peroxide	Toxic Emission	129591
4	Ethyl Alcohol	104	Hydrogen Peroxide	Vapor Cloud Explosion	17203

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM.	Chemical Name	Consequence Type	Area Covered sq.m.
4	Ethyl Alcohol	21	Sodium Metal	Fire Ball	18011
4	Ethyl Alcohol	21	Sodium Metal	Vapor Cloud Explosion	45617
4	Ethyl Alcohol	18	Toluene Diisocyanate	Fire Ball	15688
, 4	Ethyl Alcohol	18	Toluene Diisocyanate	Toxic Emission	28274
4	Ethyl Alcohol	18	Toluene Diisocyanate	Vapor Cloud Explosion	17203
34	Ethylene Oxide	104	Hydrogen Peroxide	Explosions	25239
104	Ferric Chloride Solution	107	Acetic Anhydride	Fire Ball	554
104	Ferric Chloride Solution	107	Acetic Anhydride	Toxic Emission	.3142
104	Ferric Chloride Solution	104	Hydrogen Peroxide	Explosions	15900
104	Ferric Chloride Solution	104	Hydrogen Peroxide	Toxic Emission	129591
104	Ferric Chloride Solution	107	Maleic Anhydride	Toxic Emission	?
101	Flammable Liquid	164	Hydrogen Peroxide	Explosions	15900
101	Flammable Liquid	104	Hydrogen Peroxide	Toxic Emission	9425
5	Formaldehyde Solution	107	Acetic Anhydride	Fire Ball	554
5	Formaldehyde Solution	107	Acetic Anhydride	Toxic Emission	2356
5	Formaldehyde Solution	10	Alkaline Liquid	Toxic Emission	2356
5	Formaldehyde Solution	7	Aniline Oil, Liquid	Fire Ball	3038
5	Formaldehyde Solution	7	Aniline Oil, Liquid	Toxic Emission	49488
5	Formaldehyde Solution	104	Ferric Chloride Solution	Toxic Emission	2356
5	Formaldehyde Solution	7	Hexamethylene Diamine Soln.	Fire Ball	265
5	Formaldehyde Solution	7	Hexamethylene Diamine Soln.	Toxic Emission	18850
5	Formaldehyde Solution	104	Hydrogen Peroxide	Explosions	15906
5	Formaldehyde Solution	104	Hydrogen Peroxide	Toxic Emission	9425
5	Formaldehyde Solution	10	Potassium Hydroxide, Liq	Toxic Emission	2356
5	Formaldehyde Solution	10	Sodium Hydroxide, Liq.	Toxic Emission	2356
5	Formaldehyde Solution	21	Sodium Metal	Fire Ball	3679
5	Formaldehyde Solution	21	Sodium Metal	Toxic Emission	2350
5	Formaldehyde Solution	21	Sodium Metal	Vapor Cloud Explosion	13893
5	Formaldehyde Solution	18	Toluene Diisocyanate	Fire Ball	307
5	Formaldehyde Solution	18	Toluene Diisocyanate	Toxic Emission	27489
101	Fuel Oil	104	Hydrogen Peroxide	Explosions	15906
101	Fuel Oil	104	Hydrogen Peroxide	Fire Ball	568
101	Fuel Dil	184	Hydrogen Peroxide	Toxic Emission	942
101	Fuel, Aviation	104	Hydrogen Peroxide	Explosions	15900
101	Fuel, Aviation	104	Hydrogen Peroxide	Fire Ball	552
101	Fuel, Aviation	104	Hydrogen Peroxide	Toxic Emission	942
101	Gasoline	104	Hydrogen Peroxide	Explosions	15900
101	Gasoline	104	Hydrogen Peroxide	Toxic Emission	9425
101	Gasoline, nec	104	Hydrogen Peroxide	Explosions	15906
101	Gasoline, nec	194	Hydrogen Peroxide	Toxic Emission	942
14	Glycol Ethers	104	Ferric Chloride Solution	Fire Ball	1260
14	Glycol Ethers	104	Ferric Chloride Solution	Toxic Emission	78
7	Hexamethylene Diamine Soln.	107	Acetic Anhydride	Fire Ball	806
7	Hexamethylene Diamine Soln.	107	Acetic Anhydride	Toxic Emission	12566
7	Hexamethylene Diamine Soln.	107	Acetic Anhydride	Vapor Cloud Explosion	138

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
7	Hexamethylene Diamine Soln.	26	Acrylonitrile	Fire Ball	25392
7	Hexamethylene Diamine Soln.	26	Acrylonitrile	Toxic Emission	60476
7	Hexamethylene Diamine Soln.	26	Acrylonitrile	Vapor Cloud Explosion	25862
7	Hexamethylene Diamine Soln.	17	Carbon Tetrachloride	Toxic Emission	148440
7.	Hexamethylene Diamine Soln.	17	Chloroprene, Inhibited	Fire Ball	11492
7	Hexamethylene Diamine Soln.	17	Chloroprene, Inhibited	Toxic Emission	176715
7	Hexamethylene Diamine Soln.	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
7	Hexamethylene Diamine Soln.	104	Hydrogen Peroxide	Explosions	15900
7	Hexamethylene Diamine Soln.	104	Hydrogen Peroxide	Fire Ball	362
7	Hexamethylene Diamine Soln.	104	Hydrogen Peroxide	Toxic Emission	129591
7	Hexamethylene Diamine Soln.	107	Maleic Anhydride	Fire Ball	896
7	Hexamethylene Diamine Soln.	107	Maleic Anhydride	Toxic Emission	?
7	Hexamethylene Diamine Soln.	21	Sodium Metal	Fire Ball	5357
7	Hexamethylene Diamine Soln.	21	Sodium Metal	Vapor Cloud Explosion	13893
7	Hexamethylene Diamine Soln.	18	Toluene Diisocyanate	Fire Ball	275
7	Hexamethylene Diamine Soln.	18	Toluene Diisocyanate	Toxic Emission	28274
1	Hydrochloric Acid	3	Acetic Acid, Glacial	Fire Ball	82
1	Hydrochloric Acid	3	Acetic Acid, Glacial	Toxic Emission	628
1	Hydrochloric Acid	107	Acetic Anhydride	Fire Ball	8Ø5
1	Hydrochloric Acid	107	Acetic Anhydride	Toxic Emission	1963
1	Hydrochloric Acid	107	Acetic Anhydride	Vapor Cloud Explosion	132
1	Hydrochloric Acid	19	Acetone	Fire Ball	1586
1	Hydrochloric Acid	19	Acetone	Toxic Emission	2748
1	Hydrochloric Acid	19	Acetone	Vapor Cloud Explosion	4265
1	Hydrochloric Acid	3	Acrylic Acid	Fire Ball	4.4
1	Hydrochloric Acid	3	Acrylic Acid	Toxic Ensission	78
1	Hydrochloric Acid	26	Acrylonitrile	Fire Ball	890
1	Hydrochloric Acid	26	Acrylonitrile	Toxic Emission	6047
1	Hydrochloric Acid	26	Acrylonitrile	Vapor Cloud Explosion	2586
1	Hydrochloric Acid	4	Alcoholic Beverage	Fire Ball	2667
1	Hydrochloric Acid	4	Alcoholic Beverage	Toxic Emission	1256
1	Hydrochloric Acid	4	Alcoholic Beverage	Vapor Cloud Explosion	636
1	Hydrochloric Acid	10	Alkaline Liquid	Toxic Emission	1963
1	Hydrochloric Acid	7	Aniline Dil, Liquid	Toxic Emission	235
1	Hydrochloric Acid	13	Buty! Acrylate	Fire Ball	704
1	Hydrochloric Acid	13	Butyl Acrylate	Toxic Emission	1963
1	Hydrochloric Acid	13	Butyl Acrylate	Vapor Cloud Explosion	246
1	Hydrochloric Acid	4	Butyl Alcohol	Fire Ball	174
1	Hydrochloric Acid	4	Butyl Alcohol	Toxic Emission	78
1	Hydrochloric Acid	5	Butyraldehyde	Fire Ball	2694
1	Hydrochloric Acid	5	Butyraldehyde	Toxic Emission	78
1	Hydrochloric Acid	٠5	Butyraldehyde	Vapor Cloud Explosion	7575
1	Hydrochloric Acid	17	Chloroprene, Inhibited	Fire Ball	1147
1	Hydrochloric Acid	17	Chloroprene, Inhibited	Toxic Emission	17671
1	Hydrochloric Acid	17	Chioroprene, Inhibited	Vapor Cloud Explosion	1399

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
1	Hydrochloric Acid	101	Coal Tar Distillate	Fire Ball	597
1	Hydrochloric Acid	101	Crude Bil Petroleum	Fire Ball	230
1	Hydrochloric Acid	4.	Denatured Alcohol	Fire Ball	56667
1	Hydrochloric Acid	4	Denatured Alcohol	Toxic Emission	12566
1	Hydrochloric Acid	4	Denatured Alcohol	Vapor Cloud Explosion	59396
1	Hydrochloric Acid	13	Ethyl Acetate	Fire Ball	15323
1	Hydrochloric Acid	13	Ethyl Acetate	Toxic Emission	6283
1	Hydrochloric Acid	13	Ethy! Acetate	Vapor Cloud Explosion	19879
· 1	Hydrochioric Acid	13	Ethyl Acrylate	Fire Ball	16922
1	Hydrochloric Acid	13	Ethyl Acrylate	Toxic Emission	62832
1	Hydrochloric Acid	13	Ethy! Acrylate	Vapor Cloud Explosion	19856
1	Hydrochloric Acid	4	Ethyl Alcohol	Fire Ball	5666
1	Hydrochloric Acid	4	Ethyl Alcohol	Toxic Emission	1256
1	Hydrochloric Acid	4	Ethyl Alcohol	Vapor Cloud Explosion	5939
1	Hydrochloric Acid	5	Formaldehyde Solution	Toxic Emission	235
1	Hydrochloric Acid	101	Fuel Oil	Fire Ball	56
1	Hydrochloric Acid	101	Fuel, Aviation	Fire Ball	55
1	Hydrochloric Acid	14	Glycol Ethers	Fire Ball	126
1	Hydrochloric Acid	14	Glycol Ethers	Toxic Emission	706
1	Hydrochloric Acid	7	Hexamethylene Diamine Soln.	Toxic Emission	78
1	Hydrochloric Acid	104	Hydrogen Peroxide	Explosions	1590
1	Hydrochloric Acid	104	Hydrogen Peroxide	Toxic Emission	75162
1	Hydrochloric Acid	1.4	Isopropanol	Fire Ball	2715
1	Hydrochloric Acid	4	Isopropanol	Toxic Emission	78
1	Hydrochloric Acid	4	Isopropanol	Vapor Cloud Explosion	2516
1	Hydrochloric Acid	107	Maleic Anhydride	Fire Ball	109
1	Hydrochloric Acid	107	Maleic Anhydride	Toxic Emission	1963
1	Hydrochloric Acid	4	Methyl Alcohol	Fire Ball	1663
1	Hydrochloric Acid	4	Methyl Alcohol	Toxic Emission	1885
1	Hydrochloric Acid	4	Methyl Alcohol	Vapor Cloud Explosion	725
1	Hydrochloric Acid	19	Methyl Ethyl Ketone	Fire Ball	1355
1	Hydrochloric Acid	19	Methyl Ethyl Ketone	Toxic Emission	392
1	Hydrochloric Acid	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	2435
1	Hydrochloric Acid	13	Methyl Methacrylate	Fire Ball	1621
1	Hydrochloric Acid	13	Methy! Methacry!ate	Toxic Emission	3769
1	Hydrochloric Acid	13	Methyl Methacrylate	Vapor Cloud Explosion	1790
1	Hydrochloric Acid	24	Motor Fuel Antiknock Compound	Fire Ball	2667
1	Hydrochloric Acid	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	2544
1	Hydrochloric Acid	2	Nitric Acid, Fuming	Toxic Emission	50893
1	Hydrochloric Acid	2	Oleum	Toxic Emission	28981
1	Hydrochloric Acid	101	Petroleum Naphtha	Fire Ball	52
1	Hydrochloric Acid	105	Phosphorus	Fire Ball	127
1	Hydrochloric Acid	105	Phosphorus	Toxic Emission	5026
1	Hydrochloric Acid	10	Potassium Hydroxide, Liq	Toxic Emission	1963
1	Hydrochloric Acid	10	Sodium Hydroxide, Liq.	Toxic Emission	1963

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
1	Hydrochloric Acid	21	Sodium Metal	Fire Ball	9835
1	Hydrochloric Acid	21	Sodium Meta!	Toxic Emission	19635
1	Hydrochloric Acid	21	Sodium Metal	Vapor Cloud Explosion	13893
1 '	Hydrochloric Acid	16	Styrene Monomer	Fire Ball	26658
1	Hydrochloric Acid	16	Styrene Monomer	Toxic Emission	19635
1	Hydrochloric Acid	16	Styrene Monomer	Vapor Cloud Explosion	25165
1	Hydrochloric Acid	2	Sulfuric Acid	Toxic Emission	490874
1	Hydrochloric Acid	2	Sulfuric Acid, Spent	Toxic Emission	490874
1	Hydrochloric Acid	18	Toluene Diisocyanate	Fire Ball	190
1	Hydrochloric Acid	18	Toluene Diisocyanate	Toxic Emission	28274
1	Hydrochloric Acid	13	Vinyl Acetate	Fire Ball	9096
1	Hydrochloric Acid	13	Viny! Acetate	Toxic Emission	63617
1	Hydrochloric Acid	13	Vinyl Acetate	Vapor Cloud Explosion	11347
1	Hydrochloric Acid	17	Vinyl Chloride	Toxic Emission	19635
1	Hydrocyanic Acid	19	Acetone	Vapor Cloud Explosion	5024
1	Hydrocyanic Acid	104	Chlorine	Toxic Emission	565487
1	Hydrocyanic Acid	104	Hydrogen Peroxide	Explosions	15900
1	Hydrocyanic Acid	24	Motor Fuel Antiknock Compound	Fire Ball	22052
1	Hydrocyanic Acid	2.4	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	65936
1	Hydrofluorosilicic Acid	3	Acetic Acid, Glacial	Fire Ball	821
1	Hydrofluorosilicic Acid	3	Acetic Acid, Glacial	Toxic Emission	6283
1	Hydrofluorosilicic Acid	107	Acetic Anhydride	Fire Bal!	8050
1	Hydrofluorosilicic Acid	107	Acetic Anhydride	Toxic Enission	50269
1	Hydrofluorosilicic Acid	107	Acetic Anhydride	Vapor Cloud Explosion	1329
1	Hydrofluorosilicic Acid	19	Acetone	Fire Ball	15869
1	Hydrofluorosilicic Acid	19	Acetone	Toxic Emission	27489
1	Hydrofluorosilicic Acid	19	Acetone	Vapor Cloud Explosion	42654
1	Hydrofluorosilicic Acid	3	Acrylic Acid	Fire Ball	443
1	Hydrofluorosilicic Acid	3	Acrylic Acid	Toxic Emission	785
1	Hydrofluorosilicic Acid	26	Acrylonitrile	Fire Ball	8902
1	Hydrofluorosilicic Acid	26	Acrylonitrile	Toxic Emission	47124
1	Hydrofluorosilicic Acid	26	Acrylonitrile	Vapor Cloud Explosion	25862
1	Hydrofluorosilicic Acid	4	Alcoholic Beverage	Fire Ball	26672
1	Hydrofluorosilicic Acid	4	Alcoholic Beverage	Toxic Emission	785
1	Hydrofluorosilicic Acid	4	Alcoholic Beverage	Vapor Cloud Explosion	6362
1	Hydrofluorosilicic Acid	10	Alkaline Liquid	Toxic Emission	50268
1	Hydrofluorosilicic Acid	7	Aniline Dil, Liquid	Toxic Emission	2356
1	Hydrofluorosilicic Acid	13	Buty! Acrylate	Fire Ball	7046
1	Hydrofluorosilicic Acid	13	Butyl Acrylate	Toxic Emission	50265
1.	Hydrofluorosilicic Acid	13	Butyl Acrylate	Vapor Cloud Explosion	2463
1	Hydrofluorosilicic Acid .	4	Butyl Alcohol	Fire Ball	1749
1	Hydrofluorosilicic Acid	4	Buty! Alcohol	Toxic Emission	785
1	Hydrofluorosilicic Acid	5	Butyraldehyde	Fire Ball	11096
1	Hydrofluorosilicic Acid	5	Butyraldehyde	Toxic Emission	50268
1	Hydrofluorosilicic Acid	5	Butyraldehyde	Vapor Cloud Explosion	2449

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
1	Hydrofluorosilicic Acid	17	Carbon Tetrachloride	Toxic Emission	66759
1	Hydrofluorosilicic Acid	17	Chloroprene, Inhibited	Fire Bail	11471
1	Hydrofluorosilicic Acid	17	Chloroprene, Inhibited	Toxic Emission	176715
1	Hydrofluorosilicic Acid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
1	Hydrofluorosilicic Acid	101	Coal Tar Distillate	Fire Ball	597
1	Hydrofluorosilicic Acid	101	Crude Oil Petroleum	Fire Ball	238
1	Hydrofluorosilicic Acid	4	Denatured Alcohol	Fire Ball	56667
1	Hydrofluorosilicic Acid	4	Denatured Alcohol	Toxic Emission	785
1	Hydrofluorosilicic Acid	4	Denatured Alcohol	Vapor Cloud Explosion	59396
1	Hydrofluorosilicic Acid	13	Ethyl Acetate	Fire Ball	15323
1	Hydrofluorosilicic Acid	13	Ethyl Acetate	Toxic Emission	2356
1	Hydrofluorosilicic Acid	13	Ethyl Acetate	Vapor Cloud Explosion	19879
1	Hydrofluorosilicic Acid	13	Ethyl Acrylate	Fire Ball	16922
1	Hydrofluorosilicic Acid	13	Ethyl Acrylate	Toxic Emission	49480
1	Hydrofluorosilicic Acid	13	Ethyl Acrylate	Vapor Cloud Explosion	19856
1	Hydrofluorosilicic Acid	4	Ethyl Alcohol	Fire Ball	56667
1	Hydrofluorosilicic Acid	4	Ethyl Alcohol	Toxic Emission	785
1	Hydrofluorosilicic Acid	4	Ethyl Alcohol	Vapor Cloud Explosion	59396
1	Hydrofluorosilicic Acid	5	Formaldehyde Solution	Toxic Emission	2350
1	Hydrofluorosilicic Acid	101	Fuel Oil	Fire Ball	56
1	Hydrofluorosilicic Acid	101	Fuel, Aviation	Fire Ball	555
1	Hydrofluorosilicic Acid	14	Glycol Ethers	Fire Ball	1266
1	Hydrofluorosilicic Acid	14	Glycol Ethers	Toxic Emission	78
1	Hydrofluorosilicic Acid	7	Hexamethylene Diamine Soln.	Toxic Emission	
1	Hydrofluorosilicic Acid	104	-		785
1		l' l	Hydrogen Peroxide	Toxic Emission	942
_	Hydrofluorosilicic Acid	4	Isopropanol	Fire Ball	2715
1	Hydrofluorosilicic Acid	4	Isopropanol	Toxic Emission	78
1	Hydrofluorosilicic Acid	4.	Isopropanol	Vapor Cloud Explosion	2516
1	Hydrofluorosilicic Acid	107	Maleic Anhydride	Fire Ball	109
1	Hydrofluorosilicic Acid	107	Maleic Anhydride	Toxic Emission	5026
1	Hydrofluorosilicic Acid	4	Methyl Alcohol	Fire Ball	1663
1	Hydrofluorosilicic Acid	4	Methyl Alcohol	Vapor Cloud Explosion	725
1	Hydrofluorosilicic Acid	17	Methyl Chloride	Toxic Emission	301593
1	Hydrofluorosilicic Acid	19	Methyl Ethyl Ketone	Fire Ball	13556
1	Hydrofluorosilicic Acid	19	Methyl Ethyl Ketone	Toxic Emission	3927
1	Hydrofluorosilicic Acid	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	24354
1	Hydrofluorosilicic Acid	13	Methyl Methacrylate	Fire Ball	16214
1	Hydrofluorosilicic Acid	13	Methyl Methacrylate	Toxic Emission	49486
1	Hydrofluorosilicic Acid	13	Methyl Methacrylate	Vapor Cloud Explosion	17908
1	Hydrofluorosilicic Acid	24	Motor Fuel Antiknock Compound	Fire Ball	26679
1	Hydrofluorosilicic Acid	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	25447
1	Hydrofluorosilicic Acid	2	Nitric Acid, Fuming	Toxic Emission	84430
1	Hydrofluorosilicic Acid	2	Oleum	Toxic Emission	706858
1	Hydrofluorosilicic Acid	101	Petroleum Naphtha	Fire Ball	527
1	Hydrofluorosilicic Acid	31	Phenol	Toxic Emission	314

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Conséquence Type	Area Covered sq.m.
1	Hydrofluorosilicic Acid	105	Phosphorus	Toxic Emission	785
1	Hydrofluorosilicic Acid	16	Potassium Hydroxide, Liq	Toxic Emission	50265
1	Hydrofluorosilicic Acid	10	Sodium Hydroxide, Liq.	Toxic Emission	50265
1	Hydrofluorosilicic Acid	21	Sodium Metal	Fire Ball	9835
1	Hydrofluorosilicic Acid	21	Sodium Metal	Toxic Emission	50265
1	Hydrofluorosilicic Acid	21	Sodium Metal	Vapor Cloud Explosion	13893
1	Hydrofluorosilicic Acid	16	Styrene Monomer	Fire Ball	26658
1	Hydrofluorosilicic_Acid	16	Styrene Monomer	Toxic Emission	50265
1	Hydrofluorosilicic Acid	16	Styrene Monomer	Vapor Cloud Explosion	25165
1	Hydrofluorosilicic Acid	2	Sulfuric Acid	Toxic Emission	907920
1	Hydrofluorosilicic Acid	18	Toluene Diisocyanate	Fire Ball	190
1	Hydrofluorosilicic Acid	18	Toluene Diisocyanate	Toxic Emission	50265
1	Hydrofluorositicic Acid	13	Vinyl Acetate	Fire Ball	9ø96
1	Hydrofluorosilicic Acid	13	Vinyl Acetate	Toxic Emission	50265
1	Hydrofluorosilicic Acid	13	Vinyl Acetate	Vapor Cloud Explosion	11347
104	Hydrogen Peroxide	107	Acetic Anhydride	Explosions	25239
104	Hydrogen Peroxide	107	Acetic Anhydride	Fire Ball	8050
104	Hydrogen Peroxide	107	Acetic Anhydride	Toxic Emission	129591
104	Hydrogen Peroxide	107	Acetic Anhydride	Vapor Cloud Explosion	1329
104	Hydrogen Peroxide	107	Maleic Anhydride	Explosions	25239
104	Hydrogen Peroxide	107	Maleic Anhydride	Fire Ball	1099
104	Hydrogen Peroxide	107	Maleic Anhydride	Toxic Emission	12959
104	Hydrogen Peroxide	105	Phosphorus	Explosions	25239
104	Hydrogen Peroxide	105	Phosphorus	Toxic Emission	942
1	Hydroiodic Acid	3	Acetic Acid, Glacial	Fire Ball	82
1	Hydroiodic Acid	3	Acetic Acid, Glacial	Toxic Emission	628
1	Hydroiodic Acid	107	Acetic Anhydride	Fire Ball	8056
1	Hydroiodic Acid	107	Acetic Anhydride	Toxic Emission	12566
1	Hydroiodic Acid	107	Acetic Anhydride	Vapor Cloud Explosion	1326
1	Hydroiodic Acid	19	Acetone	Fire Ball	1586
1	Hydroiodic Acid	19	Acetone	Toxic Emission	27489
1	Hydroiodic Acid	19	Acetone .	Vapor Cloud Explosion	42654
1	Hydroiodic Acid	3	Acrylic Acid	Fire Ball	t .
1	Hydroiodic Acid	3	Acrylic Acid	Toxic Emission	441
1		- 1	· ·	1	
_	Hydroiodic Acid	26	Acrylonitrile	Fire Ball	890
1	Hydroiodic Acid	26	Acrylonitrile	Toxic Emission	6047
1	Hydroiodic Acid	26	Acrylonitrile	Vapor Cloud Explosion	2586
1	Hydroiodic Acid	7	Aniline Oil, Liquid	Toxic Emission	2350
1	Hydroiodic Acid	13	Butyl Acrylate	Fire Ball	7046
1	Hydroiodic Acid	13	Butyl Acrylate	Toxic Emission	
1	Hydroiodic Acid	13	Butyl Acrylate	Vapor Cloud Explosion	246
1	Hydroiodic Acid	5	Butyraldehyde	Fire Ball	2694
1	Hydroiodic Acid	5	Butyraldehyde	Vapor Cloud Explosion	7575
1	Hydroiodic Acid	17	Carbon Tetrachloride	Toxic Emission	14844
1	Hydroiodic Acid	17	Chloroprene, Inhibited	Fire Ball	1147

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

STM #	Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered
1	Hydroiodic Acid	17	Chloroprene, Inhibited	Toxic Emission	176715
1	Hydroiodic Acid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
1	Hydroiodic Acid	101	Coal Tar Distillate	Fire Ball	597
1	Hydroiodic Acid	101	Crude Oil Petroleum	Fire Ball	230
1	Hydroiodic Acid	4	Denatured Alcohol	Toxic Emission	?
1	Hydroiodic Acid	13	Ethyl Acetate	Fire Ball	15323
1	Hydroiodic Acid	13	Ethyl Acetate	Toxic Emission	6283
1	Hydroiodic Acid	13	Ethyl Acetate	Vapor Cloud Explosion	19879
1	Hydroiodic Acid	13	Ethyl Acrylate	Fire Ball	16922
1	Hydroiodic Acid	13	Ethyl Acrylate	Toxic Emission	62832
1	Hydroiodic Acid	13	Ethyl Acrylate	Vapor Cloud Explosion	19856
1	Hydroiodic Acid	4	Ethyl Alcohol	Toxic Emission	1
1	Hydroiodic Acid	5	Formaldehyde Solution	Toxic Emission	235
1	Hydroiodic Acid	101	Fuel Oil ·	Fire Ball	56
1	Hydroiodic Acid	101	Fuel, Aviation	Fire Ball	55
1	Hydroiodic Acid	14	Glycol Ethers	Fire Ball	- 126
1	Hydroiodic Acid	1,4	Glycol Ethers	Toxic Emission	1256
1	Hydroiodic Acid	104	Hydrogen Peroxide	Toxic Emission	942
1	Hydroiodic Acid	4	Isopropanol	Toxic Emission	!
1	Hydroiodic Acid	107	Maleic Anhydride	Fire Ball	109
1	Hydroiodic Acid	107	Maleic Anhydride	Toxic Emission	1
1	Hydroiodic Acid	4	Methyl Alcohol	Fire Ball	142
1	Hydroiodic Acid	4	Methyl Alcohol	Toxic Emission	11231
1	Hydroiodic Acid	19	Methyl Ethyl Ketone	Fire Ball	1355
1	Hydroiodic Acid	19	Methyl Ethyl Ketone	Toxic Emission	392
1	Hydroiodic Acid	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	2435
1	Hydroiodic Acid	13	Methyl Methacrylate	Fire Ball	1621
1	Hydroiodic Acid	13	Methyl Methacrylate	Toxic Emission	3769
1	Hydroiodic Acid	13	Methyl Methacrylate	Vapor Cloud Explosion	1790
1	Hydroiodic Acid	24	Motor Fuel Antiknock Compound	Fire Ball	2667
1	Hydroiodic Acid	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	2544
1	Hydroiodic Acid	.2	Nitric Acid, Fuming	Toxic Emission	55213
1	Hydroiodic Acid	2	Oleum	Toxic Emission	25132
1	Hydroiodic Acid	101	Petroleum Naphtha	Fire Ball	52
1	Hydrojodic Acid	105	Phosphorus	Fire Ball	127
1	Hydroiodic Acid	105	Phosphorus	Toxic Emission	25446
1	Hydroiodic Acid	21	Sodium Metal	Fire Ball	983
1	Hydroiodic Acid	21	Sodium Metal	Toxic Emission	78
1	Hydroiodic Acid	21	Sodium Metal	Vapor Cloud Explosion	1389
1	, · · · ·	16		Fire Ball	2665
1	Hydroiodic Acid Hydroiodic Acid	16	Styrene Monomer Styrene Monomer	Toxic Emission	314
1	Hydroiodic Acid	16	Styrene Monomer	Vapor Cloud Explosion	2516
1	· ·	2	Sulfuric Acid	Toxic Emission	25446
1	Hydroiodic Acid	2 2		Toxic Emission	9503
1	Hydroiodic Acid	2	Sulfuric Acid, Spent	LUXIC CHISSION	1 9989

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
1	Hydroiodic Acid	18	Toluene Diisocyanate	Toxic Emission	28274
1	Hydroiodic Acid	13	Viny! Acetate	Fire Ball	9096
1	Hydroiodic Acid	13	Vinyl Acetate	Toxic Emission	63617
1	Hydroiodic Acid	13	Vinyl Acetate	Vapor Cloud Explosion	11347
4	Isopropanol	5	Butyraldehyde	Fire Ball	22542
4	Isopropanol	5	Butyraldehyde	Vapor Cloud Explosion	66908
4	Isopropanol	194	Chlorine	Explosions	35792
4	Isopropanol	104	Hydrogen Peroxide	Explosions	25239
4	Isopropanol	104	Hydrogen Peroxide	Fire Ball	178Ø1
4	Isopropanol	194	Hydrogen Peroxide	Toxic Emission	129591
4	Isopropanol	104	Hydrogen Peroxide	Vapor Cloud Explosion	22698
4	Isopropanol	21	Sodium Metal	Fire Ball	20305
4	Isopropanol	21	Sodium Metal	Toxic Emission	19635
4	Isopropanol	21	Sodium Metal	Vapor Cloud Explosion	52279
4	Isopropanol	18	Toluene Diisocyanate	Fire Ball	17833
4	Isopropanol	18	Toluene Diisocyanate	Toxic Emission	28274
4	Isopropanol	18	Toluene Diisocyanate	Vapor Cloud Explosion	22698
28	LPG-Butene	194	Hydrogen Peroxide	Explosions	15900
28	LPG-Butylene	104	Hydrogen Peroxide	Explosions	15900
28	LPG-Isobutylene	104	Hydrogen Peroxide	Explosions	15900
28	LPG-Propylene	104	Hydrogen Peroxide	Explosions	15900
28	LPG-Propylene	104	Hydrogen Peroxide	Toxic Emission	9425
4	Methyl Alcohol	5	Butyraldehyde	Fire Ball	8601
4	Methyl Alcohol	5	Butyraldehyde	Vapor Cloud Explosion	44722
4	Methyl Alcohol	104	Chlorine	Explosions	42580
٠ 4	Methyl Alcohol	104	Ferric Chloride Solution	Toxic Emission	6283
4	Methyl Alcohol	104	Hydrogen Peroxide	Explosions	25239
4	Methyl Alcohol	194	Hydrogen Peroxide	Fire Ball	6861
4 '	Methyl Alcohol	104	Hydrogen Peroxide	Toxic Emission	129591
4	Methyl Alcohol	164	Hydrogen Peroxide	Vapor Cloud Explosion	6163
4	Methyl Alcohol	21	Sodium Metal	Fire Ball	8861
4	Methyl Alcohol	21	Sodium Metal	Toxic Emission	27489
4	Methyl Alcohol	21	Sodium Metal	Vapor Cloud Explosion	30972
4	Methyl Alcohol	18	Toluene Diisocyanate	Fire Ball	6886
4	Methyl Alcohol	18	Toluene Diisocyanate	Toxic Emission	27489
4	Methyl Alcohol	18	Toluene Diisocyanate	Vapor Cloud Explosion	6163
17	Methyl Chloride	104	Hydrogen Peroxide	Explosions	15900
19	Methyl Ethyl Ketone	104	Ferric Chloride Solution	Fire Ball	13550
19	Methyl Ethyl Ketone	104	Ferric Chloride Solution	Toxic Emission	3927
19	Methyl Ethyl Ketone	104	Ferric Chloride Solution	Vapor Cloud Explosion	24354
19	Methyl Ethyl Ketone	104	Hydrogen Peroxide	Explosions	25239
19	Methyl Ethyl Ketone	194	Hydrogen Peroxide	Fire Ball	13301
19	Wethyl Ethyl Ketone	104	Hydrogen Peroxide	Toxic Emission	129591
19	Methyl Ethyl Ketone	104	Hydrogen Peroxide	Vapor Cloud Explosion	23503
19	Methyl Ethyl Ketone	21	Sodium Metal	Fire Ball	15859

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM .	Chemical Name	Consequence Type	Area Covered sq.m.
19	Methyl Ethyl Ketone	21	Sodium Metal	Toxic Emission	60476
19	Methyl Ethyl Ketone	21	Sodium Metal	Vapor Cloud Explosion	53493
. 13	Methyl Methacrylate	104	Ferric Chloride Solution	Fire Ball	5103
13	Methyl Methacrylate	104	Ferric Chloride Solution	Toxic Emission	2356
13	Methyl Methacrylate	104	Ferric Chloride Solution	Vapor Cloud Explosion	1521
13	Methyl Methacrylate	104	Hydrogen Peroxide	Explosions	25239
13	Methyl Methacrylate	104	Hydrogen Peroxide	Fire Ball	16214
13	Methyl Methacrylate	104	Hydrogen Peroxide	Toxic Emission	129591
13	Methyl Methacrylate	104	Hydrogen Peroxide	Vapor Cloud Explosion	17908
13	Methyl Methacrylate	21	Sodium Metal	Fire Ball	18527
13	Methyl Methacrylate	21	Sodium Metal	Toxic Emission	37699
13	Methyl Methacrylate	21	Sodium Metal	Vapor Cloud Explosion	46377
24	Motor Fuel Antiknock Compound	104	Hydrogen Peroxide	Explosions	25239
24	Motor Fuel Antiknock Compound	104	Hydrogen Peroxide	Toxic Emission	129591
2	Nitric Acid, Fuming	5	Acetaldehyde	Explosions	45623
2	Nitric Acid, Fuming	5	Acetaldehyde	Toxic Emission	520719
2	Nitric Acid, Fuming	3	Acetic Acid, Glacial	Explosions	38936
2	Nitric Acid, Fuming	3	Acetic Acid, Glacial	Fire Ball	821
2	Nitric Acid, Fuming	3	Acetic Acid, Glacial	Toxic Emission	137445
2	Nitric Acid, Fuming	107	Acetic Anhydride	Explosions	38936
. 2	Nitric Acid, Fuming	107	Acetic Anhydride	Fire Ball	8050
2.	Nitric Acid, Fuming.	107	Acetic Anhydride	Toxic Emission	552135
2	Nitric Acid, Fuming	19	Acetone	Explosions	45623
2	Nitric Acid, Fuming	19	Acetone	Fire Ball	1867
2	Nitric Acid, Fuming	19	Acetone	Toxic Emission	552135
2	Nitric Acid, Fuming	19	Acetone	Vapor Cloud Explosion	3870
2	Nitric Acid, Fuming	3	Acrylic Acid	Fire Ball	441
2	Nitric Acid, Fuming	3	Acrylic Acid	Toxic Emission	137445
2	Nitric Acid, Fuming	26	Acrylonitrile	Explosions	26389
2	Nitric Acid, Fuming	26	Acrylonitrile	Fire Ball	8902
2	Nitric Acid, Fuming	26	Acrylonitrile	Toxic Emission	552135
2	Nitric Acid, Fuming	26	Acrylonitrile	Vapor Cloud Explosion	25862
2	Nitric Acid, Fuming	4	Alcoholic Beverage	Explosions	18573
2	Nitric Acid, Fuming	4	Alcoholic Beverage	Fire Ball	21389
2	Nitric Acid, Fuming	4	Alcoholic Beverage	Toxic Emission	552135
2	Nitric Acid, Fuming	4	Alcoholic Beverage	Vapor Cloud Explosion	25730
2	Nitric Acid, Fuming	10	Alkaline Liquid	Toxic Emission	552135
2	Nitric Acid, Fuming	106	Ammonium Nitrate Soln.	Toxic Emission	137445
2	Nitric Acid, Fuming	7	Aniline Oil, Liquid	Fire Ball	3452
2 "	Nitric Acid, Fuming	7	Aniline Oil, Liquid	Toxic Emission	552135
. 2	Nitric Acid, Fuming	16	Benzene	Explosions	44479
` 2	Nitric Acid, Fuming	16	Benzene ,	Fire Ball	26041
2	Nitric Acid, Fuming	16	Benzene	Toxic Emission	552135
2	Nitric Acid, Fuming	16	Benzene	Vapor Cloud Explosion	37909
2	Nitric Acid, Fuming	28	Butadiene	Toxic Emission	137445

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

		T		<u> </u>	
ASTM		ASTM			Area Covered
# Y2 (M	Chemical Name	#	Chemical Name	Consequence Type	sq.m.
<u> </u>					
2	Nitric Acid, Fuming	13	Butyl Acrylate	Fire Ball	7Ø46
2	Nitric Acid, Fuming	13	Butyl Acrylate	Toxic Emission	552135
2	Nitric Acid, Fuming	13	Butyl Acrylate	Vapor Cloud Explosion	2463
2	Nitric Acid, Fuming	4	Butyl Alcohol	Explosions	41394
2	Nitric Acid, Fuming	4	Butyl Alcohol	Fire Ball	32060
2	Nitric Acid, Fuming	4	Butyl Alcohol	Toxic Emission	552135
2	Nitric Acid, Fuming	4	Butyl Alcohol	Vapor Cloud Explosion	47916
2	Nitric Acid, Fuming	5	Butyraldehyde	Explosions	45623
2	Nitric Acid, Fuming	5	Butyraldehyde	Fire Ball	11096
2	Nitric Acid, Fuming	5	Butyraldehyde	Toxic Emission	552135
2	Nitric Acid, Fuming	5	Butyraldehyde	Vapor Cloud Explosion	24497
2	Nitric Acid, Fuming	20	Carbon Disulfide	Toxic Emission	25Ø542
2	Nitric Acid, Fuming	20	Carbon Disulfide	Vapor Cloud Explosion	2043
2	Nitric Acid, Fuming	17	Carbon Tetrachloride	Toxic Emission	?
2	Nitric Acid, Fuming	17	Chloroprene, Inhibited	Fire Ball	32945
2	Nitric Acid, Fuming	17	Chloroprene, Inhibited	Toxic Emission	552135
2	Nitric Acid, Fuming	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
2	Nitric Acid, Fuming	101	Coal Tar Distillate	Fire Ball	597
2	Nitric Acid, Fuming	101	Coal Tar Distillate	Toxic Emission	137445
2	Nitric Acid, Fuming	101	Combustible Liquid	Toxic Emission	137445
2	Nitric Acid, Fuming	101	Compound, Cleaning	Toxic Emission	137445
2	Nitric Acid, Fuming	101	Crude Oil Petroleum	Fire Ball	230
2	Nitric Acid, Fuming	101	Crude Oil Petroleum	Toxic Emission	137445
2	Nitric Acid, Fuming	4	Denatured Alcohol	Explosions	29482
2	Nitric Acid, Fuming	4	Denatured Alcohol	Fire Ball	72304
2	Nitric Acid, Fuming	4	Denatured Alcohol	Toxic Emission	552135
2	Nitric Acid, Fuming	4	Denatured Alcohol	Vapor Cloud Explosion	180203
2	Nitric Acid, Fuming	27	Dinitrotoluene, Liquid	Toxic Emission	137445
2	Nitric Acid, Fuming	101	Distillate Fuel Oil	Toxic Emission	137445
2	Nitric Acid, Fuming	13	Ethyl Acetate	Fire Ball	12469
2	Nitric Acid, Fuming	13	Ethyl Acetate	Toxic Emission	137445
2	Nitric Acid, Fuming	13	Ethyl Acetate	Vapor Cloud Explosion	19879
2	Nitric Acid, Fuming	13	Ethyl Acrylate	Fire Ball	16922
2	Nitric Acid, Fuming	13	Ethyl Acrylate	Toxic Emission	552135
2 ·	Nitric Acid, Fuming	13	Ethyl Acrylate	Vapor Cloud Explosion	19856
2	Nitric Acid, Fuming	4	Ethyl Alcohol	Explosions	29482
2.	Nitric Acid, Fuming	4	Ethyl Alcohol	Fire Ball	72304
2	Nitric Acid, Fuming	4	Ethyl Alcohol	Toxic Emission	552135
2	Nitric Acid, Fuming	4	Ethyl Alcohol	Vapor Cloud Explosion	180203
2	Nitric Acid, Fuming	34	Ethylene Oxide	Toxic Emission	124878
2	Nitric Acid, Fuming	101	Flammable Liquid	Toxic Emission	137445
2	Nitric Acid, Fuming	5	Formaldehyde Solution	Fire Ball	157
2	Nitric Acid, Fuming	5	Formaldehyde Solution	Toxic Emission	552135
2	Nitric Acid, Fuming	101	Fuel Oil	Fire Ball	565
2	Nitric Acid, Fuming	101	Fuel Oil	Toxic Emission	137445

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
2	Nitric Acid, Fuming	101	Fuel, Aviation	Fire Ball	552
2	Nitric Acid, Fuming	101	Fuel, Aviation	Toxic Emission	137445
2	Nitric Acid, Fuming	101	Gasoline	Toxic Emission	137445
2	Nitric Acid, Fuming	101	Gasoline, nec	Toxic Emission	13744
2	Nitric Acid, Fuming	14	Glycol Ethers	Explosions	2948
2	Nitric Acid, Fuming	14	Glycol Ethers	Fire Ball	2314
2	Nitric Acid, Fuming	14	Glycol Ethers	Toxic Emission	55213
2	Nitric Acid, Fuming	14	Glycol Ethers	Vapor Cloud Explosion	1227
2	Nitric Acid, Fuming	7	Hexamethylene Diamine Soln.	Fire Ball	36
2	Nitric Acid, Fuming	7	Hexamethylene Diamine Soln.	Toxic Emission	55213
2	Nitric Acid, Fuming	104	Hydrogen Peroxide	Explosions	25239
2	Nitric Acid, Fuming	104	Hydrogen Peroxide	Toxic Emission	13744
2	Nitric Acid, Fuming	4	Isopropanol	Explosions	4565
2	Nitric Acid, Fuming	4 .	Isopropanol	Fire Ball	4261
2	Nitric Acid, Fuming	4	Isopropanol	Toxic Emission	55213
2	Nitric Acid, Fuming	4	Isopropanol	Vapor Cloud Explosion	9348
2	Nitric Acid, Fuming	28	LPG	Toxic Emission	13744
2	Nitric Acid, Fuming	28	LPG-Butene	Toxic Emission	13744
2	·	28		· /····• —····	
2	Nitric Acid, Fuming	1	LPG-Isobutylene	Toxic Emission	13744
2	Nitric Acid, Fuming	28	LPG-Propylene	Toxic Emission	13744
	Nitric Acid, Fuming	107	Maleic Anhydride	Explosions	2708
2	Nitric Acid, Fuming	107	Maleic Anhydride	Toxic Emission	13744
2	Nitric Acid, Fuming	4	Methyl Alcohol	Explosions	4899
2	Nitric Acid, Fuming	4	Methyl Alcohol	Fire Ball	4718
2	Nitric Acid, Fuming	4	Methyl Alcohol	Toxic Emission	55213
2	Nitric Acid, Fuming	4	Methyl Alcohol	Vapor Cloud Explosion	11413
2	Nitric Acid, Fuming	17	Methyl Chloride	Toxic Emission	41233
2	Nitric Acid, Fuming	19	Methyl Ethyl Ketone	Fire Ball	1330
2	Nitric Acid, Fuming	19	Methyl Ethyl Ketone	Toxic Emission	55213
2	Nitric Acid, Fuming	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	2350
2	Nitric Acid, Fuming	13	Methyl Methacrylate	Fire Ball	1621
2	Nitric Acid, Fuming	13	Methyl Methacrylate	Toxic Emission	55213
2	Nitric Acid, Fuming	13	Methyl Methacrylate	Vapor Cloud Explosion	1790
2	Nitric Acid, Fuming	24	Motor Fuel Antiknock Compound	Toxic Emission	55213
2	Nitric Acid, Fuming	4	Octyl Alcohol	Explosions	2938
2	Nitric Acid, Fuming	4	Octyl Alcohol	Toxic Emission	13744
2	Nitric Acid, Fuming	101	Oil	Toxic Emission	13744
2	Nitric Acid, Fuming	2	Oleum	Toxic Emission	25132
2	Nitric Acid, Fuming	101	Petroleum Naphtha	Fire Ball ¹	52
2	Nitric Acid, Fuming	101	Petroleum Naphtha	Toxic Emission	13744
2	Nitric Acid, Fuming	101	Petroleum Residual Fuel	Toxic Emission	13744
2	Nitric Acid, Fuming	101	Petroleum, Refined	Toxic, Emission	13744
2	Nitric Acid, Fuming	31	Phenol	Fire Ball	199
2	Nitric Acid, Fuming	31	Pheno I	Toxic Emission	55213
2	Nitric Acid, Fuming	105	Phosphorus	Toxic Emission	55213

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
2	Nitric Acid, Fuming	10	Potassium Hydroxide, Liq	Toxic Emission	552135
2	Nitric Acid, Fuming	101	Pulp Mill Liquid	Toxic Emission	137445
2	Nitric Acid, Fuming	10	Sodium Hydroxide, Liq.	Toxic Emission	552135
2	Nitric Acid, Fuming	21	Sodium Metal	Toxic Emission	552135
2	Nitric Acid, Fuming	16	Styrene Monomer	Fire Ball	26658
2	Nitric Acid, Fuming	16	Styrene Monomer	Toxic Emission	552135
2	Nitric Acid, Fuming	16	Styrene Monomer	Vapor Cloud Explosion	25165
2	Nitric Acid, Fuming	16	Toluene	Fire Ball	31952
2	Nitric Acid, Fuming	16	Toluene	Toxic Emission	552135
2	Nitric Acid, Fuming	16	Toluene	Vapor Cloud Explosion	34967
2 .	Nitric Acid, Fuming	18	Toluene Diisocyanate	Fire Ball	198
. 2	Nitric Acid, Fuming	18	Toluene Diisocyanate	Toxic Emission	552135
2	Nitric Acid, Fuming	13	Vinyl Acetate	Fire Ball	9896
2	Nitric Acid, Fuming	13	Vinyl Acetate	Toxic Emission	13744
2	Nitric Acid, Fuming	13	Vinÿl Acetate	Vapor Cloud Explosion	2232
2	Nitric Acid, Fuming	17	Vinyl Chloride	Toxic Emission	13744
2	Nitric Acid, Fuming	16	Xylene	Fire Ball	3501
2	Nitric Acid, Fuming	16	Xylene	Toxic Emission	55213
2	Nitric Acid, Fuming	16	Xvlene	Vapor Cloud Explosion	4Ø82
4	Octyl Alcohol.	104	Chlorine	Explosions	1716
4	Octyl Alcohol	104	Hydrogen Peroxide	Explosions	2523
4	Octyl Alcohol	104	Hydrogen Peroxide	Toxic Emission	12959
4	Octyl Alcohol	21	Sodium Metal	Fire Ball	983
4	Octyl Alcohol	21	Sodium Metal	Vapor Cloud Explosion	1389
4	Octyl Alcohol	18	Toluene Diisocyanate	Toxic Emission	2827
101	Oil	104	Hydrogen Peroxide	Explosions	1590
101	Oil	104	Hydrogen Peroxide	Toxic Emission	942
2	l Oleum	5	Acetaldehyde	Toxic Emission	60318
2	Oleum	3	Acetic Acid, Glacial	Fire Ball	1143
2	Oleum	3	Acetic Acid, Glacial	Toxic Emission	60318
2	Oleum	3	Acetic Acid, Glacial	Vapor Cloud Explosion	255
2	Oleum	107	Acetic Anhydride	Fire Ball	805
2	Oleum	107	Acetic Anhydride	Toxic Emission	60318
2	Oleum	107	Acetic Anhydride	Vapor Cloud Explosion	132
2	Oleum	19	Acetone	Fire Ball	186
2	Oleum	19	Acetone	Toxic Emission	60318
			l .		1
2	Oleum	19	Acetone Acrylic Acid	Vapor Cloud Explosion Fire Ball	387
2	Oleum	3	1	Toxic Emission	1473
2	Oleum	3	Acrylic Acid		60318
2	Oleum	3	Acrylic Acid	Vapor Cloud Explosion	865
2	Oleum	26	Acrylonitrile	Fire Ball	890
2	Oleum	26	Acrylonitrile	Toxic Emission	60318
2	Oleum	26	Acrylonitrile	Vapor Cloud Explosion	2586
2	Oleum	4	Alcoholic Beverage	Fire Ball	2138
2	Oleum	1 4	Alcoholic Beverage	Toxic Emission	60318

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #		Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
2	Oleum		4	Alcoholic Beverage	Vapor Cloud Explosion	25730
2	Oleum	•	10	Alkaline Liquid	Toxic Emission	603186
2	Oleum		106	Ammonium Nitrate Soln.	Explosions	35982
2	Oleum		106	Ammonium Nitrate Soln.	Toxic Emission	603186
2	Oleum	•	7	Aniline Oit, Liquid	Fire Ball	3452
2	Oleum		7	Aniline Oil, Liquid	Toxic Emission	603186
2	Oleum		16	Benzene	Fire Ball	26041
2	Oleum	•	16	Веплен	Toxic Emission	603186
2	Oleum		16	Benzene	Vapor Cloud Explosion	37909
2	Oleum		28	Butadiene	Toxic Emission	329867
2	Oleum	ř	13	Butyl Acrylate	Fire Ball	7046
2	Oleum		13.	Butyl Acrylate	Toxic Emission	603186
2	Oleum		13	Butyl Acrylate	Vapor Cloud Explosion	2463
2	Oleum		4	Butyl Alcohol	Fire Ball	32066
2	Oleum		4	Butyl Alcohol	Toxic Emission	603186
2	Oleum		4	Butyl Alcohol	Vapor Cloud Explosion	4791
2	Oleum	^	5	Butyraldehyde	Fire Ball	1109
2	Oleum		5	Butyraldehyde -	Toxic Emission	60318
2	Oleum		5	Butyraldehyde	Vapor Cloud Explosion	2449
2	Oleum		17	Carbon Tetrachloride	Toxic Emission	60318
2	Oleum		17	Chloroprene, Inhibited	Fire Ball	3294
2	Oleum	,	17	Chloroprene, Inhibited	Toxic Emission	32986
2	Oleum		17	Chloroprene, Inhibited	Vapor Cloud Explosion	5217
2	Oleum	•	101	Coal Tar Distillate	Fire Ball	59
2	Oleum		101	Coal Tar Distillate	Toxic Emission	25132
2	Oleum		101	Combustible Liquid	Toxic Emission	25132
. 2	Oleum		101	Compound, Cleaning	Toxic Emission	25132
2	Oleum		101	Crude Oil Petroleum	Fire Ball	23
2	Dleum		101	Crude Dil Petroleum	Toxic Emission	25132
2	Oleum	•	4	Denatured Alcohol	Fire Ball	7230
2	Oleum		4	Denatured Alcohol	Toxic Emission	60318
,2	Oleum	•	4	Denatured Alcohol	Vapor Cloud Explosion	18020
2	Oleum	•	27	Dinitrotoluene, Liquid	Explosions	3051
2	Oleum		27	Dinitrotoluene, Liquid	Toxic Emission	60318
2	Oleum		101	Distillate Fuel Oil	Toxic Emission	25132
2	Oleum	•	13	Ethyl Acetate	Fire Ball	1246
2	Oleum	•	13	Ethyl Acetate	Toxic Emission	60318
2	Dleum		13	Ethyl Acetate	Vapor Cloud Explosion	1387
2	Oleum		13	Ethyl Acrylate	Fire Ball	1692
2	Oleum		13	Ethyl Acrylate	Toxic Emission	60318
2	Cleum		13	Ethyl Acrylate	Vapor Cloud Explosion	1985
2	Oleum		4	Ethyl Alcohol	Fire Ball	7230
2	Oleum		4	Ethyl Alcohol	Toxic Emission	6Ø318
2	Dleum		4	Ethyl Afcohol	Vapor Cloud Explosion	18020
2	Oleum		34	Ethylene Oxide	Toxic Emission	6Ø318

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Che	mical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
2	Oleum		101	Flammable Liquid	Toxic Emission	251327
2	Oleum		5	Formaldehyde Solution	Fire Ball	157
2	Oleum		5	Formaldehyde Solution	Toxic Emission	329867
2	Oleum .		101	Fuel Oil	Fire Ball	565
2	01eum	-	101	Fuel Dil	Toxic Emission	251327
2	Oleum		101	Fuel, Aviation	Fire Ball	552
2	Oleum		101	Fuel, Aviation	Toxic Emission	251327
2	Oleum		101	Gasoline	Toxic Emission	251327
2	Oleum		101	Gasoline, nec	Toxic Emission	251327
2	Oleum		14	Glycol Ethers	Fire Ball	23147
2	Cleum		14	Glycol Ethers	Toxic Emission	603186
2	Oleum		14	Glycol Ethers	Vapor Cloud Explosion	12272
2	Oleum		7	Hexamethylene Diamine Soln.	Fire Ball	362
2	Oleum		7	Hexamethylene Diamine Soln.	Toxic Emission	329867
2	Oleum		104	Hydrogen Peroxide	Explosions	25239
2	Oleum		104	Hydrogen Peroxide	Toxic Emission	603186
2	0 enw		4	Isopropanol	Fire Ball	42615
2	Oleum		4	Isopropanol	Toxic Emission	603186
2	Oleum		4	Isopropanol	Vapor Cloud Explosion	93482
. 2	Oleum		28	LPG	Toxic Emission	329867
2	0leum		28	LPG-Butene	Toxic Emission	329867
2	Oleum		28	LPG-Isobutylene	Toxic Emission	329867
2	Oleum		28	LPG-Propylene	Toxic Emission	329867
2	Oleum	•	107	Maleic Anhydride	Toxic Emission	603186
2	Oleum		4	Methyl Alcohol	Fire Ball	47189
2	Oleum		4	Methy! Alcoho!	Toxic Emission	603186
2	Oleum		4	Methyl Alcohol	Vapor Cloud Explosion	114134
2	Oleum		17	Methyl Chloride	Toxic Emission	251327
2	Oleum		19	Methyl Ethyl Ketone	Fire Ball	13301
2	Dleum		19	Methyl Ethyl Ketone	Toxic Emission	603186
2	Oleum		19	Methyl Ethyl Ketone	Vapor Cloud Explosion	23503
2	Oleum		13	Methyl Methacrylate	Fire Ball	16214
2	Oleum		13	Methyl Methacrylate	Toxic Emission	603186
2	.0leum		13	Methyl Methacrylate	Vapor Cloud Explosion	17908
2	Oleum		24	Motor Fuel Antiknock Compound	Toxic Emission	603186
2	Oleum		4	Octyl Alcohol	Toxic Emission	603186
2	Oleum		101	Oil	Toxic Emission	251327
2	Oleum		101	Petroleum Naphtha	Fire Ball	527
· 2	Oleum		101	Petroleum Naphtha	Toxic Emission	251327
2	Oleum .		101	Petroleum Residual Fuel	Toxic Emission	251327
2	Oleum		101	Petroleum, Refined	Toxic Emission	251327
2	Oleum		31	Pheno I	Fire Ball	1991
2	Oleum		31	Pheno I	Toxic Emission	6Ø3186
2	Oleum		105	Phosphorus	Toxic Emission .	603186
2	01eum		10	Potassium Hydroxide, Liq	Toxic Emission	603186

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM		ASTM	,		Area Covered
#	Chemical Name	#	Chemical Name	Consequence Type	sq.m.
2	Oleum	34	Propylene Oxide	Toxic Emission	603186
2	Oleum	101	Pulp Mill Liquid	Toxic Emission	251327
2	Oleum	10	Sodium Hydroxide, Liq.	Toxic Emission	603186
2	Oleum	21	Sodium Metal	Toxic Emission	32986
2	0 leum	16	Styrene Monomer	Fire Ball	2665
2	Oleum	16	Styrene Monomer	Toxic Emission	60318
2	Oleum	16	Styrene Monomer	Vapor Cloud Explosion	2516
2	Oleum	16	Toluene	Fire Ball	3195
2	0 leum	16	Toluene	Toxic Emission	60318
2	Oleum ,	16	Toluene	Vapor Cloud Explosion	3496
2	0 i eum	18	Toluene Diisocyanate	Fire Ball	199
2	Dleum	18	Toluene Diisocyanate	Toxic Emission	60318
2	Oleum	13	Vinyl Acetate	Fire Ball	909
2	Oleum	13	Vinyl Acetate	Toxic Emission	60318
2	Oleum	13	Vinyl Acetate	Vapor Cloud Explosion	1134
2	Oleum .	17	Vinyl Chloride	Toxic Emission	- 25132
2	Oleum	16	Xylene	Fire Ball	3501
2	Oleum	16	Xylene	Toxic Emission	60318
2	Oleum	16	Xy I ene	Vapor Cloud Explosion	4Ø82
101	Petroleum Naphtha	104	Hydrogen Peroxide	Explosions	1590
101	Petroleum Naphtha	104	Hydrogen Peroxide	Fire Ball	52
101	Petroleum Naphtha	104	Hydrogen Peroxide	Toxic Emission	942
101	Petroleum Refined	104	Hydrogen Peroxide	Explosions	1590
101	Petroleum Residual Fuel	104	Hydrogen Peroxide	Explosions	1590
101	Petroleum Residual Fuel	104	Hydrogen Peroxide	Toxic Emission	942
101	Petroleum, Refined	104	Hydrogen Peroxide	Toxic Emission	942
31	Phenol	104	Hydrogen Peroxide	Explosions	2523
31	Pheno I	104	Hydrogen Peroxide	Fire Ball	199
31	Pheno I	104	Hydrogen Peroxide	Toxic Emission	12959
31	Phenol	105	Phosphorus	Fire Ball	10
31	Pheno!	- 105	Phosphorus	Toxic Emission	314
1	Phosphoric Acid	3	Acetic Acid, Glacial	Fire Ball	82
1	Phosphoric Acid	3	Acetic Acid, Glacial	Toxic Emission	628
1	Phosphoric Acid	107	Acetic Anhydride	Fire Ball	805
1	Phosphoric Acid	107	Acetic Anhydride	Toxic Emission	1256
1	Phosphoric Acid	107	Acetic Anhydride	Vapor Cloud Explosion	132
1	Phosphoric Acid	_ 19	Acetone	Fire Ball	1586
1	Phosphoric Acid	19	Acetone	Toxic Emission	2748
1	Phosphoric Acid	19	Acetone	Vapor Cloud Explosion	4265
1	Phosphoric Acid	3	Acrylic Acid	Fire Ball	44
1	Phosphoric Acid	3	Acrylic Acid	Toxic Emission	
1	Phosphoric Acid	26	Acrylonitrile	Fire Ball	890
1	Phosphoric Acid	26	Acrylonitrile	Toxic Emission	6047
1	Phosphoric Acid	26	Acrylonitrile	Vapor Cloud Explosion	2586
1	Phosphoric Acid	7	Aniline Oil, Liquid	Toxic Emission	235

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
1	Phosphoric Acid	5	Butyraldehyde	Fire Ball	2694ø
1	Phosphoric Acid	5	Butyraldehyde	Vapor Cloud Explosion	75758
1	Phosphoric Acid	17	Chloroprene, Inhibited	Fire Ball	11471
1	Phosphoric Acid	17	Chloroprene, Inhibited	Toxic Emission	176715
1	Phosphoric Acid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
1	Phosphoric Acid	101	Coal Tar Distillate	Fire Ball	597
1	Phosphoric Acid	. 101	Crude Oil Petroleum	Fire Ball	230
1	Phosphoric Acid	5	Formaldehyde Solution	Toxic Emission	2356
1	Phosphoric Acid	1,01	Fuel Dil	Fire Ball	565
1	Phosphoric Acid	101	Fuel, Aviation	Fire Ball,	552
1	Phosphoric Acid	104	Hydrogen Peroxide	Toxic Emission	9425
1	Phosphoric Acid	107	Maleic Anhydride	Fire Ball	1092
1	Phosphoric Acid	107	Maleic Anhydride	Toxic Emission	1
1	Phosphoric Acid	19	Methyl Ethyl Ketone	Fire Ball	13556
1	Phosphoric Acid	19	Methyl Ethyl Ketone	Toxic Emission	392
1	Phosphoric Acid	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	24354
1	Phosphoric Acid	24	Motor Fuel Antiknock Compound	Fire Ball	26676
1	Phosphoric Acid	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	2544
1	Phosphoric Acid	101	Petroleum Naphtha	Fire Ball	52
1	Phosphoric Acid	21	Sodium Metal	Fire Ball	983
1	Phosphoric Acid	21	Sodium Meta!	Toxic Emission	78
1	Phosphoric Acid	21	Sodium Metal	Vapor Cloud Explosion	1389
1 .	Phosphoric Acid	16	Styrene Monomer	Fire Ball	2665
1	Phosphoric Acid	16	Styrene Monomer	Toxic Emission	314
1	Phosphoric Acid	16	Styrene Monomer	Vapor Cloud Explosion	2516
1	Phosphoric Acid	18	Toluene Diisocyanate	Fire Ball	19
1	Phosphoric Acid	18	Toluene Diisocyanate	Toxic Emission	2827
105	Phosphorus	106	Ammonium Nitrate Soln.	Explosions	3598
1Ø5	Phosphorus	107	Maleic Anhydride	Toxic Emission	****
10	Potassium Hydroxide, Liq	107	Acetic Anhydride	Fire Ball	805
10	Potassium Hydroxide, Liq	107	Acetic Anhydride	Toxic Emission	1256
10	Potassium Hydroxide, Liq	107	Acetic Anhydride	Vapor Cloud Explosion	132
10	Potassium Hydroxide, Liq	19	Acetone	Fire Ball	1586
10	Potassium Hydroxide, Liq	19	Acetone	Toxic Emission	2748
10	Potassium Hydroxide, Liq	19	Acetone	Vapor Cloud Explosion	4265
10	Potassium Hydroxide, Liq	26	Acrylonitrile	Fire Ball	890
10	Potassium Hydroxide, Liq	26	Acrylonitrile	Toxic Emission	6047
10	Potassium Hydroxide, Liq	17	Carbon Tetrachloride	Toxic Emission	31808
10	Potassium Hydroxide, Liq	17	Chloroprene, Inhibited	Fire Ball	1147
10	Potassium Hydroxide, Liq	17	Chloroprene, Inhibited	Toxic Emission	17671
10	Potassium Hydroxide, Liq	17	Chloroprene, Inhibited	Vapor Cloud Explosion	1399
10	Potassium Hydroxide, Liq	27	Dinitrotoluene, Liquid	Explosions	3051
10	Potassium Hydroxide, Liq	13	Ethyl Acetate	Fire Ball	1246
10	Potassium Hydroxide, Liq	13	Ethyl Acetate	Toxic Emission	1885
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Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTW	Chemical Name	Consequence Type	Area Covered sq.m.
10	Potassium Hydroxide, Liq	107	Maleic Anhydride	Fire Ball	1092
10	Potassium Hydroxide, Liq	107	Maleic Anhydride	Toxic Emission	?
10	Potassium Hydroxide, Liq	19	Methyl Ethyl Ketone	Fire Ball	13550
10	Potassium Hydroxide, Liq	19	Methyl Ethyl Ketone	Toxic Emission	3927
10	Potassium Hydroxide, Liq	19	Methy! Ethyl Ketone	Vapor Cloud Explosion	24354
10	Potassium Hydroxide, Liq	105	Phosphorus ·	Fire Ball	1162
10	Potassium Hydroxide, Liq	105	Phosphorus	Toxic Emission	50265
10	Potassium Hydroxide, Liq	21	Sodium Metal	Fire Ball	9835
10	Potassium Hydroxide, Liq	21	Sodium Metal	Vapor Cloud Explosion	13893
10	Potassium Hydroxide, Liq	16	Styrene Monomer	Fire Ball	26658
10	Potassium Hydroxide, Liq	18	Styrene Monomer	Toxic Emission	3142
10	Potassium Hydroxide, Liq	18	Styrene Monomer	Vapor Cloud Explosion	25165
10	Potassium Hydroxide, Liq	18	Toluene Diisocyanate	Fire Ball	190
10	Potassium Hydroxide, Liq	18	Toluene Diisocyanate	Toxic Emission	28274
10	Potassium Hydroxide, Liq	13	Vinyl Acetate	Fire Ball	9096
10	Potassium Hydroxide, Liq	13	Vinyl Acetate	Toxic Emission	63617
10	Potassium Hydroxide, Liq	13	Vinyl Acetate	Vapor Cloud Explosion	11347
19	Potassium Hydroxide, Liq	17	Vinyl Chloride	Fire Ball	839ø
34	Propylene Oxide	104	Hydrogen Peroxide	Explosions	25239
101	Pulp Mill Liquid	104	Hydrogen Peroxide	Explosions	15900
101	Pulp Mill Liquid	104	Hydrogen Peroxide	Toxic Emission	9425
10	Sodium Hydroxide, Liq.	107	Acetic Anhydride	Fire Ball	8050
10	Sodium Hydroxide, Liq.	107	Acetic Anhydride	Toxic Emission	12566
10	Sodium Hydroxide, Liq.	107	Acetic Anhydride	Vapor Cloud Explosion	1320
10	Sodium Hydroxide, Liq.	19	Acetone	Fire Ball	15865
10	Sodium Hydroxide, Liq.	19	Acetone	Toxic Emission	27489
10	Sodium Hydroxide, Liq.	19	Acetone	Vapor Cloud Explosion	42654
10	Sodium Hydroxide, Liq.	26	Acrylonitrile	Fire Ball	8902
10	Sodium Hydroxide, Liq.	26	Acrylonitrile	Toxic Emission	60476
10	Sodium Hydroxide, Liq.	17	Carbon Tetrachloride	Toxic Emission	318086
10	Sodium Hydroxide, Liq.	17	Chloroprene, Inhibited	Fire Ball	11471
10	Sodium Hydroxide, Liq.	17	Chloroprene, Inhibited	Toxic Emission	176715
10	Sodium Hydroxide, Liq.	17	Chloropreme, Inhibited	Vapor Cloud Explosion	13996
10	Sodium Hydroxide, Liq.	27	Dinitrotoluene, Liquid	Explosions	30516
10	Sodium Hydroxide, Liq.	13	Ethyl Acetate	Fire Ball	12469
10	Sodium Hydroxide, Liq.	13	Ethyl Acetate	Toxic Emission	18850
10	Sodium Hydroxide, Liq.	13	Ethyl Acetate	Vapor Cloud Explosion	13876
10	Sodium Hydroxide, Liq.	107	Maleic Anhydride	Fire Ball	1092
10	Sodium Hydroxide, Liq.	107	Maleic Anhydride	Toxic Emission	?
10	Sodium Hydroxide, Liq	19	Methyl Ethyl Ketone	Fire Ball	13550
10	Sodium Hydroxide, Liq.	19	Methyl Ethyl Ketone	Toxic Emission	3927
10	-Sodium Hydroxide, Liq.	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	24354
10	Sodium Hydroxide, Liq.	105	Phosphorus	Fire Ball	1162
10	Sodium Hydroxide, Liq.	195	Phosphorus	Toxic Emission	50265
10	Sodium Hydroxide, Liq.	16	Styrene Monomer	Fire Ball	26658

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
10	Sodium Hydroxide, Liq.	16	Styrene Monomer	Toxic Emission	3142
10	Sodium Hydroxide, Liq.	16	Styrene Monomer	Vapor Cloud Explosion	25165
10	Sodium Hydroxide, Liq.	18	Toluene Diisocyanate	Fire Ball	196
10	Sodium Hydroxide, Liq.	18	Toluene Diisocyanate	Toxic Emission	28274
10	Sodium Hydroxide, Liq.	13	Vinyl Acetate	Fire Ball	9098
10	Sodium Hydroxide, Liq.	13	Vinyl Acetate	Toxic Emission	63617
10	Sodium Hydroxide, Liq.	13	Vinyl Acetate	Vapor Cloud Explosion	11347
10	Sodium Hydroxide, Liq.	17	Vinyl Chloride	Fire Ball	8390
21	Sodium Metal	107	Acetic Anhydride	Fire Ball	8050
21	Sodium Metal	197	Acetic Anhydride	Toxic Emission	12566
21	Sodium Metal	107	Acetic Anhydride	Vapor Cloud Explosion	1320
21	Sodium Metal	26	Acrylonitrile	Fire Ball	29538
21	Sodium Metal	26	Acrylonitrile	Toxic Emission	89478
21	Sodium Metal	26	Acrylonitrile	Vapor Cloud Explosion	55836
21	Sodium Metal	106	Ammonium Nitrate Soln.	Explosions	35982
21	Sodium Metal	106	Ammonium Nitrate Soln.	Fire Ball	9835
21	Sodium Metal	106	Ammonium Nitrate Soln.	Vapor Cloud Explosion	13893
21	Sodium Metal	27	Dinitrotoluene, Liquid	Explosions	30516
21	Sodium Metal	184	Ferric Chloride Solution	Fire Ball	9838
21	Sodium Metal	194	Ferric Chloride Solution	Vapor Cloud Explosion	13893
21	Sodium Metal	194	Hydrogen Peroxide	Explosions	25239
21	Sodium Metal	164	Hydrogen Peroxide	Fire Ball	9838
21	Sodium Metal	194	Hydrogen Peroxide	Toxic Emission	129591
21	Sodium Metal	184	Hydrogen Peroxide	Vapor Cloud Explosion	13893
21	Sodium Metal	197	Maleic Anhydride	Fire Ball	1092
21	Sodium Metal	107	Maleic Anhydride	Toxic Emission	1
21	Sodium Metal	101	Pulp Will Liquid	Fire Ball	9835
21	Sodium Metal	191	Pulp Will Liquid	Vapor Cloud Explosion	13893
16	Styrene Monomer	104	Ferric Chloride Solution	Fire Ball	1758
16	Styrene Monomer	184	Hydrogen Peroxide	Explosions	25239
16	Styrene Monomer	194	Hydrogen Peroxide	Fire Ball	26658
16	Styrene Monomer	104	Hydrogen Peroxide	Toxic Emission	129591
16	Styrene Monomer	104	Hydrogen Peroxide	Vapor Cloud Explosion	25165
16	Styrene Monomer	21	Sodium Metal	Fire Ball	7358
16	Styrene Monomer	21	Sodium Metal	Vapor Cloud Explosion	15394
2	Sulfuric Acid	3	Acetic Acid, Glacial	Fire Ball	11433
2	Sulfuric Acid	3	Acetic Acid, Glacial	Toxic Emission	131947
2	Sulfuric Acid	3	Acetic Acid, Glacial	Vapor Cloud Explosion	2552
2	Sulfuric Acid	197	Acetic Anhydride	Fire Ball	8056
2	Sulfuric Acid	107	Acetic Anhydride	Toxic Emission	12566
2	Sulfuric Acid	107	Acetic Anhydride	Vapor Cloud Explosion	1326
2	Sulfuric Acid	19.	Acetone	Fire Ball	1867
2	Sulfuric Acid	19	Acetone	Toxic Emission	18850
2	Sulfuric Acid	19	Acetone	Vapor Cloud Explosion	3879
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Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
2	Sulfuric Acid	3	Acrylic Acid	Toxic Emission	٠,
2	Sulfuric Acid	3	Acrylic Acid	Vapor Cloud Explosion	8659
2	Sulfuric Acid	26	Acrylonitrile	Fire Ball	8902
2	Sulfuric Acid	26	Acrylonitrile	Toxic Emission	91892
2	Sulfuric Acid	26	Acrylonitrile	Vapor Cloud Explosion	25862
2	Sulfuric Acid	4	Alcoholic Beverage	Fire Ball	21389
2	Sulfuric Acid	4	Alcoholic Beverage	Vapor Cloud Explosion	25730
2	Sulfuric Acid	198	Ammonium Nitrate Soln.	Explosions	35982
2	Sulfuric Acid	7	Aniline Oil, Liquid	Fire Ball	3452
2	Sulfuric Acid	7	Aniline Oil, Liquid	Toxic Emission	94248
2	Sulfuric Acid	16	Benzene	Fire Ball	26041
2	Sulfuric Acid	16	Benzene	Toxic Emission	75398
2	Sulfuric Acid	18	Benzene	Vapor Cloud Explosion	37909
2	Sulfuric Acid	13	Butyl Acrylate	Fire Ball	7046
2	Sulfuric Acid	13	Butyl Acrylate	Toxic Emission	. ?
2	Sulfuric Acid	13	Butyl Acrylate	Vapor Cloud Explosion	2463
2	Sulfúric Acid	4	Butyl Alcohol	Fire Ball	20262
2	Sulfuric Acid	4	Butyl Alcohol	Toxic Emission	28274
2	Sulfuric Acid	4 1	Butyl Alcohol	Vapor Cloud Explosion	28055
2	Sulfuric Acid	5	Butyraldehyde	Fire Ball	11096
2	Sulfuric Acid	5	Butyraldehyde	Vapor Cloud Explosion	24497
2	Sulfuric Acid	17	Carbon Tetrachloride	Toxic Emission	544281
2	Sulfuric Acid	17	Chioroprene, Inhibited	Fire Ball	32945
2	Sulfuric Acid	17	Chloroprene, Inhibited	Toxic Emission	48Ø664
2	Sulfuric Acid	17	Chloroprene, Inhibited	Vapor Cloud Explosion	52173
2	Sulfuric Acid	4	Denatured, Alcohol	Fire Ball	72304
2	Sulfuric Acid	4	Denatured Alcohol	Vapor Cloud Explosion	180203
2	Sulfuric Acid	27	Dinitrotoluene, Liquid	Explosions	3Ø516
2	Sulfuric Acid	13	Ethyl Acetate	Fire Ball	15361
2	Sulfuric Acid	13	Ethyl Acetate	Toxic Emission	2356
2	Sulfuric Acid	13	Ethyl Acetate	Vapor Cloud Explosion	19879
2	Sulfuric Acid	13	Ethyl Acrylate	Fire Ball	16922
2	Sulfuric Acid	13	Ethyl Acrylate	Toxic Emission	52832
2	Sulfuric Acid	13	Ethyl Acrylate	Vapor Cloud Explosion	19856
2	Sulfuric Acid	4	Ethyl Alcohol	Fire Ball	72384
2	Sulfuric Acid	4	Ethyl Alcohol	Vapor Cloud Explosion	180203
2	Sulfuric Acid	5	Formaldehyde Solution	Fire Ball ·	157
2	Sulfuric Acid	5	Formaldehyde Solution	Toxic Emission	18850
2	Sulfuric Acid	14	Glycol Ethers	Fire Ball	23147
2	Sulfuric Acid	14	Glycol Ethers	Toxic Emission	12566
2	Sulfuric Acid	14	Glycol Ethers	Vapor Cloud Explosion	12272
. 2	Sulfuric Acid	7	Hexamethylene Diamine Soln.	Fire Ball	362
2	Sulfuric Acid	194	Hydrogen Peroxide	Toxic Emission	129591
2 -	Sulfuric Acid	4	Isopropanol	Fire Ball	17801
2	Sulfuric Acid	4	Isopropanol	Toxic Emission	19635

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTNI #	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covere sq.m.
2	Sulfuric Acid	4	Isopropanol	Vapor Cloud Explosion	2269
2	Sulfuric Acid	107	Maleic Anhydride	Toxic Emission	(
2	Sulfuric Acid	4	Methyl Alcohol	Fire Ball	4718
2	Sulfuric Acid	4	Methyl Alcohol	Toxic Emission	2827
2	Sulfuric Acid	4	Methyl Alcohol	Vapor Cloud Explosion	11413
2	Sulfuric Acid	17	Methyl Chloride	Toxic Emission	1649
2	Sulfuric Acid	19	Methyl Ethyl Ketone	Fire Ball	1330
2	Sulfuric Acid	19	Methy! Ethyl Ketone	Toxic Emission	6847
2	Sulfuric Acid	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	2350
2	Sulfuric Acid	13	Methyl Methacrylate	Fire Ball	1621
2	Sulfuric Acid	13	Methyi Methacrylate	Toxic Emission	3769
2	Sulfuric Acid	13	Methyl Methacrylate	Vapor Cloud Explosion	1798
2	Sulfuric Acid	31	Phenol	Fire Ball	199
2	Sulfuric Acid	31	Phenal	Toxic Emission	3848
2	Sulfuric Acid	105	Phosphorus	Toxic Emission	(
2	Sulfuric Acid	21	Sodium Metal	Toxic Emission	17671
2	Sulfuric Acid	16	Styrene Monomer	Fire Ball	266
2	Sulfuric Acid	16	Styrene Monomer	Toxic Emission	314
2	Sulfuric Acid	18	Styrene Monomer	Vapor Cloud Explosion	2510
2	Sulfuric Acid	16	Toluene	Fire Bail	319
2	Sulfuric Acid	16	Toluene	Toxic Emission	628
2	Sulfuric Acid	16	Toluene	Vapor Cloud Explosion	349
2	Sulfuric Acid	18	Toluene Diisocyanate	Fire Ball	19
2	Sulfuric Acid	18	Toluene Diisocyanate	Toxic Emission	282
2	Sulfuric Acid	13	Vinyl Acetate	Fire Ball	909
2	Sulfuric Acid	13	Vinyl Acetate	Toxic Emission	ľ
2	Sulfuric Acid	13	Vinyl Acetate	Vapor Cloud Explosion	113
2	Sulfuric Acid	17	Vinyl Chloride	Toxic Emission	950
2	Sulfuric Acid	16	Xy lene	Fire Ball	350
2	Sulfuric Acid	18	Xylene	Toxic Emission	70
2	Sulfuric Acid	16	Xylene	Vapor Cloud Explosion	408
2	Sulfuric Acid, Spent	3	Acetic Acid, Glacial	Fire Ball	114
2	Sulfuric Acid, Spent	3	Acetic Acid, Glacial	Toxic Emission	1319
2	Sulfuric Acid, Spent	3	Acetic Acid, Glacial	Vapor Cloud Explosion	25
2	Sulfuric Acid, Spent	107	Acetic Anhydride	Fire Ball	80
2	Sulfuric Acid, Spent	107	Acetic Anhydride	Toxic Emission	125
2	Sulfuric Acid, Spent	197	Acetic Anhydride	Vapor Cloud Explosion	13:
2	Sulfuric Acid, Spent	19	Acetone	Fire Ball	158
2	Sulfuric Acid, Spent	19	Acetone	Toxic Emission	274
2	Sulfuric Acid, Spent	19	Acetone	Vapor Cloud Explosion	426
2	Sulfuric Acid, Spent	3	Acrylic Acid	Fire Ball	147
2	Sulfuric Acid, Spent	3	Acrylic Acid	Toxic Emission	
2	Sulfuric Acid, Spent	3	Acrylic Acid	Vapor Cloud Explosion	86
2	Sulfuric Acid, Spent	26	Acrylonitrile	Fire Ball	89
2	Sulfuric Acid, Spent	26	Acrylonitrile	Toxic Emission	753

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM #	Chemical Name	ASTN	Chemical Name	Consequence Type	Area Covered sq.m.
2	Sulfuric Acid, Spent	26	Acrylonitrile	Vapor Cloud Explosion	25862
2	Sulfuric Acid, Spent	4	Alcoholic Beverage	Fire Ball	25978
2	Sulfuric Acid, Spent	4.	Alcoholic Beverage	Vapor Cloud Explosion	31103
2	Sulfuric Acid, Spent	7	Aniline Oil, Liquid	Fire Ball	3452
2	Sulfuric Acid, Spent	7	Aniline Dil, Liquid	Toxic Exission	37699
2	Sulfuric Acid, Spent	16	Benzene	Fire Ball	35623
2	Sulfuric Acid, Spent	16	Benzene	Toxic Emission	9425
2	Sulfuric Acid, Spent	16	Benzene	Vapor Cloud Explosion	58Ø57
2	Sulfuric Acid, Spent	13	Butyl Acrylate	Fire Ball	965
2	Sulfuric Acid, Spent	13	Butyl Acrylate	Toxic Emission	?
2	Sulfuric Acid, Spent	. 4	Butyl Alcohol	Fire Ball	20262
2	Sulfuric Acid, Spent	4	Butyl Alcohol	Toxic Emission	28274
2	Sulfuric Acid, Spent	4	Butyl Alcohol	Vapor Cloud Explosion	28055
2	Sulfuric Acid, Spent	5	Butyraldehyde	Fire Ball	11096
2	Sulfuric Acid, Spent	5	Butyraldehyde	Vapor Cloud Explosion	24497
2	Sulfuric Acid, Spent	17	Carbon Tetrachloride	Toxic Emission	148440
2	Sulfuric Acid, Spent	17	Chloroprene, Inhibited	Fire Ball	11471
2	Sulfuric Acid, Spent	17	Chloroprene, Inhibited	Toxic Emission	178715
2	Sulfuric Acid, Spent	17	Chloroprene, Inhibited	Vapor Cloud Explosion	13996
2	Sulfuric Acid, Spent	4	Denatured Alcohol	Fire Ball	38266
2	Sulfuric Acid, Spent	4	Denatured Alcohol	Vapor Cloud Explosion	73062
2	Sulfuric Acid, Spent	27	Dinitrotoluene, Liquid	Explosions	30516
2	Sulfuric Acid, Spent	13	Ethyl Acetate	Fire Ball	15361
2	Sulfuric Acid, Spent	13	Ethyl Acetate	Toxic Emission	2356
2	Sulfuric Acid, Spent	13	Ethyl Acetate	Vapor Cloud Explosion	19879
2	Sulfuric Acid, Spent	13	Ethyl Acrylate	Fire Ball	5929
2	Sulfuric Acid, Spent	13	Ethyl Acrylate	Toxic Emission	6283
2	Sulfuric Acid, Spent	13	Ethyl Acrylate	Vapor Cloud Explosion	2124
2	Sulfuric Acid, Spent	4	Ethyl Alcohol	Fire Ball	38266
2	Sulfuric Acid, Spent	4	Ethyl Alcohol	Vapor Cloud Explosion	73062
2	Sulfuric Acid, Spent	5	Formaldehyde Solution	Toxic Emission	2356
2	Sulfuric Acid, Spent	14	Glycol Ethers	Fire Ball	1266
2	Sulfuric Acid, Spent	14	Glycol Ethers	Toxic Emission	785
2			I *	Fire Ball	
-	Sulfuric Acid, Spent	7	Hexamethylene Diamine Soln.	Fire Ball	382
2	Sulfuric Acid, Spent	4	Isopropano l	1	17801
2	Sulfuric Acid, Spent	4	Isopropanol	Toxic Emission	19635
2	Sulfuric Acid, Spent	4	Isopropanol	Vapor Cloud Explosion	22698
2	Sulfuric Acid, Spent	107	Maleic Anhydride	Toxic Emission	?
2	Sulfuric Acid, Spent	4	Methyl Alcohol	Fire Ball	29676
2	Sulfuric Acid, Spent	4	Methyl Alcohol	Toxic Emission	28274
2	Sulfuric Acid, Spent	4	Methyl Alcohol	Vapor Cloud Explosion	55757
2	Sulfuric Acid, Spent	19	Methyl Ethyl Ketone	Fire Ball	13550
2	Sulfuric Acid, Spent	19	Methyl Ethyl Ketone	Toxic Emission	3927
2	Sulfuric Acid, Spent	19	Methyl Ethyl Ketone	Vapor Cloud Explosion	24354
2	Sulfuric Acid, Spent	13	Methyl Methacrylate	Fire Ball	5103

Table J1. Summary of Consequences (Net Area Above the Lethal Limit) from Incompatible Binary Combinations (Cont.)

ASTM	Chemical Name	ASTM	Chemical Name	Consequence Type	Area Covered sq.m.
2	Sulfuric Acid, Spent	13	Methyl Methacrylate	Toxic Emission	2356
2	Sulfuric Acid, Spent	13	Methyl Methacrylate	Vapor Cloud Explosion	1521
2	Sulfuric Acid, Spent	24	Motor Fuel Antiknock Compound	Fire Ball	2667@
2	Sulfuric Acid, Spent	24	Motor Fuel Antiknock Compound	Vapor Cloud Explosion	25447
2	Sulfuric Acid, Spent	31	Pheno I	Fire Ball	106
2	Sulfuric Acid, Spent	31	Phenoi	Toxic Emission	3142
2	Sulfuric Acid, Spent	21	Sodium Metal	Fire Ball	9835
2	Sulfuric Acid, Spent	21	Sodium Metal	Vapor Cloud Explosion	13893
2	Sulfuric Acid, Spent	16	Styrene Monomer	Fire Ball	1758
2	Sulfuric Acid, Spent	16	Toluene	Fire Ball	5856
2	Sulfuric Acid, Spent	16	Toluene	Toxic Emission	2356
2	Sulfuric Acid, Spent	16	Toluene	Vapor Cloud Explosion	962
2	Sulfuric Acid, Spent	18	Toluene Diisocyanate	Fire Ball	190
2	Sulfuric Acid, Spent	18	Toluene Diisocyanate	Toxic Emission	28274
2	Sulfuric Acid, Spent	13	Vinyl Acetate	Fire Ball	14405
2	Sulfuric Acid, Spent	13	Vinyl Acetate	Toxic Emission	?
2	Sulfuric Acid, Spent	13	Vinyl Acetate	Vapor Cloud Explosion	22258
2	Sulfuric Acid, Spent	16	Xylene	Fire Ball	2237
16	Toluene	.104	Hydrogen Peroxide	Explosions	25239
16	Toluene	184	Hydrogen Peroxide	Fire Ball	31952
18	Toluene	184	Hydrogen Peroxide	Toxic Emission	129591
16	Toluene	184	Hydrogen Peroxide	Vapor Cloud Explosion	34987
18	Toluene Diisocyanate	196	Ammonium Nitrate Soln.	Toxic Emission	785
18	Toluene Diisocyanate	104	Hydrogen Peroxide	Explosions	25239
18	Toluene Diisocyanate	104	Hydrogen Peroxide	Fire Ball	196
18	Toluene Diisocyanate	164	Hydrogen Peroxide	Toxic Emission	129591
18	Toluene Diisocyanate	105	Phosphorus	Toxic Emission	785
18	Toluene Diisocyanate	21	Sodium Metal	Fire Ball	9838
18	Toluene Diisocyanate	21	Sodium Metal	Toxic Emission	785
18	Toluene Diisocyanate	21	Sodium Metal	Vapor Cloud Explosion	13893
13	Vinyl Acetate	194	Ferric Chloride Solution	Fire Ball	14484
13	Vinyl Acetate		Ferric Chloride Solution	Toxic Emission	ישררו
13	Vinyl Acetate Vinyl Acetate	194	Ferric Chloride Solution		22327
13	Vinyl Acetate	104		Vapor Cloud Explosion	
13	•	1	Hydrogen Peroxide	Explosions	25239
	Vinyl Acetate	104	Hydrogen Peroxide	Fire Ball	9096
13	Vinyl Acetate	164	Hydrogen Peroxide	Toxic Emission	129591
13	Vinyl Acetate	104	Hydrogen Peroxide	Vapor Cloud Explosion	11347
13	Vinyl Acetate	21	Sodium Metal	Fire Ball	11223
13	Vinyl Acetate	21	Sodium Metal	Toxic Emission	0041
13	Vinyl Acetate	21	Sodium Metal	Vapor Cloud Explosion	28414
17	Vinyl Chloride	194	Hydrogen Peroxide	Explosions	25239
17	Vinyl Chloride	104	Hydrogen Peroxide	Toxic Emission	9425
18	Xylene	104	Hydrogen Peroxide	Explosions	25239
16	Xylene	104	Hydrogen Peroxide	Fire Ball	35015
16	Xylene	184	Hydrogen Peroxide	Toxic Emission	129591
16	Xylene	184	Hydrogen Peroxide	Vapor Cloud Explosion	4682

Table J2. Summary of Consequences (Net Area Above the Lethal Limit) from Spills of Single Chemicals

	<u> </u>		r
}			Area
ASTM	<u>.</u>		Covered
#	Chemical Name	Consequence Type	sq.m.
-	Anaka 7 datu da	F: D 11	50100
5	Acetaldehyde	Fire Ball	50198
5	Acetaldehyde	Toxic Emission	95033
] 2	Acetaldehyde	Vapor Cloud Explosion	108103
5 5 3 3	Acetic Acid, Glacial	Fire Ball	167
	Acetic Acid, Glacial	Toxic Emission	785
19	Acetone	Fire Ball	16644
19	Acetone	Toxic Emission	785
19	Acetone	Vapor Cloud Explosion	16742
3	Acrylic Acid	Fire Ball	118
26	Acrylonitrile	Fire Ball	4573
26	Acrylonitrile	Toxic Emission	3142
26	Acrylonitrile	Vapor Cloud Explosion	154
10	Anhydrous Ammonia	Fire Ball	49807
10	Anhydrous Ammonia	Toxic Emission	1320254
10	Anhydrous Ammonia	Vapor Cloud Explosion	42273
7	Aniline Oil, Liquid	Toxic Emission	785
16	Benzene	Fire Ball	9474
16	Benzene	Toxic Emission	3142
16	Benzene	Vapor Cloud Explosion	2206
28	Butadiene	Fire Ball	121871
28	Butadiene	Toxic Emission	50265
28	Butadiene	Vapor Cloud Explosion	255176
29	Butane	Fire Ball	124699
29	Butane	Vapor Cloud Explosion	265120
13	Butyl Acrylate	Fire Ball	171
4	Butyl Alcohol	Fire Ball	142
5 5	Butyraldehyde	Fire Ball	9607
	Butyraldehyde	Vapor Cloud Explosion	5675
20 20	Carbon Disulfide Carbon Disulfide	Fire Ball	30928
20	Carbon Disulfide	Toxic Emission	63617
	=	Vapor Cloud Explosion	15394
17	Carbon Tetrachloride	Toxic Emission	28274
104	Chlorine	Toxic Emission	5026548
17	Chloroprene, Inhibited	Fire Ball	15824
17	Chloroprene, Inhibited	Toxic Emission	50265
17	Chloroprene, Inhibited	Vapor Cloud Explosion	6362
101	Coal Tar Distillate	Fire Ball	133
29	Cyclohexane	Fire Ball	14235
29	Cyclohexane	Toxic Emission	785
29	Cyclohexane	Vapor Cloud Explosion	6504
1 4	Denatured Alcohol	Fire Ball Fire Ball	1764
13	Ethyl Acetate		4224
13	Ethyl Acetate	Toxic Emission	785 254
13	Ethyl Acetate	Vapor Cloud Explosion Fire Ball	1418
13	Ethyl Acrylate	THE DATE	1410

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Table J2. Summary of Consequences (Net Area Above the Lethal Limit) from Spills of Single Chemicals (Cont.)

			Area
ASTM			Covered
#	Chemical Name	Consequence Type	sq.m.
			
13	Ethyl Acrylate	Toxic Emission	785
4	Ethyl Alcohol	Fire Ball	1764
34	Ethylene Oxide	Fire Ball	54260
34	Ethylene Oxide	Toxic Emission	490874
34	Ethylene Oxide	Vapor Cloud Explosion	130100
5	Formaldehyde Solution	Toxic Emission	785
101	Fuel Oil	Fire Ball	187
101	Fuel, Aviation	Fire Ball	183
14 29	Glycol Ethers	Fire Ball	356
29	Hexane	Fire Ball	22317
29	Hexane Hexane	Toxic Emission	3142
1 1		Vapor Cloud Explosion Fire Ball	17671
1 1	Hydrocyanic Acid	Toxic Emission	73186
l il	Hydrocyanic Acid Hydrocyanic Acid		6082123
1 1	Hydrogen Fluoride	Vapor Cloud Explosion Toxic Emission	99538 5674502
104	Hydrogen Peroxide	Toxic Emission	3142
4	Isopropanol	Fire Ball	1221
28	LPG	Toxic Emission	28274
28	LPG-Butene	Fire Ball	124352
28	LPG-Butene	Toxic Emission	50265
28	LPG-Butene	Vapor Cloud Explosion	263298
28	LPG-Butylene	Toxic Emission	50265
29	LPG-Isobutane	Fire Ball	124502
29	LPG-Isobutane	Vapor Cloud Explosion	264208
28	LPG-Isobutylene	Fire Ball	123261
28	LPG-Isobutylene	Vapor Cloud Explosion	259672
29	LPG-Propane	Fire Ball	125027
29	LPG-Propane	Toxic Emission	63617
29	LPG-Propane	Vapor Cloud Explosion	266948
28	LPG-Propylene	Fire Ball	97827
28	LPG-Propylene	Vapor Cloud Explosion	187805
4	Methyl Alcohol	Fire Ball	6914
4	Methyl Alcohol	Toxic Emission	785 ⁻
4	Methyl Alcohol	Vapor Cloud Explosion	1075
17	Methyl Chloride	Fire Ball	43200
17	Methyl Chloride	Toxic Emission	78540
17	Methyl Chloride	Vapor Cloud Explosion	24606
19	Methyl Ethyl Ketone	Fire Ball	6355
19	Methyl Ethyl Ketone	Toxic Emission	3142
19	Methyl Ethyl Ketone	Vapor Cloud Explosion	1662
13	Methyl Methacrylate	Fire Ball	1570
13	Methyl Methacrylate	Toxic Emission	785
24	Motor Fuel Antiknock Compound	Toxic Emission	785
2	Nitric Acid, Fuming	Toxic Emission	63617

Table J2. Summary of Consequences (Net Area Above the Lethal Limit) from Spills of Single Chemicals (Cont.)

ASTM #	Chemical Name	Consequence Type	Area Covered sq.m.
2 101	Oleum Petroleum Naphtha	Toxic Emission Fire Ball	201062 371
34	Propylene Oxide	Fire Ball	59482
34 34 34	Propylene Oxide	Toxic Emission	226980
34	Propylene Oxide	Vapor Cloud Explosion	156228
16	Styrene Monomer	Fire Ball	312
105	Sulfur Dioxide	Toxic Emission	1590431
16	Toluene	Fire Ball	1258
16	Toluene	Toxic Emission	785
13	Vinyl Acetate	Fire Ball	7138
13	Vinyl Acetate	Vapor Cloud Explosion	1521
16 13 13 17 17	Vinyl Chloride	Fire Ball	64672
17	Vinyl Chloride	Vapor Cloud Explosion	71631
16	Xylene	Fire Ball	699

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit)

ASTM		ACTM		Area
#1	Chemical Name #1	ASTM #2	Chamical Name #2	Covered
	Chemical Name #1	#4	Chemical Name #2	sq.m.
1	Hydrofluorosilicic Acid	2	Sulfuric Acid	907920
2	0]eum	4	Ethyl Alcohol	855693
2	Oleum	4	Denatured Alcohol	855693
1 1	Hydrofluorosilicic Acid	2	Nitric Acid, Fuming	844303
2 2 1	Nitric Acid, Fuming	4	Ethyl Alcohol	834124
2	Nitric Acid, Fuming	4	Denatured Alcohol	834124
	Hydrochloric Acid	104	Hydrogen Peroxide	767526
2 2 2 2	Oleum	4	Methyl Alcohol	764509
2	Nitric Acid, Fuming	4	Methyl Alcohol	762448
4	Oleum	4	[sopropano]	739283
1	Nitric Acid, Fuming	4	Isopropanol	733888
2	Hydrofluorosilicic Acid Oleum	2 4	Oleum	706858
2	Oleum	16	Butyl Alcohol Xylene	683162
2 2	Nitric Acid, Fuming	4	Ayrene Butyl Alcohol	679029 673505
2	Oleum	16	Toluene	670105
20	Carbon Disulfide	104	Hydrogen Peroxide	668480
2	Oleum	16	Benzene	667136
2	Nitric Acid, Fuming	16	Benzene	660564
2	Oleum	16	Styrene Monomer	655009
2	Oleum	4	Alcoholic Beverage	650305
2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	01eum -	19	Methyl Ethyl Ketone	639990
2	Oleum	13	Ethyl Acrylate	639964
2	Oleum	106	Ammonium Nitrate Soln.	639168
2	Oleum	5	Butyraldehyde	638779
2	Oleum	14	Glycol Ethers	638605
2	Oleum	26	Acrylonitrile	637950
2	Oleum	13	Methyl Methacrylate	637308
2	Oleum	27	Dinitrotoluene, Liquid	633702
2 2	Nitric Acid, Fuming	5	Butyraldehyde	633351
4	Oleum	13	Ethyl Acetate	629531
2	Oleum	104	Hydrogen Peroxide	628425
4	Nitric Acid, Fuming Oleum	16	Xylene	627978
6	Oleum	3	Acrylic Acid	626584
2	Nitric Acid, Fuming	13 16	Vinyl Acetate	623629
5	Nitric Acid, Fuming	4	Toluene	619054
5	Oleum	3	Alcoholic Beverage Acetic Acid, Glacial	617827 617171
5	Nitric Acid, Fuming	14	Glycol Ethers	617036
2	Nitric Acid, Fuming	26	Acrylonitrile	613288
2	Oleum	13	Butyl Acrylate	612695
2	Oleum	107	Acetic Anhydride	612556
2	01eum	19	Acetone	608923
2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	01eum	7	Aniline Oil, Liquid	606638
] 2	Oleum	31	Pheno1	605177

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

				
.				Area
ASTM		ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
2	Nitric Acid, Fuming	16	Styrene Monomer	603958
2	Nitric Acid, Fuming	19	Acetone	603495
2222222222222221	01eum	18	Toluene Diisocyanate	603376
2	Oleum	24	Motor Fuel Antiknock Comp	
2	01eum	17	Carbon Tetrachloride	603186
2	Oleum	10	Sodium Hydroxide, Liq.	603186
2	Oleum	107	Maleic Anhydride	603186
2	Oleum	10	Alkaline Liquid, n.o.s.	603186
2	Oleum	4	Octyl Alcohol	603186
2	Oleum	5	Acetaldehyde	603186
2	Oleum	10	Potassium Hydroxide, Liq	603186
2	Oleum	34	Propylene Oxide	603186
2	Oleum	34	Ethylene Oxide	603186
2	Oleum	105	Phosphorus	603186
2	Nitric Acid, Fuming	107	Acetic Anhydride	599121
2	Nitric Acid, Fuming	17	Chloroprene, Inhibited	599076
2	Nitric Acid, Fuming	19	Methyl Ethyl Ketone	588939
2	Nitric Acid, Fuming	13	Ethyl Acrylate	588913
2	Nitric Acid, Fuming	13	Methyl Methacrylate	586257
2	Nitric Acid, Fuming Sulfuric Acid	5	Acetaldehyde	566342
1	Hydrocyanic Acid	17 104	Chloroprene, Inhibited Chlorine	565782 565487
2	Nitric Acid, Fuming	13	Butyl Acrylate	561644
2	Nitric Acid, Fuming	7	Aniline Oil, Liquid	555587
2	Nitric Acid, Fuming	31	Phenol	554126
2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	Nitric Acid, Fuming	7	Hexamethylene Diamine Sol	552497
2	Nitric Acid, Fuming	18	Toluene Diisocyanate	552325
2	Nitric Acid, Fuming	5	Formaldehyde Solution	552292
2	Nitric Acid, Fuming	10	Alkaline Liquid, n.o.s.	552135
2	Nitric Acid, Fuming	105	Phosphorus	552135
2	Nitric Acid, Fuming	21	Sodium Metal	552135
2	Nitric Acid, Fuming	24	Motor Fuel Antiknock Comp	552135
2	Nitric Acid, Fuming	10	Sodium Hydroxide, Liq.	552135
2	Nitric Acid, Fuming	10	Potassium Hydroxide, Liq	552135
1	Hydroiodic Acid	2	Nitric Acid, Fuming	552135
2	Sulfuric Acid	17	Carbon Tetrachloride	544281
10	Anhydrous Ammonia	17	Methyl Chloride	519315
1	Hydrochloric Acid	2	Nitric Acid, Fuming	508938
1	Hydrochloric Acid	2	Sulfuric Acid	490874
	Hydrochloric Acid	2	Sulfuric Acid, Spent	490874
1 2 2	Oleum .	17	Chloroprene, Inhibited	414985
	Nitric Acid, Fuming	17	Methyl Chloride	412334
20	Carbon Disulfide	104	Ferric Chloride Solution	388772
2	Oleum	7	Hexamethylene Diamine Sol	330229
2	Oleum	5	Formaldehyde Solution	330024
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Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

	·]		Area
ASTM	Ob	ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
2	Oleum	28	Butadiene	329867
	01eum	21	Sodium Metal	329867
2 2 2 2 2	Oleum	28	LPG	329867
2	Oleum	28	LPG-Butene	329867
2	Oleum	28	LPG-Isobutylene	329867
2	Oleum	28	LPG-Propylene	329867
10	Sodium Hydroxide, Liq.	17	Carbon Tetrachloride	318086
10	Alkaline Liquid, n.o.s.	17	Carbon Tetrachloride	318086
10	Potassium Hydroxide, Liq	17	Carbon Tetrachloride	318086
l î l	Hydrofluorosilicic Acid	17	Methyl Chloride	301593
Ī	Hydrochloric Acid	2	Oleum	289812
17	Chloroprene, Inhibited	.24	Motor Fuel Antiknock Comp	261833
17	Chloroprene, Inhibited	104	Ferric Chloride Solution	261833
i	Hydroiodic Acid	105	Phosphorus	255742
l īl	Hydroiodic Acid	2	Sulfuric Acid	254469
20	Carbon Disulfide	21	Sodium Metal	252602
2	Nitric Acid, Fuming	20	Carbon Disulfide	252585
5	Sulfuric Acid	4	Ethyl Alcohol	252507
2	Sulfuric Acid	4	Denatured Alcohol	252507
2	Oleum	101	Coal Tar Distillate	251924
2	Oleum	101	Fuel Oil	251924
2	Oleum	101	Fuel, Aviation	251879
2 2 2 2 2 2 2	Oleum	101	Petroleum Naphtha	251854
2	Oleum	101	Crude Oil Petroleum	251557
2	Oleum	101	Pulp Mill Liquid	251327
<u>2</u>	Oleum	101	Combustible Liquid	251327
2	Oleum	101	Compound, Cleaning	251327
2	Oleum	101	Gasoline, nec	251327
2 2 2 2 2 2 2 2	Oleum	101	Gasoline	251327
2	Nitric Acid, Fuming	2	Oleum	251327
ī	Hydroiodic Acid	2	Oleum	251327
1 2	Oleum	101	Flammable Liquid	251327
2	Oleum	101	Distillate Fuel Oil	251327
2	Oleum	17	Methyl Chloride	251327
2 2 2 2 2 2	Oleum	101	Petroleum Residual Fuel	251327
1 2	01eum	101	Petroleum, Refined	251327
2	Oleum	101	0il	251327
2	Oleum	17	Vinyl Chloride	251327
17	Carbon Tetrachloride	20	Carbon Disulfide	250542
5	Butyraldehyde	104	Hydrogen Peroxide	241684
3	Acrylic Acid	17	Chloroprene, Inhibited	238922
17	Chloroprene, Inhibited	21	Sodium Metal	231305
16	Xylene	104	Hydrogen Peroxide	230673
17	Chloroprene, Inhibited	104	Hydrogen Peroxide	227421
3	Acetic Acid, Glacial	17	Chloroprene, Inhibited	226871
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Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

				,
]		Area
ASTM	,	ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
				
20	Carbon Disulfide	34	Propylene Oxide	225409
16	Toluene	104	Hydrogen Peroxide	221749
16	Benzene	104	Hydrogen Peroxide	218780
7	Aniline Oil, Liquid	17	Chloroprene, Inhibited	206971
16	Styrene Monomer	104	Hydrogen Peroxide	206653
3 4	Acetic Acid, Glacial	21	Sodium Metal	203636
	Butyl Alcohol	104	Hydrogen Peroxide	203147
7	Hexamethylene Diamine Sol	17	Chloroprene, Inhibited	202203
10	Potassium Hydroxide, Liq	17	Chloroprene, Inhibited	202182
2	Sulfuric Acid, Spent	17	Chloroprene, Inhibited	202182
10	Alkaline Liquid, n.o.s.	17	Chloroprene, Inhibited	202182
10	Sodium Hydroxide, Liq.	17	Chloroprene, Inhibited	202182
1 1	Hydroiodic Acid	17	Chloroprene, Inhibited	202182
i i	Phosphoric Acid	17	Chloroprene, Inhibited	202182
l i	Hydrofluorosilicic Acid	17		
1	Hydrochloric Acid	17	Chloroprene, Inhibited	202182
4			Chloroprene, Inhibited	202182
3	Isopropanol	104	Hydrogen Peroxide	195329
	Acetic Acid, Glacial	5	Butyraldehyde	194662
19	Methyl Ethyl Ketone	104	Hydrogen Peroxide	191634
13	Ethyl Acrylate	104	Hydrogen Peroxide	191608
2	Sulfuric Acid	4	Methyl Alcohol	189597
26	Acrylonitrile	104	Hydrogen Peroxide	189594
13	Methyl Methacrylate	104	Hydrogen Peroxide	188952
4	Ethyl Alcohol	104	Hydrogen Peroxide	187691
4	Denatured Alcohol	104	Hydrogen Peroxide	187691
13	Ethyl Acetate	104	Hydrogen Peroxide	181175
21	Sodium Metal	104	Hydrogen Peroxide	178558
3 2 2	Acrylic Acid	104	Hydrogen Peroxide	178228
2	Nitric Acid, Fuming	3	Acetic Acid, Glacial	177202
2	Sulfuric Acid	21	Sodium Metal	176715
13	Vinyl Acetate	104	Hydrogen Peroxide	175273
2	Nitric Acid, Fuming	13	Ethyl Acetate	169793
2	Nitric Acid, Fuming	13	Vinyl Acetate	168868
3	Acetic Acid, Glacial	104	Hydrogen Peroxide	168815
4	Methyl Alcohol	104	Hydrogen Peroxide	167854
2	Nitric Acid, Fuming	4	Octyl Alcohol	166834
2 2	Nitric Acid, Fuming	107	Maleic Anhydride	164533
13	Butyl Acrylate	107		
104		104	Hydrogen Peroxide	164339
	Hydrogen Peroxide		Acetic Anhydride	164200
2	Nitric Acid, Fuming	104	Hydrogen Peroxide	162684
27	Dinitrotoluene, Liquid	104	Hydrogen Peroxide	161644
19	Acetone	104	Hydrogen Peroxide	160567
3	Acetic Acid, Glacial	107	Acetic Anhydride	158610
7	Aniline Oil, Liquid	104	Hydrogen Peroxide	158282
31	Phenol	104	Hydrogen Peroxide	156821
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Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

ASTM					Area
The Chemical Name #1 #2 Chemical Name #2 \$q.m.	ASTM		ASTM		
Hydrogen Peroxide 107		Chemical Name #1		Chemical Name #2	
18 Tolueme Diisocyanate 104 Hydrogen Peroxide 155020 154830 24 Motor Fuel Antiknock Comp 104 Hydrogen Peroxide 154830 164831 164831 164937				onemical mane #2	34.111.
18 Tolueme Diisocyanate 4 Octyl Alcohol 24 Motor Fuel Antiknock Comp 3 Acetic Acid, Glacial 7 Hexamethylene Diamine Sol 17 Carbon Tetrachloride 18 Toluene Diisocyanate 18 Toluene 18440 18 Toluene Diisocyanate 18 Toluene 184840 18 Acetic Acid, Glacial 19 Carbon Tetrachloride 184840 18 Acetic Acid, Glacial 19 Carbon Tetrachloride 184840 18 Acetic Acid, Glacial 19 Carbon Tetrachloride 184840 18 Acetic Acid, Glacial 19 Carbon Tetrachloride 184840 19 Carbon Tetrachloride 18 Acetic Acid, Glacial 19 Carbon Tetrachloride 18 Acetic Acid, Glacial 10 Potassium Hydroxide, Liq 145932 1459	104	Hydrogen Peroxide	107	Maleic Anhydride	155922
4 Octyl Alcohol 24 Motor Fuel Antiknock Comp 3 Acetic Acid, Glacial 4 Hydrogen Peroxide 154830 7 Hexamethylene Diamine Sol 7 Carbon Tetrachloride 17 Carbon Tetrachloride 18 Carbon Tetrachloride 19 Carbon Tetrachloride 19 Carbon Tetrachloride 19 Carbon Tetrachloride 10 C	18	Toluene Diisocyanate		Hydrogen Peroxide	
Acetic Acid, Glacial 18	4	Octyl Alcohol			
Acetic Acid, Glacial Hydroidic Acid, Spent Hydroidic Acid, Spent Aniline Oil, Liquid Acetic Acid, Glacial Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming Acetic Acid, Fuming	24				
7 Hexamethylene Diamine Sol 17 Carbon Tetrachloride 148440 11 Hydroiodic Acid 17 Carbon Tetrachloride 148440 12 Acetic Acid, Glacial 17 Carbon Tetrachloride 148440 13 Acetic Acid, Glacial 5 Sulfuric Acid 3 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Sodium Hydroxide, Liq 145932 Sulfuric Acid, Spent 3 Acetic Acid, Glacial 10 Sodium Hydroxide, Liq 145932 Sulfuric Acid, Spent 3 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Sodium Hydroxide, Liq 145932 Acetic Acid, Glacial 10 Acetic Acid, Fuming 101 Carbon Tetrachloride 148440 Acetic Acid, Glacial 148250 Acetic Acid, Glacial 148250 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 145932 Acetic Acid, Glacial 145932 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 10 Acetic Acid, Glacial 145932 Acetic Acid, Glacial 10 Ace	3				
1 Hydroiodic Acid	7				
1 Hydroiodic Acid	2				
1 Hydroiodic Acid	17	Carbon Tetrachloride			
21 Sodium Metal 26 Acrylonitrile 145850 104 Ferric Chloride Solution 2 Sulfuric Acid 16 Benzene 139348 2 Nitric Acid, Fuming 101 Coal Tar Distillate 138042 2 Nitric Acid, Fuming 101 Fuel 0il 137997 2 Nitric Acid, Fuming 101 Petroleum Naphtha 137972 2 Nitric Acid, Fuming 101 Crude Oil Petroleum 137675 2 Nitric Acid, Fuming 101 Flammable Liquid 137445 2 Nitric Acid, Fuming 101 Distillate Fuel Oil 137445 2 Nitric Acid, Fuming 101 Distillate Fuel Oil 137445 2 Nitric Acid, Fuming 101 Gasoline 137445 2 Nitric Acid, Fuming 28 LPG-Propylene 137445 2 Nitric Acid, Fuming 27 Dinitrotoluene, Liquid 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Gasoline 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 106 Ammonium Nitrate Soln. 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 LPG 3 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 LPG 3 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 Butadiene 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445	-		17		
21 Sodium Metal 26 Acrylonitrile 145850 104 Ferric Chloride Solution 2 Sulfuric Acid 16 Benzene 139348 2 Nitric Acid, Fuming 101 Coal Tar Distillate 138042 2 Nitric Acid, Fuming 101 Fuel 0il 137997 2 Nitric Acid, Fuming 101 Petroleum Naphtha 137972 2 Nitric Acid, Fuming 101 Crude Oil Petroleum 137675 2 Nitric Acid, Fuming 101 Flammable Liquid 137445 2 Nitric Acid, Fuming 101 Distillate Fuel Oil 137445 2 Nitric Acid, Fuming 101 Distillate Fuel Oil 137445 2 Nitric Acid, Fuming 101 Gasoline 137445 2 Nitric Acid, Fuming 28 LPG-Propylene 137445 2 Nitric Acid, Fuming 27 Dinitrotoluene, Liquid 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Gasoline 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 106 Ammonium Nitrate Soln. 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 LPG 3 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 LPG 3 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 Butadiene 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445	7				
21 Sodium Metal 26 Acrylonitrile 145850 104 Ferric Chloride Solution 2 Sulfuric Acid 16 Benzene 139348 2 Nitric Acid, Fuming 101 Coal Tar Distillate 138042 2 Nitric Acid, Fuming 101 Fuel 0il 137997 2 Nitric Acid, Fuming 101 Petroleum Naphtha 137972 2 Nitric Acid, Fuming 101 Crude Oil Petroleum 137675 2 Nitric Acid, Fuming 101 Flammable Liquid 137445 2 Nitric Acid, Fuming 101 Distillate Fuel Oil 137445 2 Nitric Acid, Fuming 101 Distillate Fuel Oil 137445 2 Nitric Acid, Fuming 101 Gasoline 137445 2 Nitric Acid, Fuming 28 LPG-Propylene 137445 2 Nitric Acid, Fuming 27 Dinitrotoluene, Liquid 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Gasoline 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 101 Petroleum Residual Fuel 137445 2 Nitric Acid, Fuming 106 Ammonium Nitrate Soln. 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 LPG 3 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 LPG 3 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 2 Nitric Acid, Fuming 28 Butadiene 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Petroleum, Refined 137445	′3		1/2		
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Nitric Acid, Fuming 28 LPG-Isobutylene 137445 Nitric Acid, Fuming 28 LPG-Isobutylene 137445 Nitric Acid, Fuming 101 Gasoline, nec 137445 Nitric Acid, Fuming 106 Ammonium Nitrate Soln. 137445 Nitric Acid, Fuming 28 LPG-Butene 137445 Nitric Acid, Fuming 101 Pulp Mill Liquid 137445 Nitric Acid, Fuming 28 LPG 137445 Nitric Acid, Fuming 28 LPG 137445 Nitric Acid, Fuming 28 Butadiene 137445 Nitric Acid, Fuming 17 Vinyl Chloride 137445 Nitric Acid, Fuming 101 Petroleum, Refined 137445 Nitric Acid, Fuming 101 Compound, Cleaning 137445	2				
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2 Nitric Acid, Fuming 101 Petroleum, Refined 137445 2 Nitric Acid, Fuming 101 Compound, Cleaning 137445 13 Ethyl Acrylate 21 Sodium Metal 130821	2				
2 Nitric Acid, Fuming 101 Compound, Cleaning 137445 13 Ethyl Acrylate 21 Sodium Metal 130821	2			Petroleum, Refined	
13 Etnyl Acrylate 21 Sodium Metal / 130821	2			Compound, Cleaning	137445
	13	Etnyl Acrylate	21	Sodium Metal /	130821

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

ASTM #1	Chemical Name #1	ASTM #2	Chemical Name #2	Area Covered sq.m.
26	Acrylonitrile	31	Phenol	
19	Methyl Ethyl Ketone	21	Sodium Metal	130246 129828
	Sulfuric Acid	16	Toluene	129751
2 2 1	Sulfuric Acid	104	Hydrogen Peroxide	129591
$\overline{1}$	Hydrochloric Acid	4	Ethyl Alcohol	128629
	Hydrochloric Acid	4	Denatured Alcohol	128629
1 2 2 1	Sulfuric Acid	26	Acrylonitrile	126656
2	Nitric Acid, Fuming	34	Ethylene Oxide	124878
1	Hydrofluorosilicic Acid	4	Denatured Alcohol	116848
1	Hydrofluorosilicic Acid	4	Ethyl Alcohol	116848
7	Aniline Oil, Liquid	26	Acrylonitrile	116697
1 1	Hydroiodic Acid	4	Methyl Alcohol	113740
1 2 7	Sulfuric Acid, Spent	4	Methyl Alcohol	113707
7	Hexamethylene Diamine Sol	26	Acrylonitrile	111730
2	Sulfuric Acid, Spent	4	Denatured Alcohol	111328
2	Sulfuric Acid, Spent	4	Ethyl Alcohol	111328
2 2 2 5 1 2 1 1 5	Sulfuric Acid, Spent	26	Acrylonitrile	110162
5	Butyraldehyde	7	Hexamethylene Diamine Sol	
1 1	Hydrochloric Acid	5	Butyraldehyde	103483
2	Sulfuric Acid, Spent	16	Benzene	103105
1 1	Hydroiodic Acid	5	Butyraldehyde	102698
1 1	Phosphoric Acid	5	Butyraldehyde	102698
5	Butyraldehyde	21	Sodium Metal	102698
4	Alcoholic Beverage	5	Butyraldehyde	102692
13	Methyl Methacrylate	21	Sodium Metal	102603
1	Hydrofluorosilicic Acid	16	Styrene Monomer	102088
4	Butyl Alcohol	21	Sodium Metal	101070
1	Hydroiodic Acid	13	Ethyl Acrylate	99610
1	Hydrochloric Acid	13	Ethyl Acrylate	99610
2 4	Sulfuric Acid	13	Ethyl Acrylate	99610
2	Butyl Alcohol Sulfuric Acid	5 7	Butyraldehyde	99054
	Sulfuric Acid	19	Aniline Oil, Liquid	97700
5	Butyraldehyde	7	Methyl Ethyl Ketone Aniline Oil, Liquid	97280 96076
1 1		26		
2 5 1 1 1 2 5 4	Phosphoric Acid Hydrochloric Acid	26	Acrylonitrile Acrylonitrile	95240 95240
1 1	Hydroiodic Acid	26	Acrylonitrile	95240
1 1	Hydroiodic Acid	20	Sulfuric Acid, Spent	95033
5	Sulfuric Acid	17	Vinyl Chloride	95033
5	Butyraldehyde	107	Acetic Anhydride	93406
4	Isopropanol	21	Sodium Metal	92219
4	Isopropanol	5	Butyraldehyde	89450
1	Hydrocyanic Acid	24	Motor Fuel Antiknock Comp	87982
1 1	Hydrofluorosilicic Acid	13	Ethyl Acrylate	86258
2	Sulfuric Acid, Spent	19	Acetone	86008

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

			·	Area
ASTM		ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
				<u> </u>
10	Alkaline Liquid, n.o.s.	19	Acetone	86008
10	Sodium Hydroxide, Liq.	19	Acetone	86008
10	Potassium Hydroxide, Liq	19	Acetone	86008
1 1	Hydrochloric Acid	19	Acetone	86008
1 1	Phosphoric Acid	19	Acetone	86008
1	Hydrofluorosilicic Acid	19	Acetone	86008
1 1	Hydroiodic Acid	19	Acetone	86008
	Hydrofluorosilicic Acid	5	Butyraldehyde	85858
3	Acrylic Acid	21	Sodium Metal	85559
1 3 5	Acetaldehyde	104	Hydrogen Peroxide	84233
i	Hydroiodic Acid	13	Vinyl Acetate	84060
10	Alkaline Liquid, n.o.s.	13	Vinyl Acetate Vinyl Acetate	84060
1 1	Hydrochloric Acid	13	Vinyl Acetate	
10	Potassium Hydroxide, Liq	13		84060
10	Sodium Hydroxide, Liq.	13	Vinyl Acetate	84060
1 1	Hydrofluorosilicic Acid		Vinyl Acetate	84060
1 1	Hydrofluorosilicic Acid	13	Methyl Methacrylate	83602
4	Hydrofluorosilicic Acid Ethyl Alcohol	26	Acrylonitrile	81888
4		5	Butyraldehyde	79420
7	Denatured Alcohol	5	Butyraldehyde	79420
2	Sulfuric Acid	16	Xylene	76628
4 2 2 3 5 3	Butyl Alcohol	18	Toluene Diisocyanate	76625
2	Sulfuric Acid	4	Butyl Alcohol	76591
4	Sulfuric Acid, Spent	4	Butyl Alcohol	76591
ا ع	Acrylic Acid	7	Aniline Oil, Liquid	76337
ا ي	Butyraldehyde	31	Phenol	75749
3	Acrylic Acid	5	Butyraldehyde	75105
13	Ethyl Acetate	21	Sodium Metal	74806
1 1	Hydrofluorosilicic Acid	21	Sodium Metal	73993
1 1	Hydroiodic Acid	13	Methyl Methacrylate	71821
1 2 7	Hydrochloric Acid	13	Methyl Methacrylate	71821
2	Sulfuric Acid	13	Methyl Methacrylate	71821
7	Aniline Oil, Liquid	21	Sodium Metal	71678
] 1]	Hydrochloric Acid	16	Styrene Monomer	71458
1	Hydrofluorosilicic Acid	13	Vinyl Acetate	70708
10	Alkaline Liquid, n.o.s.	26	Acrylonitrile	69378
10	Potassium Hydroxide, Liq	26	Acrylonitrile	69378
10	Sodium Hydroxide, Liq.	26	Acrylonitrile	69378
4	Isopropanol	18	Toluene Diisocyanate	68805
4	Methyl Alcohol	21	Sodium Metal	67322
1 1	Hydrofluorosilicic Acid	17	Carbon Tetrachloride	66759
3	Acetic Acid, Glacial	7	Aniline Oil, Liquid	64424
] 3	Acetic Acid, Glacial	7	Hexamethylene Diamine Sol	64249
5	Butyraldehyde	18	Toluene Diisocyanate	64059
1 3 3 5 4	Ethyl Alcohol	21	Sodium Metal	63628
4	Denatured Alcohol	21	Sodium Metal	63628

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

				
		}		Area
ASTM	• • • •	ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
4	Denatured Alcohol	18	Taluana Diicaayanata	61165
4	Ethyl Alcohol	18	Toluene Diisocyanate Toluene Diisocyanate	61165
2	Sulfuric Acid, Spent	4	Isopropanol	60134
2	Sulfuric Acid	4	Isopropanol	60134
1	Hydrofluorosilicic Acid	13	Butyl Acrylate	59774
21	Sodium Metal	106	Ammonium Nitrate Soln.	59710
1	Hydrofluorosilicic Acid	107	Acetic Anhydride	59635
10	Anhydrous Ammonia	24	Motor Fuel Antiknock Comp	58934
3	Acrylic Acid	18	Toluene Diisocyanate	57761
2	Sulfuric Acid, Spent	4	Alcoholic Beverage	57081
19	Acetone	21	Sodium Metal	56291
10	Potassium Hydroxide, Liq	16	Styrene Monomer	54965
2	Sulfuric Acid	16	Styrene Monomer	54965
1	Phosphoric Acid	16	Styrene Monomer	54965
il	Hydroiodic Acid	16	Styrene Monomer	54965
10	Alkaline Liquid, n.o.s.	16	Styrene Monomer	54965
10	Sodium Hydroxide, Liq.	16	Styrene Monomer	54965
7	Aniline Oil, Liquid	24	Motor Fuel Antiknock Comp	
4	Methyl Alcohol	5	Butyraldehyde	53323
il	Hydrochloric Acid	4	Isopropanol	53105
ī	Hydrofluorosilicic Acid	4	Isopropanol	53105
5	Formaldehyde Solution	7	Aniline Oil, Liquid	52518
ĭ	Phosphoric Acid	24	Motor Fuel Antiknock Comp	
ī	Hydroiodic Acid	24	Motor Fuel Antiknock Comp	
	Sulfuric Acid, Spent	24	Motor Fuel Antiknock Comp	
2	Hydrochloric Acid	24	Motor Fuel Antiknock Comp	
ī	Hydrofluorosilicic Acid	24	Motor Fuel Antiknock Comp	L .
ī	Hydrochloric Acid	105	Phosphorus	51538
5	Butyraldehyde	104	Ferric Chloride Solution	51437
10	Potassium Hydroxide, Liq	105	Phosphorus	51427
10	Sodium Hydroxide, Liq.	105	Phosphorus	51427
10	Alkaline Liquid, n.o.s.	105	Phosphorus	51427
1	Hydrofluorosilicic Acid	107	Maleic Anhydride	51357
34	Propylene Oxide	104	Hydrogen Peroxide	50564
i	Hydrofluorosilicic Acid	18	Toluene Diisocyanate	50455
ī	Hydrofluorosilicic Acid	10	Sodium Hydroxide, Lig.	50265
i	Hydrofluorosilicic Acid	10	Alkaline Liquid, n.o.s.	50265
ī	Hydrofluorosilicic Acid	10	Potassium Hydroxide, Liq	50265
	Aniline Oil, Liquid	107	Acetic Anhydride	49327
7 3 2 2	Acrylic Acid	107	Acetic Annydride	48326
2	Sulfuric Acid	14	Glycol Ethers	47985
2	Sulfuric Acid	4	Alcoholic Beverage	47119
1	Hydrochloric Acid	4	Alcoholic Beverage	45600
10	Sodium Hydroxide, Liq.	13	Ethyl Acetate	45195
10	Potassium Hydroxide, Liq	13	Ethyl Acetate	45195
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Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

		,		
				Area
ASTM		ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
		 		<u> </u>
10	Alkaline Liquid, n.o.s.	13	Ethyl Acetate	45195
1 1	Hydrochloric Acid	21	Sodium Metal	43363
1	Hydrochloric Acid	4	Methyl Alcohol	42740
4	Methyl Alcohol	104	Chlorine	42580
3	Acrylic Acid	5	Formaldehyde Solution	42352
19	Methyl Ethyl Ketone	104	Ferric Chloride Solution	41831
10	Sodium Hydroxide, Liq.	19	Methyl Ethyl Ketone	41831
2	Sulfuric Acid, Spent	19	Methyl Ethyl Ketone	41831
10	Potassium Hydroxide, Liq	19	Methyl Ethyl Ketone	41831
10	Alkaline Liquid, n.o.s.	19	Methyl Ethyl Ketone	41831
1	Hydroiodic Acid	19	Methyl Ethyl Ketone	41831
Ī	Phosphoric Acid	19	Methyl Ethyl Ketone	41831
1	Hydrofluorosilicic Acid	19	Methyl Ethyl Ketone	41831
ī	Hydrochloric Acid	19	Methyl Ethyl Ketone	41831
1	Hydrochloric Acid	13	Ethyl Acetate	41485
l î	Hydroiodic Acid	13	Ethyl Acetate	41485
7	Aniline Oil, Liquid	18	Toluene Diisocyanate	41239
2	Sulfuric Acid, Spent	7	Aniline Oil, Liquid	41151
4	Methyl Alcohol	18	Toluene Diisocyanate	
2	Sulfuric Acid	31	Phenol	40538 40476
7	Aniline Oil, Liquid	107		
13	Vinyl Acetate	21	Maleic Anhydride Sodium Metal	39832
3		5		39637
3	Acetic Acid, Glacial	13	Acetaldehyde	37699
2 2	Sulfuric Acid, Spent Sulfuric Acid		Ethyl Acetate	37596
1		13	Ethyl Acetate	37596
	Hydrofluorosilicic Acid	13	Ethyl Acetate	37558
13	Vinyl Acetate	104	Ferric Chloride Solution	36731
2	Sulfuric Acid, Spent	13	Vinyl Acetate	36663
4	Ethyl Alcohol	104	Chlorine	36161
4	Denatured Alcohol	104	Chlorine	36161
104	Chlorine	106	Ammonium Nitrate Soln.	35982
105	Phosphorus	106	Ammonium Nitrate Soln.	35982
2	Sulfuric Acid	106	Ammonium Nitrate Soln.	35982
4	Isopropanol	104	Chlorine	35792
2 2 5 5 5	Sulfuric Acid, Spent	5	Butyraldehyde	35593
] 2	Sulfuric Acid	5	Butyraldehyde	35593
5	Butyraldehyde	10	Potassium Hydroxide, Liq	35593
5	Butyraldehyde	10	Alkaline Liquid, n.o.s.	35593
	Butyraldehyde	10	Sodium Hydroxide, Liq.	35593
17	Vinyl Chloride	104	Hydrogen Peroxide	34664
104	Hydrogen Peroxide	105	Phosphorus	34664
13	Butyl Acrylate	21	Sodium Metal	34543
3	Acetic Acid, Glacial	26	Acrylonitrile	34460
5	Acetaldehyde	21	Sodium Metal	34309
3	Acrylic Acid	26	Acrylonitrile	33921
L	<u> </u>	<u> </u>		<u> </u>

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ACTH		ACTM		Area
ASTM #1	Chemical Name #1	ASTM	Chemical Name #2	Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
1	Hydrofluorosilicic Acid	4	Alcoholic Beverage	33819
26	Acrylonitrile	104	Ferric Chloride Solution	33068
3	Acrylic Acid	7	Hexamethylene Diamine Sol	31674
3 2	Sulfuric Acid, Spent	27	Dinitrotoluene, Liquid	30516
10	Potassium Hydroxide, Liq	27	Dinitrotoluene, Liquid	30516
5	Butyraldehyde	27	Dinitrotoluene, Liquid	30516
27	Dinitrotoluene, Liquid	105	Phosphorus	30516
2	Sulfuric Acid	27	Dinitrotoluene, Liquid	30516
27	Dinitrotoluene, Liquid	104	Chlorine	30516
10	Alkaline Liquid, n.o.s.	27	Dinitrotoluene, Liquid	30516
5	Acetaldehyde	27		30516
10		27	Dinitrotoluene, Liquid	
21	Sodium Hydroxide, Liq. Sodium Metal		Dinitrotoluene, Liquid	30516
		27	Dinitrotoluene, Liquid	30516
10	Anhydrous Ammonia	17	Methyl Chloride	30012
1	Hydrochloric Acid	13	Butyl Acrylate	29144
1 1	Hydrochloric Acid	107	Acetic Anhydride	29005
7	Hexamethylene Diamine Sol	18	Toluene Diisocyanate	28546
4	Alcoholic Beverage	18	Toluene Diisocyanate	28500
1 1	Phosphoric Acid	18	Toluene Diisocyanate	28464
1	Hydrochloric Acid	18	Toluene Diisocyanate	28464
1	Hydroiodic Acid	18	Toluene Diisocyanate	28464
2 2	Sulfuric Acid, Spent	18	Toluene Diisocyanate	28464
	Sulfuric Acid	18	Toluene Diisocyanate	28464
10	Potassium Hydroxide, Liq	18	Toluene Diisocyanate	28464
10	Alkaline Liquid, n.o.s.	18	Toluene Diisocyanate	28464
[10]	Sodium Hydroxide, Liq.	18	Toluene Diisocyanate	28464
4	Octyl Alcohol	18	Toluene Diisocyanate	28274
5	Formaldehyde Solution	18	Toluene Diisocyanate	27796
4	Butyl Alcohol	104	Chlorine	27423
101	Coal Tar Distillate	104	Hydrogen Peroxide	25922
101	Fuel Oil	104	Hydrogen Peroxide	25890
101	Fuel, Aviation	104	Hydrogen Peroxide	25877
101	Petroleum Naphtha	104	Hydrogen Peroxide	25852
101	Crude Oil Petroleum	104	Hydrogen Peroxide	25555
101	Distillate Fuel Oil	104	Hydrogen Peroxide	25325
101	Flammable Liquid, n.o.s.	104	Hydrogen Peroxide	25325
101	Gasoline	104	Hydrogen Peroxide	25325
101	Combustible Liquid	104	Hydrogen Peroxide	25325
28	LPG-Propylene	104	Hydrogen Peroxide	25325
28	LPG-Butylene	104	Hydrogen Peroxide	25325
28	LPG-Butene	104	Hydrogen Peroxide	25325
28	LPG-Isobutylene	104	Hydrogen Peroxide	25325
4	Alcoholic Beverage	104	Hydrogen Peroxide	25325
101	Compound, Cleaning	104	Hydrogen Peroxide	25325
5	Formaldehyde Solution	104	Hydrogen Peroxide	25325
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Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

	· · · · · · · · · · · · · · · · · · ·			
i i				Area
ASTM		ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
101	D. L. 1. D. 1. J 7			
101	Petroleum Residual Fuel	104	Hydrogen Peroxide	25325
101	Oil	104	Hydrogen Peroxide	25325
101	Petroleum Refined	104	Hydrogen Peroxide	25325
101	Gasoline, nec	104	Hydrogen Peroxide	25325
34	Ethylene Oxide	104	Hydrogen Peroxide	25239
2	Sulfuric Acid	19	Acetone	24587
1 1	Phosphoric Acid	21	Sodium Metal	24513
18	Toluene Diisocyanate	21	Sodium Metal	24513
1	Hydroiodic Acid	21	Sodium Metal	24513
1	Hydrofluorosilicic Acid	4	Methyl Alcohol	23890
4	Alcoholic Beverage	21	Sodium Metal	23732
21	Sodium Metal	101	Pulp Mill Liquid	23728
10	Alkaline Liquid, n.o.s.	21	Sodium Metal	23728
2	Sulfuric Acid, Spent	21	Sodium Metal	23728
4	Octyl Alcohol	21	Sodium Metal	23728
10	Potassium Hydroxide, Liq	21	Sodium Metal	23728
21	Sodium Metal	104	Ferric Chloride Solution	23728
3	Acrylic Acid	10	Alkaline Liquid, n.o.s.	23398
2	Sulfuric Acid, Spent	. 3	Acrylic Acid	23398
3	Acrylic Acid	10	Potassium Hydroxide, Liq	23398
2 3 3 2	Acrylic Acid	10	Sodium Hydroxide, Liq.	23398
	Sulfuric Acid	3	Acrylic Acid	23398
4	Alcoholic Beverage	104	Chlorine	22780
16	Styrene Monomer	21	Sodium Metal	22752
7	Hexamethylene Diamine Sol		Acetic Anhydride	22013
21	Sodium Metal	107	Acetic Anhydride	21936
$\begin{bmatrix} 1\\1 \end{bmatrix}$	Hydroiodic Acid	107	Acetic Anhydride	21936
27	Phosphoric Acid	107	Acetic Anhydride	21936
10	Dinitrotoluene, Liquid	107	Acetic Anhydride	21936
2	Alkaline Liquid, n.o.s.	107	Acetic Anhydride	21936
	Sulfuric Acid, Spent	107	Acetic Anhydride	21936
10	Sodium Hydroxide, Liq.	107	Acetic Anhydride	21936
2	Sulfuric Acid	107	Acetic Anhydride	21936
10	Potassium Hydroxide, Liq	107	Acetic Anhydride	21936
1 2	Hydrochloric Acid	107	Maleic Anhydride	20727
2	Sulfuric Acid	13	Vinyl Acetate	20443
10	Anhydrous Ammonia	104	Chlorine	20017
5	Formaldehyde Solution	21	Sodium Metal	19919
1 1	Hydrochloric Acid	17	Vinyl Chloride	19635
i	Hydrochloric Acid Hydrochloric Acid	10	Sodium Hydroxide, Liq.	19635
1	Hydrochloric Acid	10 10	Potassium Hydroxide, Liq	19635
7	Hexamethylene Diamine Sol	21	Alkaline Liquid, n.o.s. Sodium Metal	19635
5	Formaldehyde Solution	7	Hexamethylene Diamine Sol	19250 19115
2	Sulfuric Acid	5	Formaldehyde Solution	19115
			- Tormarachyae Soracion	13007

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

			,	
				Area
ASTM	Ob == 1 == 3 N #4	ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
4	Octyl Alcohol	104	Chlorine	17168
2	Sulfuric Acid	17	Methyl Chloride	16493
101	Pulp Mill Liquid	104	Hydrogen Peroxide	15900
1	Hydrocyanic Acid	104	Hydrogen Peroxide	15900
17	Methyl Chloride	104	Hydrogen Peroxide	15900
28	Butadiene	104	Hydrogen Peroxide	15900
2	Sulfuric Acid, Spent	13	Ethyl Acrylate	14336
1]	Hydroiodic Acid	14	Glycol Ethers	13832
13	Ethyl Acrylate	104	Ferric Chloride Solution	12212
10	Anhydrous Ammonia	19	Methyl Ethyl Ketone	12055
2	Sulfuric Acid, Spent	5	Butyraldehyde	11096
10	Anhydrous Ammonia	13	Vinyl Acetate	9777
1	Hydroiodic Acid	13	Butyl Acrylate	9509
2	Sulfuric Acid	13	Butyl Acrylate	9509
1	Phosphoric Acid	104	Hydrogen Peroxide	9425
1 2 1 1 2 2	Hydrofluorosilicic Acid	104	Hydrogen Peroxide	9425
1	Hydroiodic Acid	104	Hydrogen Peroxide	9425
2	Sulfuric Acid, Spent	16	Toluene	9174
2	Sulfuric Acid, Spent	13	Methyl Methacrylate	8980
13	Methyl Methacrylate	104	Ferric Chloride Solution	8980
10	Sodium Hydroxide, Liq.	17	Vinyl Chloride	8390
10	Alkaline Liquid, n.o.s.	17	Vinyl Chloride	8390
10	Potassium Hydroxide, Liq	17	Vinyl Chloride	8390
1	Hydrochloric Acid	14	Glycol Ethers	8335
1	Hydrofluorosilicic Acid	3	Acetic Acid, Glacial	7104
1	Hydrochloric Acid		Acetic Acid, Glacial	7104
1 1 3 1	Acetic Acid, Glacial	106	Ammonium Nitrate Soln.	7104
	Phosphoric Acid	3	Acetic Acid, Glacial	7104
1	Hydroiodic Acid	3	Acetic Acid, Glacial	7104
10	Anhydrous Ammonia	13	Ethyl Acetate	6981
4	Methyl Alcohol	104	Ferric Chloride Solution	6283
1	Hydrocyanic Acid	19	Acetone	5024
104	Ferric Chloride Solution	107	Acetic Anhydride	3696
106	Ammonium Nitrate Soln.	107	Acetic Anhydride	3696
2	Sulfuric Acid, Spent	31	Pheno1	3248
31	Phenol	105	Phosphorus	3248
1	Hydrofluorosilicic Acid	31	Phenol	3142
5	Formaldehyde Solution	107	Acetic Anhydride	2910
1	Hydrochloric Acid	4	Butyl Alcohol	2534
5 1 1 2 1	Hydrofluorosilicic Acid	4	Butyl Alcohol	2534
2	Sulfuric Acid, Spent	5	Formaldehyde Solution	2356
1	Phosphoric Acid	5	Formaldehyde Solution	2356
1	Hydrofluorosilicic Acid	5	Formaldehyde Solution	2356
1	Hydroiodic Acid	55555	Formaldehyde Solution	2356
1	Hydrochloric Acid		Formaldehyde Solution	2356

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Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

				,
[·			Area
ASTM	0 1 1 2 • • • • • • • • • • • • • • • • • • •	ASTM		Covered
#1	Chemical Name #1	#2	Chemical Name #2	sq.m.
5	Formaldehyde Solution	10	Dotaccium Uudnovido lia	2256
5	Formaldehyde Solution	10	Potassium Hydroxide, Liq Sodium Hydroxide, Liq.	2356 2356
ĭ	Hydroiodic Acid	7	Aniline Oil, Liquid	2356
5	Formaldehyde Solution	10	Alkaline Liquid, n.o.s.	2356
ĭ	Hydrochloric Acid	7	Aniline Oil, Liquid	2356
ī	Hydrofluorosilicic Acid	7	Aniline Oil, Liquid	2356
1	Phosphoric Acid	7	Aniline Oil, Liquid	2356
7	Aniline Oil, Liquid	104	Ferric Chloride Solution	2356
5	Formaldehyde Solution	104	Ferric Chloride Solution	2356
2	Sulfuric Acid, Spent	16	Xylene	2237
14	Glycol Ethers	104	Ferric Chloride Solution	2051
2	Sulfuric Acid, Spent	14	Glycol Ethers	2051
1 1	Hydrofluorosilicic Acid	14	Glycol Ethers	2051
2	Sulfuric Acid, Spent	16	Styrene Monomer	1758
16	Styrene Monomer	104	Ferric Chloride Solution	1758
1	Hydrofluorosilicic Acid	3	Acrylic Acid	1226
1 1	Hydrochloric Acid	3	Acrylic Acid	1226
10	Sodium Hydroxide, Liq.	107	Maleic Anhydride	1092
10	Alkaline Liquid, n.o.s.	107	Maleic Anhydride	1092
1 1	Phosphoric Acid	107	Maleic Anhydride	1092
1 1	Hydroiodic Acid	107	Maleic Anhydride	1092
27	Dinitrotoluene, Liquid	107	Maleic Anhydride	1092
21	Sodium Metal	107	Maleic Anhydride	1092
10	Potassium Hydroxide, Liq	107	Maleic Anhydride	1092
13	Butyl Acrylate	104	Ferric Chloride Solution	965
2	Sulfuric Acid, Spent	13	Butyl Acrylate	965
7	Hexamethylene Diamine Sol	107	Maleic Anhydride	896
1 1	Hydrofluorosilicic Acid	7	Hexamethylene Diamine Sol	785
18	Toluene Diisocyanate	106	Ammonium Nitrate Soln.	785
18	Toluene Diisocyanate	105	Phosphorus	785
	Hydrofluorosilicic Acid	105	Phosphorus	785
1 1	Hydrochloric Acid	7	Hexamethylene Diamine Sol	
i	Hydrochloric Acid Hydroiodic Acid	101	Coal Tar Distillate	597
	Phosphoric Acid	101 101	Coal Tar Distillate Coal Tar Distillate	597 507
i	Hydrofluorosilicic Acid	101	Coal Tar Distillate	597
1	Phosphoric Acid	101	Fuel Oil	597
1	Hydrofluorosilicic Acid	101	Fuel Oil	565 565
1 1	Hydrochloric Acid	101	Fuel Oil	565
1	Hydroiodic Acid	101	Fuel Oil	565
li	Phosphoric Acid	101	Fuel, Aviation	552
i	Hydroiodic Acid	101	Fuel, Aviation	552
î	Hydrochloric Acid	101	Fuel, Aviation	552
î	Hydrofluorosilicic Acid	101	Fuel, Aviation	552
l î	Hydrochloric Acid	101	Petroleum Naphtha	527
				l

Table J3. Consequence Rankings by Binary Combination (Sort by Area Above the Lethal Limit) (Cont.)

AST M #1	Chemical Name #1	ASTM #2	Chemical Name #2	Area Covered sq.m.
1	Hydrofluorosilicic Acid	101	Petroleum Naphtha	527
1	Phosphoric Acid	101	Petroleum Naphtha	527
1 3	Hydroiodic Acid	101	Petroleum Naphtha	527
3	Acrylic Acid	106	Ammonium Nitrate Soln.	441
1	Hydroiodic Acid	3	Acrylic Acid	441
1	Phosphoric Acid	3	Acrylic Acid	441
1 2 2	Sulfuric Acid, Spent	7	Hexamethylene Diamine Sol	362
2	Sulfuric Acid	7	Hexamethylene Diamine Sol	362
1	Hydrochloric Acid	101	Crude Oil Petroleum	230
1	Hydrofluorosilicic Acid	101	Crude Oil Petroleum	230
1	Phosphoric Acid	101	Crude Oil Petroleum	230
1	Hydroiodic Acid	101	Crude Oil Petroleum	230

Note: The "Area Covered" is the total area around the spill site that is above the lethal limit resulting from mixing of the chemicals. It is a sum of the lethal areas from toxic emissions, fireballs, condensed phase explosions, and unconfined vapor cloud explosions (where applicable).

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Table J4. Consequence Rankings Based on ASTM Groups (Sort by Total Area Above the Lethal Limit)

ASTM	ASTM	Total	% of	Cumul-
#	#	Area	Total	ative %
222212222141222222222 1022214122222222 20132612277134233450 1010	4 101 16 13 2 17 10 5 19 28 7 17 10 3 26 104 104 104 104 104 104 104 104 104 104	5591603 2918261 2849900 2778061 2696272 2120946 1732982 1713810 1345527 1168280 1092263 1067651 1011634 948752 744028 688594 673108 672578 665625 652839 606315 603719 601514 577661 541223 528626 485253 480118 460350 439807 431011 416090 406298 353027 258190 253995 251680 251327 249034 232897 217664 202262 202081 197786	11.469 5.985 5.845 5.845 5.698 5.530 4.350 3.554 3.554 3.554 3.554 3.554 3.515 2.760 2.240 2.190 2.190 2.190 1.381 1.379 1.385 1.385 1.1084 0.995 0.9854 0.995 0.9854 0.530 0.5515 0.515 0.416 0.416 0.416 0.416	11.469 17.454 23.299 28.997 34.527 38.877 42.432 45.947 48.706 51.103 53.343 55.533 57.608 59.554 61.080 62.492 63.872 65.252 66.617 67.956 69.200 70.438 71.672 72.856 75.051 76.046 77.031 77.975 78.877 79.761 80.614 81.448 82.725 83.776 84.292 84.808 85.318 85.796 84.292 84.808 85.796 87.487 87.893

Table J4. Consequence Rankings Based on ASTM Groups (Sort by Area Above the Lethal Limit) (Cont.)

Table J4. Consequence Rankings Based on ASTM Groups (Sort by Area Above the Lethal Limit) (Cont.)

ASTM	ASTM	Total	% of	Cumul-
#	#	Area	Total	ative %
7 10 34 5 7 10 3 5 21 10 7 10 105 104 104 1 27 21 18 1 21 27 21 16 10 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	21 18 104 31 18 107 26 27 106 24 21 106 105 3 105 27 21 107 107 107 107 107 107 106 107 105 31 106 105	45464 42696 37902 37875 34893 34542 34191 30516 29855 29467 27237 23728 17991 17332 15875 15258 15258 15257 12109 11864 11514 11574	0.093 0.088 0.078 0.072 0.071 0.070 0.063 0.061 0.060 0.056 0.049 0.037 0.037 0.037 0.033 0.031 0.025 0.024 0.024 0.024 0.024 0.024 0.023 0.011 0.010 0.008 0.003 0.001	98.845 98.932 99.010 99.088 99.159 99.300 99.363 99.424 99.540 99.589 99.626 99.663 99.663 99.731 99.762 99.793 99.868 99.819 99.843 99.915 99.938 99.938 99.938 99.938 99.938 99.938 99.938 99.938 99.938 99.938 99.999 99.999

Note: The "Total Area" is a sum of the lethal areas around the spill site for all combinations of specific chemicals within the ASTM groups.

Table J5. Risk Rankings by Binary Combination

ASTM	Chemical Name #1	Car Moves/yr	ASTM	Chemical Name #2	Car Moves/yr #2	Area Covered sq.m.	Risk *	Normal- ized Risk	% of Total Risk	Cumul- ative
1	Hydrochloric Acid	10973		Sulfuric Acid	46265	490874	2.49E+14	611024	10.394	10.39
2	Sulfuric Acid	46265	4	Methyl Alcohol	20739	189597	1.82E+14	446049	7.588	17.98
2	Oleum	3243	10	Sodium Hydroxide, Liq.	75841	693186	1.48E+14	362799	6.171	24.15
2	Sulfuric Acid	46265	4	Denatured Alcohol	12087	2525Ø7	1.41E+14	346222	5.889	30.04
2	Sulfuric Acid	46265	17	Vinyl Chloride	20081	95033	8.83E+13	216482	3.682	33.72
2	Oleum	3243	4	Methyl Alcohol	20739	764509	5.14E+13	126075	2.145	35.87
16	Sodium Hydroxide, Liq.	75641	16	Styrene Monomer	10859	54985	4.51E+13	110699	1.883	37.75
10	Anhydrous Ammonia	42526	104	Chlorine	46686	20017	3.97E+13	97443	1.658	39.41
1	Hydrofluorosilicic Acid	930	2	Sulfuric Acid	48265	907920	3.91E+13	95784	1.629	41.04
2	Sulfuric Acid	46265	17	Carbon Tetrachloride	1343	544281	3.38E+13	82921	1.411	42.45
2	0 leum	3243	4	Denatured Alcohol	12087	855693	3.35E+13	82242	1.399	43.85
2	Sulfuric Acid	46265	16	Benzene	5Ø67	139348	3.27E+13	80097	1.362	45.21
3	Acetic Acid, Glacial	2953	10	Sodium Hydroxide, Liq.	75641	145932	3.26E+13	79925	1.360	46.57
1	Hydrocyanic Acid	1226	104	Chlorine	46686	585487	3.24E+13	79382	1.350	47.92
10	Sodium Hydroxide, Liq.	75841	17	Carbon Tetrachloride	1343	318Ø86	3.23E+13	7923Ø	1.348	49.27
2	Oleum	3243	161	Fuel Oil	39248	251892	3.21E+13	78612	1.337	50.61
2	Sulfuric Acid	46265	26	Acrylonitrile	4811	126656	2.82E+13	69123	1.176	51.78
10	Sodium Hydroxide, Liq.	75641	13	Vinyl Acetate	4431	84060	2.82E+13	69081	1.175	52.96
2	Sulfuric Acid	46265	16	Styrene Monomer	10859	54965	2.76E+13	677Ø8	1.152	54.11
1	Hydrochloric Acid	10973	2	Sulfuric Acid, Spent	4809	49Ø874	2.59E+13	63513	1.080	55.19
10	Anhydrous Ammonia	42526	17	Methyl Chloride	1168	519315	2.58E+13	63247	1.076	56.26
10	Sodium Hydroxide, Liq.	75641	26	Acrylonitrile	4811	69378	2.52E+13	61905	1.053	57.32
2	Oleum	3243	28	LPG	21857	329867	2.34E+13	57331	0.975	58.29
2	Sulfuric Acid	46265	17	Chloroprene, Inhibited	889	565782	2.33E+13	57058	0.971	59.26
2	0 i eum '	3243	16	Styrene Monomer	1Ø859	655009	2.31E+13	56558	Ø.962	60.23
4	Denatured Alcohol	12087	164	Chlorine	46686	36161	2.84E+13	50033	Ø.851	61.98
2	Sulfuric Acid	46265	3	Acetic Acid, Glacial	2953	145932	1.99E+13	48885	0.832	61.91
10	Sodium Hydroxide, Liq.	75641	19	Acetone	2971	86008	1.93E+13	47392	Ø.8Ø6	62.71
2	Sulfuric Acid	46265	16	Xy lene	5415	76628	1.92E+13	47071	0.801	63.52
2	Nitric Acid, Fuming	441	10	Sodium Hydroxide, Liq.	75641	552135	1.84E+13	45180	Ø.768	64.28
1	Hydrochloric Acid	10973	4	Denatured Alcohol	12087	128629	1.71E+13	41831	0.712	
1	Phosphoric Acid	28008	16	Styrene Monomer	10859	54965	1.67E+13	40989	0.697	65.69
2	0 i eum	3243	17	Vinyl Chloride	20081	251327	1.64E+13	40131	0.683	66.37
1	Hydrochloric Acid	10973	10	Sodium Hydroxide, Liq.	75641	19635	1.63E+13	3996Ø	0.680	67.05
1	Hydrochloric Acid	10973	104	Hydrogen Peroxide	1646	767526	1.39E+13	33991	0.578	67.63
10	Sodium Hydroxide, Liq.	75641	105	Phosphorus	3533	51427	1.37E+13	33698	0.573	68.21
10	Sodium Hydroxide, Liq.	75641	17	Chloroprene, Inhibited	889	202182	1.36E+13	33336	0.567	68.77
1	Phosphoric Acid	28008	26	Acrylonitrile	4811	9524Ø	1.28E+13	31466	Ø.535	69.31
2	01eum	3243	34	Ethylene Oxide	6535	603186	1.28E+13			69.84
10	Sodium Hydroxide, Liq.	75641	17	Vinyl Chloride	20081	8390	1.27E+13	31247		
2	Sulfuric Acid	48285	. 18	Toluene	2084	129751	1.25E+13			
2	Oleum	3243	16	Xy I ene	5415	679Ø29	1.19E+13	29238	0.497	
2	Bleum	3243	31	Phenoi	5974	605177	1.17E+13	28748	0.489	71.88
2	Sulfuric Acid, Spent	4809	4	Methyl Alcohol	20739	113707	1.13E+13	27806	0.473	72.35
2	Sulfuric Acid	46265	31	Phenol	5974	49476	1.12E+13	27430	0.467	72.82

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car			Car	Area		Normal-	% of	Cumul-
ASTM		Moves/yr	ASTM		Moves/yr	Covered .		i zed	Total	ative
#	Chemical Name #1	#1	ŧ	Chemical Name #2	‡ 2	sq.≡.	Risk *	Risk	Risk	x
2	Dieum	3243	101	Petroleum Naphtha	13541	251854	1.11E+13	27118	0.461	73.28
2	Gleum	3243	16	Benzene	5Ø87	667136	1.10E+13	2688Ø	Ø.457	73.74
1	Hydroiodic Acid	897	2	Sulfuric Acid	46265	254469	1.06E+13	25893	0.440	74.18
1	Hydrochloric Acid	10973	2	Ûleu∎	3243	289812	1.03E+13	25287	Ø.43Ø	74.61
2	Sulfuric Acid	46265	4	Ethyl Alcohol	867	252507	1.01E+13	24834	Ø.422	75.03
2	Dieum	3243	26	Acrylonitrile	4811	63795Ø	9.95E+12	24405	0.415	75.45
2	Sulfuric Acid	46265	164	Hydrogen Peroxide	1646	129591	9.87E+12	24197	0.412	75.86
1	Hydrochloric Acid	10973	4	Methyl Alcohoi	20739	42746	9.73E+12	23848	0.406	76.26
2	Sulfuric Acid	46265	13	Methyl Methacrylate	2927	71821	9.73E+12	23847	0.406	76.67
2	Oleum	3243	34	Propylene Oxide	4913	603186	9.61E+12	23564	0.401	77.87
2	0 leum	3243	13	Vinyl Acetate	4431	623629	8.96E+12	21973	0.374	77.44
2	Oleum	3243	101	Petroleum, Refined	10858	251327	8.69E+12	21300	Ø.362	77.81
2	Oleum	3243	28	Butadiene	8Ø3Ø	329867	8.59E+12	21063	0.358	78.17
1	Hydrochloric Acid	10973	16	Styrene Monomer	1Ø859	71458	8.51E+12	20877	Ø.355	78.52
2	Cleum	3243	101	Crude Oil Petroleum	9995	251557	8.15E+12	19993	0.340	78.86
2	Sulfuric Acid	46265	4	Isopropanol	2848	60134	7.92E+12	19428	0.330	79.19
1	Phosphoric Acid	28008	19	Acetone	2971	86008	7.16E+12	17548	Ø.299	79.49
2	Sulfuric Acid	46265	21	Sodium Metal	855	176715	6.99E+12	17140	Ø.292	79.78
2	Nitric Acid, Fuming	441	4	Methyl Alcohol	20739	762448	6.97E+12	17Ø98	8.291	80.97
2	Cleum	3243	105	Phosphorus	3533	693186	6.91E+12	16945	Ø.288	80.36
2	Oleum	3243	5	Acetaldehyde	3520	6Ø3186	6.89E+12	16883	0.287	80.65
2	0 leum	3243	4	Isopropanol	2848	739283	6.83E+12	16742	0.285	80.93
2	Sulfuric Acid, Spent	4809	4	Denatured Alcohol	12987	111328	8.47E+12	15867	9.270	81.20
2	Oleum	3243	101	Petroleus Residual Fuel	7907	251327	6.44E+12	15802	0.269	81.47
2	Cleum	3243	101	Gasoline	7878	251327	6.26E+12	15340	0.261	81.73
2	Oleum	3243	28	LPG-Propylene	5707	329867	6.11E+12	14969	Ø.255	81.99
2	01eum	3243	13	Methyl Methacrylate	2927	6373Ø8	6.05E+12	14833	Ø.252	82.24
2	0 leum	3243	7	Hexamethylene Diamine So	5580	330229	5.98E+12	14652	0.249	82.49
2	Sulfuric Acid	46265	19	Methyl Ethyl Ketone	1318	97286	5.93E+12	14545	Ø.247	82.74
2	0 i eum	3243	3	Acetic Acid, Glacial	2953	617171	5.91E+12	14492	0.247	82.98
2	Sulfuric Acid	46265	13	Ethyl Acrylate	1281	99610	5.9ØE+12	14475	0.246	83.23
2	0 leum	3243	19	Acetone	2971	608923	5.87E+12	14385	0.245	83.47
10	Sodium Hydroxide, Liq.	75641	18	Toluene Diisocyanate	2610	28464	5.62E+12	13779	0.234	83.71
2	Oleum	3243	101	Distillate Fuel Oil	6549	251327	5.34E+12	13088	Ø.223	83.93
4	Methyl Alcohol	2630	104	Chlorine	48886	42580	5.23E+12	12819	Ø.218	84.15
2	0 leum	3243	10	Potassium Hydroxide, Liq	2622	603186	5.13E+12	12576	0.214	84.36
2	0 leu s	3243	18	Toluene Diisocyanate	2619	6Ø3376	5.11E+12	12522	0.213	84.58
1	Phosphoric Acid	28008	17	Chloroprene, Inhibited	889	202182	5.03E+12	12343	0.210	84.79
2	O leum	3243	101	Fuel, Aviation	6157		5.03E+12			
1	Hydrochloric Acid	10973	26	Acrylonitrile	4811		5.Ø3E+12		Ø.21Ø	
2	Sulfuric Acid	46265	7	Aniline Oil, Liquid	1Ø88		4.92E+12		0.205	
4	Isopropano l	2848	164	Chlorine	46686		4.76E+12		0.198	
. 2	Cleum	3243	16	Toluene	2084		4.53E+12			
2	Nitric Acid, Fuming	441	4	Denatured Alcohol	12987		4.45E+12			
1	Hydrochloric Acid	10973	17	Vinyl Chloride	20081		4.33E+12			

Table J5. Risk Rankings by Binary Combination (Cont.)

ASTM #	Chemical Name #1	Car Moves/yr #1	ASTM #	Chemical Name #2	Car Moves/yr #2	Area Covered sq.m.	Risk *	Normal- ized Risk	% of Total Risk	Cumul- ative
2	Cleus	3243	28	LPG-Butene	3945	329867	4.22E+12	10348	0.176	86.34
2	Sulfuric Acid	46265	13	Vinyl Acetate	4431	20443	4.19E+12	10276	0.175	86.51
10	Sodium Hydroxide, Liq.	75841	19	Methy! Ethy! Ketone	1318	41831	4.17E+12	10225	0.174	86.69
1	Hydrochloric Acid	10973	13	Vinyl Acetate	4431	84Ø8Ø	4.89E+12	10021	Ø.17Ø	86.86
2	Oleum	3243	13	Butyl Acrylate	1957	612695	3.89E+12	9534	Ø.162	87.02
2	0 i eum	3243	106	Ammonium Nitrate Soln.	1869	639168	3.86E+12	9453	Ø.161	87.18
26	Acrylonitrile	4811	31	Phenol	5974	136246	3.74E+12	9179	Ø.156	87.34
4	Denatured Alcohol	12087	104	Hydrogen Peroxide	1848	187691	3.73E+12	9156	Ø.156	87.49
2	Sulfuric Acid	46265	4	Butyl Alcohol	1052	76591	3.73E+12	9140	Ø.155	87.65
16	Styrene Monomer	10859	184	Hydrogen Peroxide	1648	206653	3.69E+12	9057	Ø.154	87.80
10	Sodium Hydroxide, Liq.	75641	13	Ethyl Acetate	1066	45195	3.64E+12	8935	Ø.152	87.95
1	Hydrofluorosilicic Acid	93Ø	10	Sodium Hydroxide, Liq.	75841	50265	3.54E+12	8670	Ø.147	88.10
2	Sulfuric Acid	46265	18	Toluene Diisocyanate	2618	28484	3.44E+12	8428	Ø.143	88.25
2.	Sulfuric Acid	48265	19	Acetone	2971	24587	3.38E+12	8286	0.141	88.39
2	Oleum	3243	104	Hydrogen Peroxide	1846	828425	3.35E+12	8225	0.140	88.53
2	Oleum	3243	5	Formaldehyde Solution	3Ø58	330024	3.27E+12	8025	Ø.137	88.66
2	Oleum	3243	4	Octyl Alcohol	1671	603186	3.27E+12	8Ø15	Ø.136	88.80
104	Chlorine	46686	198	Ammonium Nitrate Soln.	1860	35982	3.12E+12	7661	Ø.13Ø	88.93
2	Sulfuric Acid	46265	108	Ammonium Nitrate Soln.	1866	35982	3.16E+12	7592	Ø.129	89.06
2	0 leum	3243	107	Acetic Anhydride	1533	612556	3.05E+12	7467	Ø.127	89.19
7	Hexamethylene Diamine So	5580	26	Acrylonitrile	4811	111730	3.00E+12	7354	Ø.125	89.31
2	Nitric Acid, Fuming	441	16	Styrene Monomer	10859	603958	2.89E+12	7092	Ø.121	89.43
1	Hydrochloric Acid	10973	19	Acetone	2971	86008	2.8ØE+12	6875	Ø.117	89.55
2	Sulfuric Acid	46265	4	Alcoholic Beverage	1276	47119	2.78E+12	6820	0.118	89.66
2	0 leum	3243	19	Methyl Ethyl Ketone	1318	639990	2.74E+12	6797	Ø.114	89.78
2	Oleum	3243	4	Alcoholic Beverage	1276	650305	2.69E+12	6598	Ø.112	89.89
2	Sulfuric Acid	46265	5	Formaldehyde Solution	3058	19007	2.69E+12	6593	0.112	90.00
2	Oleum	3243	13	Ethyl Acrylate	1281	839964	2.66E+12	6519	0.111	90.11
2	Oleum	3243	17	Carbon Tetrachloride	1343	603186	2.63E+12	6441	9.119	90.22
2	Oleum	3243	101	Combustible Liquid	3139	251327	2.56E+12	8273	Ø.107	90.33
2	Sulfuric Acid, Spent	4809	26	Acrylonitrile	4811	110162	2.55E+12	6249	0.106	98.44
10				Acetic Anhydride	1533		2.54E+12	6237	0.106	90.54
2	Sulfuric Acid, Spent	4809	16	Benzene	5067		2.51E+12	6160	0.105	
10	Anhydrous Ammonia	42526	24	Motor Fuel Antiknock Comp		58934	2.51E+12	6145	0.105	90.75
1	Phosphoric Acid	28008	5	Butyraldehyde	870	102698	2.50E+12	6136	0.104	90.86
1	Hydrochloric Acid	10973	2	Nitric Acid, Fuming	441	508938	2.46E+12	6039	0.103	90.96
2	Cleum	3243	4	Ethyl Alcohol	867	855693	2.41E+12	5899	0.100	91.06
2	Nitric Acid, Fuming	441	101	Fuel Cil	39248	138010	2.39E+12	5857	Ø.100	91.16
5	Butyraldehyde	870	10	Sodium Hydroxide, Liq.	75641	35593	2.34E+12	5743	0.098	91.26
2	Oleum	3243	4	Butyl Alcohol	1052	683162	2.33E+12	5715	0.097	91.35
1	Hydrochloric Acid	10973	13	Methyl Methacrylate	2927	71821	2.31E+12	5858 5640	0.096	
2	Oleum Oleum	3243	107	Maleic Anhydride	1168	603186	2.28E+12	5602	Ø. Ø95	91.55
2	Oleum	3243	13	Ethyl Acetate	1066 1088	629531 6Ø6638	2.18E+12	5336 E049	Ø. Ø91	91.64
1	Hydrofluorosilicic 'Acid	3243 93ø	7	Aniline Oil, Liquid Oleum	3243	706858	2.14E+12 2.13E+12	5248 5227	Ø.Ø89 Ø.Ø89	

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car			Car	Area		Normal-	% of	Cumul-
ASTM		Moves/yr	ASTM		Moves/yr	Covered		i zed	Total	ative
#	Chemical Name #1	‡ 1	#	Chemical Name #2	#2	sq.∎.	Risk *	Risk	Risk	*
1	Hydroiodic Acid	897	4	Methyl Alcohol	20739	113740	2.12E+12	5188	0.088	91.90
2	0 leum	3243	3	Acrylic Acid	1031	626584	2.10E+12	5137	Ø. Ø87	91.99
1	Phosphoric Acid	28008	18	Toluene Diisocyanate	2610	28484	2.Ø8E+12	5102	0.087	92.08
2	Sulfuric Acid, Spent	4809	3	Acetic Acid, Glacial	2953	145932	2.Ø7E+12	5Ø81	0.086	92.16
16	Xy I ene	5415	184	Hydrogen Peroxide	1646	23Ø673	2.Ø6E+12	5041	0.086	92.25
2	Sulfuric Acid	46265	14	Glycol Ethers	982	47985	2.00E+12	4916	Ø.Ø84	92.33
10	Sodium Hydroxide, Liq.	75641	27	Dinitrotoluene, Liquid	888	3Ø518	2.00E+12	4901	0.083	92.42
1	Hydrochloric Acid	10973	105	Phosphorus	3533	51538	2.88E+12	4899	0.083	92.50
1	Hydrochloric Acid	10973	17	Chloroprene, Inhibited	889	202182	1.97E+12	4836	0.082	92.58
2	0 leum	3243	24	Motor Fuel Antiknock Comp	1000	603186	1.96E+12	4796	0.082	92.66
4	Denatured Alcohol	12Ø87	18	Toluene Diisocyanate	2619	61165	1.93E+12	4731	Ø.080	92.74
2	Oleum	3243	10	Alkaline Liquid, n.o.s.	975	603186	1.91E+12	4676	9.989	92.82
2	0 leum	3243	14	Glycol Ethers	902	638605	1.87E+12	4580	0.078	92.90
2	Sulfuric Acid	46265	13	Ethyl Acetate	1968	37598	1.85E+12	4548	0.077	92.98
16	Anhydrous Ammonia	42526	13	Vinyl Acetate	4431	9777	1.84E+12	4517	Ø.Ø77	93.85
2	0 leum	3243	101	Flammable Liquid	2256	251327	1.84E+12	4509	0.077	93.13
3	Acrylic Acid	1031	10	Sodium Hydroxide, Liq.	75641	23398	1.82E+12	4474	Ø.076	93.21
16	Benzene	5Ø87	104	Hydrogen Peroxide	1646	218789	1.82E+12	4474	0.076	93.28
2	Dieum	3243	5	Butyraidehyde	870	638779	1.8ØE+12	4419	0.075	93.36
2	0 leum	3243	27	Dinitrotoluene, Liquid	866	6337 6 2	1.78E+12	4364	0.074	93.43
101	Fuel Oil	39248	184	Hydrogen Peroxide	1646	2589#	1.67E+12	4101	0.070	93.50
1	Hydrochloric Acid	10973	4	Isopropano!	2848	53105	1.66E+12		0.069	
10	Potassium Hydroxide, Liq		16	Styrene Monomer	10859	54965	1.56E+12		9.065	93.64
2	Sulfuric Acid	46265	107	Acetic Anhydride	1533	21936	1.56E+12		0.065	
1	Phosphoric Acid	28ØØ8	19	Methyl Ethyl Ketone	1318	41831	1.54E+12		0.084	
31	Pheno I	- 5974	194	Hydrogen Peroxide	1646	156821	1.54E+12		0.064	
26	Acrylonitrile	4811	194	Hydrogen Peroxide	1646	189594	1.50E+12		0.063	
2	Nitric Acid, Fuming	441	16	Xy lene	5415		1.5ØE+12		0.083	
10	Anhydrous Ammonia	42526	17	Methyl Chloride	1168	30012	1.49E+12		0.062	
2	Nitric Acid, Fuming	441	16	Benzene	5067	660564	1.48E+12		0.062	
4	Ethyl Alcohol	867	164	Chlorine	46686	36161			0.061	
2	Nitric Acid, Fuming	441	31	Pheno I	5974	•	1.46E+12		0.061	
1	Phosphoric Acid	28008	24	Motor Fuel Antiknock Comp		52117				
2	Sulfuric Acid	46265	5	Butyraldehyde	879		1.43E+12		0.080	
1	Hydrochloric Acid	10973	13	Ethyl Acrylate	1281		1.40E+12			
2	Nitric Acid, Fuming	441	7	Hexamethylene Diamine So					0.057	
4	Alcoholic Beverage	1276	104	Chlorine	46686	22780	1.36E+12			
4	Butyl Alcohol	1052	164	Chlorine	46686		1.35E+12			
2	Oleum	3243	28	LPG-Isobuty lene	1258				0.056	
7	Hexamethylene Diamine So		184	Hydrogen Peroxide	1848		1.34E+12			
4	Octyl Alcohol	1671	104	Chlorine	46686					
3	Acetic Acid, Glacial	2953	5	Formaldehyde Solution	3058		1.34E+12			
2	Nitric Acid, Fuming	441	28	LPG	21857					
1	Hydrofluorosilicic Acid	930	4	Denatured Alcohol	12087					
2	Nitric Acid, Fuming	441	26	Acrylonitrile	4811	613288	1.3ØE+12	3190	0.054	94.94

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car			Car	Area		Normal-	% of	Cunu I -
ASTM		Moves/yr	ASTM		Moves/yr	Covered		i zed	Total	ative
	Chemical Name #1	#1	#	Chemical Name #2	#2 ————	sq.∎.	Risk *	Risk	Risk	<u>*</u>
13	Vinyl Acetate	4431	184	Hydrogen Peroxide	1646	175273	1.28E+12	3134	0.053	94.99
27	Dinitrotoluene, Liquid	866	104	Chlorine	46686	30516	1.23E+12	3Ø25	0.051	95.04
2	Sulfuric Acid, Spent	4809	19	Acetone	2971	88008	1.23E+12	3013	0.051	95.09
1	Hydrochloric Acid	10973	4	Ethyl Alcohol	887	128629	1.22E+12	3001	0.051	95.15
2	Sulfuric Acid	46265	27	Dinitrotoluene, Liquid	866	- 30516	1.22E+12	2998	0.051	95.20
2	Nitric Acid, Fuming	441	17	Vinyl Chloride	20081	137445	1.22E+12	2984	Ø.Ø51	95.25
2	Oleum	3243	17	Chloroprene, Inhibited	889	414985	1.20E+12	2934	0.050	95.30
3	Acetic Acid, Glacial	2953	18	Toluene Diisocyanate	2619	153739	1.18E+12	29ø5	0.049	95.35
,2	Oleum	3243	101	Gasoline, nec	1416	251327	1.15E+12	2818	0.048	95.39
2	Gleum	3243	101	Oil	1407	251327	1.15E+12	2812	0.048	95.44
17	Vinyl Chloride	20081	184	Hydrogen Peroxide	1646	34664	1.15E+12	2809	0.048	95.49
3	Acetic Acid, Glacial	2953	10	Potassium Hydroxide, Liq		145932	1.13E+12	2770	0.047	
20	Carbon Disulfide	1013	34	Propylene Oxide	4913	225409	1.12E+12	2751	0.047	95.58
10	Potassium Hydroxide, Liq	2622	17	Carbon Tetrachloride	1343	318086	1.12E+12	2746	0.047	95.63
2	Sulfuric Acid	46265	3	Acrylic Acid	1931	23398	1.12E+12	2737	0.047	95.68
20	Carbon Disulfide	1013	184	Hydrogen Peroxide	1646	668480	1.11E+12	2733	0.046	95.72
7	Hexamethylene Diamine So		17	Carbon Tetrachloride	1343	148440	1.11E+12	2728	Ø. 9 46	95.77
3	Acetic Acid, Glacial	2953	7	Hexamethylene Diamine Sol		64249	1.06E+12	2596	0.044	95.81
2	Oleum	3243	101	Compound, Cleaning	1281	251327	1.04E+12	2560	0.844	95.86
1	Hydrofluorosilicic Acid	930	16	Styrene Monomer	1Ø859	102088	1.Ø3E+12	2528	0.043	95.90
2	Öleum	3243	101	Coal Tar Distillate	1259	251924	1.03E+12	2522	0.043	95.94
7	Hexamethylene Diamine So		17	Chloroprene, Inhibited	889	202203	1.00E+12	2459	0.042	95.99
1	Hydrochloric Acid	10973	5	Butyraldehyde	878	103483	9.88E+11	. 2422	8.841	96.83
10	Potassium Hydroxide, Liq	2622	13	Vinyl Acetate	4431	84969	9.77E+11	2395	0.041	96.07
2	Sulfuric Acid, Spent	4869	17	Carbon Tetrachloride	1343	148440	9.59E+11	2351	0.040	96.11
2	Cleum	3243	17	Methyl Chloride	1168	251327	9.52E+11	2334	0.040	96.15
2	Oleum Pharabasia Asid	3243	101	Pulp Mill Liquid	1159	251327	9.45E+11	2316	0.039	96.19
1	Phosphoric Acid	28008	107	Acetic Anhydride	1533	21936	9.42E+11	2309	0.039	96.23
2	Nitric Acid, Fuming	441	4	Isopropanol	2848	733888	9.22E+11	2260	Ø.Ø38	96.26
4	Isopropano! Oleum	2848	104	Hydrogen Peroxide	1646	195329	9.16E+11	2245	0.038	96.30
2		3243	21	Sodium Metal	855	329867	9.15E+11	2243	Ø.Ø38	96.34
13	Methyl Methacrylate	2927		Hydrogen Peroxide		188952	9.10E+11		0.038	
2	Sulfuric Acid	46265	17	Methyl Chloride	1168		8.91E+11	2185		96.42
2	Nitric Acid, Funing	441	5	Acetaldehyde	3520		8.79E+11	2156	0.037	
10	Potassium Hydroxide, Liq	2622	26	Acrylonitrile	4811	69378	8.75E+11	2146	0.037	
2	Sulfuric Acid, Spent	4809	17	Chloroprene, Inhibited	889		8.64E+11	2119	Ø.Ø36	96.53
2	Sulfuric Acid	46265	13	Butyl Acrylate	1957	9509	8.61E+11	2111	0.036	96.56
2	Nitric Acid, Fuming	441	195	Phosphorus	3533	552135	8.60E+11	2109	Ø. Ø38	98.60
4	Denatured Alcohol	12087	5	Butyraldehyde Petroleum Naphtha	87Ø	79420	8.35E+11	2048	Ø. Ø35	96.63
2	Nitric Acid, Fuming	441	191	•	13541		8.24E+11	2020	0.034	96.67
3	Sulfuric Acid, Spent Acetic Acid, Glacial	4809	104	Isopropanol	2848 1848	60134 168815	8.24E+11 8.21E+11	2019	Ø. Ø34	96.70 06.72
1	Hydrochloric Acid	2953	104 18	Hydrogen Peroxide Toluene Diisocyanate	2610	28464	8.15E+11	2012 1999	0.034 0.034	96.73 96.77
1	Hydroiodic Acid	10973 897	105	Phosphorus	3533		8.10E+11	1999	Ø.034	96.80
2	Nitric Acid, Fuming	441		Acetone	2971		7.91E+11	1939	Ø.033	
		771	13	veroue	2911		1.015-11	1303		30.07

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car	·		Car	Area		Normal-	% of	Cumul-
ASTM	Charical Name #1	Moves/yr	_	Chant - 1 No 10	Moves/yr		D: 1 .	ized	Total	ative
#	Chemical Name #1	#1	<u> </u>	Chemical Name #2	#2 ———	sq.#.	Risk *	Risk	Risk	×
19	Acetone	2971	184	Hydrogen Peroxide	1646	160567	7.85E+11	1925	Ø.Ø33	98.87
2	Sulfuric Acid, Spent	4809	13	Vinyl Acetate	4431	36663	7.81E+11	1916	0.033	98.90
16	Toluene	2984	194	Hydrogen Peroxide	1646	221749	7.61E+11	1865	0.032	96.93
2	Nitric Acid, Fuming	441	13	Methyl Methacrylate	2927	586257	7.57E+11	1855	0.032	96.96
2	Nitric Acid, Fuming	441	5	Formaldehyde Solution	3058	552292	7.45E+11	1826	0.031	97.00
20	Carbon Disulfide	1013	104	Ferric Chloride Solution	1883	388772	7.42E+11	1818	0.031	97.03
1	Hydroiodic Acid	897	2	Oleum	3243	251327	7.31E+11	1793	0.030	97.06
4	Methyl Alcohol	2630	104	Hydrogen Peroxide	1646	167854	7.27E+11	1782	0.030	97.09
3	Acetic Acid, Glacial	2953	107	Acetic Anhydride	1533	158610	7.18E+11	1761	0.030	97.12
10	Anhydrous Ammonia	42526	19	Methyl Ethyl Ketone	1318	12055	6.76E+11	1657	0.028	97.15
10	Potassium Hydroxide, Liq	2622	19	Acetone	2971	86008	6.7ØE+11	1643	0.028	97.17
18	Toluene Diisocyanate	2610	184	Hydrogen Peroxide	1646	155020	6.66E+11	1633	Ø. Ø28	97.26
4	Denatured Alcohol	12087	21	Sodium Metal	855	63628	6.58E+11	1612	0.027	97.23
2	Nitric Acid, Fuming	441	191	Petroleum, Refined	19658	137445	6.46E+11	1584	0.027	97.26
1	Hydrochloric Acid	10973	4	Alcoholic Beverage	1276	45800	6.38E+11	1565	0.027	97.28
2	Nitric Acid, Fuming	. 441	10	Potassium Hydroxide, Liq	2622	552135	6.38E+11	1565	0.027	97.31
2	Nitric Acid, Fuming	441	18	Toluene Diisocyanate	2610	552325	6.36E+11	1559	0.027	97.33
1	Hydrochloric Acid	10973	13	Butyl Acrylate	1957	29144	6.26E+11	1535	0.026	97.36
1	Phosphoric Acid	28008	101	Fuel Oil	39248	565	6.21E+11	1523	0.026	97.39
7	Aniline Oil, Liquid	1988	26	Acrylonitrile	4811	116697	6.11E+11	1498	0.025	97.41
2	Nitric Acid, Fuming	441	191	Crude Oil Petroleum	9995	137675	8.07E+11		Ø.Ø25	97.44
1	Hydrochloric Acid	10973	19	Methy! Ethy! Ketone	1318	41831	6.05E+11	1483	0.025	
21	Sodium Metal	855	26	Acrylonitrile	4811	145850	6.00E+11	1471	0.025	97.49
3	Acetic Acid, Glacial	2953	17	Chloroprene, Inhibited	889	226871	5.96E+11		0.025	
1	Phosphoric Acid	28008	3	Acetic Acid, Glacial	2953	7184	5.88E+11	1441	6.825	97.54
1	Phosphoric Acid	28008	21	Sodium Metal	855	24513	5.87E+11	1439	0.024	97.56
10	Alkaline Liquid, n.o.s.	975	18	Styrene Monomer	10859	54965	5.82E+11		0.024	97.59
101	Petroleum Naphtha	13541	194	Hydrogen Peroxide	1646	25852	5.76E+11	1413	0.024	97.61
1	Hydrochloric Acid	10973	24	Motor Fuel Antiknock Com	p 1000	52117	5.72E+11	1402	0.024	97.63
2	Nitric Acid, Fuming	441	16	To luene	2084		5.69E+11	1395	0.024	97.66
1	Hydrochloric Acid	10973	10	Potassium Hydroxide, Liq		19635	5:65E+11		Ø.Ø24	
5		3058	19	eserial interest and	75641	2356	5.45E+11	1336	0.023	97.70
1.	•	897	16	Styrene Monomer	10859		5.35E+11	1313	0.022	97.73
13	Butyl Acrylate	1957	194	Hydrogen Peroxide	1646		5.29E+11	- 1298		97.75
5	Butyraidehyde	870	7	Hexamethylene Diamine So		107035	5.2ØE+11	1274	Ø.Ø22	97.77
3	Acetic Acid, Glacial	2953	21	Sodium Metal	855		5.14E+11	1261	Ø. Ø21	,97.79
4	Isopropanol	2848	18	Toluene Diisocyanate	2619	68805	5.11E+11	1254	0.021	97.81
3	Acetic Acid, Glacial	2953	5	Butyraidehyde	879	194662	5.00E+11	1226	0.021	97.83
3	Acetic Acid, Glacial	2953	26	Acrylonitrile	4811		4.9ØE+11	1200	0.020	97.85
5	Acetaldehyde	3520	164	Hydrogen Peroxide	1646		4.88E+11		0.020	
1	Hydrochloric Acid	10973	107	Acetic Anhydride	1533	29005	4.88E+11		0.020	
2	Nitric Acid, Fuming	441	28	Butadiene	8030		4.87E+11		0.020	
1	Hydrochloric Acid	10973	13	Ethyl Acetate	1966	41485	4.85E+11		0.020	
2	Nitric Acid, Fuming	441	13	Butyl Acrylate	1957		4.85E+11		0.020	
2	Nitric Acid, Fuming	441	101	Petroleum Residual Fuel	7907	137445	4.79E+11	1175	0.020	97.98

		Car			Car	Area		Normal-	% of	Cumul-
ASTM		Moves/yr	ASTM		Moves/yr	Covered		i zed	Total	ative
#	Chemical Name #1	#1	<u> </u>	Chemical Name #2	‡ 2	sq.m.	Risk +	Risk	Risk	*
10	Potassium Hydroxide, Liq	2622	105	Phosphorus	3533	51427	4.78E+11	1168	0.020	98.00
10	Potassium Hydroxide, Liq	2622	17	Chloroprene, Inhibited	889	202182	4.71E+11	1156	0.020	98.02
2	Nitric Acid, Fuming	441	101	Gasoline	7676	137445	4.65E+11	1141	0.019	98.03
2	Sulfuric Acid, Spent	4809	4	Ethyl Alcohol	867	111328	4.64E+11	1138	0.019	98.05
1	Hydrofluorosilicic Acid	930	4	Methyl Alcohol	20739	23890	4.61E+11	1130	0.019	98.07
184	Ferric Chloride Solution	1883	184	Hydrogen Peroxide	1646	145491	4.51E+11	1106	0.019	98.09
101	Petroleum Refined	10658	104	Hydrogen Peroxide	1646	25325	4.44E+11	1Ø89	0.019	98.11
16	Potassium Hydroxide, Liq	2622	17	Vinyl Chloride	20081	8390	4.42E+11	1083	0.018	98.13
17	Chloroprene, Inhibited	889	104	Ferric Chloride Solution	1883	261833	4.38E+11	1075	Ø.Ø18	98.15
1	Phosphoric Acid	28008	184	Hydrogen Peroxide	1648	9425	4.35E+11	1965	0.018	98.17
4	Octyl Alcohol	1871	164	Hydrogen Peroxide	1646	15483Ø	4.26E+11	1944	0.018	98.18
101	Crude Oil Petroleum	9995	194	Hydrogen Peroxide	1646	25555	4.20E+11	1031	0.018	98.20
3	Acetic Acid, Glacial	2953	10	Alkaline Liquid, n.o.s.	975	145932	4.20E+11	1030	0.018	98.22
10	Alkaline Liquid, n.o.s.	975	17	Carbon Tetrachloride	1343	318Ø86	4.17E+11	1021	0.017	98.24
7	Hexamethylene Diamine So	l 558Ø	18	Toluene Diisocyanate	2610	28548	4.16E+11	1019	0.017	98.25
19	Methyl Ethyl Ketone	1318	184	Hydrogen Peroxide	1646	191634	4.18E+11	1019	0.017	98.27,
104	Hydrogen Peroxide	1646	107	Acetic Anhydride	1533	164200	4.14E+11	1016	0.017	98.29
1	Hydroiodic Acid .	897	⁻ 26	Acrylonitrile	4811	95240	4.11E+11	1998	0.017	98.30
1	Hydroiodic Acid	. 897	2	Sulfuric Acid, Spent	4809	95033	4.10E+11	1005	0.017	98.32
34	Propylene Oxide	4913	164	Hydrogen Peroxide	1646	50564	4.09E+11	1003	0.017	98.34
1	Hydrochloric Acid	10973	21	Sodium Metal	855	43363	4.07E+11	998	0.017	98.36
2	Nitric Acid, Fuming	441	107	Acetic Anhydride	1533	599121	4.05E+11	993	0.017	98.37
13	Ethyl Acrylate	1281	194	Hydrogen Peroxide	1648	191808	4.84E+11	~ 991	0.017	98.39
2	Nitric Acid, Fuming	441	101	Distillate Fuel Oil	8549	137445	3.97E+11	973	0.017	98.41
5	Butyra i dehyde	870	31	PhenoI	5974	75749	3.94E+11	985	Ø. Ø16	98.42
3	Acetic Acid, Glacial	2953	5	Aceta I dehyde	3520	37699	3.92E+11	961	Ø.Ø16	98.44
2	Sulfuric Acid, Spent	4809	4	Butyl Alcohol	1052	76591	3.87E+11	95ø	Ø.Ø16	98.45
2	Nitric Acid, Fuming	441	101	Fuel, Aviation	6157	137997	3.75E+11	919	0.016	98.47
1	Hydrofluorosilicic Acid	930	26	Acrylonitrile	4811	81888	3.66E+11	898	0.015	98.49
10	Alkaline Liquid, n.o.s.	975	13	Vinyl Acetate	4431	84080	3.63E+11	898	0.015	98.50
2	Nitric Acid, Fuming	441	34	Ethylene Oxide	6535	124878	3.60E+11	882	Ø.Ø15	98.52
2	Nitric Acid, Fuming	441	2	01eum	3243	251327	3.59E+11	881	0.015	98.53
2	Sulfuric Acid, Spent	4809	18	Toluene Diisocyanate	2610	28464	3.57E+11	876	0.015	98.55
4	Butyl Alcohol	1Ø52	104	Hydrogen Peroxide	1646	203147	3.52E+11	863	0.015	98.56
2	Sulfuric Acid, Spent	4809	4	Alcoholic Beverage	1276	57Ø81	3.50E+11	859	0.015	98.58
2	Nitric Acid, Fuming	441	4	Alcoholic Beverage	1278	617827	3.48E+11	852	0.015	98.59
1	Hydrofluorosilicic Acid	930	2	Nitric Acid, Fuming	441	8443Ø3	3.46E+11	849	0.014	98.60
5	Butyraidehyde	870	194	Hydrogen Peroxide	1848	241684	3.46E+11	849	0.014	98.62
2	Nitric Acid, Fuming	441	28	LPG-Propylene	5707	137445	3.46E+11	848	0.014	98.63
2	Nitric Acid, Fuming	441	19	Methy! Ethy! Ketone	1318	588939	3.42E+11	839	0.014	98.65
17	Carbon Tetrachloride	1343	20	Carbon Disulfide	1013	250542	3.41E+11	836	Ø.Ø14	98.66
1	Hydroiodic Acid	897	13	Vinyl Acetate	4431		3.34E+11		0.614	98.68
17	Chioroprene, Inhibited	889	194	Hydrogen Peroxide	1646	227421	3.33E+11	816	0.014	98.69
2	Nitric Acid, Fuming	441	13	Ethyl Acrylate	1281	588913	3.33E+11	816	0.014	98.70
2	Nitric Acid, Fuming	441	13	Vinyl Acetate	4431	168868	3.30E+11	809	0.014	98.72

Table J5. Risk Rankings by Binary Combination (Cont.)

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car			Саг	Area		Normal-	% of	Cumul-
ASTM		Moves/yr	ASTM		Moves/yr	Covered		ized	Total	ative
#	Chemical Name #1	#1	#	Chemical Name #2	‡ 2	sq.m.	Risk ∗	Risk	Risk	x
101	Petroleum Residual Fuel	7907	184	Hydrogen Peroxide	1646	25325	3.30E+11	808	Ø. Ø14	98.73
1	Hydrofluorosilicic Acid	93Ø	17	Methyl Chioride	1168	301593	3.28E+11	803	0.014	98.74
5	Formaldehyde Solution	3Ø58	7	Hexamethylene Diamine Sol	5580	19115	3.26E+11	800	0.014	98.76
10	Alkaline Liquid, n.o.s.	975	26	Acrylonitrile	4811	89378	3.25E+11	798	0.014	98.77
101	Gasoline	7676	164	Hydrogen Peroxide	1848	25325	3.20E+11	785	0.013	98.78
2	Nitric Acid, Fuming	441	4	Ethyl Alcohol	867	834124	3.19E+11	782	0.013	98.86
13	Ethyl Acetate	1066	104	Hydrogen Peroxide	1646	181175	3.18E+11	779	0.013	98.81
10	Anhydrous Ammonia	42526	13	Ethyl Acetate	1088	6981	3.18E+11	778	0.013	98.82
2	Nitric Acid, Fuming	441	4	Butyl Alcohol	1052	6735Ø5	3.12E+11	766	0.013	98.84
13	Vinyl Acetate	4431	164	Ferric Chloride Solution	1883	36731	3.Ø6E+11	751	Ø.013	98.85
3	Acrylic Acid	1031	194	Hydrogen Peroxide	1646	178228	3.Ø2E+11	742	0.013	98.86
104	Hydrogen Peroxide	1646	107	Maleic Anhydride	1168	155922	3.00E+11	735	0.013	98.88
26	Acrylonitrile	4811	104	Ferric Chloride Solution	1883	33068	3.00E+11	735	0.012	98.89
1	Hydrofluorosilicic Acid	936	13	Vinyl Acetate	4431	79798	2.91E+11	714	0.012	98.90
7	Aniline Oil, Liquid	1088	184	Hydrogen Peroxide	1646	158282	2.83E+11	695	0.012	98.91
4	Methyl Alcohol	263Ø	18	Toluene Diisocyanate	2610	40538	2.78E+11	682	Ø.Ø12	98.92
101	Distillate Fuel Oil	6549	104	Hydrogen Peroxide	1646	25325	2.73E+11	669	0.011	98.94
34	Ethylene Oxide	6535	194	Hydrogen Peroxide	1648	25239	2.71E+11	666	0.011	98.95
4	Ethyl Alcohol	867	104	Hydrogen Peroxide	1646	187691	2.68E+11	657	8.611	98.98
2	Nitric Acid, Fuming	441	7	Aniline Oil, Liquid	1Ø88	555587	2.67E+11	654	0.011	98.97
1	Hydrochloric Acid	10973	197	Maleic Anhydride	1168	20727	2.66E+11	651	0.011	98.98
2	Sulfuric Acid, Spent	4809	19	Methyl Ethyl Ketone	1318	41831	2.65E+11	650	0.011	98.99
101	Fuel, Aviation	6157	184	Hydrogen Peroxide	1646	25877	2.82E+11	643	0.011	99.00
13	Methyl Methacrylate	2927	21	Sodium Metal	855	102603	2.57E+11	630	0.011	99.01
24	Motor Fuel Antiknock Com	p 1000	104	Hydrogen Peroxide	1646	154830	2.55E+11	625	0.011	99.02
21	Sodium Metal	855	184	Hydrogen Peroxide	1646	178558	2.51E+11	616	0.010	99.03
2	Sulfuric Acid, Spent	4869	24	Motor Fuel Antiknock Comp	p 1000	52117	2.51E+11	615	0.010	99.04
10	Alkaline Liquid, n.o.s.	975	19	Acetone	2971	86008	2.49E+11	611	0.010	99.05
2	Nitric Acid, Fuming	441	14	Glycol Ethers	982	617036	2.45E+11	602	9.010	99.08
2	Nitric Acid, Fuming	441	24	Motor Fuel Antiknock Comp	p 1000	552135	2.43E+11	597	0.010	99.07
1	Hydrochloric Acid	10973	101	Fuel Oil	39248	585	2.43E+11	597	0.010	99.08
2	Nitric Acid, Fuming	441	5	Butyraldehyde	870	633351	2.43E+11	596	0.010	99.10
2	Nitric Acid, Fuming	- 441	28	LPG-Butene	3945	137445	2.39E+11	586	0.010	99.11
28	LPG-Propylene	5707	184	Hydrogen Peroxide	1646	25325	2.38E+11	583	0.010	99.11
1	Hydrofluorosilicic Acid	930	19	Acetone	2971	86008	2.38E+11	583	0.010	99.12
2	Nitric Acid, Fuming	441	19	Alkaline Liquid, n.o.s.	975	552135	2.37E+11		0.010	99.13
105	Phosphorus	3533	106	Ammonium Nitrate Soln.	1860	35982	2.36E+11	586	Ø. Ø1Ø	99.14
2	Nitric Acid, Fuming	441	17	Chloroprene, Inhibited	889	599076	2.35E+11	576	0.010	99.15
17	Chloroprene, Inhibited	889	24	Motor Fuel Antiknock Com			2.33E+11		0.010	
2	Nitric Acid, Fuming	441	3	Acetic Acid, Glacial	2953		2.31E+11		0.010	
27	Dinitrotoluene, Liquid	866	164	Hydrogen Peroxide	1646		2.30E+11		8.819	
1	Hydrochloric Acid	10973	3	Acetic Acid, Glacial	2953	7104	2.30E+11		0.010	•
1	Hydroiodic Acid	897	19	Acetone	2971	86008	2.29E+11		0.010	
1	Hydrofluorosilicic Acid	930	13	Methyl Methacrylate	2927	83602	2.28E+11	558	0.009	99.21
4	Isopropanol	2848	21	Sodium Metal	855	92219	2.25E+11	551	Ø. ØØ9	99.22

Table J5. Risk Rankings by Binary Combination (Cont.)

ASTM	Chemical Name #1	Car Moves/yr #1	ASTM	Chemical Name #2	Car Moves/yr	Area Covered	Risk *	Normal- ized Risk	% of Total Risk	Cumul- ative
—										
5	Formaldehyde Solution	3Ø58	18	Toluene Diisocyanate	2610	27798	2.22E+11	544	0.009	-
4	Isopropanol	2848	5	Butyraldehyde	870	89450	2.22E+11	543	0.009	99.24
3	Acrylic Acid	1031	17	Chloroprene, Inhibited	889		2.19E+11	537	0.009	99.25
20	Carbon Disulfide	1913	21	Sodium Metal	855	252602	2.19E+11	536	0.009	99.26
1	Hydroiodic Acid	897	2	Nitric Acid, Fuming	441	552135	2.18E+11	536	0.009	99.27
7	Aniline Oil, Liquid	1088	17	Carbon Tetrachloride	1343	•	2.17E+11	532	0.009	99.28
2	Sulfuric Acid, Spent	48Ø9	7	Aniline Oil, Liquid	1088	41151	2.15E+11	528	0.009	99.29
2	Nitric Acid, Fuming	441	17	Methyl Chloride	1168	412334	2.12E+11	521	0.009	99.29
16	Styrene Monomer	10859	21	Sodium Metal	855	22752	2.11E+11	518	0.009	,
4	Buty! Alcohol	1052	18	Toluene Diisocyanate	2610	76625	2.10E+11	516	6.009	
28	Butadiene	8030	184	Hydrogen Peroxide	1646	15900	2.10E+11	515	0.009	99.32
1	Hydrochloric Acid	10973	10	Alkaline Liquid, n.o.s.	975	19635	2.1ØE+11	515	0.009	99.33
2	Nitric Acid, Fuming	441	21	Sodium Metal	855	552135	2.Ø8E+11	510	0.009	99.34
3	Acetic Acid, Glacial	2953	7	Aniline Oil, Liquid	1088	84424	2.07E+11	508	0.009	99.35
. 1	Phosphoric Acid	28008	5	Formaldehyde Solution	3Ø58	2356	2.02E+11	495	0.008	99.35
104	Hydrogen Peroxide	1646	105	Phosphorus	3533	34664	2.Ø2E+11	494	0.008	99.36
7	Aniline Oil, Liquid	1988	17	Chloroprene, Inhibited	889	206971	2.00E+11	491	0.908	99.37
1	Phosphoric Acid	28008	101	Petroleum Naphtha	13541	527	2.00E+11	490	Ø. ØØ8	99.38
10	Potassium Hydroxide, Liq		18	Toluene Diisocyanate	2610	28464	1.95E+11	478	0.008	
2	Sulfuric Acid, Spent	4809	13	Ethyl Acetate	1966	37596	1.93E+11	473	0.008	
2	Nitric Acid, Fuming	441	161	Combustible Liquid	3139	137445	1.90E+11	467	0.008	
1	Hydroiodic Acid	897	13	Methyl Methacrylate	2927	71821	1.89E+11	462	0.008	
7	Hexamethylene Diamine So		107	Acetic Anhydride	1533	22013	1.88E+11	462	9.998	
3	Acrylic Acid	1031	7	Hexamethylene Diamine So		31674	1.82E+11	447	0.008	
1	Hydroiodic Acid	897	17	Carbon Tetrachloride	1343	148446	1.79E+11	438	9.997	
10	Alkaline Liquid, n.o.s.	975	105	Phosphorus	3533	51427	1.77E+11	434	0.007	
17	Chloroprene, Inhibited	889	21	Sodium Metal	855	2313Ø5	1.76E+11		0.007	
10	Alkaline Liquid, n.o.s.	975	17	Chloroprene, Inhibited	889	202182	1.75E+11	430	0.007	99.48
5	Formaldehyde Solution	3058	7	Aniline Dil, Liquid	1088	52518	1.75E+11	428	0.007	99.46
17	Carbon Tetrachloride	1343	21	Sodium Metal	855	148440	1.7ØE+11	418	0.007	99.47
3	Acrylic Acid	1031	26	Acrylonitrile	4811	33921	1.68E+11	413	0.007	99.48
1	Hydrofluorosilicic Acid	936	17	Chloroprene, Inhibited	889	202182	1.87E+11	410	0.007	99.49
10	Alkaline Liquid, n.o.s.	975	17	Vinyl Chloride	20081	8390	1.84E+11	403	0.007	99.49
2	Sulfuric Acid, Spent	4809	197	Acetic Anhydride	1533	21936	1.62E+11	397	0.007	99.50
1	Hydroiodic Acid	897	17	Chloroprene, Inhibited	889	202182	1.61E+11	395	0.007	99.51
3	Acrylic Acid	1031	18	Toluene Diisocyanate	2610	57761	1.55E+11	381	0.006	99.51
4	Methyl Alcohol	2630	21	Sodium Metal	855	67322	1.51E+11	371	Ø.006	99.52
13	Vinyl Acetate	4431	21	Sodium Metal	855	39637	1.50E+11	368	0.006	99.52
2	Sulfuric Acid, Spent	4809	5	Butyraidehyde	870	35593	1.49E+11	365	Ø.006	99.53
19	Methyl Ethyl Ketone	1318	21	Sodium Metal	855	129828	1.46E+11	359	Ø. ØØ8	99.54
5	Butyra I dehyde	870	18	Toluene Diisocyanate	2610	64059	1.45E+11	357	0.006	99.54
10	Potassium Hydroxide, Liq	2622	19	Methyl Ethyl Ketone	1318	41831	1.45E+11	354	0.006	99.55
13	Ethyl Acrylate	1281	21	Sodium Metal	855	130821	1.43E+11	351	9.998	99.58
19	Acetone	2971	21	Sodium Metal	855	56291	1.43E+11	351	0.006	99.56
. 1	Hydrofluorosilicic Acid	930	4	Isopropanol	2848	531Ø5	1.41E+11	345	0.008	99.57

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car			Car	Area	**	Normal-	% of	Cumu I –
ASTM		Moves/yr	ASTM		Moves/yr	Covered		ized	Total	ative
#	Chemical Name #1	#1	#	Chemical Name #2	#2	sq.∎.	.Risk +	Risk	Risk	×
4	Ethyl Alcohol	867	18	Toluene Diisocyanate	2610	61165	1.38E+11	339	0.006	99.57
2	Nitric Acid, Fuming	441	161	Flammable Liquid	2258	137445	1.37E+11	335	Ø.996	99.58
3	Acrylic Acid	1 0 31	5	Formaldehyde Solution	3Ø58	42352	1.34E+11	327	0.006	99.58
101	Combustible Liquid	3139	164	Hydrogen Peroxide	1646	25325	1.31E+11	321	0.005	99.59
5	Formaldehyde Solution	3Ø58	104	Hydrogen Peroxide	1646	25325	1.27E+11	313	0 .005	99.59
2	Sulfuric Acid, Spent	48Ø9	27	Dinitrotoluene, Liquid	866	3Ø516	1.27E+11	312	0.005	99.60
2	Sulfuric Acid, Spent	48Ø9	13	Methyl Methacrylate	2927	8986	1.26E+11	310	8.005	99.61
10	Potassium Hydroxide, Liq	2622	13	Ethyl Acetate	1966	45195	1.26E+11	310	0.005	99.61
5	Butyraidehyde	879	107	Acetic Anhydride	1533	93408	1.25E+11	305	0.005	99.62
4	Octyl Alcohol	1671	18	Toluene Diisocyanate	2619	28274	1.23E+11	302	0.005	99.62
2	Nitric Acid, Fuming	441	4	Octyl Alcohol	1871	166834	1.23E+11	301	0.005	99.63
1	Hydrofluorosilicic Acid	930	10	Potassium Hydroxide, Liq	2622	50265	1.23E+11	301	0.005	99.63
1	Hydrofluorosilicic Acid	930	18	Toluene Diisocyanate	2610	50455	1.22E+11	366	0.005	99.64
4	Methyl Alcohol	2630	5	Butyraldehyde	876	53323	1.22E+11	299	0.005	99.64
2	Nitric Acid, Fuming	441	184	Hydrogen Peroxide	1646	162684	1.18E+11	290	0.005	99.85
7	Aniline Bil, Liquid	1088	18	Toluene Diisocyanate	2610	41239	1.17E+11	287	0.005	99.65
28	LPG-Butene	2806	104	Hydrogen Peroxide	1646	25325	1.17E+11	287	0.005	99.66
2	Sulfuric Acid, Spent	4809	- 3	Acrylic Acid	1631	23398	1.16E+11		0.005	99.66
1	Hydroiodic Acid	897	13	Ethyl Acrylate	1281	99610	1.14E+11	281	0.005	99.67
4	Alcoholic Beverage	1278	. 5	Butyraldehyde	870	1ø2692	1.14E+11	280	0.005	99.67
2	Nitric Acid, Fuming	441	20	.Carbon Disulfide	1913	252585	1.13E+11	277	0.005	99.88
2	Nitric Acid, Fuming	441	106	Ammonium Nitrate Soln.	1860	137445	1.13E+11	276	0.005	99.68
1	Hydrofluorosilicic Acid	936	13	Butyl Acrylate	1957	59774	1.09E+11		0.005	99.68
1	Hydrocyanic Acid	1226	24	Motor Fuel Antiknock Comp	1000	87982	1.Ø8E+11	264	0.004	99.69
19	Methyl Ethyl Ketone	1318	194	Ferric Chloride Solution	1883	41831	1.84E+11	255	0.004	99.69
5	Acetaldehyde	3520	21	Sodium Metal	855	34309	1.03E+11	253	0.004	99.70
1	Hydrofluorosilicic Acid	930	13	Ethyl Acrylate	1281	86258	1.03E+11	252	0.004	99.70
2	Sulfuric Acid, Spent	4809	21	Sodium Metal	855	23728	9.76E+1Ø	239	0.004	99.71
10	Sodium Hydroxide, Liq.	75641	107	Maleic Anhydride	1168	1092	9.65E+10	237	8.884	99.71
1	Phosphoric Acid	28008	101	Fuel, Aviation	6157	552	9.52E+1Ø	233	0.004	
21	Sodium Metal	855	106	Ammonium Nitrate Soln.	1860		9.5ØE+1Ø	233	0.004	99.72
4		1276	18	Toluene Diisocyanate	2610	28500	9.49E+10	233	0.004	99.72
1	•	930	4	Ethyl Alcohol	867	116848	9.42E+10	231	0.004	99.73
101	1	2258	104	Hydrogen Peroxide	1646		9.40E+10	231	0.004	99.73
2	Sulfuric Acid	46265	7	Hexamethylene Diamine So	l 558ø	362	9.35E+10	229	0.004	99.73
27	Dinitrotoluene, Liquid	868	105	Phosphorus -	3533	3Ø516	9.34E+10	229	0.004	99.74
2	Sulfuric Acid, Spent	4899	31	Pheno I	5974	3248	9.33E+10	229	0.004	99.74
5	Aceta i dehyde	3520	27	Dinitrotoluene, Liquid	866	3Ø516	9.30E+10	228	0.004	99.75
2	Sulfuric Acid, Spent	4809	16	Toluene	2984		9.19E+10		8.894	
7	Hexamethylene Diamine So		21	Sodium Metal	855		9.18E+10		0.884	
2	Sulfuric Acid, Spent	4809	16	Styrene Monomer	10859		9.18E+10		Ø.004	
5	Butyraldehyde	87Ø	7	Aniline Oil, Liquid	1088		9.09E+10		0.004	
4	Butyl Alcohol	1052	21	Sodium Metal	855		9.09E+10		0.004	
4	Butyl Alcohol	1052	5	Butyraldehyde	870		9.07E+10		0.004	
2	Sulfuric Acid, Spent	4809	13	Ethyl Acrylate	1281	14338	8.83E+18	217	0.004	99.77

Table J5. Risk Rankings by Binary Combination (Cont.)

ASTM	Chemical Name ∰1	Car Moves/yr #1	ASTM	Chemical Name #2	Car Moves/yr #2	Area Covered sq.m.	Risk *	Normal- ized Risk	% of Total Risk	Cumul- ative
10	Potassium Hydroxide, Liq	2622	107	Acetic Anhydride	1533	21936	8.82E+10	216	0.004	99.78
3	Acrylic Acid	1031	7	Aniline Oil, Liquid	1088	76337	8.58E+10	210	0.004	99.78
2	Nitric Acid, Fuming	441	101	Gasoline, nec	1416	137445	8.55E+10	210	0.004	99.78
2	Nitric Acid, Fuming	441	101	Oil	1407	137445	8.53E+10	209	0.004	99.79
1	Hydrofluorosilicic Acid	93Ø	107	Acetic Anhydride	1533	59635	8.5ØE+1Ø	208	0.004	99.79
2	Nitric Acid, Fuming	441	107	Maleic Anhydride	1168	164533	8.47E+10	208	0.004	99.79
5	Butyraldehyde	87Ø	104	Ferric Chloride Solution	1883	51437	8.43E+1Ø	287	8.994	99.80
1	Hydrofluorosilicic Acid	93Ø	17	Carbon Tetrachloride	1343	66759	8.34E+16	284	0.003	99.80
1	Hydrochloric Acid	10973	14	Glycol Ethers	962	8335	8.25E+10	202	0.003	99.80
7	Aniline Oil, Liquid	1088	107	Acetic Anhydride	1533	49327	8.23E+10	282	0.003	99.81
5	Butyra I dehyde	87Ø	10	Potassium Hydroxide, Liq	2622	35593	8.12E+10	199	0.003	99.81
1	Hydroiodic Acid	897	5	Butyraldehyde	87Ø	102698	8.Ø1E+1Ø	197	0.003	99.81
2	Nitric Acid, Fuming	441	13	Ethyl Acetate	1086	189793	7.98E+10	198	0.003	99.82
1	Hydrochloric Acid	10973	5	Formaldehyde Solution	3058	2356	7.91E+10	194	0.003	99.82
1	Hydrochloric Acid	1Ø973	101	Petroleum Naphtha	13541	527	7.83E+10	192	0.003	99.82
2	Nitric Acid, Fuming	441	161	Compound, Cleaning	1281	137445	7.76E+10	190	0.003	99.83
2	Nitric Acid, Fuming	441	101	Coal Tar Distillate	1259	138042	7.66E+10	188	0.003	99.83
5	Butyraldehyde	87Ø	21	Sodium Metal	855	102698	7.64E+1Ø	187	0.003	99.83
3	Acrylic Acid	1931	107	Acetic Anhydride	1533	48326	7.64E+10	187	0.003	99.84
2	Nitric Acid, Fuming	441	28	LPG-Isobutylene	1258	137445	7.63E+10	187	0.003	99.84
3	Acrylic Acid	1031	21	Sodium Metal	855	85559	7.54E+10	185	0.003	99.84
10	Alkaline Liquid, n.o.s.	975	18	Toluene Diisocyanate	2616	28464	7.24E+10	178	0.003	99.85
1	Phosphoric Acid	28008	7	Aniline Oil, Liquid	1088	2356	7.18E+10	178	0.003	99.85
2	Nitric Acid, Fuming	441	101	Pulp Mill Liquid	1159	137445	7.03E+10	172	Ø.00 <u>3</u>	99.85
1	Hydrofluorosilicic Acid	930	5	Butyra Idehyde	870	85858	6.95E+10	170	0.003	99.85
10	Potassium Hydroxide, Liq	2622	27	Dinitrotoluene, Liquid	866	3Ø516	6.93E+1Ø	179	0.003	99.86
31	Pheno I	5974	105	Phosphorus	3533	3248	6.86E+1Ø	168	0.003	99.86
13	Ethyl Acetate	1966	21	Sodium Metal	855	74806	6.82E+10	167	0.003	99.86
3	Acrylic Acid	1031	5	Butyraldehyde	879	75105	8.74E+10	185	9.003	99.87
7	Aniline Oil, Liquid	1088	21	Sodium Metal	855	71678	6.67E+1Ø	163	0.003	99.87
1	Hydroiodic Acid	897	18	Toluene Diisocyanate	2610	28464	6.66E+1Ø	163	Ø.Ø03	99.87
1	Phosphoric Acid	28008	191	Crude Oil Petroleum	9995	230	8.44E+10	158	0.003	99.87
3	Acrylic Acid	1031	10	Potassium Hydroxide, Liq	2622	23398	6.33E+1Ø	155	0.003	99.88
2	Nitric Acid, Fuming	441	3	Acrylic Acid	1031	137886	6.27E+10	154	0.003	99.88
4	Ethyl Alcohol	867	5	Butyraldehyde	870	79420	5.99E+10	147	0.002	99.88
7	Aniline Oil, Liquid	1Ø88	24	Motor Fuel Antiknock Com	p 1000	54473	5.93E+10	145	0.002	99.88
1	Hydrofluorosilicic Acid	930	21	Sodium Metal	855	73993	5.88E+10	144	0.002	99.89
161	Gasoline, nec	1410	194	Hydrogen Peroxide	1646	25325	5.88E+1Ø	144	0.002	99.89
101	Oil	1407	104	Hydrogen Peroxide	1646	25325	5.87E+10	144	0.002	99.89
2	Sulfuric Acid, Spent	4809	18	Xylene	5415	2237	5.83E+1Ø	143	0.002	99.89
13	Butyl Acrylate	1957	21	Sodium Metal	855	34543	5.78E+10	142	0.002	99.90
1	Hydrofluorosilicic Acid	930	107	Maleic Anhydride	1168	51357	5.58E+1Ø	137	0.002	99.90
18	Toluene Diisocyanate	2619	21	Sodium Metal	855	24513	5.47E+1Ø	134	0.002	99.98
10	Alkaline Liquid, n.o.s.	975	19	Methyl Ethyl Ketone	1318	41831	5.38E+1Ø	132	0.002	99.90
101	Coal Tar Distillate	1259	104	Hydrogen Peroxide	1646	25922	5.37E+1Ø	132	0.002	99.91

Table J5. Risk Rankings by Binary Combination (Cont.)

ACTU		Car	ACTV		Car	Area		Normal-		Cumu I -
ASTM	Chemical Name #1	Moves/yr	V2 IN	Chemical Name #2	Moves/yr	covered sq.m.	Risk *	ized Risk	Total Risk	ative %
										
101	Compound, Cleaning	1281		Hydrogen Peroxide	1646	25325	5.34E+1Ø	131	0.002	99.91
19	Potassium Hydroxide, Liq	2622	21	Sodium Metal	855	23728	5.32E+10	130	0.002	99.91
4	Alcoholic Beverage	1276	184	Hydrogen Peroxide	1848	25325	5.32E+1Ø	130	0.002	99.91
2	Nitric Acid, Fuming	441	27	Dinitrotoluene, Liquid	866	137445	5.25E+1Ø	129	0.002	99.91
28	LPG-Isobuty lene	1258	104	Hydrogen Peroxide	1848	25325	5.24E+1Ø	129	Ø.ØØ2	99.92
5	Formaldehyde Solution	3058	21	Sodium Metal	855	19919	5.21E+10	128	6.662	99.92
1	Hydrofluorosilicic Acid	930	19	Methyl Ethyl Ketone	1318	41831	5.13E+10	126	0.002	99.92
7	Aniline Oil, Liquid	1088	107	Maleic Anhydride	1168	39832	5.86E+18	124	0.002	99.92
13	Methyl Methacrylate	2927	104	Ferric Chloride Solution	1883	8980	4.95E+10	121	0.002	99.93
1	Hydroiodic Acid	897	19	Methyl Ethyl Ketone	1318	41831	4.95E+1Ø	121	0.002	99.93
1	Hydrofluorosilicic Acid	930	24	Motor Fuel Antiknock Comp	-	52117	4.85E+10	119	0.002	99.93
1	Hydrochloric Acid	10973	7	Hexamethylene Diamine Sol		785	4.81E+1Ø	118	0.002	
28	LPG-Butylene	1139	104	Hydrogen Peroxide	1646	25325	4.75E+1Ø	116	0.002	
4	Ethyl Alcohol	867	21	Sodium Metal	855		4.72E+10	116	0.002	
10	Alkaline Liquid, n.o.s.	975	13	Ethyl Acetate	1066	45195	4.78E+18	115	0.002	
1	Hydroiodic Acid	897	24	Motor Fuel Antiknock Comp		•	4.67E+10		0.002	
2	Sulfuric Acid, Spent	4809	5	Butyraldehyde	879	11096	4.84E+10	114	0.002	
1	Hydrofluorosilicic Acid	930	10	Alkaline Liquid, n.o.s.	975	•	4.56E+10	112	0.002	99.94
1	Hydrofluorosilicic Acid	930	4	Alcoholic Beverage	1276		4.01E+10	98	0.002	
1	Hydroiodic Acid	897	13	Ethyl Acetate	1066		3.97E+10	97	0.002	
3	Acetic Acid, Glacial	2953	106	Ammonium Nitrate Soln.	1860		3.90E+10	96	0.002	
21	Sodium Metal	855	104	Ferric Chloride Solution	1883	23728	3.82E+1Ø	94	0.002	
1	Hydrochloric Acid	10973	101	Fuel, Aviation	8157		3.73E+1Ø	91	0.002	
1	Hydrofluorosilicic Acid	930	13	Ethyl Acetate	1966		3.72E+10	91	Ø.Ø82	
16	Styrene Monomer	10859	104	Ferric Chloride Solution	1883		3.59E+1Ø	88	0.001	99.95
1	Phosphoric Acid	28008	107	Maleic Anhydride	1168		3.57E+10	88	0.001	99.96
2	Sulfuric Acid, Spent	4809	5	Formaldehyde Solution	3Ø58		3.46E+1Ø	85	0.001	99.96
4	Octyl Alcohol	1871	21	Sodium Metal	855		3.39E+1Ø	83	0.001	99.98
10	Alkaline Liquid, n.o.s.	975	167	Acetic Anhydride	1533		3.28E+10	8Ø	0.001	99.96
1	Hydrocyanic Acid	1226	164	Hydrogen Peroxide	1646		3.21E+10		0.001	99.96
4	Methyl Alcohol	2630	184	Ferric Chloride Solution	1883		3.11E+10	76	0.001	99.96
17	Methyl Chloride	1168	194	Hydrogen Peroxide	1648	15900	3.08E+10			99.96
101	Pulp Mill Liquid	1159	104	Hydrogen Peroxide	1646	15900	3.03E+10	74	9.961	99.97
5	Butyra i dehyde	870	19	Alkaline Liquid, n.o.s.	975	35593	3.02E+10	. 74	0.001	99.97
1	Hydroiodic Acid	897	107	Acetic Anhydride	1533	21936	3.02E+10	74	0.001	99.97
13	Ethyl Acrylate	1281	104	Ferric Chloride Solution	1883	12212	2.95E+10	72	0.001	99.97
1	Hydrochloric Acid	10973	4	Butyl Alcohol	1052	2534	2.93E+1Ø	72	0.001	99.97
27	Dinitrotoluene, Liquid	866	107	Acetic Anhydride	1533	21936	2.91E+1Ø	71	0.001	99.97
21	Sodium Metal	855	107	Acetic Anhydride	1533	21936	2.88E+1Ø	76	0.001	99.97
1	Hydrochloric Acid	10973	7	Aniline Oil, Liquid	1Ø88	2356	2.81E+10	69	0.001	99.97
4	Alcoholic Beverage	1276	21	Sodium Metal	855	23732	2.59E+10	63	0.001	99.97
10	Alkaline Liquid, n.o.s.	975	27	Dinitrotoluene, Liquid	866		2.58E+1Ø	63	0.001	99.98
1	Hydrochloric Acid	10973	101	Crude Gil Petroleum	9995		2.52E+10	62	0.001	99.98
3	Acrylic Acid	1031	10	Alkaline Liquid, n.o.s.	975		2.35E+10		0.001	
21	Sodium Metal	855	101	Pulp Mill Liquid	1159	-23728	2.35E+10	58	0.001	99.98

Table J5. Risk Rankings by Binary Combination (Cont.)

		Car		•	Car	Area		Normal-		Cumul-
ASTM	AL 1 1 N	Moves/yr	_	4 1 1 1 1 1 8 4	Moves/yr		5 ' '	ized	Total	ative
#	Chemical Name #1	#1		Chemical Name #2	#2 	sq.■.	Risk *	Risk	Risk ———	<u> </u>
5	Butyraldehyde	870	27	Dinitrotoluene, Liquid	866	30516	2.30E+10	56	0.001	99.98
21	Sodium Metal	855	27	Dinitrotoluene, Liquid	866	30516	2.26E+1Ø	55	0.001	99.98
1	Phosphoric Acid	28008	101	Coal Tar Distillate	1259	597	2.11E+1Ø	52	0.001	99.98
1	Hydrofluorosilicic Acid	930	101	Fuel Oil	39248	565	2.Ø6E+1Ø	51	0.001	99.98
1	Hydroiodic Acid	897	161	Fuel Oil	39248	565	1.99E+10	49	0.001	99.98
10	Alkaline Liquid, n.o.s.	975	21	Sodium Metal	855	23728	1.98E+1Ø	49	0.001	99.98
1	Hydrofluorosilicic Acid	930	3	Acetic Acid, Glacial	2953	7104	1.95E+10	48	0.001	99.99
5	Formaldehyde Solution	3Ø58	10	Potassium Hydroxide, Liq	2622	2356	1.89E+10	46	0.001	99.99
1	Hydroiodic Acid	897	3	Acetic Acid, Glacial	2953	7104	1.88E+10	46	0.001	99.99
1	Hydroiodic Acid	897	21	Sodium Metal	855	24513	1.88E+1Ø	46	0.001	99.99
1	Hydrocyanic Acid	1226	19	Acetone	2971	5824	1.83E+1Ø	. 45	9.991	99.99
1	Hydrofluorosilicic Acid	930	31	Phenol	5974	3142	1.75E+10	43	0.001	99.99
1	Hydroiodic Acid	897	13	Butyl Acrylate	1957	95Ø9	1.67E+16	41	Ø.ØØ1	99.99
1	Hydrofluorosilicic Acid	930	104	Hydrogen Peroxide	1848	9425	1.44E+16	35	0.001	99.99
1	Hydroiodic Acid	897	184	Hydrogen Peroxide	1648	9425	1.39E+10	34	0.001	99.99
1	Hydrochloric Acid	1Ø973	3	Acrylic Acid	1031	1226	1.39E+1Ø	·· 34	0.001	99.99
5	Formaldehyde Solution	3058	197	Acetic Anhydride	1533	2910	1.36E+1Ø	33	0.001	99.99
5	Formaldehyde Solution	3058	184	Ferric Chloride Solution	1883	2358	1.36E+1Ø	33	0.001	99.99
1	Phosphoric Acid	28008	3	Acrylic Acid	1031	441	1.27E+10	31	0.001	99.99
1	Hydroiodic Acid	897	14	Glycol Ethers	902	13832	1.12E+10	27	0.000	99.99
104	Ferric Chloride Solution	1883	197	Acetic Anhydride	1533	3696	1.07E+10	26	9.999	99.99
106	Ammonium Nitrate Soln.	1860	107	Acetic Anhydride	1533	3696	1.05E+10	26	0.000	99.99
2	Sulfuric Acid, Spent	4809	7	Hexamethylene Diamine So	5580	362	9.71E+09	₽ 24	0.000	99.99
2	Sulfuric Acid, Spent	4809	13	Butyl Acrylate	1957	985	9.08E+09	22	Ø.000°	100.00
2	Sulfuric Acid, Spent	4809	14	Glycol Ethers	902	2051	8.90E+09	22	0.000	100.00
1	Hydrochloric Acid	10973	101	Coal Tar Distillate	1259	597	8.25E+Ø9	20	0.000	100.00
18	Toluene Diisocyanate	2610	105	Phosphorus	3533	785	7.24E+09	18	0.000	100.00
5	Formaldehyde Solution	3Ø58	10	Alkaline Liquid, n.o.s.	975	2356	7.02E+09	17	0.000	100.00
1	Hydrofluorosilicic Acid	930	5	Formaldehyde Solution	3058	2356	8.7ØE+Ø9	16	0.000	100.00
1	Hydrofluorosilicic Acid	930	101	Petroleum Naphtha	13541	527	6.64E+69	16	8.888	100.00
1	Hydroiodic Acid	897	5	Formaldehyde Solution	3058	2356	6.46E+Ø9	16	0.000	100.00
1	.	897	161	Petroleum Naphtha	13541	527	6.4ØE+Ø9	16	9.000	100.00
7	Hexamethylene Diamine So		107	Maleic Anhydride	1168	898	5.84E+Ø9	14		100.00
7	Aniline Oil, Liquid	1988	164	Ferric Chloride Solution	1883	2358	4.83E+89	. 12		100.00
1	Hydrofluorosilicic Acid	930	7	Hexamethylene Diamine So		785	4.07E+09			100.00
18	Toluene Diisocyanate	2610	106	Ammonium Nitrate Soln.	1860	785	3.81E+09			100.00
13	Butyl Acrylate	1957	164	Ferric Chloride Solution	1883	965	3.56E+09			100.00
14	Glycol Ethers	902	104	Ferric Chloride Solution	1883	2051	3.48E+09			100.00
10	Potassium Hydroxide, Liq	2622	107	Maleic Anhydride	1168	1092	3.34E+Ø9			100.00
1	Hydrofluorosilicic Acid	930	101	Fuel, Aviation	6157	552	3.16E+Ø9			100.00
1	Hydroiodic Acid	897	101	Fuel, Aviation	6157	552	3.05E+09	7		100.00
1	Hydrofluorosilicic Acid	930	105	Phosphorus	3533	785	2.58E+09			100.00
1	Hydrofluorosilicic Acid,	930	4	Butyl Alcohol	1052		2.48E+Ø9			100.00
1	Hydrofluorosilicic Acid	930	7	Aniline Oil, Liquid	1088	2356	2.38E+Ø9			100.00
1	Hydroiodic Acid	897	7		1088	2356	2.3ØE+Ø9	6		100.00

Table J5. Risk Rankings by Binary Combination (Cont.)

ASTM		Car Moves/yr	ASTM		Car Moves/yr	Area Covered		Normal- ized	% of Total	Cumul- ative
#	Chemical Name #1	#1	#	Chemical Name #2	#2	sq.∎.	Risk *	Risk	Risk	×
1	Hydrofluorosilicic Acid	930	181	Crude Oil Petroleum	9995	230	2.14E+89	5	0.000	100.00
1	Hydroiodic Acid	897	101	Crude Oil Petroleum	9995	236	2.Ø6E+Ø9	5	0.000	100.00
1	Hydrofluorosilicic Acid	930	14	Glycol Ethers	952	2051	1.72E+09	4	9.666	100.00
10	Alkaline Liquid, n.o.s.	975	107	Maleic Anhydride	1168	1892	1.24E+69	3	0.000	100.00
1	Hydrofluorosilicic Acid	930	3	Acrylic Acid	1031	1226	1.18E+09	3	0.000	100.00
1	Hydroiodic Acid	897	107	Maleic Anhydride	1168	1892	1.14E+89	3	0.000	100.00
27	Dinitrotoluene, Liquid	868	107	Maleic Anhydride	1168	1092	1.1ØE+Ø9	3	6.666	100.00
21	Sodium Metal	855	107	Maleic Anhydride	1168	1092	1.Ø9E+Ø9	3	0.000	100.00
3	Acrylic Acid	1031	106	Ammonium Nitrate Soln.	1869	441	8.46E+Ø8	2	9.000	100.00
1	Hydrofluorosilicic Acid	93Ø	101	Coal Tar Distillate	1259	597	6.99E+08	2	0.000	100.00
1	Hydroiodic Acid	897	101	Coal Tar Distillate	1259	597	6.74E+Ø8	2	0.000	100.00
1	Hydroiodic Acid	897	3	Acrylic Acid	1031	441	4.Ø8E+Ø8	1	0.000	100.00

^{*} Risk = Car myes/yr for \$1 chemical x Car moves/yr for \$2 chemical x area covered.

The Normalized Risk is the Risk divided by the risk from the combination with the lowest risk (i.e. hydroiodic acid with acrylic acid). It is a relative measure of the risk between the binary combinations.

Table J6. Risk Ranking by Specific Chemical

		Normal-	% of	Cumu I -
	Total	ized	Total	ative
Chemical Name	Risk	Risk	Risk	*
Sulfuric Acid	1.02E+15	951	21.18	21.18
Oleum	6.15E+14	575	12.79	33.98
Sodium Hydroxide, Liq.	4.32E+14	464	8.99	42.94
Hydrochloric Acid	3.88E+14	363	8.07	51.02
Methyl Alcohol	2.78E+14	253	5.63	56.68
Denatured Alcohol	2.32E+14	217	4.82	61.47
Styrene Monomer	1.32E+14	123	2.74	64.23
Vinyl Chloride	1.25E+14	117	2.60	66.8
Chlorine	1.12E+14	105	2.34	69.14
Acrylonitrile	9.75E+13	91	2.03	71.17
Carbon Tetrachloride	7.34E+13	69	1.53	72.79
Anhydrous Ammonia	7.24E+13	68	1.51	74.2
Acetic Acid, Glacial	7.10E+13	66	1.48	75.89
Nitric Acid, Fuming	6.67E+13	62	1.39	77.07
Hydrogen Peroxide	6.54E+13	61	1.38	78.4
Sulfuric Acid, Spent	5.97E+13	58	1.24	79.68
Phosphoric Acid	5.32E+13	50	1.11	80.79
Vinyl Acetate	5.21E+13	49	1.08	81.87
Hydrofluorosilicic Acid	5.09E+13	48	1.06	82.93
Chloroprene, Inhibited	5.Ø3E+13	47	1.05	83.98
Benzen e	4.94E+13	46	1.03	85.0
Acetone	4.29E+13	46	Ø.89	85.96
Fuel Cil	3.78E+13	z 35	Ø.77	86.6
Xy lene	3.47E+13	33	0.72	87.39
Hydrocyanic Acid	3.25E+13	30	Ø.68	88.07
Phenol	3.Ø2E+13	28	0.63	88.76
Methyl Chloride	2.97E+13	28	Ø.82	89.3
Phosphoru s	2.56E+13	24	Ø.53	89.8
Isopropanol	2.49E+13	23	₫.52	90.3
Toluene Diisocyanate	2.48E+13	23	Ø.52	90.89
LPG	2.47E+13	23	Ø.51	91.4
Methyl Methacrylate	2.06E+13	19	0.43	91.8
Petroleum Naphtha	1.92E+13	18	0.40	
Toluene	1.85E+13	17	Ø.38	92.6
Hydroiodic Acid	1.74E+13		Ø.36	92.9
Wethyl Ethyl Ketone	1.72E+13	16	0.36	
Hexamethylene Diamine Solut		16	Ø.35	
Ethyl Alcohol	1.66E+13	16	Ø.35	94.0
Potassium Hydroxide, Liq	1.50E+13	14	0.31	94.3
Sodium Metal	1.44E+13	13	0.30	94.6
Butyraldehyde	· 1.36E+13	13	Ø. 28	
Ethylene Oxide	1.34E+13		Ø.28	
Ethyl Acrylate	1.12E+13		Ø.23	95.4
Propylene Oxide	1.11E+13		Ø.23	
Acetic Anhydride	1.11E+13		₫.23	
Ammonium Nitrate Soln.	1.06E+13		Ø.22	96.1

Table J8. Risk Ranking by Specific Chemical (Cont.)

Chemical Name	Total Risk	Normal- ized Risk	% of Total Risk	
Formaldehyde Solution	1.00E+13	9	Ø.21	96.33
Aniline Bil, Liquid	9.9ØE+12	9	0.21	96.53
Petroleum Refined	9.78E+12	9	8.28	98.74
Ethyl Acetate	9.39E+12	9	0.20	96.93
Butadiene	9.29E+12	9	Ø.19	97.13
Crude Oil Petroleum	9.27E+12	9	Ø.19	97.32
Butyl Alcohol	8.88E+12	8	0.18	97.50
Aceta I dehyde	8.84E+12	8	Ø.18	97.69
Alcoholic Beverage	8.49E+12	8	Ø.18	97.86
Motor Fuel Antiknock Compound	7.74E+12	7	Ø.16	98.03
Petroleum Residual Fuel	7.25E+12	7	0.15	98.18
Gasoline	7.04E+12	7	0.15	98.32
Dinitrotoluene, Liquid	7.00E+12	7	0.15	98.47
Acrylic Acid	6.8ØE+12	6	8.14	98.61
LPG-Propylene	6.69E+12	6	0.14	98.75
Buty! Acrylate	6.59E+12	6	8.14	98.89
Distillate Fuel Oil	6.91E+12	. 6	Ø.13	99.01
Fuel, Aviation	5.8ØE+12	5	9.12	99.13
Alkaline Liquid, n.o.s.	5.59E+12	5	Ø.12	99.25
Octyl Alcohol	5.31E+12	5	0.11	99.36
LPG-Butene	4.62E+12	4	0.10	99.46
Glycol Ethers	4.22E+12	4	0.09	99.54
Carbon Disulfide	3.65E+12	3	0.08	99.82
Maleic Anhydride	3.19E+12	3	0.07	99.69
Combustible Liquid	2.88E+12	3	Ø. Ø6	99.75
Ferric Chloride Solution	2.65E+12	. 2	0.08	99.80
Flammable Liquid, n.o.s.	2.87E+12	. 2	8.84	99.84
LPG-Isobuty lene	1.47E+12		0.03	
Gasoline, nec	1.29E+12		0.03	99.90
Oil	1.29E+12	_	0.03	
Coal Tar Distillate	1.19E+12		0.02	
Compound, Cleaning	1.18E+12	_	0.02	
Pulp Mill Liquid	1.07E+12			100.00

Table J7. Risk Ranking by ASTM Group

				
		Normal-	% of	Cumu i -
ASTM	Totai	ized	Total	ative
No.	Risk	Risk	Risk	×
2	1.76E+15	481	36.59	36.59
4	5.86E+14	155	11.79	48.37
1	5.42E+14	148	11.28	59.65
10	5.25E+14	144	10.93	70.58
17	2.78E+14	76	5.79	76.37
16	2.34E+14	84	4.88	81.25
164	1.8E+14	49	3.76	85.00
101	1.12E+14	31	2.34	87.34
13	9.98E+13	27	2.08	89.42
26	9.75E+13	27	2.63	91.45
3	7.78E+13	21	1.62	93.97
19	6.01E+13	16	1.25	94.32
28	4.68E+13	13	0.97	95.29
5	3.24E+13	. 9	Ø.68	95.97
31	3.Ø2E+13	8	0.63	96.60
7	2.66E+13	7	Ø.55	97.15
105	2.56E+13	7	0.53	97.69
18	2.48E+13	7	0.52	98.20
34	2.48E+13	7	0.51	98.71
21	1.44E+13	4	0.30	99.01
107	1.43E+13	4	Ø.3Ø	99.31
106	1.08E+13	3	0.22	99.53
24	7.74E+12	2	Ø.16	99.69
27	7E+12	2	Ø.15	99.84
14	4.22E+12	1	0.09	99.92
20	3.65E+12	1	Ø.08	100.00

The Normalized Risk is the Risk divided by the risk from the ASTM group with the lowest risk (ASTM Group #20). It is a relative measure of the contribution from each ASTM Group to the total risk.